# ACUTE EXPOSURE GUIDELINE LEVELS (AEGLs) FOR CADMIUM (CAS Reg. No. 7440-43-9)

Cd

# **INTERIM**

1	
2	
3	ACUTE EXPOSURE GUIDELINE LEVELS (AEGLs)
4	FOR
5	CADMIUM
6	(CAS Reg. No. 7440-43-9)
7	
8	
9	
10	
11	PROPOSED
12	
13	
14	
15	
16	
17	
18	

1 2

3

15 16

17 18

19

20 21 22

23

24

25

26

30

### PREFACE

Under the authority of the Federal Advisory Committee Act (FACA) P. L. 92-463 of
1972, the National Advisory Committee for Acute Exposure Guideline Levels for Hazardous
Substances (NAC/AEGL Committee) has been established to identify, review and interpret
relevant toxicologic and other scientific data and develop AEGLs for high priority, acutely toxic
chemicals.

AEGLs represent threshold exposure limits for the general public and are applicable to emergency exposure periods ranging from 10 minutes to 8 hours. Three levels — AEGL-1, AEGL-2 and AEGL-3 — are developed for each of five exposure periods (10 and 30 minutes, 1 hour, 4 hours, and 8 hours) and are distinguished by varying degrees of severity of toxic effects. The three AEGLs are defined as follows:

**AEGL-1** is the airborne concentration (expressed as parts per million or milligrams per cubic meter [ppm or mg/m3]) of a substance above which it is predicted that the general population, including susceptible individuals, could experience notable discomfort, irritation, or certain asymptomatic, non-sensory effects. However, the effects are not disabling and are transient and reversible upon cessation of exposure.

**AEGL-2** is the airborne concentration (expressed as ppm or mg/m3) of a substance above which it is predicted that the general population, including susceptible individuals, could experience irreversible or other serious, long-lasting adverse health effects or an impaired ability to escape.

AEGL-3 is the airborne concentration (expressed as ppm or mg/m3) of a substance
 above which it is predicted that the general population, including susceptible individuals,
 could experience life-threatening health effects or death.

31 Airborne concentrations below the AEGL-1 represent exposure levels that could produce 32 mild and progressively increasing but transient and nondisabling odor, taste, and sensory 33 irritation or certain asymptomatic, non-sensory effects. With increasing airborne concentrations 34 above each AEGL, there is a progressive increase in the likelihood of occurrence and the severity 35 of effects described for each corresponding AEGL. Although the AEGL values represent threshold levels for the general public, including susceptible subpopulations, such as infants, 36 37 children, the elderly, persons with asthma, and those with other illnesses, it is recognized that 38 individuals, subject to unique or idiosyncratic responses, could experience the effects described 39 at concentrations below the corresponding AEGL

- 40
- 41

1	TABLE OF CONTENTS	
2	PREFACE	
3	LIST OF TABLES	6
4	EXECUTIVE SUMMARY	7
5	1. INTRODUCTION	10
6	2. HUMAN TOXICITY DATA	11
7 8	2.1. Acute Lethality	<i>11</i>
9 10 11	<ul> <li>2.2. Nonlethal Toxicity</li> <li>2.2.1. Odor Threshold/Odor Awareness</li></ul>	
12	2.3. Neurotoxicity	
13	2.4. Developmental/Reproductive Toxicity	
14	2.5. Genotoxicity	
15	2.6. Carcinogenicity	
16	2.7. Summary	
17	3. ANIMAL TOXICITY DATA	16
18 19	<i>3.1. Acute Lethality</i> 3.1.1. Rat	<i>16</i> 16
20 21 22	3.2.       Nonlethal Toxicity         3.2.1.       Rabbit.         3.2.2.       Rat.	
23	3.3. Developmental/Reproductive Toxicity	20
24	3.4. Genotoxicity	
25 26 27	3.5.         Repeated Dose           3.5.1.         Rat           3.5.2.         Mouse	
28	3.6. Chronic Toxicity/Carcinogenicity	
29	3.7. Summary	
30	4. SPECIAL CONSIDERATIONS	
31	4.1. Metabolism and Disposition	
32	4.2. Mechanism of Toxicity	
33	4.3. Structure Activity Relationships	
34 35 36 37 38	<ul> <li>4.4. Other Relevant Information</li></ul>	34 34 34 35 35
39	5. DATA ANALYSIS FOR AEGL-1	
40	5.1. Summary of Human Data Relevant to AEGL-1	

### INTERIM: Sep-2010

1	5.2.	Summary of Animal Data Relevant to AEGL-1	
2	5.3.	Derivation of AEGL-1	
3	6. DA	ATA ANALYSIS FOR AEGL-2	
4	6.1.	Summary of Human Data Relevant to AEGL-2	
5	6.2.	Summary of Animal Data Relevant to AEGL-2	
6	6.3.	Derivation of AEGL-2	
7	7. DA	ATA ANALYSIS FOR AEGL-3	
8	7.1.	Summary of Human Data Relevant to AEGL-3	
9	7.2.	Summary of Animal Data Relevant to AEGL-3	
10	7.3.	Derivation of AEGL-3	
11	8. SU	IMMARY OF AEGLS	
12	8.1.	AEGL Values and Toxicity Endpoints	
13	8.2.	Comparison with Other Standards and Guidelines	
14	<i>8.3</i> .	Data Adequacy and Research	
15	9. RE	FERENCES	40
16	APPEND	DIX A: Derivation of AEGL Values	44
17	APPEND	DIX B: Time-Scaling Calculations	
18	APPEND	DIX C: Carcinogenicity Assessment	51
19	APPEND	DIX D: Derivation Summary for Cadmium AEGLs	52
20	APPEND	DIX E: Category Plot for Cadmium	55
21			

### LIST OF TABLES

1	LIST OF TABLES	
2		
3	TABLE 1. Summary of AEGL Values Cadmium	9
4	TABLE 2. Chemical and Physical Properties	10
5	TABLE 3. Chemical and Physical Properties of Cadmium Compounds	11
6	TABLE 4. Blood Cd Concentrations of Workers and Air Cd Concentrations	13
7	TABLE 5. Summary of Acute Inhalation Data in Laboratory Animals	20
8	TABLE 6. Histopathological Findings in rats Exposed for 2 wks to Cadmium Oxide	23
9	TABLE 7. Histopathological Findings in Rats Exposed for 13 wks to Cadmium Oxide	24
10	TABLE 8. Histopathological Findings in Mice Exposed for 2 wks to Cadmium Oxide	26
11	TABLE 9. Histopathological Findings in Mice Exposed for 13 wks to Cadmium Oxide	27
12	TABLE 10. Summary of Repeat Dose Inhalation Data in Laboratory Animals	28
13	TABLE 11. Observations after Inhalation Exposures of Rats to Various Cd Compounds	30
14	TABLE 12. AEGL-1 Values for Cadmium	36
15	TABLE 13. AEGL-2 Values for Cadmium	37
16	TABLE 14. AEGL-3 Values for Cadmium	38
17	TABLE 15. Summary of AEGL Values	38
18	TABLE 16. Extant Standards and Guidelines for Cadmium	39
19		

1

2

9

### **EXECUTIVE SUMMARY**

Cadmium (Cd) is a metal used in a variety of consumer and industrial materials with a high percentage used in the production of nickel-cadmium batteries and in electroplating. It is also in pigments used in plastics, ceramics and glasses and is used as a stabilizer for polyvinyl chloride (PVC). World production between 1990 and 2000 was ~19,000 tons/year (Morrow, 2001). Estimated U.S. production of cadmium was about 1450 metric tons in 2003 and 700 metric tons in 2006 (ATSDR 2008).

10 Human exposure to cadmium can be from inhalation of cadmium containing particles, 11 inhalation of cigarette smoke or inhalation from fumes/dust in an occupational setting. In case reports of accidental acute exposure, cadmium caused respiratory irritation, dyspnea, alveolar 12 13 damage, pneumonitis, and death. Chronic occupational exposure caused decreased lung 14 function. Cadmium and cadmium compounds are characterized as "probable human carcinogens" based on evidence of carcinogenicity in humans (U.S. EPA 1994). Respiratory 15 16 cancers were increased in workers at a Cd-nickel battery factory, however, chronic Cd exposure 17 was not found to lead to lung carcinogenicity. In animal inhalation studies, cadmium oxide is 18 used as it is the most common airborne form of cadmium. The size of the cadmium particle often 19 determines the extent of absorption and distribution. Cadmium, in various forms, caused 20 respiratory irritation, pulmonary edema, rales, pneumonitis, lacrimation, increased alveolar 21 macrophages, and death in rabbits and rats exposed for 1-6 hours. Rats and mice exposed for 90 days or less exhibited pulmonary inflammation and edema, pulmonary hyperplasia, nasal and 22 23 respiratory epithelium degeneration, and renal lesions. In carcinogenicity studies, rats exposed 24 to Cd had an increased incidence of primary lung carcinomas.

25

The AEGL-1 values are based on the experimental concentration,  $0.55 \text{ mg Cd/m}^3$ , that 26 27 caused slight respiratory irritation in rats (Takenaka et al. 2004). After a 6 hour exposure, 28 increased neutrophils and multifocal alveolar inflammation were observed. At the next higher 29 experimental exposure, pneumonitis was observed (Grose et al. 1987). Although the exposure 30 was a whole-body exposure, the size of the ultrafine particles (51 nM MMAD, 1.7 GSD) would 31 mimic a gaseous state and the majority of the aerosol would be inhaled and not deposited on the fur. An interspecies uncertainty factor of 3 was applied because at acute exposures, cadmium is 32 33 a direct-acting respiratory irritant as indicated by the signs of irritation in rabbits and rats. This 34 mode of action is not expected to differ among species. Rabbits and rats exposed for 2 hours to 35  $0.25-4.5 \text{ mg/m}^3$  displayed similar histological and biochemical pulmonary effects including 36 pneumonitis, increased lung weight, pulmonary inflammatory cell influx, and decreased 37 glutathione peroxidase activity (Grose et al. 1987). Rats exposed to cadmium (0.00169-5.3 mg/m<sup>3</sup>) from 1-6 hours (Buckley and Bassett 1987; Oberdörster et al. 1987; Takenaka et al. 38 39 2004) exhibited the same effects as those observed in the Grose et al. (1987) study. An 40 intraspecies uncertainty factor of 3 was applied because at acute exposures, cadmium is a direct-41 acting respiratory irritant in humans, and this mode of action is not expected to differ among 42 individuals. After a five hour exposure to cadmium, workers experienced cough, throat 43 irritation, dyspnea, and pulmonary edema (Beton et al. 1966) which are signs of respiratory 44 irritation. The concentration-exposure time relationship for many irritant and systemicallyacting vapors and gases may be described by  $C^n x t = k$ , where the exponent, n, ranges from 0.8 45 to 3.5 (ten Berge et al. 1986). To obtain conservative and protective AEGL values in the absence 46 of an empirically derived chemical-specific scaling exponent, temporal scaling was performed 47 48 using n=3 when extrapolating to shorter time points and n=1 when extrapolating to longer time points using the  $C^n x t = k$  equation. The 30-minute AEGL-1 value was adopted as the 10-49

minute value due to the added uncertainty of extrapolating from a 6-hour time point to 10 1 2 minutes (NRC 2001).

3

4 The AEGL-2 values are based on the experimental concentration,  $5.3 \text{ mg Cd/m}^3$ , that 5 caused overt respiratory irritation and pathology in rats (Buckley and Bassett 1987). The 3 hour 6 exposure resulted in reduced weight gain and increased lung weight, protein content, DNA 7 content, number of cuboidal alveolar cells, number of inflammatory cells, and focal areas of 8 interstitial thickening. An interspecies uncertainty factor of 3 was applied because at acute 9 exposures, cadmium is a direct-acting respiratory irritant as indicated by the signs of irritation in rabbits and rats. This mode of action is not expected to differ among species. Rabbits and rats 10 exposed for 2 hours to  $0.25-4.5 \text{ mg/m}^3$  displayed similar histological and biochemical pulmonary 11 effects including pneumonitis, increased lung weight, pulmonary inflammatory cell influx, and 12 13 decreased glutathione peroxidase activity (Grose et al. 1987). Rats exposed to cadmium 14 (0.00169-5.3 mg/m<sup>3</sup>) from 1-6 hours (Buckley and Bassett 1987; Oberdörster et al. 1987; Takenaka et al. 2004) exhibited the same effects as those observed in the Grose et al. (1987) 15 16 study. An intraspecies uncertainty factor of 3 was applied because at acute exposures, cadmium 17 is a direct-acting respiratory irritant in humans, and this mode of action is not expected to differ 18 among individuals. After a five hour exposure to cadmium, workers experienced cough, throat 19 irritation, dyspnea, and pulmonary edema (Beton et al. 1966) which are signs of respiratory 20 irritation. The concentration-exposure time relationship for many irritant and systemicallyacting vapors and gases may be described by  $C^n x t = k$ , where the exponent, n, ranges from 0.8 21 22 to 3.5 (ten Berge et al. 1986). To obtain conservative and protective AEGL values in the absence 23 of an empirically derived chemical-specific scaling exponent, temporal scaling was performed using n=3 when extrapolating to shorter time points and n=1 when extrapolating to longer time 25 points using the  $C^n x t = k$  equation. 26

24

27 The AEGL-3 values are based on the 2 hour LC<sub>50</sub> for cadmium fume in rats, 112 mg/m<sup>3</sup> 28 (Rusch et al. 1986). The  $LC_{50}$  was divided by 3 to estimate a threshold of lethality. An 29 interspecies uncertainty factor of 3 was applied because at acute exposures, cadmium is a direct-30 acting respiratory irritant as indicated by the signs of irritation in rabbits and rats. This mode of action is not expected to differ among species. Rabbits and rats exposed for 2 hours to 0.25-4.5 31  $mg/m^3$  displayed similar histological and biochemical pulmonary effects including pneumonitis, 32 33 increased lung weight, pulmonary inflammatory cell influx, and decreased glutathione 34 peroxidase activity (Grose et al. 1987). Rats exposed to cadmium  $(0.00169-5.3 \text{ mg/m}^3)$  from 1-6 35 hours (Buckley and Bassett 1987; Oberdörster et al. 1987; Takenaka et al. 2004) exhibited the 36 same effects as those observed in the Grose et al. (1987) study. An intraspecies uncertainty 37 factor of 3 was applied because at acute exposures, cadmium is a direct-acting respiratory irritant 38 in humans, and this mode of action is not expected to differ among individuals. After a five hour 39 exposure to cadmium, workers experienced cough, throat irritation, dyspnea, and pulmonary 40 edema (Beton et al. 1966) which are signs of respiratory irritation. The concentration-exposure 41 time relationship for many irritant and systemically-acting vapors and gases may be described by 42  $C^n x t = k$ , where the exponent, n, ranges from 0.8 to 3.5 (ten Berge et al. 1986). To obtain 43 conservative and protective AEGL values in the absence of an empirically derived chemical-44 specific scaling exponent, temporal scaling was performed using n=3 when extrapolating to shorter time points and n = 1 when extrapolating to longer time points using the  $C^n x t = k$ 45 46 equation. 47

- 48
- 49

	TABLE 1. Summary of AEGL Values Cadmium						
Classification	10-min	30-min	1-hr	4-hr	8-hr	Endpoint (Reference)	
AEGL-1 (Nondisabling)	0.13 mg/m <sup>3</sup>	0.13 mg/m <sup>3</sup>	0.10 mg/m <sup>3</sup>	0.063 mg/m <sup>3</sup>	0.041 mg/m <sup>3</sup>	Respiratory irritation, 0.55 mg Cd/m <sup>3</sup> for 6 hr (Takenaka et al. 2004)	
AEGL–2 (Disabling)	1.4 mg/m <sup>3</sup>	0.96 mg/m <sup>3</sup>	0.76 mg/m <sup>3</sup>	0.40 mg/m <sup>3</sup>	0.20 mg/m <sup>3</sup>	Overt respiratory tract irritation and pathology, 5.3 mg/m <sup>3</sup> CdO for 3 hr Buckley and Bassett. 1987)	
AEGL–3 (Lethal)	8.5 mg/m <sup>3</sup>	5.9 mg/m <sup>3</sup>	4.7 mg/m <sup>3</sup>	1.9 mg/m <sup>3</sup>	0.93 mg/m <sup>3</sup>	Threshold of lethality based on the 2-hr rat $LC_{50}$ for Cd fumes, 112 mg/m <sup>3</sup> (Rusch et al. 1986)	

### References

- ATSDR (Agency for Toxic Substances and Disease Registry). 2008. Draft toxicological profile for cadmium. U.S. Department of Health and Human Services, Atlanta, GA.
- Beton, D.C., G. S. Andrews, H.J. Davies, L. Howells and G.F. Smith. 1966. Acute cadmium fume poisoning. Five cases with one death from renal necrosis. Brit. J. Industr. Med 23: 292-301.
- Buckley, B.J. and D.J.P. Bassett. 1987 Pulmonary cadmium oxide toxicity in the rat. J. Toxicol. Environ. Health. 21: 233-250.
- Grose, E.C., J.H. Richards, R.H. Jaskot, M.G. Ménache, J.A. Graham and W.C. Dauterman. 1987. A comparative study of the effects of inhaled cadmium chloride and cadmium oxide: pulmonary response. J. Toxicol. Environ. Health, 21:219-232.
- Morrow, H. 2001. Cadmium and Cadmium Alloys. Kirk-Othmer Encyclopedia of Chemical Technology, 4<sup>th</sup> Ed., M. Howe-Grant, ed. Available. Online. http://mrw.interscience.wiley.com/emrw/978047123896.
- NRC (National Research Council). 2001. Standing Operating Procedures for Developing Acute Exposure Guideline Levels for Hazardous Chemicals. The National Academies Press, Washington, DC.
- Rusch, G.M., J.S. O'Grodnick and W.E. Rinehart. 1986. Acute inhalation study in the rat of comparative uptake, distribution and excretion for different cadmium containing materials. Am. Ind. Hyg. Assoc. J. 47(12): 754-763.
- Takenaka, S., E. Karg, W.G. Kreyling, B. Lentner, H. Schultz, A. Ziesenis, P. Schramel and J. Heyder. 2004. Fate and toxic effects of inhaled ultrafine cadmium oxide particles in the rat lung. Inhal. Toxicol. 16 (suppl.1): 83-92.

