

High Moisture Sources And Stack Testing

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Sampling Methods For Stacks With High Moisture Content

High moisture sources:

- ◆ Ammonium nitrate prilling facilities
- ◆ Lime hydrators
- ◆ Evaporators
- ◆ Coke oven quench towers

The Problem

$$\Delta H = D_n \left[\frac{\pi K_p C_p}{4K_m} \right] \left[\frac{M_d(1-B_{ws})^2}{M_d(1-B_{ws}) + 18(B_{ws})} \right] \left[\frac{T_m P_s}{T_s P_m} \right] \Delta P$$

Sampling Problems

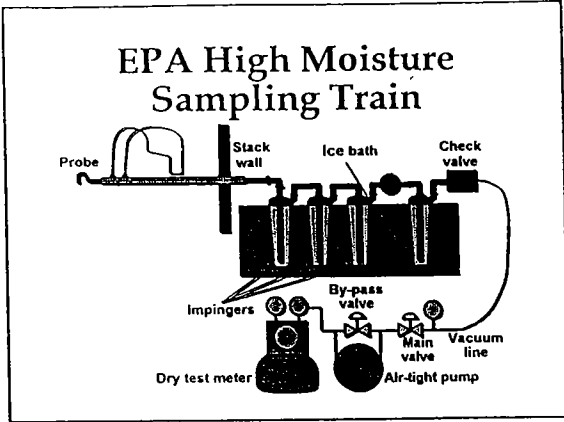
- ◆ Erroneous readings due to low flow rate through orifice
- ◆ Inaccurate control of sampling rate due to small volume of gas passing through control valves
- ◆ Non-isokinetic sampling due to fluctuations in moisture content of stack gas

Solution

- ◆ Place orifice meter before impingers
- ◆ Total sample volume passes through orifice meter
- ◆ Moisture content measurement unnecessary
- ◆ Isokinetics not affected by moisture changes

JACA Corporation Method

- ◆ Advantages
- ◆ Prevention of condensation at orifice
- ◆ Protection from particulate fouling
- ◆ Disadvantages
- ◆ Orifice meter pressure does not remain constant



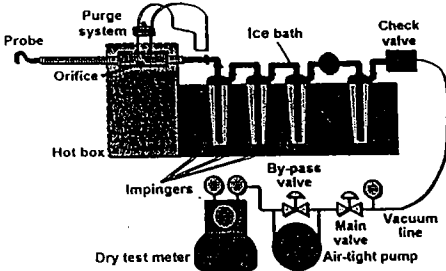
EPA Method

- ◆ Developed for use at ammonium nitrate facilities
- ◆ Consists of an in-situ orifice with a changeable orifice plate
- ◆ Filter located before silica gel impinger
- ◆ Probe heated to prevent condensation

EPA Method Isokinetic Sampling Rate Equation

$$\Delta H = D_n^4 \left[\frac{K_p C_p}{4K_m} \right]^2 \Delta P$$

High Moisture Sampling Train



Problems Common To All Three Methods

- ◆ Entrained water droplets
- ◆ Condensation in manometer lines
- ◆ Improper condensation in impingers

SW-848, Method 0023A

Sampling and Analysis for
Polychlorinated DiBenzo-p-Dioxins
(PCDDs) and Polychlorinated
Dibenzofurans (PCDFs)

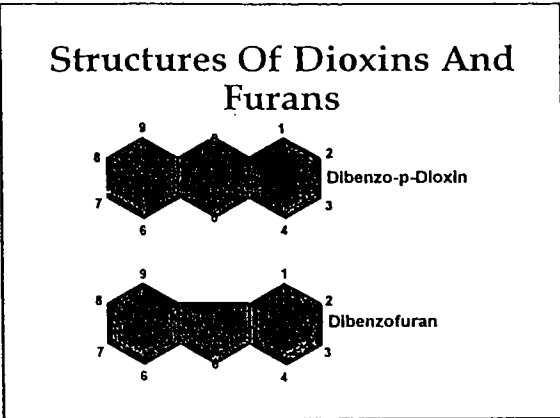
William T. Winberry Jr.