### 1. INTRODUCTION

4 Cadmium is used in a variety of consumer and industrial materials with a high percentage 5 used in the production of nickel-cadmium batteries and in electroplating. It is also in pigments 6 used in plastics, ceramics and glasses and as a stabilizer for polyvinyl chloride (PVC). The 7 demand for cadmium has decreased since the 1990s as lithium-ion and nickel metal hydride 8 batteries become more popular (ATSDR 2008). World production between 1990 and 2000 was 9 ~19,000 tons/year (Morrow 2001). Estimated U.S. production of cadmium was about 1450 metric tons in 2003 and 700 metric tons in 2006 (ATSDR 2008). Human exposure to cadmium 10 11 can be from consumption of food, drinking water, incidental ingestion of soil or dust, inhalation of cadmium containing particles, inhalation of cigarette smoke, or inhalation from fumes/dust in 12 13 an occupational setting. Cadmium is usually not present in the environment as pure metal but as 14 a mineral combined with other elements such as oxygen (cadmium oxide), chlorine (cadmium chloride) or sulfur (cadmium sulfate/sulfide) (ATSDR 2008). These forms are also solids but 15 16 some are water soluble.

17

1 2

3

TABLE 2. Chemical and Physical Properties					
Parameter	Value	References			
Synonyms	Colloidal cadmium				
Chemical formula	Cd	HSDB 2005			
Molecular weight	112.41 g	HSDB 2005			
CAS Reg. No.	7440-43-9	HSDB 2005			
Physical state	Silver-white, blue-tinged, lustrous metal; solid	HSDB 2005			
Solubility in water	Insoluble in water	HSDB 2005			
Vapor pressure	1 mm Hg @ 394°C	ATSDR 2008			
Vapor density (air =1)	-	-			
Liquid density (water =1)	-	-			
Melting point	321°C	HSDB 2005			
Boiling point	765°C	HSDB 2005			
Flammability limits	Powder flammable in air; Auto-ignites at 250 °C	ACGIH 1996; NIOSH 2005			
Conversion factors	1 ppm = $4.6 \text{ mg/m}^3$ 1 mg/m <sup>3</sup> = $0.22 \text{ ppm}$	Calculated by reviewer using 1 ppm = $mg/m^3 \times 24.45/112.4$			

TABLE 3. Chemical and Physical Properties of Cadmium Compounds						
	Cd carbonate	Cd chloride	Cd oxide	Cd sulfate	Cd sulfide	
Synonyms	Carbonic acid, cadmium salt	Dichlorocadmium	Cadmium fume; cadmium monoxide	Sulfuric acid, cadmium salt	Cadmium yellow; Cadmium orange	
Chemical formula	CdCO <sub>3</sub>	CdCl <sub>2</sub>	CdCO	$CdSO_4$	CdS	
Molecular wt.	172.42	183.32	128.41	208.47	144.47	
CAS No.	513-78-0	10108-64-2	1306-19-0	10124-36-4	1306-23-6	
Physical state	White powder or rhombohedral leaflets	Colorless, rhombohedral crystals	Dark brown infusible powder or cubic crystals	Colorless monoclinic crystals	Light yellow or orange cubic or hexagonal structure	
Solubility in water	Insoluble	Soluble	Insoluble	Soluble	Soluble at 1.3 mg/L @ 18 °C	
Vapor pressure	_	10 mm Hg @ 656 °C	1 mm Hg @ 1000 °C	_	_	
Vapor density (air =1)	4.26 g/cm <sup>3</sup> @ 4°C	3.3 g/cm <sup>3</sup> @ 20°C	Crystals- 8.15 g/cm <sup>2</sup> ; amorphous powder- 6.95 g/cm <sup>3</sup>	4.69 g/cm <sup>3</sup>	Hexagonal structure- 4.82 g/cm <sup>3</sup> ; Cubic structure- 4.5 g/cm <sup>3</sup>	
Liquid density (water =1)	_	_	_	_	_	
Melting point	Decomposes (< 500 °C)	568 °C	_	1000 °C	1750 °C	
Boiling point	_	960 °C	Sublimes at 1559 °C	-	Sublimes in N <sub>2</sub> @ 980 °C	
Flammability limits				_		
Conversion factors	$\frac{1 \text{ mg/m}^3}{0.14 \text{ ppm}}$	$\frac{1 \text{ mg/m}^3}{0.13 \text{ ppm}}$	$\frac{1 \text{ mg/m}^3}{0.19 \text{ ppm}}$	$\frac{1 \text{ mg/m}^3}{0.12 \text{ ppm}}$	$\frac{1 \text{ mg/m}^3}{0.17 \text{ ppm}}$	

Data from ATSDR, 2008

- No data available

8

#### HUMAN TOXICITY DATA 2.

2.1. **Acute Lethality** 7

# 2.1.1. Case Reports

9 Numerous case reports are available on acute inhalation exposure to Cd; while some 10 report the Cd levels in tissues, very few have the actual Cd exposure concentrations. Panchal and Vaideeswar (2006) reported on a man exposed to an unknown concentration of Cd oxide 11 fumes who developed a cough, dyspnea and finally died. Fernández et al. (1996) reported on a 12 man exposed to Cd fumes for approximately 60-75 minutes while flame-cutting an alloy 13 14 containing about 10% Cd. The man developed dyspnea 4 hours after finishing work, was hospitalized, and on day 15 after exposure, the blood and urine concentration of cadmium was 15 16  $0.34 \,\mu g/100 \,\text{mL}$  (control =  $0.11 \,\mu g/100 \,\text{mL}$ ) and creatinine was  $17.6 \,\mu g/g$  (control =  $0.2 \,\mu g/g$ ). 17 He had multi-organ failure and died after 19 days. Autopsy showed diffuse alveolar damage to the lungs with beginning intra-alveolar fibrosis. Concentrations of cadmium in tissues were 823 18 19 ng/g liver, 3571 ng/g kidney and 1143 ng/g lung. The author stated that exposures to 200-500  $\mu g/m^3$  usually result in "metal fume fever" that lasts for one to two days, so this patient was 20 almost certainly exposed to a much higher concentration. 21

A man was exposed for ~5 hours to a brownish-yellowish smoke during a copper smelting process after which he complained of fatigue and nausea (Yamamoto et al. 1983). The clinical signs continued to worsen with hypoxemia developing. On day 2, Cd was  $6 \mu g/mL$  in the blood and on day 5 was 332  $\mu g/mL$  in the urine. Twelve days after the accident, the man died, and the cadmium content of the right upper lung lobe was found to be 1.06  $\mu g/g$ .

6

7 Five workers were accidentally exposed to Cd fumes for 5 hours in a tank while using an 8 oxyacetylene burner to melt off bolts made of cadmium (Beton et al., 1966). Cadmium oxide 9 was released due to the heat of the burner. None of the men except for the burner wore any type of respiratory protection. All men experienced coughing and slight irritation of the throat during 10 11 exposure with dyspnea developing 4-10 hours later. One man died on post-exposure day 5; all others had degrees of pulmonary edema that resolved over time. The man that died was found to 12 13 have severe pulmonary edema, alveolar metaplasia of the lungs and bilateral cortical necrosis of 14 the kidneys. The lungs contained 0.25 g CdO per 100 g wet specimen. The author speculated that if 11% of the inhaled CdO was retained in the lungs (% retention was estimated based on earlier 15 work in five animal species), approximately 51.7 mg CdO fume must have been inhaled. 16 Working for 5 hours with a ventilatory rate of 20 L/min, the concentration of CdO in the air 17 would have been about 8.6 mg/m<sup>3</sup> or 2,580 minute-mg/m<sup>3</sup>. 18

19 20

22 23

24

26

### 2.2. Nonlethal Toxicity

21 **2.2.1.** Odor Threshold/Odor Awareness

Cadmium is odorless (HSDB 2005).

### 25 2.2.2. Epidemiologic Studies

27 Jakubowski et al. (2004) looked at long-term occupational exposure and lung function of 28 79 workers (median age:  $50.4 \pm 8.9$  years; 35 men and 44 women) to Cd in a cadmium battery 29 factory (mean period of  $17.4 \pm 9.1$  years). For comparison, 159 non-exposed workers ( $48.4 \pm 4.2$ 30 years; 91 men and 68 women) were used as controls. Subjects were divided into four groups 31 depending on their cumulative cadmium exposure calculated either by cadmium levels in the blood x time or cadmium levels in the air x time. The range of Cd-Blood ( $\mu$ g/L) x time (years) 32 33 was  $< 25, 25-500, > 500-1000 \text{ or} > 1000 \text{ and for Cd-Air (mg Cd/m<sup>3</sup>) x time (years) was <math>< 0.01$ . 34 0.10-1.5, > 1.5-4.0 or > 4. Lung function was evaluated using a LUNGTEST 500 spirometer and measuring the following: forced expiratory volume in one second (FEV<sub>1</sub>), forced vital capacity 35 (FVC), peak expiratory flow (PEF), mid expiratory flow (MEF) at 25, 50 and 75%, vital capacity 36 37 (VC), inspiratory capacity (IC) and the percentage of the FEV<sub>1</sub>/FVC ratio. Statistically significant decreases in FEV<sub>1</sub> (85% of predicted values, p=0.0208), PEF (76% of predicted 38 values, p=0.0488), MEF 25% (103 % of predicted values, p=0.0404), MEF 50% (86% of 39 40 predicted values, p=0.0169) and MEF 75% (78% of predicted values, p= 0.0248) were observed 41 in the workers exposed to  $>1000 \,\mu$ g/L x years as measured by Cd-blood concentration compared 42 to controls. Workers in the group exposed to > 4 mg Cd-air x time also had a significant decrease 43 in MEF 50% and slight decrease in the FEV<sub>1</sub>. The results indicated that long-term exposure 44 could cause some decrease in lung function suggestive of mild airway obstruction. Table 4 45 provides the Cd concentrations in the blood of workers and Cd concentrations found at the plant.

### **INTERIM: Sep-2010**

TABLE 4. Blood Cd Concentrations of Workers and Air Cd Concentrations				
Parameter	Ν	Geometric mean ± GSD	Range	
1983				
$Cd \times B (\mu g/L)$	43	$31.8 \pm 2.14$	9.11 - 166	
$Cd x A (mg/m^3)$			0.08 - 0.51	
1986-1988				
$Cd \times B (\mu g/L)$	91	$29.1 \pm 2.01$	4.1 - 120.3	
$Cd x A (mg/m^3)$			0.03 - 0.38	
1998-1999				
Cd x B (µg/L)	116	9.2 ± 2.14	0.5 - 42.1	
$Cd x A (mg/m^3)$			0.03 - 0.032	

Data from Jakubowski et al. (2004)

B=Blood; A=Air

1 2 3

4

5

6 7

8 9

10 11

12 13

14

Lauwerys et al. (1979) followed eleven male workers in a small factory producing cadmium salts and monitored the cadmium levels in their blood and urine as well as the air exposure levels for 13 months. Nine of eleven men wore personal air monitors and the exposure levels (excluding outliers) ranged from 88 to  $6276 \,\mu g/m^3$ , equivalent to 0.088 to  $6.276 \,m g/m^3$ . Most of the exposures were to cadmium oxide. Health effects were not evaluated.

### 2.3. Neurotoxicity

No data were located.

### 2.4. Developmental/Reproductive Toxicity

15 Fifty seven Japanese women, 58.1% of whom had delivered infants of gestational age of more than 30 weeks, and their infants were tested to determine if maternal urinary levels of Cd 16 17 had any effect on the infant growth, gestational age at birth and/or the Cd level in the breast milk 18 (Nishijo et al., 2002). All subjects lived in areas close to a Cd polluted region where *itai-itai* 19 disease, the most severe manifestation of chronic cadmium poisoning, is still being eliminated. 20 No differences were found in the women's socioeconomic status, nutrition status or prenatal care. Maternal urine and breast milk samples were taken on the fifth or eighth day post-partum 21 22 and information about occupational or environmental exposure to Cd, including smoking, was 23 obtained. Eight women were smokers (6 in the low Cd group and 2 in the high Cd group); 24 however, 6/8 had stopped smoking early in pregnancy making smoking less relevant to current 25 Cd levels. Women were divided into two groups based on the samples, those with urinary Cd of  $< 2 \mu g/g$  Creatinine (Cr) (n= 45) and those with  $\geq 2 \mu g/g$  Cr (n= 12). This value of  $2 \mu g/g$  Cr 26 27 was derived from previous studies stating that those exposed to  $2 \mu g/g$  Cr were found to have 10% proteinuria. In the infants from mothers with higher Cd levels, the mean gestational age at 28 29 birth (37 weeks vs. 39.1 weeks), height at birth (47.2 cm vs. 49.2 cm) and weight at birth (2663 g 30 vs. 3099g) were significantly lower (p < 0.01 and 0.05) than the infants of mothers with Cd levels  $< 2 \mu g/g$  Cr. The number of infants delivered by Cesarean section (7 % vs. 4%) was also higher 31 in the high Cd mothers compared to the mothers having lower Cd levels. Multiple regression 32 analysis indicated that an increase of maternal urinary Cd was related to a decrease in gestational 33 age after adjustment for maternal age. Samples of breast milk also had a higher mean 34 concentration of Cd in mothers in the high Cd group. Overall, the study suggested that Cd 35 36 exposure may increase the possibility of pre-term births thus indirectly causing decreased birth 37 weight.

$\frac{1}{2}$	2.5 Conotovicity	
23	2.5. Genotoxicity	
4	Data are summarized from IARC 2003	
5		(-10) = $(-10)$ = $(-10)$
07	Human lemale <i>trat-trat</i> patients	(n=12) exposed to cadmium through the diet had higher
8	matched control subjects $(n-9)$ Howe	ver there were no differences in the frequencies of cells
9	with structural aberrations in cultures f	rom blood of four female <i>itai-itai</i> patients compared to
10	four control subjects.	tom crood of road remain was was parents compared to
11	, and the second s	
12	No differences in chromosomal	aberrations of lymphocytes were observed between five
13	alkaline battery factory workers and th	ree office workers. The battery workers had been
14	employed for 5-24 years and the average	ge cadmium concentration in personal air samples was
15	estimated to be $0.70 \text{ mg/m}^3$ . Blood cad	Imium concentration in the battery workers was 37.7 ng/g
16	and 2.3 ng/g in the office workers.	
17		
18	No differences were found in c	romosomal or chromatid aberration frequency in workers
19	exposed from six weeks to 34 years in	a cadmium pigment plant when compared to controls,
20	a administrative and faboratory personne	r at the same plant.
$\frac{21}{22}$	No difference was observed in	the incidence of chromosomal aberrations in workers
23	exposed to cadmium dusts for 6-25 year	ars when compared to the office worker controls.
24		
25	A small increase in the incidence	ce of chromosomal aberrations was observed in smelter
26	workers when compared to non-smelte	r control subjects. It was not determined if smoking
27	habits were included as a source of cad	mium exposure.
28		
29	Abnormal metaphase rates were	significantly higher in peripheral blood lymphocytes in
30	male workers exposed to cadmium fun	ies and dusts compared to the age-matched controls.
31 22	Codmium oblorido induced sist	ar abromatid ay abanga in human lumphaaytaa in vitra
32 33		er chromatid exchange in numan fymphocytes in vitro.
33 34	2.6. Carcinogenicity	
35	2.0. Curenogenety	
36	Sorahan and Esmen (2004) rep	orted on a cohort study occurring from 1947-2000
37	involving 926 males that worked at a C	d-nickel battery factory. The aim was to investigate
38	mortality of the workers in relation to a	cumulative cadmium hydroxide exposure. All those
39	included in the study were required to	have worked at the factory for a minimum of 12 months.
40	Work histories were available from 194	47-1986 and the factory closed in 1992. Two approaches
41	were used to analyze the data: indirect	standardization and Poisson regression. Based on the
42	serial mortality rates for England and V	Vales, significantly increased mortality was observed for
43	cancers of the pharynx (observed 4, ex $(0.05)$ and $(0.05)$	pected U. /; standardized mortality ratio (SMR) 559,
44 15	p < 0.05), non-mailgnant diseases of the	respiratory tract (observed 61, expected 43; SMR 142, p<
43 16	5 = 0.05, and non-mangnant diseases of the $243$ p = 0.05. Non significantly increases	sed SMRs were observed for lung concer (observed 45
	$2 \pm 3$ , $p \ge 0.03$ ). NON-significantly increases	concer (chapter d) as a concerted 7.5. SMD 116). The results

48 do not indicate that chronic cadmium hydroxide exposure leads to carcinogenicity in the lung,

but that it does lead to increase in non-malignant diseases of the respiratory and genitourinary
systems and pharyngeal cancers.

4 A mortality study of 602 white males with at least 6 months of production work in the 5 same factory between 1940 and 1969 was performed to determine the effects of Cd exposure 6 (Thun et al., 1985). Workers were followed until 1978. Cause-specific mortality rates for seven 7 causes of death were compared between the workers and average US white males. Cadmium 8 inhalation exposure concentrations measured in the plant ranged from 0.007 to 1.5  $mg/m^3$  from 9 pre-1950 to 1976, depending on the area of work. Most of the Cd exposure was to Cd oxide. The study population was obtained from employment histories from the personnel files. Mortality 10 11 was analyzed with the use of the modified life-table system developed by NIOSH. Of all the workers, 83% had over 20 years of follow-up. For respiratory cancer, the expected rate was 12 13 12.15 and the actual rate was 20. All 20 had over 2 years of employment and all mortalities were 14 due to cancers of the lung, trachea and bronchus. Six deaths (expected 4.45) from genitourinary cancer were observed with one due to renal cancer (expected 0.9), two due to cancers of the 15 16 bladder or other urinary organs (expected 1.10) and three due to prostate cancer (expected 2.20). 17 Nine deaths were from non-malignant gastrointestinal disease (NMGID) and the expected 18 number was 2.35. Even with adjustments to account for the lack of knowledge of the worker's 19 smoking habits and the fact that arsenic was used in the same factory prior to 1926, there is still 20 an increase in the respiratory cancers. 21

22 Kjellström et al. (1979) reported on the mortality and cancer incidence of Swedish 23 workers exposed to Cd for more than 5 years. Data were collected from 269 Cd-Ni battery factory workers and 94 Cd-Cu alloy factory workers. At the Cd-Ni factory, the levels of Cd in 24 the air were: before  $1947: >1 \text{ mg Cd/m}^3$ ; in the 1950's:  $200 \mu \text{g Cd/m}^3$ ; between 1962-1974: 5025  $\mu$ g Cd/m<sup>3</sup>; and 1979: < 5  $\mu$ g Cd/m<sup>3</sup>. In this same factory, similar levels of nickel hydroxide were 26 found in the air. In the Cd-Cu factory, cadmium concentrations were not obtained until the 27 28 1960s, when the levels were 100-400  $\mu$ g Cd/m<sup>3</sup>; since the 1970s, the value has been about 50  $\mu$ g  $Cd/m^3$ . An internal reference group from the Cd-Cu factory was used; workers that were 29 30 involved in processes not exposing them to any Cd. A life-table method with the national average cancer incidence rates for men in different age groups was used to help determine any 31 32 correlation between Cd exposure and the cancer incidence. Among the cadmium nickel workers, 33 the risk ratio for nasopharyngeal cancer was 10 (2 cases) which was statistically significantly 34 higher than 1, the expected value. However, part of this increase could be attributed to the nickel hydroxide dust they were exposed to as well as the cadmium oxide dust. There was an increased 35 tendency for mortality from prostate cancer (4 cases) in the Cd-Cu alloy workers; however, the 36 37 risk ratio when calculated was not statistically significantly increased (2.4).

38

The U.S. EPA (1994) listed cadmium and cadmium compounds as probably human
carcinogens based on limited evidence in occupational epidemiologic studies and sufficient
evidence of carcinogenicity in rats and mice by inhalation, injection, and subcutaneous injection.
An inhalation unit risk factor was calculated based on lung, trachea, and bronchus cancer death
data in human males (Thun et al. 1985).

44

The International Agency for Research on Cancer (IARC 2003) concluded that there is
 sufficient evidence in humans for carcinogenicity of cadmium and cadmium compounds and
 categorized cadmium and cadmium compounds as being carcinogenic to humans.

1

2

14

18

31

### 2.7. Summary

3 Case reports, occupational studies, and epidemiological studies showed how inhalation of 4 cadmium affected humans. Although the case reports did not include Cd exposure 5 concentrations, they did show that acute accidental exposure to Cd caused respiratory irritation, 6 dyspnea, alveolar damage, pneumonitis, and death. Chronic exposure to Cd in occupational 7 settings caused decreased lung function and nephrotoxicity. The results of carcinogenicity 8 studies in Cd workers were equivocal, which may be due to concurrent exposures to other metals 9 in the workplace. Respiratory cancers were increased in workers at a Cd-nickel battery factory; 10 although, chronic Cd exposure was not statistically correlated with lung cancer. The U.S. EPA 11 (1994) and IARC (2003) list cadmium and cadmium compounds as being carcinogenic to humans. Maternal urine Cd concentration was associated with decreased gestational age and 12 13 lower weight at birth.

- 15 3. ANIMAL TOXICITY DATA
- 16 **3.1.** Acute Lethality
- 17 **3.1.1. Rat**

Male Crl:CD(SD)Br rats, 24/group, were exposed by nose-only inhalation to 0, 0.25, 0.45 19 20 or 4.5 mg/m<sup>3</sup> of both CdCl<sub>2</sub> and CdO for 2 hours (Grose et al. 1987). The exposure concentrations were given as mg Cd/m<sup>3</sup>. Animals were killed immediately or 72 hours post-21 exposure. The following parameters were determined: Cd content in the lungs, lung weight, body 22 23 weight, biochemical responses and histopathological lesions in the lungs. Concentrations were 24 measured using 47 mm cellulose acetate filters and analyzed by atomic absorption and found to 25 be consistent in the chambers. Three exposed rats died during the study. Two rats in the 0.45 26 mg/m<sup>3</sup> CdO group died; one rat had cardiovascular failure associated with pulmonary congestion 27 and the other had an undetermined cause of death. The exposure group of the third rat was not 28 reported and cause of death was undetermined. It is believed that the rats died from causes 29 related to complications with the exposure apparatus and not from exposure to cadmium. No 30 deaths were reported in rats of the high-dose group.

Twenty six adult Sprague-Dawley CD rats/sex/group were exposed for 2 hours to 97 32  $mg/m^3$  of cadmium red pigment, 99 mg/m<sup>3</sup> cadmium yellow pigment, 132 mg/m<sup>3</sup> cadmium 33 carbonate, or 112 mg/m<sup>3</sup> cadmium fume (Rusch et al. 1986). The exposure concentrations are 34 35 based on cadmium content. An air control group was also included. Animals were exposed in a glass and stainless steel exposure chamber that had a total volume of one cubic meter and an 36 37 effective volume of 760 L. The Cd carbonate and pigments were sieved through a 60-mesh sieve and hand-packed into a Wright dust feed mechanism. Dry air was passed through the dust feed 38 39 and diluted with room air before being delivered to the chamber. Samples from the chamber 40 were taken with a dust monitor. The Cd fume was derived from a 10% aqueous solution of 41 cadmium acetate dehydrate. An aerosol was created by putting the metered solution and dry air 42 through a nebulizer. No mortality occurred in groups exposed to air (control), Cd red or Cd 43 yellow pigment. In the cadmium carbonate group, one female rat died on Day 4 and one male 44 and one female rat died on day 13. All animals were moribund prior to death. Blood and food 45 matter were found in the gastrointestinal tract, and the lung and liver of the animals were 46 enlarged and discolored. In the cadmium fume group, 25/52 rats died. Six males and one female 47 died on day 2, seven males and five females died on day 3, three females died on day 4, two 48 females died on day 5, and one male rat died on day 6. Lung and liver discoloration and

1 congestion were observed in the animals that died from the Cd fume group. Based on the study, 2 the  $LC_{50}$  for cadmium fume was 112 mg/m<sup>3</sup>.

3 4

### **3.2.** Nonlethal Toxicity

### 5 **3.2.1. Rabbit**

6 A maximum of eight male DLA:New Zealand White rabbits (~30 days old) were exposed 7 8 by nose-only inhalation to 0 (controls), 0.25, 0.45 or 4.5 mg/m<sup>3</sup> of both CdCl<sub>2</sub> and CdO for 2 9 hours (Grose et al. 1987). The exposure concentrations were given as mg Cd/m<sup>3</sup>. Animals were killed immediately or 72 hours post-exposure. The following parameters were determined: Cd 10 11 content in the lungs, lung weight, body weight, biochemical responses and histopathological lesions in the lungs. Concentrations were measured and found to be consistent in the chambers. 12 13 Rabbits exposed to  $0.45 \text{ mg/m}^3$  CdO had a greater number of alveolar macrophages present when compared to controls and those exposed to  $CdCl_2$ . At 4.5 mg/m<sup>3</sup>, the lungs of rabbits had 14 moderate to severe multifocal interstitial pneumonitis that was more severe in the CdO group 15 16 with the presence of fibrocytic-type cells as well as pneumocytes. Rabbits exposed 4.5  $mg/m^3$  of 17 either chemical had increased lung weight and lung-to-body weight ratios. In the rabbit, CdCl<sub>2</sub> 18 had an inhibitory effect on pulmonary GSH peroxidase activity at the lowest and highest 19 concentrations. The two highest concentrations of CdO inhibited GSH peroxidase activity. The 20 activity of GSH transferase was increased after treatment with 0.45 mg/m<sup>3</sup> CdCl<sub>2</sub>. The authors hypothesized that the changes in GSH peroxidase and transferase activity could be a response to 21 22 protect cells against lipid peroxidation.

### 24 3.2.2. Rat

23

25

26 Twenty-four female Fischer 344 rats were exposed for 6 hours to ultrafine particles of CdO at a concentration of 70  $\mu$ g Cd/m<sup>3</sup> in whole-body chambers (330 L volume; ventilation 27 28 exchange of 20 times/hour) (Takenaka et al. 2004). The MMAD was 40 nm and the GSD 1.6. 29 Four rats were sacrificed immediately after exposure and on days 1, 4, and 7 for morphology and 30 elemental analysis. Eight rats were sacrificed on day 0 for lung lavage. An additional 16 rats were exposed to 550  $\mu$ g Cd/m<sup>3</sup> in a similar manner. The MMAD was 51 nm and the GSD was 31 1.7. When converted, 70  $\mu$ g Cd/m<sup>3</sup> is equivalent to 0.07 mg/m<sup>3</sup> and 550  $\mu$ g Cd/m<sup>3</sup> is equivalent 32 to 0.550 mg/m<sup>3</sup>. Eight rats were sacrificed on day 0 for lung lavage and four rats were sacrificed 33 34 on days 0 and 1 for morphology and elemental analysis. Twelve animals for each exposure were 35 used as controls, and exposed to clean air only. Just after exposure, Cd in the lungs of rats exposed to 0.07 mg/m<sup>3</sup> was 19% of the total inhaled dose and this remained the same at the other 36 37 time points. A slight but significant increase of Cd in the liver was observed only in the rats 38 sacrificed 7 days after exposure. The lung lavage indicated no exposure-related morphological 39 changes in the lungs or inflammatory responses in the low-dose rats. In rats exposed to the higher concentration,  $0.550 \text{ mg/m}^3$ , Cd content was similar in the lungs on day 0 and 1 but was 40 41 significantly elevated in the liver and kidneys on both days, and 2/4 of the rats had increased 42 blood Cd. Lung layage of the rats in the high-dose group showed increased neutrophils, and 43 multifocal alveolar inflammation was observed histologically. 44

Four groups of ten male Long Evans rats were exposed to CdO dust (0.00195 mg/m<sup>3</sup>),
CdO fume (0.00169 mg/m<sup>3</sup>), CdS (0.00180 mg/m<sup>3</sup>) or sham-exposed in a nose-only inhalation
chamber for 1 hour (Oberdörster et al. 1987). The exposure concentration was not reported as
mg Cd/m<sup>3</sup>, therefore, the concentration of Cd in this study is unknown. The CdO dust was ballmilled for 24 hours and had a mass median aerodynamic diameter (MMAD) of 0.51 µm. The

CdO fume was generated by an electric arc burning off metallic Cd electrodes, producing 1 2 particles of 0.4 µm. Pure CdS with a MMAD of 0.45 µm was used. Twenty-four hours post-3 exposure, the animals were sacrificed, and the lungs were lavaged, and the cellular components of the lavage fluid were analyzed. Lung epithelial permeability was also determined by 4 measuring the activity of <sup>99m</sup>Tc-DTPA in the lung lavage fluid; this substance was injected 5 6 intravenously 10 minutes prior to sacrifice. An increase in activity in the lung lavage fluid 7 indicates a loss of epithelial integrity. Administration of both CdO dust and CdO fume resulted 8 in a significant decrease in the number of alveolar macrophages and a significant increase in 9 numbers of polymorphonuclear neutrophils, with a more pronounced effect observed with the 10 CdO dust. Epithelial permeability was increased with exposure to CdO dust, but not CdO fume. 11 Inhalation exposure to CdS resulted in no differences in rats compared to those in the shamexposed group. The study report provided very little detail but did help to show that water 12 13 solubility does not always correlate with increased toxicity.