Applicability

- ◆ This method is applicable to the determination of PCDDs and PCDFs from stationary sources
- ◆ Simultaneous sampling and analysis for polychlorinated biphenyls (PCBs), polynuclear aromatic hydrocarbons (PAHs), and semi-volatile organic compounds (SVOCs) can also be performed

What Are Dioxins?

- ◆ Dioxin is a family of 210 different molecules with one or two basic structure:
 - The dioxin structure with two oxygen atoms
 - The furan structure with one oxygen atom



Differences Between Molecules

- ◆ The differences between the molecules lie in the number and attachment positions of chlorine atoms.

Chlorinated Dioxins And Furans

- ◆ 7-2,3,7,8-substituted chlorinated dioxins
- ◆ 75 Total chlorinated dibenzo-p-dioxins

Chlorinated Dioxins And Furans

- ◆ 10-2,3,7,8-substituted chlorinated furans
- ◆ 135 Total chlorinated dibenzofurans

Congeners

- ◆ Molecules with different chemical formulas and the same basic structure are referred to as congeners. Generally, the most common one is octachlorodibenzo-p-dioxin (OCDD), with chlorine in all eight available positions.

Interference's

- ◆ If not using high resolution GC/MS, then interferences from polychlorinated biphenyls and polychlorinated diphenyl ethers could effect low resolution techniques.

Interference's

- ◆ Very high amounts of other organics compounds in the matrix will interfere with the analysis
- ◆ Contamination in solvents, reagents, glassware, and other sampling processing hardware (all glassware must be cleaned thoroughly before use)

Method 0023A Design Requirements

- ◆ Gas flow measurement system (EPA Methods 2-4)
- ◆ Modified Method 5 sampling train, retaining heated filter
- ◆ Addition of condenser and XAD-2 resin trap

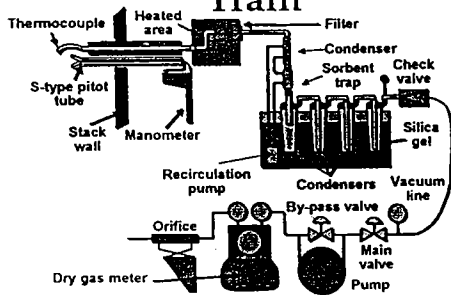
Method 0023A Sampling Train

- ◆ Probe with nozzle
- ◆ Pitot tube/temperature sensor array
- ◆ Heated filter assembly
- ◆ Condenser/XAD-2 resin trap assembly

Method 0023A Sampling Train

- ◆ Condensing impingers
- ◆ Silica gel
- ◆ Pump/dry gas meter/orifice assembly

Method 0023A Sampling Train



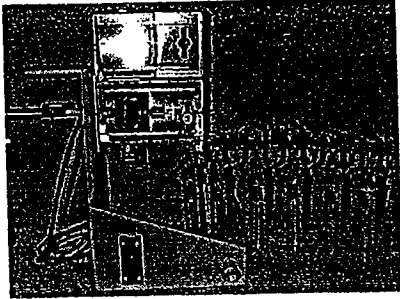
XAD-2 Resin Trap

- ◆ XAD-2 is a cross-linked styrene-divinylbenzene(-CH₂-CH-CH₂-CH-CH₂)-Organic Polymeric Adsorbent
- ◆ Amberlite XAD-2 physical characteristics:

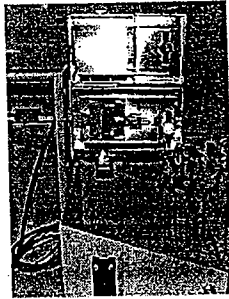
Mesh Size: 20-60
Bulk Density: 1.08 g/mL
Surface Area: 300 m²/g
Temp. Max: 190° C

Demo

Method 0051 Sample Train



Method 0051 Sample Train



Operational Requirements

- ◆ Single point (> 1 meter) integrated sampling
- ◆ Purging of probe for 5 minutes prior to sampling
- ◆ Probe/filter at 120° C (248° F)
- ◆ Sampling at 2 L/min for 60 minutes

Operational Requirements

- ◆ Leak-free system
- ◆ Performance evaluation (PE) sample

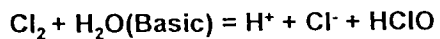
Method 0051 Chemistry

- ◆ Impingers 2 & 3 remove HCl from sample gas stream by the following equation:



Method 0051 Chemistry

- ◆ Impingers 4 & 5 remove Cl_2 from the stack gas stream by the following reaction:



Method 0051 Impinger Arrangement

- ◆ 1st impinger (Optional)
 - empty
- ◆ 2nd and 3rd impingers
 - 15 mL of 0.05 M H₂SO₄

Method 0051 Impinger Arrangement

- ◆ 4th and 5th Impingers
 - 15 mL of 0.1 M NaOH
- ◆ 6th Impinger/drying tube
 - 30 g silica gel

Method 0051 Operation

- ◆ Preliminary field determination (sample location, recovery area) same as FRM 6
- ◆ Sample train preparation (charging of filter and impingers) same as FRM 6

Method 0051 Operation

- ◆ Pre-leak check (-10 " Hg) stopcock, probe, and filter prior to inserting probe into stack
- ◆ Pre-leak check (-10 " Hg) impinger and meter box assembly before testing

Method 0051 Operation

- ◆ Purge probe/filter assembly at 2 L/min for 5 minutes
- ◆ Position probe/Teflon-elbow pointing downstream
- ◆ Sample collection in general accordance with FRM 6
- ◆ Sample at 2 L/min for 60 minutes

Method 0051 Operation

- ◆ The last impinger of the 0.1 M NaOH must be maintained strongly basic during sampling. Monitor pH of solution frequently. To resolve:
 - Use stronger base (0.5M)
 - Add volume to impinger (30 mL)
 - Recharge impinger during sampling

Method 0051 Operation

- ◆ Leak check before/after component changes during sample run
- ◆ Post-leak check in accordance with FRM 6

Sample Train Recovery Containers

- ◆ 1- Combined impingers 1,2 and 3 (measured) and water rinses
- ◆ 2- Combined impingers 4 and 5 (measured) and water rinses (add 2 mL of $\text{Na}_2\text{S}_2\text{O}_3$)
- ◆ 3- Silica gel contents (note color, weigh)

Sample Train Recovery

- ◆ (Overhead of Method 0051 Sample Train Recovery Scheme)

Analysis

- ◆ Analysis of recovery reagents for Cl- by SW-846, Method 9057, Ion Chromatography

Method 0051 QA/QC Requirements

- ◆ (Overhead of Method 0051 QA/QC Requirements)

Method 0051 Key Points

- ◆ System uses a modified FRM 6 sampling train
- ◆ Sampling is performed at 2 L/min for 60 minutes

Method 0051 Key Points

- ◆ Methodology quantitates HCl from Cl₂ emissions
- ◆ Must maintain last impinger of NaOH strongly basic

Method 0051 Key Points

- ◆ Two tier pre-test leak check (-10 " Hg):
 - Stopcock, probe, and filter
 - Impinger and meter box assembly
- ◆ Can also determine moisture of stack gas

Method 0051 Key Points

- ◆ Glass lined probe/Teflon elbow and Teflon filter assembly only
- ◆ Methodology limited to relatively dry, particulate-free gas streams

**Method 0051 Observer
Checklist**

◆ (Overhead of Method 0051 Observer
Checklist)

Sample Train Recovery Containers

- ◆ 5A- Impinger 4 (measured) and HNO₃ rinse of impinger 4 (100 mL)
- ◆ 5B- KMNO₄ impingers contents (measured) and KMNO₄ (100 mL) + water (100 mL)
- ◆ 5C- 8 M HCl (25 mL) rinse of the two impingers and transferred to container containing water (200 mL)

Sample Train Recovery Containers

- ◆ 6- Silica gel contents (note color, weigh)
- ◆ 7- Acetone blank (100 mL)
- ◆ 8A- HNO₃ reagent blank (300 mL)
- ◆ 8B- Water reagent blank (100 mL)
- ◆ 9- 5 % HNO₃/10% H₂O₂ reagent blank (200 mL)

Sample Train Recovery Containers

- ◆ 10- KMNO₄ reagent blank (100 mL)
- ◆ 11- 8 M HCl reagent blank (200 mL of water + 25 mL of 8 M HCl)
- ◆ 12- Filter blank