14

Male Wistar rats (16/group) were exposed to 0, 0.5, or 5.3 mg/m<sup>3</sup> CdO aerosols for 3 15 hours in a laminar flow exposure chamber (Buckley and Bassett 1987). The animals were 16 17 observed for up to 15 days post exposure. Interim sacrifices took place 2, 4, 7, and 15 day post 18 exposure and 4 rats/group were sacrificed at each necropsy. The CdO aerosols were generated 19 by oxidizing cadmium acetate aerosols as they passed through a heated quartz tube. The 20 chamber atmosphere was sampled using 0.22 µm pore diameter polycarbonate filters and 21 determined by gravimetric analysis of aerosol samples. The geometric standard deviations were 22 2.31 and 3.18 with mass median aerodynamic diameters of 0.26 and 0.33 µm, respectively, for 23 the low and high dose concentrations. Body weight of rats exposed to 0 or  $0.5 \text{ mg/m}^3$  increased 24 from exposure through the end of the observation period, however, rats exposed to  $5.3 \text{ mg/m}^3$  did 25 not gain weight until day 7 post exposure. Body weight was similar among all groups on day 15 post exposure. At 0.5 mg/m<sup>3</sup>, foci of petechial hemorrhage were occasionally observed 2 and 4 26 days post exposure and were consistently observed at  $5.3 \text{ mg/m}^3$ . Mild hypercellularity at 27 28 bronchoalveolar junctions and adjacent alveoli and inflammatory cell influx of mononuclear cells were observed at  $0.5 \text{ mg/m}^3$  prior to day 7 post exposure. Morphology of the lungs was 29 normal in rats exposed to  $0.5 \text{ mg/m}^3$  7 days post exposure. Focal areas of interstitial thickening, 30 increases in cuboidal alveolar cells, and numerous inflammatory cells (interstitial mononuclear 31 cells, alveolar macrophages, eosinophils, and basophils) were observed at 5.3 mg/m<sup>3</sup> in the 32 33 bronchoalveolar junctions and peripheral alveoli. Lung weight (dry) and protein content of the lungs were significantly increased in rats exposed to 5.3 mg/m<sup>3</sup> compared to those of control on 34 35 study days 2-15 post exposure, and DNA content was increased days 4-15. Glutathione 36 peroxidase, glutathione reductase, glucose-6-phosphate dehydrogenase, and 6-phosphogluconate 37 dehydrogenase activity were significantly increased in rats exposed to 5.3 mg/m<sup>3</sup> compared to 38 the activity levels in control rats.

39

40 Male Crl:CD(SD)Br rats, 24/group, were exposed by nose-only inhalation to 0, 0.25, 0.45 41 or 4.5 mg/m<sup>3</sup> of both CdCl<sub>2</sub> and CdO for 2 hours (Grose et al. 1987). The exposure concentrations were given as mg Cd/m<sup>3</sup>. Animals were killed immediately or 72 hours post-42 43 exposure. The following parameters were determined: Cd content in the lungs, lung weight, body 44 weight, biochemical responses and histopathological lesions in the lungs. Concentrations were measured and found to be consistent in the chambers. Three rats exposed to cadmium died 45 during the study; only one rat (0.45 mg/m<sup>3</sup> CdO group) had cardiovascular failure associated 46 with pulmonary congestion and both others had unknown causes of death. No effects were 47 observed in the lungs at 0 and 72 hours in rats in the 0.45 mg/m<sup>3</sup> group. Groups of rats exposed 48 49 to wither 4.5 mg/m<sup>3</sup> CdCl<sub>2</sub> or CdO had no lesions observed immediately after exposure, but at 72

hours there was moderate to severe multifocal interstitial pneumonitis that was more severe in
 the CdO group. It was characterized by the presence of fibrocytic-type cells, alveoli edema,

goblet cell hyperplasia, as well as hyperplastic pneumocytes. The pneumonitis observed in rats

4 exposed to CdCl<sub>2</sub> presented as thickening of the alveolar walls, edema, hemorrhage, and

5 increases in neutrophils and alveolar macrophages. There was no difference in Cd deposition in

6 the lungs in animals exposed to 0.25 or 0.45  $mg/m^3$  CdCl<sub>2</sub> or CdO at either 0 or 72 hours. While

7 there was an increase in Cd deposition in the lungs of the rats exposed to  $4.5 \text{ mg/m}^3$ , at 72 hours

there was significantly less Cd in the lungs of rats exposed to  $4.5 \text{ mg/m}^3$ CdCl<sub>2</sub> than in those exposed to 1 ppm CdO. Rats exposed to 0.25 mg/m<sup>3</sup> CdCl<sub>2</sub> had a 13% decrease in lung-to-body

weight ratio. Rats exposed to  $0.45 \text{ mg/m}^3$  group CdCl<sub>2</sub> had a decrease in body weight and at 4.5

 $11 \text{ mg/m}^3 \text{ a } 20\%$  decrease in body weight and increased lung and lung-to-body weight ratio 72

12 hours after exposure. In the 0.45 or 4.5  $mg/m^3$  CdO exposed rats, there was an increase in lung

13 weight but no effect on body weight. Cadmium  $(CdCl_2)$  had an inhibitory effect (27%) on

pulmonary GSH peroxidase activity at the lowest concentration and at 0.45 mg/m<sup>3</sup>. Cadmium
 inhibited GSH peroxidase activity at the two highest dose levels.

16

17 Twenty six adult Sprague-Dawley CD rats/sex/group were exposed in a single 2-hour exposure to 97 mg/m<sup>3</sup> of cadmium red pigment, 99 mg/m<sup>3</sup> cadmium yellow pigment, 132 mg/m<sup>3</sup> 18 cadmium carbonate, or 112 mg/m<sup>3</sup> cadmium fume (Rusch et al. 1986). The exposure 19 20 concentrations are based on cadmium content. An air control group was also included. Animals 21 were exposed in a glass and stainless steel exposure chamber that had a total volume of one 22 cubic meter and an effective volume of 760L. The Cd carbonate and pigments were sieved 23 through a 60-mesh sieve and hand-packed into a Wright dust feed mechanism. Dry air was 24 passed through the dust feed and diluted with room air before being delivered to the chamber. 25 Samples from the chamber were taken with a dust monitor. The Cd fume was derived from a 26 10% aqueous solution of cadmium acetate dehydrate. An aerosol was created by putting the 27 metered solution and dry air through a nebulizer.

28

29 The rats exposed to Cd fume exhibited clinical signs (hypoactivity and closed eyes) 30 during the exposure. Following exposure, excessive lacrimation was observed in rats exposed to Cd red and yellow, and dry rales and body tremors were observed in the animals exposed to Cd 31 32 carbonate. Those exposed to Cd fume had dry rales, labored breathing, and excessive 33 lacrimation. Animals exposed to Cd fume also had decreased body weight compared to the 34 controls, with all others maintaining weight similar to that of controls. No gross abnormalities 35 were observed in those sacrificed immediately following exposure. Renal discoloration was 36 observed in the rats in the Cd red pigment group. Pulmonary edema, observed as increased lung 37 weight was seen in the Cd carbonate exposed group starting at 24 hours post-exposure. 38 Exposure to Cd carbonate or Cd fume resulted in an increased incidence of lung discoloration 39 and erosions in the stomach. Blood levels indicated that Cd was absorbed to a greater degree in 40 the carbonate and fume groups when compared to the pigment groups. Urine and feces samples 41 were collected at 0-24 hrs, 24-48 hrs, 48-72 hrs, 6-7 days and 29-30 days post-treatment. 42 Samples indicated that the highest levels of cadmium were excreted in the first 24 hour period 43 and urinary excretion was similar in all three groups while 80% of the red and yellow pigments 44 were excreted in the feces within the first 24 hours. Tissues were collected for cadmium analysis 45 in all but the Cd fume-exposed animals and these were not collected due to the moribund 46 condition of the animals. The Cd carbonate-exposed animals had the greatest amount of 47 cadmium measured in both the kidneys and liver; Cd levels increased initially in the liver and 48 then dropped off but continued to increase in the kidney. 49

### INTERIM: Sep-2010

	TABLE 5.         Summary of Acute Inhalation Data in Laboratory Animals				
Species	Concentration (mg Cd/m <sup>3</sup> )	Exposure Time	Effect	Reference	
Rabbit	$     \frac{\text{CdCl}_2}{0.25} \\     0.45 \\     4.5     $	2 hr	<ul> <li>↓ Pulmonary GSH peroxidase activity 25%</li> <li>↑ GSH transferase activity</li> <li>Moderate-severe multifocal interstitial pneumonitis; ↑lung weight</li> </ul>	Grose et al. 1987	
Rabbit	CdO 0.25 0.45 4.5	2 hr	No effects ↑ Alveolar macrophages; ↓ pulmonary GSH peroxidase activity Moderate-severe multifocal interstitial pneumonitis; ↑lung weight; ↓ pulmonary GSH peroxidase activity	Grose et al. 1987	
Rat	<u>CdO</u> 0.07 0.550	6 hr	No morphological changes or inflammatory response ↑ Neutrophils and multifocal alveolar inflammation	Takenaka et al. 2004	
Rat	0.00195 Cd dust 0.00169 Cd fume 0.00180 CdS	1 hr	<ul> <li>↓ Alveolar macrophages; ↑PMNs; ↑epithelial permeability</li> <li>↓ Alveolar macrophages; ↑PMNs No effect</li> </ul>	Oberdörster et al. 1987	
Rat	<u>CdO</u> 0.5 5.3	3 hr	Transient mild hypercellularity at bronchoalveolar junctions and adjacent alveoli, inflammatory cell influx Interstitial thickening, ↑ cuboidal alveolar cells, ↑inflammatory cells, ↑dry lung weight, ↑protein content, ↑DNA content, ↑ GP, GR, G6PD, 6PGD activity	Buckley and Bassett 1987	
Rat	<u>CdCl<sub>2</sub></u> 0.25 0.45 4.5	2 hr	<ul> <li>↓ Pulmonary GSH peroxidase activity 27%</li> <li>↓ bw; ↓ pulmonary GSH peroxidase activity</li> <li>↓ bw 20%; ↑lung weight; ↓ pulmonary GSH peroxidase activity; pneumonitis</li> </ul>	Grose et al. 1987	
Rat	<u>CdO</u> 0.25 0.45 4.5	2 hr	<ul> <li>No effects</li> <li>↑ Lung weight; ↓ pulmonary GSH peroxidase activity; 2 deaths-cardiovascular failure (1)</li> <li>↑ Lung weight; ↓ pulmonary GSH peroxidase activity; pneumonitis</li> </ul>	Grose et al. 1987	
Rat	97 Cd red 99 Cd yellow 132 Cd carbonate 112 Cd fume	2 hr	Lacrimation; renal discoloration Lacrimation Dry rales, body tremors; 5.8% mortality Hypoactivity; closed eyes; lacrimation; dry rales; ↓ bw; ↑ lung discoloration and stomach erosion; 48% mortality; LC <sub>50</sub>	Rusch et al. 1986	

### 3.3. Developmental/Reproductive Toxicity

Four female Hartley guinea pigs/group were exposed in late gestation (days 50-55) to 0 or  $0.05 \text{ mg/m}^3$  cadmium chloride for 4 hours/day for 1 or 5 consecutive days by nose-only inhalation (Trottier et al. 2002). Cadmium aerosol was generated in a nebulizer using a solution

of Cd made from CdCl dissolved in distilled water. Total airflow to the chamber was 22 L 1 2

air/min and the concentration of Cd in the chamber was monitored by obtaining air samples 3 through filters during the exposure. The mean Cd concentration was  $53.2 \pm 4.6 \,\mu g/m^3$  and the

MMAD was 0.3µm. Twenty-four hours after the last exposure, females were euthanized and the 4

- 5 tissues processed. Tissue Cd content was determined by graphite furnace atomic absorption 6 spectrophotometry and placental metallothionein and cadmium content were also determined.
- 7

17

8 Inhalation exposure did not affect the maternal body weight, fetal body weight, maternal 9 or fetal organ weight, or placental weight when compared to controls. Maternal rats had a significant increase (p<0.01) in lung Cd compared to controls after only a single day and a 10 11 significant increase in both lung and liver after 5 days. In the fetus, brain, liver and heart Cd levels were significantly increased compared to controls after 1 day of exposure, and levels in 12 13 brain and liver remained elevated after 5 days. Maternal blood Cd increased from 25.6 to 57.3 14 pg/mg of protein after 5 days of treatment but the fetal blood had no change. The levels of Cd and MT in the placenta did not change upon exposure, but the placental cadmium decreased after 15 16 the 5-day exposure.

18 Thirty two pregnant Sprague-Dawley female rats and thirty-three pregnant Swiss (CD-1) mice were exposed to 0, 0.05, 0.5 or  $2 \text{ mg/m}^3$  cadmium oxide by whole-body inhalation for 6 19 20 hours/day, 7 days/week on gestation days (GD) 4-19 in the rats and GD 4-17 in mice (NTP 1995). Generation of the aerosol was by the same methods described in Section 3.5.1. One rat 21 22 exposed to  $2 \text{ mg/m}^3$  died on GD 17. Clinical signs observed in rats were dyspnea in all those exposed and hypoactivity at  $2 \text{ mg/m}^3$ , in females. The female rats exposed to  $2 \text{ mg/m}^3$  had 23 significant decreases (p < 0.01) in body weight (14% less than controls) and body weight gain 24 25 (41% less than controls). The high dose female rats also had decreases in absolute and relative 26 liver weight and absolute kidney weight, compared to controls. Developmental toxicity was observed at  $2 \text{ mg/m}^3$  in rats, and included a decrease in the weight of live fetuses. This 27 28 concentration also caused significant increases in the mean percent of fetuses per litter with 29 reduced ossification of the pelvis (12 vs. 2.4 in controls) and sternebrae (25 vs. 4.4 in controls). 30 Dyspnea and hypoactivity were observed in mice exposed to concentrations  $\geq 0.5 \text{ mg/m}^3$ cadmium oxide. At  $\ge 0.5$  mg/m<sup>3</sup>, the number of pregnant mice was significantly decreased. Fetal 31 body weight following exposure to  $0.5 \text{ mg/m}^3$  was significantly less than control fetal body 32 weight. Five mice in the  $2 \text{ mg/m}^3$  group were euthanized moribund before the end of the study. 33 34 Maternal body weight gain, absolute and relative gravid uterine weights, and absolute liver 35 weight were significantly lower compared to control values, and relative kidney weight was significantly increased compared to control kidney weight in female mice exposed to 2 mg/m<sup>3</sup>. 36 37 The total number of resorptions per litter was significantly increased in this group, and the fetal 38 body weight and percentage of live male fetuses per litter were significantly decreased in the 2  $mg/m^3$  group. 39

40

41 Male and female Fischer 344 rats (10 weeks old) were exposed whole body to 0, 0.3, 1.0 or 2.0 mg CdCl<sub>2</sub>/m<sup>3</sup> for 6 hours/day, 5 days/week for 62 exposures (Kutzman et al. 1986). 42 43 Twenty male rats were used for multiple pulmonary endpoint assessments, eight males for pathology only, eight male and eight females for reproductive studies and ten males for 44 cytogenetic endpoints. Rats were exposed in a 1.4 m<sup>3</sup> stainless steel and Lucite chamber. 45 Airflow was equivalent to 15 chamber volumes/hr. Laskin-type nebulizers were used to generate 46 47 the aerosol. An optical particle size analyzer was used to characterize the size distribution of the

48 aerosolized particles. RAM-1 aerosol mass monitors were used to continuously monitor the

1 chamber atmosphere. Chamber atmospheres were 0.33 ( $\pm$  0.02), 1.06 ( $\pm$  0.04) and 2.13 ( $\pm$  0.11) 2 mg Cd Cl<sub>2</sub>/m<sup>3</sup> in the 0.3, 1.0 and 2.0 mg/m<sup>3</sup> chambers, respectively.

3

All rats exposed to 2.0  $mg/m^3$  lost weight rapidly and died within the first 45 days and 4 5 were observed to have rapid and shallow breathing and appeared unkempt prior to death. The 6 females averaged a higher survival (40 days; n=10) compared to the males (32 days; n=57). At 7  $1.0 \text{ mg/m}^3$ , five males died and all animals at 0.3 mg/m<sup>3</sup> survived. Some exposed males and 8 females were allowed to breed with unexposed mates and there were no decreases in 9 reproductive potential. No findings were associated with treatment in the number of viable 10 embryos, late deaths, early deaths (resorptions), number of corpea lutea, or pre-implantation 11 losses. At necropsy, there was a dose-dependent increase in organ-to-body weight ratio for the lungs, heart, spleen, kidneys, and testis. Also, the liver and brain weight-to-body ratio was 12 13 increased in the high-dose, compared to the controls. Lesions of type II hyperplasia, alveolar 14 macrophages, and polymorphonuclear leukocytes were observed at the terminal bronchioles of both the low- and mid-dose rats. Areas of fibrosis were also observed in the mid-dose rats. 15 16 Similar lesions were identified in those that died in the high-dose group. Based on the histopathological findings, the LOAEL was 0.3 mg/m<sup>3</sup> CdCl<sub>2</sub> and the NOAEL could not be 17 18 determined.

19

21

### 20 **3.4.** Genotoxicity

Cadmium oxide was not mutagenic in *Salmonella typhimurium* strains TA98, TA100,
 TA1535 or TA1537 with or without metabolic activation and did not produce micronuclei in
 erythrocytes of mice exposed by inhalation for 13 weeks (NTP, 1995).

25

Cadmium chloride induced DNA strand breaks in *Escherichia coli* and induced
 differential toxicity in *Bacillus subtilis* and *E. coli* strains. It also induced gene conversion in
 *Saccharomyces cerevisiae*, but did not induce reverse mutation in *S. cerevisiae*. Unscheduled
 DNA synthesis and DNA strand breaks were observed in primary cultures of rat hepatocytes, but
 not in primary cultures of rat Leydig cells. Calcium chloride was mutagenic to Chinese hamster
 V79 cells and mouse lymphoma L5178Y cells (IPCS 1992).

### 33 **3.5. Repeated Dose**

34 3.5.1. Rat 35 36 Male and female F344/N rats (6 wks old) were exposed to cadmium oxide aerosol 37 (99.4% purity; mass median aerodynamic diameter (MMAD) =  $1.1-1.6 \mu$ gm) for 6 hours/day, 5 days/week for 2 weeks (n = 5/sex/group) or 13 weeks (n = 10/sex/group) (NTP 1995). Animals 38 were exposed to 0, 0.1, 0.3, 1, 3 or 10 mg/m<sup>3</sup> for the 2 week studies and 0, 0.025, 0.05, 0.1, 0.25 39 or  $1 \text{ mg/m}^3$  for 13 weeks. Animals were exposed in whole-body chambers that had a total 40 41 volume of 2.3 m<sup>3</sup>. Chemical concentration, airflow, temperature and relative humidity were 42 controlled and monitored with an automated system. Overall, concentration within the chamber 43 was adequate. Cadmium oxide dust was mixed with compressed air to create an aerosol. The 44 MMAD of the aerosol particles was measured in each exposure chamber before the studies 45 began, once in the 2 week study and monthly in the 13 week study. Blood was obtained to measure hematology and clinical chemistry parameters, urine collected and histopathology 46 47 performed. 48

In the rats exposed for 2 weeks, all those exposed to  $10 \text{ mg/m}^3$  died by day 6; no other 1 2 deaths occurred. Clinical signs of hypoactivity, dehydration, ruffled fur, dyspnea and nasal 3 discharge were observed in rats at concentrations  $\geq 1 \text{ mg/m}^3$ . Lesions in the lungs including alveolar histiocytic infiltrate, focal inflammation and fibrosis were observed in all of the treated 4 rats with a dose-dependent increase in severity. Effects on the nasal and respiratory epithelium 5 were observed in those exposed to  $1 \text{ mg/m}^3$  and greater. Based on the findings, the doses were 6 set for the 13 week study. For the two-week study, a LOAEL of 0.1  $mg/m^3$  Cd oxide in rats was 7 8 established based on histopathological findings; a NOAEL could not be determined.

9

TABLE 6. Histopathological Findings in rats Exposed for 2 wks to Cadmium Oxide						
	(#	Affected/Total	# examined)			
	0 mg/m <sup>3</sup>	$0.1 \text{ mg/m}^3$	$0.3 \text{ mg/m}^3$	1 mg/m <sup>3</sup>	$3 \text{ mg/m}^3$	10 mg/m <sup>3</sup>
MALES						
Lung						
Alveolar infiltrate	0/5	5/5 (2.0) <sup>a</sup>	5/5 (2.0)	5/5 (3.0)	5/5 (3.8)	5/5 (4.0)
Focal inflamm/fibrosis	0/5	5/5 (1.0)	5/5 (2.0)	5/5 (3.0)	5.5 (2.8)	5/5 (4.0)
Nose						
Olfactory epithelium						
Degeneration	0/5	0/5	0/5	2/5 (1.0)	5/5 (2.0)	5/5 (2.2)
Respiratory epithelium						
Squamous metaplasia	0/5	0/5	0/5	1/5 (1.0)	0/5	5/5 (1.0)
Inflammation	0/5	0/5	0/5	1/5 (1.0)	5/5 (1.4)	3/5 (1.7)
FEMALES						
Lung						
Alveolar infiltrate	0/5	5/5 (2.0)	5/5 (2.2)	5/5 (3.0)	5/5 (4.0)	5/5 (4.0)
Focal inflamm/fibrosis	0/5	3/5 (1.0)	5/5 (2.0)	5/5 (3.0)	5.5 (3.0)	5/5 (4.0)
Nose						
Olfactory epithelium						
Degeneration	0/5	0/5	0/5	4/5 (1.3)	4/5 (2.3)	4/4 (3.0)
Respiratory epithelium						
Squamous metaplasia	0/5	0/5	0/5	0/5	4/5 (1.5)	4/4 (1.5)
Inflammation	0/5	0/5	0/5	0/5	4/5 (2.3)	3/4 (1.0)

Data from NTP (1995)

<sup>a</sup> Number in parenthesis is severity code: 1= minimal, 2= mild, 3= moderate and 4= marked

10

11 12

All rats survived the 13 week study and all treated rats had a nasal discharge that increased in frequency as the concentration increased. Rats at 1 mg/m<sup>3</sup> consistently had lower body weight throughout the study compared to controls, but it was within 10%. No significant

15 findings were observed in the hematology, clinical chemistry or urine parameters. Males and

16 females had statistically significant increases ( $p \le 0.01$  or 0.05) in organ weight at concentrations

17  $\geq 0.05 \text{ mg/m}^3$ , compared to controls including relative kidney weight, and relative and absolute

18 lung weight; however, there were no treatment-related microscopic findings in the kidney or

19 liver. Blood pressure was measured in the animals and there were no findings observed with

treatment. Grossly, the only treatment-related finding was enlargement and paleness of the
 tracheobronchial and mediastinal lymph nodes. Microscopic lesions were identified in the lungs

in all treated animals except those in the  $0.025 \text{ mg/m}^3$  group. The lesions were similar to those

identified in the 2 week study. Based on the histopathological findings, the LOAEL for rats

treated for 13 weeks  $0.05 \text{ mg/m}^3$  cadmium and the NOAEL was  $0.025 \text{ mg/m}^3$ .

TABLE 7. Histop	athological	l Findings in <b>F</b>	ats Exposed for	r 13 wks to Cao	dmium Oxide	)			
	(# Affected/Total # examined)								
	0	0.025			0.25				
	mg/m <sup>3</sup>	mg/m <sup>3</sup>	0.05 mg/m <sup>3</sup>	0.1 mg/m <sup>3</sup>	mg/m <sup>3</sup>	1 mg/m <sup>3</sup>			
Males									
Lung									
Alveolar infiltrate	0/10	0/10	10/10 (1.0) <sup>a</sup>	10/10 (2.0)	10/10 (3.0)	10/10 (3.0)			
Inflammation	0/10	0/10	0/10	0/10	10/10 (2.6)	10/10 (2.1)			
Fibrosis	0/10	5/5 (1.0)	5/5 (2.0)	5/5 (3.0)	5.5 (2.8)	5/5 (4.0)			
Tracheobronchial lymph node									
Inflammation	0/9	0/7	0/5	3/7 (1.0)	9/9 (3.0)	10/10 (3.1)			
Nose									
Olfactory epithelium									
Degeneration	0/10	0/10	0/10	0/10	1/10 (1.0)	10/10 (3.0)			
Respiratory epithelium									
Inflammation	0/10	0/10	0/10	0/10	7/10 (1.0)	9/10 (2.6)			
		Fen	nales						
Lung									
Alveolar infiltrate	0/10	0/10	10/10 (1.0)	10/10 (2.1)	10/10 (3.0)	10/10 (3.0)			
Inflammation	0/10	0/10	0/10	0/10	10/10 (1.6)	10/10 (3.5)			
Fibrosis	0/10	0/10	0/10	10/10 (1.0)	10/10 (2.0)	10/10 (2.1)			
Tracheobronchial lymph node									
Inflammation	0/7	0/4	0/8	6/8 (1.2)	6/9 (2.8)	10/10 (3.5)			
Nose									
Olfactory epithelium									
Degeneration	0/10	0/10	0/10	0/10	2/10 (1.0)	10/10 (2.8)			
Respiratory epithelium									
Inflammation	0/10	0/10	0/10	3/10 (1.0)	10/10 (1.6)	10/10 (1.8)			

Data from NTP (1995)

<sup>a</sup> Number in parenthesis is severity code: 1= minimal, 2= mild, 3= moderate and 4= marked

2 3

4 Sixty male Wistar (CHbb:THOM) rats, approximately 8 weeks old, with a mean body 5 weight of 250 g were exposed nose-only 6 hours/day for up to 10 days to 0 (air control), 6 0.3 mg/m<sup>3</sup> CdCl<sub>2</sub> or 0.2, 1.0 or 8.0 mg/m<sup>3</sup> CdS (Klimisch 1993). Four animals from each group were sacrificed on days 2, 10, 11, 12, 13, 17, 38, 66 and 94. The study was designed to expose 7 8 rats to soluble (CdCl<sub>2</sub>) and insoluble (CdS) forms of Cd. Measurements of lung, renal and fecal 9 Cd were obtained. Cadmium chloride was generated as an aerosol by taking an aqueous solution 10 and putting it through an injection pump to a binary nozzle atomizer before diluting with air. Cadmium sulfide was generated as a dust using a rotating brush-type generator and passing the 11 12 dust directly into the inhalation chamber. All concentrations were sampled and found to be 13 within an acceptable range. Upon sacrifice, the lungs and kidneys were removed and weighed 14 before being processed for Cd content. Feces were collected daily and pooled together.

15

No adverse clinical signs or mortalities occurred during the study or post-exposure 16 period. Pooled data from days 0-10 showed a statistically significant increase in mean lung 17 18 weight and lung to body weight ratio in the  $CdCl_2$ -exposed rats. Rats exposed to 8.0 mg/m<sup>3</sup> CdS 19 also had increased mean lung weight. No effects were observed on kidney weight. Cadmium 20 accumulated in the kidney of all treated animals but at a much greater proportion in those 21 exposed to CdCl<sub>2</sub>. Cadmium in feces was observed mostly during the exposure and for a few days post-exposure, with the highest amount observed in rats exposed to the highest dose of CdS. 22 23 The analyzed Cd content of test atmospheres for both compounds was 0.17 µg/L; however, the 24 Cd lung content in the CdCl<sub>2</sub>-exposed rats was about two times higher than the CdS exposed rats. This could have been caused by either the greater availability of the CdCl<sub>2</sub> aerosol 25

1 compared to the CdS dust or the fact that  $CdCl_2$  is held in the lung longer. Overall the study 2 showed a much higher bioavailability of  $CdCl_2$  compared to the soluble CdS upon inhalation 3 exposure.

4

5 Three month old female Wistar rats (n = 13-14) were exposed by inhalation to CdO for 5 6 hours/day, 5 days/week for 20 weeks at concentrations of 0.02, 0.16 or 1 mg/m<sup>3</sup> (Barański and 7 Sitarek 1987). A control group was exposed to clean air only. The aerosol concentration was 8 determined by drawing air in an inhalation chamber through Sartorious membrane filters 9 SM11306. Fractions of aerosols with particle sizes below 4.7 µm were determined with the Anderson Impactor 2000. Mortality occurred at 1 mg/m<sup>3</sup>, although histopathology and gross 10 findings were not described in the report. At  $1 \text{ mg/m}^3$ , mortality was 15% at the end of 7-8 11 weeks, 38% at the end of 13-14 weeks and 100% at the end of the study. This was compared to 12 0, 7 and 14% of the controls at the same time points, respectively. A significant decrease (p< 13 14 (0.05) in body weight gain was observed in the rats exposed to 1 mg/m<sup>3</sup> throughout the study; at  $0.16 \text{ mg/m}^3$ , exposed rats had a decrease in body weight gain compared to controls, but it was 15 16 not significant. All treated rats showed an increase in the length of estrous during treatment when compared to pre-treatment, but at 0.02 and 0.16  $mg/m^3$ , rats showed this change only at the end 17 of the study (weeks 19-20). At 1.0 mg/m<sup>3</sup>, exposed rats had an increase in estrous length during 18 the entire study. Based on the increased estrous length, the LOAEL was  $0.02 \text{ mg/m}^3$  CdO and the 19 20 NOAEL could not be determined.

21

Twelve adult female Wistar rats/group were exposed continuously to 0, 25 or 50  $\mu$ g Cd/m<sup>3</sup> 22 for 90 days and to 100  $\mu$ g Cd/m<sup>3</sup> for 63 days (Prigge 1978), equivalent to 0, 0.025, 0.052 and 23  $0.105 \text{ mg/m}^3$  administered as CdO. The rats were removed from the chamber for only 10-20 24 25 minutes daily to allow for cleaning and food was changed daily to prevent oral contamination. 26 Inhalation chambers were 50 x 50 x 90 cm with a total volume of 225L. Cadmium was nebulized 27 from a 0.2% solution of cadmium acetate. The volume flow into the chambers was about 80 28 L/min equating to about 21 changes/hour. Aerosol concentrations were checked and found to be 29 within range of the expected values. Size distribution of the particles had a mean aerodynamic 30 diameter of 0.19 µm, and the geometric standard deviation was 1.5. Both are in the range of respirable fine dust in humans. Urine and blood samples were obtained after exposure. A 31 significant dose-dependent decrease (p<0.05) in body weight was observed in the rats exposed 32 to 0.052 and 0.105 mg/m<sup>3</sup>, and 5/12 died at 0.105 mg/m<sup>3</sup> between days 45 and 60. A significant 33 34 increase in lung weight was observed in all of the treated animals. A slight increase was 35 observed in hematocrit or hemoglobin with treatment, but there was no effect on serum iron or 36 alkaline phosphatase activity. A significant (p<0.05) decrease in the partial pressure of  $O_2$  and an 37 increase in the partial pressure of  $CO_2$  were observed in the rats exposed to 0.105 mg/m<sup>3</sup>. Proteinuria was not observed in any of the treated females. Inhalation uptake of cadmium 38 39 resulted in increased liver cadmium levels but they were not as high as those observed after oral 40 administration; uptake of cadmium in the kidney was similar with both methods of 41 administration. Histopathology showed a few areas of swellings in the kidney tubuli, no lesions 42 in the liver, emphysematic areas and cell proliferation in the bronchi, bronchiole, alveoli, and 43 histiocytic cell granulomas in the lungs in the treated animals. Based on the histopathological findings, the inhalation LOAEL for cadmium by inhalation in rats was  $25 \,\mu g/m^3 \text{ Cd} (0.025)$ 44  $mg/m^3$ ) and the NOAEL was not established. 45

# 1 **3.5.2. Mouse** 2

3 Male and female B6C3F<sub>1</sub> mice (6 wks old) were exposed to cadmium oxide aerosol (99.4% 4 purity;  $MMAD = 1.1-1.6 \mu gm$ ) for 6 hours/day, 5 days/week for 2 weeks (n = 5/sex/group) or 13 weeks (n= 10/sex/group) (NTP 1995). Animals were exposed to 0, 0.1, 0.3, 1, 3 or 10 mg/m<sup>3</sup> for 5 the 2 week studies and 0, 0.025, 0.05, 0.1, 0.25 or  $1 \text{ mg/m}^3$  for 13 weeks. Animals were exposed 6 in whole-body chambers that had a total volume of  $2.3 \text{ m}^3$ . Chemical concentration, airflow, 7 8 temperature and relative humidity were controlled and monitored with an automated system. 9 Overall, concentration within the chamber was adequate. Cadmium oxide dust was mixed with 10 compressed air to create an aerosol. The mass median aerodynamic diameter (MMAD) of the 11 aerosol particles was measured in each exposure chamber before the studies began, once in the 2 week study and monthly in the 13 week study. Blood was obtained to measure hematology and 12 13 clinical chemistry parameters, urine collected and histopathology performed.

14

Findings similar to those reported in rats (NTP, 1995) were observed in the mice. In the 2 15 week study, all mice in the 10  $mg/m^3$  group died and death was due to severe respiratory 16 toxicity. Hypoactivity, abnormal posture, rapid breathing, ataxia, nasal discharge, and ruffled fur 17 18 were observed at 1, 3, and 10 mg/m<sup>3</sup>. Absolute and relative lung weights were significantly increased at concentrations  $\ge 0.3$  mg/m<sup>3</sup>. Alveolar macrophage infiltration, fibrosis, focal 19 20 inflammation, and necrosis of alveolar duct epithelium were observed. As in the rats, severity of 21 effects increased with increasing concentration. Histopathologic lesions are listed in Table 8. In the 2 week study, a NOAEL could not be established and the LOAEL for mice was  $0.1 \text{ mg/m}^3$ 22 23 cadmium based on microscopic lung lesions.

24

TABLE 8. Histo	pathological	Findings in M	ice Exposed for	r 2 wks to Ca	dmium Oxid	e	
# Affected/Total # examined							
	$0 \text{ mg/m}^3$	0.1 mg/m <sup>3</sup>	$0.3 \text{ mg/m}^3$	1 mg/m <sup>3</sup>	$3 \text{ mg/m}^3$	10 mg/m <sup>3</sup>	
		MALI	ES				
Lung							
Alveolar infiltrate	0/5	5/5 (1.2) <sup>a</sup>	5/5 (2.0)	5/5 (3.0)	5/5 (3.0)	0/5	
Focal inflamm/fibrosis	0/5	0/5	5/5	5/5	5/5	0/5	
Necrosis	0/5	0/5	1/5 (1.0)	5/5 (2.0)	5/5 (2.2)	5/5 (3.0)	
Acute inflammation	0/5	0/5	0/5	0/5	5/5	5/5 (4.0)	
Nose							
Olfactory epithelium							
Degeneration	0/5	0/5	0/5	5/5 (1.4)	5/5 (3.0)	3/5 (3.0)	
Tracheobronchial lymph node							
hyperplasia	0/5	3/5 (1.0)	5/5 (1.0)	5/5 (2.0)	5/5 (2.0)	1/5 (3.0)	
		FEMAI	LES				
Lung							
Alveolar infiltrate	0/5	5/5 (1.0)	5/5 (1.8)	5/5 (3.0)	5/5 (3.0)	0/5	
Focal inflamm/fibrosis	0/5	0/5	5/5	5/5	5.5	0/5	
Necrosis	0/5	0/5	1/5 (1.0)	5/5 (2.0)	5/5 (2.0)	5/5 (3.0)	
Acute inflammation	0/5	0/5	0/5	0/5	0/5	5/5 (4.0)	
Nose							
Olfactory epithelium							
Degeneration	0/5	0/5	0/5	5/5 (2.0)	5/5 (3.0)	5/5 (3.0)	
Tracheobronchial lymph node							
hyperplasia	0/5	4/5 (1.0)	4/5 (1.0)	4/4 (1.5)	5/5 (1.8)	0/4	

Data from Table 16, p. 69 in NTP 1995.

<sup>a</sup> Number in parenthesis is severity code: 1= minimal, 2= mild, 3= moderate and 4= marked

1

2 In the 13 week study, one mouse from the control group died with no deaths in the treated 3 animals. No clinical signs of toxicity were observed. Lung lesions were similar to those of the

4 rat and consisted of macrophage infiltrates, hyperplasia, inflammation and fibrosis, although

5 fibrosis was more prevalent in the rat as shown in Table 9. The absolute and relative lung weight

6 of both sexes, absolute and relative kidney and thymus weights in male rats, and absolute and

- 7 relative kidney, liver, and spleen weights in female rats in all treatment groups were increased
- 8 compared to control weights. No microscopic changes were found in the liver, kidney, spleen, or
- 9 thymus. In the 13 week study, a NOAEL could not be established and the LOAEL for mice was
- 10  $0.025 \text{ mg/m}^3$  cadmium based on microscopic lung lesions.
- 11

TABLE 9. Histopathological Findings in Mice Exposed for 13 wks to Cadmium Oxide							
		(# Affected/]	Fotal # examine	ed)			
	0						
	mg/m <sup>3</sup>	$0.025 \text{ mg/m}^3$	0.05 mg/m <sup>3</sup>	0.1 mg/m <sup>3</sup>	$0.25 \text{ mg/m}^3$	1 mg/m <sup>3</sup>	
		Μ	IALES				
Lung							
Alveolar infiltrate	0/10	$9/10(1.1)^{a}$	10/10 (1.0)	10/10 (2.0)	10/10 (2.0)	10/10 (3.0)	
Alveolar hyperplasia	0/10	1/10 (1.0)	10/10 (1.0)	10/10 (1.8)	10/10 (1.8)	10/10 (2.0)	
Inflammation	0/10	0/10	0/10	8/10 (3.0)	10/10 (2.2)	10/10 (2.7)	
Fibrosis	0/10	0/10	2/10 (1.0)	10/10 /1.0)	10/10 (1.0)	10/10 (1.0)	
Tracheobronchial lymph node							
hyperplasia	0/6	0/8	4/9 (1.0)	9/9 (2.3)	8/10 (2.4)	9/10 (2.7)	
Larynx squamous metaplasia	0/9	10/10 (1.0)	10/10 (1.0)	10/10 (1.0)	10/10 (1.0)	9/10 (1.0)	
Nose							
Olfactory epithelium							
Degeneration	0/10	0/10	0/10	4/10 (1.0)	10/10 (1.7)	10/10 (2.0)	
Respiratory epithelium							
hyaline droplets	0/10	0/10	0/10	0/10	2/10 (1.0)	10/10 (1.0)	
		FE	MALES				
Lung							
Alveolar infiltrate	0/10	9/10 (1.0)	10/10 (1.0)	10/10 (2.0)	10/10 (2.0)	10/10 (3.0)	
Alveolar hyperplasia	0/10	0/10	0/10	10/10 (1.4)	10/10 (2.0)	10/10 (2.0)	
Inflammation	0/10	0/10	0/10	6/10 (2.3)	8/10 (2.1)	8/10 (2.9)	
Fibrosis	0/10	0/10	1/10 (1.0)	10/10 (1.0)	10/10 (1.0)	10/10 (1.0)	
Tracheobronchial lymph node							
hyperplasia	0/6	0/6	2/9 (1.0)	8/9 (1.5)	9/10 (2.0)	10/10 (2.4)	
Larynx squamous metaplasia	0/10	10/10 (1.0)	10/10 (1.0)	10/10 (1.0)	10/10 (1.0)	10/10 (1.0)	
Nose							
Olfactory epithelium							
Degeneration	0/10	0/10	0/10	1/10 (1.0)	10/10 (1.0)	10/10 (1.0)	
Respiratory epithelium							
hyaline droplets	0/10	0/10	0/10	0/10	2/10 (1.0)	10/10 (1.0)	

Data from Table 19, p. 76 in NTP 1995.

<sup>a</sup> Number in parenthesis is severity code: 1= minimal, 2= mild, 3= moderate and 4= marked

- The repeat dose studies are summarized in Table 10.

TABLE 10.         Summary of Repeat Dose Inhalation Data in Laboratory Animals					
Species	Concentration	Exposure	Effect	Reference	
species	(mg/m³)	Time		Reference	
	CdO aerosol	6 hr/d	All: Alveolar histiocytic infiltrate, pulmonary		
	0.1	5 d/wk	focal inflammation, pulmonary fibrosis		
	0.3	2 wk	0.1: LOAEL		
Rat	1		$\geq$ 1: Hypoactivity, dehydration, dyspnea, nasal	NTP 1995	
	3		discharge, olfactory epithelium degeneration,		
	10		inflammation		
			10: 100% mortality		
	CdO aerosol	6 hr/d	0.025: NOAEL		
	0.025	5 d/wk	0.05: LOAEL		
	0.05	13 wk	$\geq$ 0.05: $\uparrow$ organ weight, pulmonary fibrosis,		
	0.1		alveolar infiltrate		
-	0.25		$\geq 0.1$ : Tracheobronchial lymph node inflammation,		
Rat	1		respiratory epithelium inflammation, pulmonary	NTP 1995	
			fibrosis, alveolar infiltrate		
			$\geq 0.25$ : Iracheobronchiai lymph node		
			infiltrate requirestery arithalium inflormation		
			alfactory epithelium deconcretion		
	0.2 C4C12	6 hr/d	$0.2 \text{ CdCl}$ : $\uparrow$ absolute and relative lung weight		
		0 m/a	$0.5 \text{ CdCl}_2$ .   absolute and relative lung weight		
Pat	0.2	10 u	8.0 CuS.   absolute lung weight	Klimisch	
Rat	1.0			1993	
	8.0				
	CdO aerosol	5 hr/d	0.02: LOAEL		
	0.02	5 d/wk	All: $\uparrow$ increased length of estrous (0.02, 0.16 at end	D (1) 1	
Rat	0.16	20 wk	of the study)	Baranski and	
	1		1: 15% mortality at 7-8 wk, 38% mortality at 13-14	Sitarek 1987	
			wk, ↓ body weight		
	mg Cd/m <sup>3</sup>	Continuous	All: Focal kidney tubuli swelling, emphysematic		
	0.025	63 d	areas and cell proliferation in bronchi		
Dat	0.052		0.025: LOAEL, ↑ lung weight,	Drigge 1078	
Kat	0.105		$0.052$ : $\downarrow$ body weight, $\uparrow$ lung weight,	Tingge 1978	
			0.105: 42% mortality between days 45 and 60, $\downarrow$		
			body weight, $\downarrow pO_2$ , $\uparrow pCO_2$ , $\uparrow lung weight$		
	CdO aerosol	6 hr/d	0.1: LOAEL		
	0.1	5 d/wk	$\geq 0.1$ : Alveolar infiltrate, tracheobronchial lymph		
	0.3	2 wk	node hyperplasia		
Mouse	1		$\geq 0.3$ : $\uparrow$ absolute and relative lung weight, alveolar	NTP 1995	
	3		auct epithelium necrosis		
	10		21: Hypoactivity, abnormal posture, ataxia, rapid broothing, alfactory anithalium deconcration		
			10: 100% mortality severe respiratory toxicity		
	CdO aerosol	6 hr/d	$0.025 \cdot I \cap AEI$ alveolar infiltrate larvay		
	0.025	5 d/wk	squamous metaplasia		
	0.05	13 wk	> 0.05 alveolar infiltrate larvnx squamous		
Mouse	0.1	15 WK	metaplasia, tracheobronchial lymph node	NTP 1995	
1.10000	0.25		hyperplasia		
	1		$\geq 0.1$ : Olfactory epithelium degeneration		
			$\geq$ 0.25: Respiratory epithelium hyaline droplets		

### 1 **3.6.** Chronic Toxicity/Carcinogenicity

2 3 As a follow-up to the Takenaka et al. (1983) study described below Oldiges et al. (1989) 4 exposed thirty four groups of twenty male and female SPF Wistar rats/sex/group (8 wks old) to a 5 variety of Cd compounds in aerosol, dust or fume form (Table 11). The animals were exposed 6 22 hours a day, 7 days a week for 18 months. A few groups had discontinuous exposure for 40 7 hours a week for 6 months. Inhalation and observation periods were terminated at mortality rates 8 of more than 25% during the inhalation period and 75% during the observation period to assure a 9 carcinogenic result. In addition, an aerosol combination with the antagonistic zinc oxide aerosol was used in some of the exposures. Results were similar to those from the Takenaka et al. (1983) 10 11 study in that no primary lung tumors were identified in the control rats, but were identified in the 12 Cd exposed animals. Lung tumors rats were increased for all Cd compounds. The inhalation 13 period was shorter for the water insoluble Cd compounds CdS and CdO because of mortality, but 14 primary lung tumors were still observed in the rats of these groups.

### INTERIM: Sep-2010

-	
	L
-	

Alfecter out		TABLE 11. Observations after Inhalation Exposures of Rats to Various Cd Compounds								
Ormona No.Concentratio (mg/m)Duration (mg/m)Image (mg/m)Permionis (mg/m)Image (mg/m)Permionis (mg/m)ICompout (mg/m)-No00 <td></td> <td></td> <td></td> <td># Affected/</td> <td>Fotal # exa</td> <td>mined</td> <td></td> <td></td> <td></td> <td></td>				# Affected/	Fotal # exa	mined				
	Crown		Concentration	Duration	(month)	Lung	P	rimary lu	ung tum	ors <sup>a</sup>
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	No.	Compound	$(mg/m^3)$	Exposure	Study	Nodules	A	В	С	D
Males           1         Control         -         -         31         0/20         0         0         0           3         CdCl <sub>2</sub> 0.03         18         30 <sup>b</sup> 18/20         2         12         0         1           5         CdCL <sub>2</sub> 0.09         6         30 <sup>b</sup> 12/20         3         5         3         0           7         CdSO <sub>4</sub> 0.09         14 <sup>b</sup> 31         10'20         2         2         11/20           9         CdS         0.09         18         30 <sup>b</sup> 16/20         4         9         2         2           11         CdS         0.27         16 <sup>b</sup> 30 <sup>b</sup> 11/19         1         8         2         3           13         CdS         0.81         7 <sup>b</sup> 30 <sup>b</sup> 6'16         1         2         1         3           17         CdS         0.27         6         27 <sup>b</sup> 2/20         3         0         0         0         0         1/20         1/20         1/20         1/20         1/20         1/20         1/20         1/20         1/20         1/20 <td></td> <td></td> <td></td> <td>Exposure</td> <td>Study</td> <td>(<b>n</b>/<b>n</b>)</td> <td></td> <td>n</td> <td>n/n</td> <td></td>				Exposure	Study	( <b>n</b> / <b>n</b> )		n	n/n	
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$		1	1	]	Males	1	•			r
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	1	Control	—	-	31	0/20	0	0	0	0
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$								0.	/20	
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	3	CdCl <sub>2</sub>	0.03	18	30 <sup>b</sup>	18/20	2	12	0	1
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $					h			15	5/20	-
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	5	$CdCL_2$	0.09	6	30°	12/20	3	5	3	0
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	7	C4EO	0.00	1.4 <sup>b</sup>	21	10/20	2		0	2
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	/	$CusO_4$	0.09	14	51	10/20	2	11	U/20	Z
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	9	CdS	0.09	18	30 <sup>b</sup>	16/20	4	9	2	2
$\begin{array}{c c c c c c c c c c c c c c c c c c c $		Cub	0.07	10	50	10/20		17	7/20	2
Image: Constraint of the second se	11	CdS	0.27	16 <sup>b</sup>	30 <sup>b</sup>	11/19	1	8	2	3
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$								14	1/19	
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	13	CdS	0.81	7 <sup>b</sup>	30 <sup>b</sup>	7/20	2	5	2	2
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	15	Cub	0.01	,	50	1120	2	11	/20	2
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	15	CdS	2.42	1 <sup>b</sup>	20 <sup>b</sup>	6/16	1	2	1	2
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	15	Cus	2.45	4	50	0/10	1	2 7	/16	3
$\begin{array}{ c c c c c c c c c c c c c c c c c c c$	17	CdS	0.27	6	27 <sup>b</sup>	2/20	3	0	0	0
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	17	Cub	0.27	0	27	2,20	5	3	/20	Ŭ
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	19	CdO dust	0.03	18	31	15/20	4	6	1	4
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$								15	5/20	
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	21	CdO dust	0.09	7 <sup>b</sup>	31 <sup>b</sup>	11/15	2	5	0	2
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $								9.	/17	
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	23	CdO dust	0.09	6	31	8/20	1	2	1	0
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$		G 10, 1	0.02	10	ach	10/10	-	4	/20	
$ \begin{array}{ c c c c c c c c } \hline & & & & & & & & & & & & & & & & & & $	25	CdO dust	0.03	18	29°	13/18	5	12	0	I
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$					h			13	0/18	
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	26	CdO dust	0.01	18	29	13/20	6	5	0	1
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$								12	2/20	r
$ \begin{array}{ c c c c c c c } \hline & & & & & & & & & & & & & & & & & & $	27	CdO dust	0.03	18	31	1/19	0	0	0	0
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $								0.	/19	
$ \begin{array}{ c c c c c c c c c c c c } \hline & & & & & & & & & & & & & & & & & & $	28	CdO dust	0.03	18	31	1/19	0	0	0	0
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$								0	/19	
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	29	CdO dust	0.03	18	31	8/19	2	1	0	0
$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	20	G 10, 1, 4	0.00	10	21	5 (1 <b>7</b>	2	3,	/20	0
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	30	CdO dust	0.09	18	31	5/17	3	1	0	0
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	31	CdO/7nO		18	31	1/20	0	4	/1/	0
$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	51	dust		10	51	1/20	0	0	0	0
$\begin{array}{c c c c c c c c c c c c c c c c c c c $	22	CdO/7nO		19	21	11/20	1	2	/20	2
Females         0/20         0 <th< td=""><td>55</td><td>dust</td><td></td><td>10</td><td>51</td><td>11/20</td><td>1</td><td>3</td><td>/20</td><td>2</td></th<>	55	dust		10	51	11/20	1	3	/20	2
$\begin{array}{c c c c c c c c c c c c c c c c c c c $		aust	<u> </u>	F	emales	I	1	0,	20	
$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	2	Control	_	-	31	0/20	0	0	0	0
4         CdCl <sub>2</sub> 0.03         18         31         15/20         4         7         0         2           13/18         15/20         13/18         15/20         13/18         15/20         13/18         15/20         13/18         15/20         13/18         15/20		-						0.	/20	
13/18	4	CdCl <sub>2</sub>	0.03	18	31	15/20	4	7	0	2
		-						13	3/18	

	TABLE 11. Observations after Inhalation Exposures of Rats to Various Cd Compounds								
# Affected/Total # examined									
Crown		Concentration	Duration	(month)	Lung	Р	rimary l	ung tum	ors <sup>a</sup>
No	Compound	$(mg/m^3)$	Fynosuro	Study	Nodules	Α	В	С	D
110.		(ing/in )	Exposure	Study	( <b>n</b> / <b>n</b> )		1	n/n	
6	CdCl <sub>2</sub>	0.09	6	29 <sup>b</sup>	6/18	0	1	2	0
							3	8/18	
8	$CdSO_4$	0.09	18	29 <sup>b</sup>	17/20	4	6	2	6
							1	8/20	
10	CdS	0.09	NR	NR	17/20	0	9	1	5
			,	,			1	5/20	
12	CdS	0.27	16 <sup>°</sup>	30 <sup>b</sup>	17/19	1	8	2	5
			h	h			1	6/19	
14	CdS	0.81	10 <sup>6</sup>	29 <sup>6</sup>	11/20	3	5	1	4
							1	3/20	
16	CdS	2.43	3 <sup>b</sup>	31	9/19	3	3	0	0
							e	5/19	
18	CdS	0.27	6	29 <sup>b</sup>	6/20	1	1	0	1
								8/20	
20	CdO dust	0.03	18	31	15/20	3	7	1	4
							1	5/20	
22	CdO dust	0.09	11 <sup>b</sup>	3 <sup>b</sup>	14/15	2	8	1	0
							1	1/16	
24	CdO dust	0.09	6	31	6/20	0	2	1	0
								8/20	
32	CdO/ZnO		18	31	4/20	0	0	0	0
	dust								
							-	0/20	
34	CdO/ZnO		18	31	11/20	1	3	1	2
	dust						7	7/20	

NR = not reported

Data from Oldiges et al. 1989

<sup>a</sup> Type of tumors: A- bronchioalveolar adenomas; B- adenocarcinomas; C- squamous cell tumors; D- combined forms

<sup>b</sup> The exposure and the study were stopped after 25% and 75% morality, respectively.

1 2 2

Forty male inbred Wistar rats/group (6 weeks old) were exposed to CdCl<sub>2</sub> aerosol at
concentrations of 0 (n=41), 12.5, 25 or 50 µg/m<sup>3</sup> for 23 hrs/day, 7 days/week for 18 months
(Takenaka et al. 1983). After exposure, animals were kept for another 13 months under normal
laboratory conditions. Equivalent concentrations were 0.0125, 0.025 or 0.050 mg/m<sup>3</sup>,
respectively. Exposures took place in a 225 L inhalation chamber and the aerosol was generated

by atomizing a solution of  $CdCl_2$  with an ultrasonic atomizer. Air flow through the atomizer was

9 700 L/min and the aerosol flow through the inhalation chamber was 80 L/min. The aerosol was

10 diluted at the lower concentrations with filtered air. Air samples were drawn through membrane

11 filters twice weekly from the intake and exhaust of the chamber to check chamber concentration.

12 The mass of Cd on the filters was determined by atomic absorption spectrometry. The mass

13 median diameter was 0.55 µm and the geometric standard deviation was 1.8. Animals were

14 weighed every 3 months and any animals dying were examined as soon as possible.

Histopathology was performed as well as measurement of the Cd content in the lungs, livers and
 kidneys.

During and after the exposure, there was no difference in body weight between the

3 4

5 control and treated rats. Mean survival time at  $0.050 \text{ mg/m}^3$  was slightly below that of the others 6 but the difference was not significant. Compared to the controls  $(0.03 \mu g/g)$ , the Cd 7 concentration in the lungs was higher in treated rats, 5.6  $\mu$ g/g in the low-dose group, 4.7  $\mu$ g/g in 8 the mid-dose group and  $10.4 \mu g/g$  in the high-dose group. Similar values were found in the liver, 9 and the kidney values were 0.3, 13.5, 16.4 and 33.6 µg/g Cd in the control, low-, mid- and highdose groups, respectively. After the 13 month waiting period, the study reported that 0/38 in the 10 control group; 6/39 (15.4%) at 0.0125 mg/m<sup>3</sup>, 20/38 (52.6%) at 0.025 mg/m<sup>3</sup>, and 25/35 (71.4%) 11 at 0.050 mg/m<sup>3</sup> had primary lung carcinomas. These tumors were identified as adenocarcinomas, 12 13 epidermoid carcinomas and combined epidermoid and adenocarcinomas. Some of the rats also 14 had metastases in the regional lymph nodes and the kidneys. Other tumors were noted but either lacked a dose response or did not appear to be treatment-related. Limited data were provided on 15

16 all other histopathological findings.

### 18 **3.7.** Summary

19

17

20 Cadmium, in various forms, caused respiratory irritation, pulmonary edema, rales, pneumonitis, lacrimation, increased alveolar macrophages, and death in rabbits and rats exposed 21 for 1-6 hours. Rats and mice exposed for 90 days or less exhibited pulmonary inflammation and 22 23 edema, pulmonary hyperplasia, and nasal and respiratory epithelium degeneration. Kidney 24 weight was also increased with limited microscopic changes observed. In carcinogenicity 25 studies, rats exposed to Cd had an increased incidence of primary lung carcinomas. 26 Reproductive studies in rats resulted in decreased weight of live fetuses and reduced ossification 27 of the pelvis and sternebrae. In mice, the number of pregnant mice decreased with exposure to 28 Cd and fetal body weight was decreased.

29

32

### 30 4. SPECIAL CONSIDERATIONS

### 31 **4.1. Metabolism and Disposition**

33 Cadmium can be absorbed by inhalation, oral and dermal routes of exposure; however, 34 dermal absorption is relatively insignificant (ATSDR 2008). In humans, acute oral exposure to 35 cadmium causes severe nausea, vomiting, diarrhea and salivation. Cadmium fumes may produce a metal fume fever, or at higher doses, pulmonary edema, inflammation and emphysema (U.S. 36 37 EPA, 1986). In humans, cadmium adsorption in the gastrointestinal tract has been reported to be 38 low, only about 3-8%, and dependent upon several blood and dietary factors. In contrast, 30-64% 39 of inhaled cadmium can be absorbed (Morrow, 2001). For the non-smoker, 95% of the total 40 cadmium intake comes from ingestion of terrestrial foods or meat from animals that ate plants 41 grown in cadmium-containing soil. For a smoker, 50% of cadmium intake is through cigarettes. 42 Based on a 50% absorption rate, a person smoking one pack a day will absorb about 1-3 µg 43 cadmium (Wittman and Hu 2002).

44

Järup et al. (1983) determined the half-life of cadmium in the blood of workers after exposure had ended. Five men were exposed to cadmium for 4-7 years and then followed up to 13 years after cessation of exposure. The estimated total inhaled amounts of cadmium during their exposures ranged from 399 to 865 mg x hr/m<sup>3</sup>. Two of the five men had developed renal

49 tubular damage. The best fit with the data was a two compartment model with the biological

1 half-life of cadmium in the blood being 75-128 days in the first compartment and 7.4 to 16.0 2 years in the second. 3

4 Once absorbed, cadmium is distributed to most tissues in the body but concentrates in the 5 liver and kidneys of all animals (ATSDR 2008). Once in the blood, cadmium is not known to 6 undergo any direct metabolic conversion, but the sulfhydryl groups in albumin and 7 metallothionein have a high affinity for cadmium so it can adhere to plasma proteins (mostly 8 albumin), plasma metallothionein or directly to the erythrocyte. Since cadmium is not easily 9 absorbed in the GI tract, most is excreted in the feces. In the lungs, some clearance occurs but 10 absorption can take place. In chronic exposure, approximately 1/3 of the cadmium in the body is 11 stored in the kidney and biological half-lives are on the order of 10-30 years (Wittman and 12 Hu 2002). 13

14 Absorbed cadmium is excreted from the body primarily in urine. The excretion rate is low and as stated above, the half-life in humans can be 10-30 years (U.S. EPA 1986; Wittman 15 16 and Hu 2002).

18 Hadley et al. (1980) dosed male Wistar rats by intratracheal instillation with 15  $\mu$ g <sup>109</sup>CdO having a particle size of approximately 1.0 µm. Rats were sacrificed 1, 2, 4, 6, 12, 24, 19 20 48, 168, and 336 hours after instillation. At approximately 6 hours post instillation, 50% instilled <sup>109</sup>CdO was no longer present in the lungs and 70% of that which had left the lungs was 21 found in the liver. At 24 hours, 80% instilled <sup>109</sup>CdO had left the lung. At the end of the 2 week 22 period, 20% instilled <sup>109</sup>CdO remained in the lungs. Approximately 8% of the instilled <sup>109</sup>CdO 23 was distributed to the kidney over 2 weeks, and the cumulative total eliminated in the urine and 24 25 feces was less than 10% of total body burden.

#### 27 4.2.

28

26

17

### **Mechanism of Toxicity**

29 Acute exposure to cadmium by the inhalation route has produced inflammatory cell 30 influx, pneumonitis, respiratory irritation, and pulmonary edema in humans, rabbits, and rats (Beton et al. 1966; Rusch et al. 1986; Grose et al. 1987). Chronic exposure to cadmium by the 31 32 oral or inhalation route has produced renal proximal tubule damage, proteinuria, polyuria and 33 glycosuria. Cadmium-induced renal injury initially presents as tubular proteinuria which can be 34 quantified by measurement of low molecular weight proteins such as  $\beta_2$ -microglobulin, retinol 35 binding protein and protein HC. With continued exposure, the progression continues and glomerular damage occurs with a characteristic decrease in the glomerular filtration rate. For the 36 37 most part, this damage is irreversible (Wittman and Hu 2002). Pneumonitis, inflammation, and fibrosis have been observed following chronic inhalation exposure (Prigge 1978; IPCS 1992; 38 39 Klimisch 1993). Very little is known, however, regarding the biochemical mechanisms by which 40 cadmium causes toxicity at the cellular level (U.S. EPA 1986).

41

43

#### 42 4.3. **Structure Activity Relationships**

- 44 No data were located.
- 45

1

2

3 4

5

6

7

### 4.4. Other Relevant Information

### 4.4.1. Species Variability

Acute exposure to cadmium caused pneumonitis, increased lung weight, and pulmonary inflammatory cell influx in rabbits and rats (Rusch et al. 1986; Buckley and Bassett 1987; Grose et al. 1987; Takenaka et al. 2004).

8 In a 4-week study, male Fisher 344 rats and Balb-c mice (number of animals not 9 specified) were exposed to  $CdCl_2$  aerosols at a concentration of 0 or 100 µg Cd/m<sup>3</sup> for 6 hours/day, 5 days/week to determine if the amount of metallothionein (MT) produced was 10 11 substantially different between the species (Oberdörster et al. 1994). The study found a significant species difference in the pulmonary response which may explain the pulmonary 12 13 carcinogenicity observed in long-term Cd exposure in rats, whereas there is none in mice. 14 Cadmium exposure significantly increased MT in both species in the total lung, persisting for a 28 day post-exposure period, however, the baseline MT level was higher in mice. Mice showed 15 16 a 3 fold increase in MT in the lavaged lung compared to rats, the increase in mice persisted for 17 28-days. Histochemical staining showed that the epithelial cells of mice in the conducting 18 airways and alveolar region had a greater induction of MT compared to those of rats, and that 19 mice had more effects on proliferative cells compared to rats. Overall, the study suggested 20 species difference in carcinogenicity susceptibility to Cd exposure and recommended further 21 human research to help determine the extent of MT induction in humans.

22

23 To help further distinguish species differences observed in pulmonary inflammation, 24 McKenna et al. (1997) exposed forty-three 15-week old male Wistar Furth rats, and sixty 17-25 week old mice/strain (C57 and DBA) to  $1.0 \text{ mg/m}^3$  (0.22 ppm) of CdO fumes for 3 hours by nose-only inhalation. Control animals were exposed to a mixture of air and argon. Animals were 26 27 sacrificed at 0, 1, 3 or 5 days post-exposure and control animals at one time point only. Overall, 28 there were interspecies and interstrain differences observed, further indicating the wide range of 29 effects that could be possible with human exposures. The main findings were that C57 mice had 30 higher cell proliferation in lung tissue and neutrophil influx in the bronchoalveolar lavage fluid (BALF) compared to DBA mice and Wistar rats, DBA mice had a higher percentage of Cd dose 31 in lung and higher levels of biochemical indices of injury in BALF, and rats responded to Cd 32 33 inhalation with a more transient response in BALF and a higher degree of acute inflammatory 34 lesions in the lung than mice.

35

### 4.4.2. Susceptible Populations

36 37

In acute inhalation studies, cadmium caused respiratory irritation and adverse pulmonary 38 39 effects, therefore, those with lung-associated diseases or conditions (such as asthma and COPD) 40 could be affected more. Children may be slightly more sensitive to cadmium by inhalation 41 exposure as the rate of respiration is usually higher than that of adults. Exposure to cadmium as 42 a child may also increase susceptibility to renal toxicity later in life (ATSDR 2008). Women 43 with low iron stores or iron deficiency typically show increased cadmium intake; however, very 44 little is transferred through the placenta or breast milk (Schoeters et al., 2006). Those with pre-45 existing renal and/or hepatic conditions would be more at risk in chronic exposures. Smokers would also be at more risk of toxicity as they are already receiving doses of cadmium from 46 47 cigarettes. Other factors found to affect susceptibility to cadmium include: diet, age, sex and 48 genetic background (U.S. EPA 1986). The factors may result in increased absorption, decreased 49 detoxification or excretion, or compromised organ function. People living close to sources of

cadmium pollution, especially in industrialized areas, are subjected to higher exposures than
 those in non-industrialized areas (ATSDR 2008).

### 4.4.3. Concentration-Exposure Duration Relationship

6 The concentration-exposure time relationship for many irritant and systemically-acting 7 vapors and gases may be described by  $C^n x t = k$ , where the exponent, n, ranges from 0.8 to 3.5 8 (ten Berge et al., 1986). To obtain conservative and protective AEGL values in the absence of an 9 empirically derived chemical-specific scaling exponent, temporal scaling was performed using 10 n=3 when extrapolating to shorter time points and n = 1 when extrapolating to longer time points 11 using the  $C^n x t = k$  equation.

12 13

14

17

20 21

22

4

5

### 4.4.4. Concurrent Exposure Issues

In occupational settings, workers are exposed to other metals in addition to cadmium.
 Smokers inhale cadmium along with other chemicals and compounds.

### 18 5. DATA ANALYSIS FOR AEGL-1

### 19 5.1. Summary of Human Data Relevant to AEGL-1

No data were located.

# 23 5.2. Summary of Animal Data Relevant to AEGL-1 24

Female Fischer 344 rats exposed (whole body) to 0.07 mg/m<sup>3</sup> Cd for 6 hours showed no 25 morphological changes in the lungs or inflammatory response (Takenaka et al. 2004). Rabbits 26 and rats exposed (nose-only) to  $0.25 \text{ mg/m}^3 \text{ CdCl}_2$  or CdO for 2 hours had inhibited pulmonary 27 28 GSH peroxidase activity (Grose et al. 1987). Rabbits and rats exposed (nose-only) to 0.45 mg/m<sup>3</sup> CdCl<sub>2</sub> or CdO for 2 hours had increased inflammatory response (Grose et al. 1987). 29 Deaths were observed in rats exposed to  $0.45 \text{ mg/m}^3$  but no deaths were observed in rats exposed 30 to 4.5 mg/m<sup>3</sup>, and those deaths may be associated with the exposure apparatus. The mortality 31 observed was also inconsistent with the data for rats exposed to similar concentrations. After 32 exposure to 0.5 mg/m<sup>3</sup> CdO for 3 hours, rats had increased inflammatory cell influx and transient 33 hypercellularity, but no mortality (Buckley and Bassett 1987). After whole-body exposure to 34 35  $0.55 \text{ mg/m}^3$  CdO for 6 hours, rats had increased neutrophils and multifocal alveolar 36 inflammation and no mortality (Takenaka et al. 2004).

37

# 38 5.3. Derivation of AEGL-139

The AEGL-1 values are based on the experimental concentration,  $0.55 \text{ mg Cd/m}^3$ , that 40 41 caused slight respiratory irritation in rats (Takenaka et al. 2004). After a 6 hour exposure, 42 increased neutrophils and multifocal alveolar inflammation were observed. At the next higher 43 experimental exposure, pneumonitis was observed (Grose et al. 1987). Although the exposure 44 was a whole-body exposure, the size of the ultrafine particles (51 nM MMAD, 1.7 GSD) would 45 mimic a gaseous state and the majority of the aerosol would be inhaled and not deposited on the fur. An interspecies uncertainty factor of 3 was applied because at acute exposures, cadmium is 46 a direct-acting respiratory irritant as indicated by the signs of irritation in rabbits and rats. This 47 48 mode of action is not expected to differ among species. Rabbits and rats exposed for 2 hours to 49  $0.25-4.5 \text{ mg/m}^3$  displayed similar histological and biochemical pulmonary effects including

- 1 pneumonitis, increased lung weight, pulmonary inflammatory cell influx, and decreased
- 2 glutathione peroxidase activity (Grose et al. 1987). Rats exposed to cadmium (0.00169-5.3
- mg/m<sup>3</sup>) from 1-6 hours (Buckley and Bassett 1987; Oberdörster et al. 1987; Takenaka et al.
   2004) exhibited the same effects as those observed in the Grose et al. (1987) study. An
- 5 intraspecies uncertainty factor of 3 was applied because at acute exposures, cadmium is a direct-
- 6 acting respiratory irritant in humans, and this mode of action is not expected to differ among
- 7 individuals. After a five hour exposure to cadmium, workers experienced cough, throat
- 8 irritation, dyspnea, and pulmonary edema (Beton et al. 1966) which are signs of respiratory
- 9 irritation. The concentration-exposure time relationship for many irritant and systemically-
- 10 acting vapors and gases may be described by  $C^n x t = k$ , where the exponent, n, ranges from 0.8
- 11 to 3.5 (ten Berge et al., 1986). To obtain conservative and protective AEGL values in the
- 12 absence of an empirically derived chemical-specific scaling exponent, temporal scaling was
- performed using n=3 when extrapolating to shorter time points and n = 1 when extrapolating to longer time points using the  $C^n x t = k$  equation. The 30-minute AEGL-1 value was adopted as
- 15 the 10-minute value due to the added uncertainty of extrapolating from a 6-hour time point to 10
- 16 minutes (NRC 2001). The calculations for the AEGL-1 values are in Appendix A.
- 17

TABLE 12. AEGL-1 Values for Cadmium						
10-min 30-min 1-hr 4-hr 8-hr						
$0.13 \text{ mg/m}^3$	$0.13 \text{ mg/m}^3$	$0.10 \text{ mg/m}^3$	$0.063 \text{ mg/m}^3$	$0.041 \text{ mg/m}^3$		

18 19

20

21

22 23

24 25

26

### 6. DATA ANALYSIS FOR AEGL-2

### 6.1. Summary of Human Data Relevant to AEGL-2

No human data relevant to AEGL-2 were located.

### 6.2. Summary of Animal Data Relevant to AEGL-2

Rabbits and rats exposed for 2 hours to 4.5 mg/m<sup>3</sup> CdCl<sub>2</sub> or CdO exhibited increased lung weight and moderate-severe pneumonitis characterized by alveolar wall thickening, hemorrhage, and edema (Grose et al. 1987). At 5.3 mg/m<sup>3</sup>, focal areas of interstitial thickening, an increase in cuboidal alveolar cells, numerous inflammatory cells (interstitial mononuclear cells, alveolar macrophages, eosinophils, and basophils), and increase in lung weight and protein content were observed (Buckley and Bassett 1987).

33

35

### 34 6.3. Derivation of AEGL-2

The AEGL-2 values are based on the experimental concentration,  $5.3 \text{ mg Cd/m}^3$ , that 36 caused overt respiratory irritation and pathology in rats (Buckley and Bassett 1987). The 3 hour 37 38 exposure resulted in reduced weight gain and increased lung weight, protein content, DNA 39 content, number of cuboidal alveolar cells, number of inflammatory cells, and focal areas of 40 interstitial thickening. An interspecies uncertainty factor of 3 was applied because at acute 41 exposures, cadmium is a direct-acting respiratory irritant as indicated by the signs of irritation in rabbits and rats. This mode of action is not expected to differ among species. Rabbits and rats 42 exposed for 2 hours to  $0.25-4.5 \text{ mg/m}^3$  displayed similar histological and biochemical pulmonary 43 44 effects including pneumonitis, increased lung weight, pulmonary inflammatory cell influx, and 45 decreased glutathione peroxidase activity (Grose et al. 1987). Rats exposed to cadmium (0.00169-5.3 mg/m<sup>3</sup>) from 1-6 hours (Buckley and Bassett 1987; Oberdörster et al. 1987; 46

1 Takenaka et al. 2004) exhibited the same effects as those observed in the Grose et al. (1987)

2 study. An intraspecies uncertainty factor of 3 was applied because at acute exposures, cadmium

3 is a direct-acting respiratory irritant in humans, and this mode of action is not expected to differ 4 among individuals. After a five hour exposure to cadmium, workers experienced cough, throat

among individuals. After a five hour exposure to cadmium, workers experienced cough, throa
 irritation, dyspnea, and pulmonary edema (Beton et al. 1966) which are signs of respiratory

6 irritation. The concentration-exposure time relationship for many irritant and systemically-

7 acting vapors and gases may be described by  $C^n x t = k$ , where the exponent, n, ranges from 0.8

8 to 3.5 (ten Berge et al. 1986). To obtain conservative and protective AEGL values in the absence

9 of an empirically derived chemical-specific scaling exponent, temporal scaling was performed

10 using n=3 when extrapolating to shorter time points and n = 1 when extrapolating to longer time

11 points using the  $C^n x t = k$  equation. The calculations for the AEGL-2 values are in Appendix 12 A.

12

TABLE 13. AEGL-2 Values for Cadmium							
10-min	10-min 30-min 1-hr 4-hr 8-hr						
$1.4 \text{ mg/m}^3$	$0.96 \text{ mg/m}^3$	$0.76 \text{ mg/m}^3$	$0.40 \text{ mg/m}^3$	$0.20 \text{ mg/m}^3$			

### 14 15

16

17

18

23

### 7. DATA ANALYSIS FOR AEGL-3

### 7.1. Summary of Human Data Relevant to AEGL-3

Workers (n=5) exposed for 5 hours to Cd oxide fumes inhaled an estimated 8.6 mg/m<sup>3</sup>.
One worker died 5 days after the exposure with severe pulmonary edema, alveolar metaplasia of
the lungs, and bilateral cortical necrosis of the kidneys (Beton et al. 1966). The other four
workers had pulmonary edema that resolved over time.

# 24 **7.2.** Summary of Animal Data Relevant to AEGL-325

26 Mortality occurred in rats exposed to cadmium carbonate (5.8%) or cadmium fume (48%) for 2 hours. The LC<sub>50</sub> for the study was  $112 \text{ mg/m}^3$  (Rusch et al. 1986). One rat exposed 27 to 0.45 mg/m<sup>3</sup> Cd died of cardiovascular failure associated with pulmonary congestion (Grose et 28 al. 1987). Two other rats died of unknown causes in the same study. Although death occurred at 29 30  $0.45 \text{ mg/m}^3$ , this value was not used to derive AEGL-3. The deaths may be associated with the 31 exposure apparatus and may not be the result of exposure to cadmium. Lack of mortality in rats exposed to similar concentrations (0.5 mg/m<sup>3</sup>, Buckley and Bassett 1987; 0.55 mg/m<sup>3</sup>, Takenaka 32 33 et al. 2004) and at a higher experimental dose within the same study,  $4.5 \text{ mg/m}^3$ , further support 34 the dismissal of the mortality data from Grose et al. (1987) from being considered for derivation 35 of AEGL-3 values.

### 36 37

38

### 7.3. Derivation of AEGL-3

39 The AEGL-3 values are based on the 2 hour  $LC_{50}$  for cadmium fume in rats, 112 mg/m<sup>3</sup> 40 (Rusch et al. 1986). The  $LC_{50}$  was divided by 3 to estimate a threshold of lethality. An 41 intraspecies uncertainty factor of 3 was applied because in acute exposures, cadmium is a direct-42 acting respiratory irritant. An interspecies uncertainty factor of 3 was applied because at acute 43 exposures, cadmium is a direct-acting respiratory irritant as indicated by the signs of irritation in 44 rabbits and rats. This mode of action is not expected to differ among species. Rabbits and rats exposed for 2 hours to 0.25-4.5 mg/m<sup>3</sup> displayed similar histological and biochemical pulmonary 45 effects including pneumonitis, increased lung weight, pulmonary inflammatory cell influx, and 46

decreased glutathione peroxidase activity (Grose et al. 1987). Rats exposed to cadmium 1

2  $(0.00169-5.3 \text{ mg/m}^3)$  from 1-6 hours (Buckley and Bassett 1987; Oberdörster et al. 1987;

- 3 Takenaka et al. 2004) exhibited the same effects as those observed in the Grose et al. (1987) 4 study. An intraspecies uncertainty factor of 3 was applied because at acute exposures, cadmium
- 5 is a direct-acting respiratory irritant in humans, and this mode of action is not expected to differ
- 6 among individuals. After a five hour exposure to cadmium, workers experienced cough, throat
- 7 irritation, dyspnea, and pulmonary edema (Beton et al. 1966) which are signs of respiratory
- 8 irritation. The concentration-exposure time relationship for many irritant and systemically-
- 9 acting vapors and gases may be described by  $C^n x t = k$ , where the exponent, n, ranges from 0.8
- to 3.5 (ten Berge et al. 1986). To obtain conservative and protective AEGL values in the absence 10
- 11 of an empirically derived chemical-specific scaling exponent, temporal scaling was performed
- 12
- using n=3 when extrapolating to shorter time points and n=1 when extrapolating to longer time 13 points using the  $C^n x t = k$  equation. The calculations for the AEGL-3 values are in Appendix A.
- 14

TABLE 14. AEGL-3 Values for Cadmium						
10-min 30-min 1-hr 4-hr 8-hr						
$8.5 \text{ mg/m}^3$	$5.9 \text{ mg/m}^3$	$4.7 \text{ mg/m}^3$	$1.9 \text{ mg/m}^3$	$0.93 \text{ mg/m}^3$		

15 16

17

19

#### 8. **SUMMARY OF AEGLS**

#### **AEGL Values and Toxicity Endpoints** 18 8.1.

20 Cadmium was shown to be an irritating metal that caused developmental, respiratory, and 21 renal effects following multiple exposures. The derived AEGL values would prevent these effects. The AEGL-1 values were based upon the experimental concentration that caused slight 22 respiratory irritation in rats, 0.55 mg Cd/m<sup>3</sup>, following a 6 hour exposure. The AEGL-2 values 23 24 were based upon the experimental concentration that caused overt respiratory tract irritation and 25 pathology in rats,  $5.3 \text{ mg/m}^3$ , following a 3 hour exposure. The AEGL-3 values were a based 26 upon the estimated threshold of lethality from cadmium fumes from the 2-hour LC<sub>50</sub>, 112 mg/m<sup>3</sup>, 27 in rats. AEGL values are summarized in Table 15.

28

TABLE 15. Summary of AEGL Values							
Classification			Exposure Duration	on			
Classification	10-min	30-min	1-hr	4-hr	8-hr		
AEGL-1 (Notable Discomfort)	0.13 mg/m <sup>3</sup>	0.13 mg/m <sup>3</sup>	0.10 mg/m <sup>3</sup>	0.063 mg/m <sup>3</sup>	0.041 mg/m <sup>3</sup>		
AEGL-2 (Disabling)	1.4 mg/m <sup>3</sup>	0.96 mg/m <sup>3</sup>	0.76 mg/m <sup>3</sup>	$0.40 \text{ mg/m}^3$	$0.20 \text{ mg/m}^3$		
AEGL-3 (Lethal)	8.5 mg/m <sup>3</sup>	5.9 mg/m <sup>3</sup>	4.7 mg/m <sup>3</sup>	$1.9 \text{ mg/m}^3$	0.93 mg/m <sup>3</sup>		

29 30

32

#### 31 8.2. **Comparison with Other Standards and Guidelines**

33 AEGL values for cadmium are compared to other guidelines and standards in Table 16. The

34 OSHA values were established to protect against lung cancer and kidney dysfunction (OSHA

2005). The IDLH was based upon acute inhalation toxicity in workers (NIOSH 1996). The 35

36 ACGIH TWA inhalable particulate value was established to minimize kidney dysfunction and

the respirable fraction value was set to minimize pulmonary accumulation of cadmium that could act as a carcinogen (ACGIH 1996).

2 3

1

	TABLE 16.	Extant Standards	and Guidelines fo	r Cadmium	
Guideline			<b>Exposure Duratio</b>	n	
Guidenne	10 min	30 min	1 hr	4 hr	8 hr
AEGL-1	$0.13 \text{ mg/m}^3$	$0.13 \text{ mg/m}^3$	$0.10 \text{ mg/m}^3$	$0.063 \text{ mg/m}^3$	$0.041 \text{ mg/m}^3$
AEGL-2	$1.4 \text{ mg/m}^3$	$0.96 \text{ mg/m}^3$	$0.76 \text{ mg/m}^3$	$0.40 \text{ mg/m}^3$	$0.20 \text{ mg/m}^3$
AEGL-3	$8.5 \text{ mg/m}^3$	$5.9 \text{ mg/m}^3$	$4.7 \text{ mg/m}^{3}$	$1.9 \text{ mg/m}^3$	$0.93 \text{ mg/m}^3$
PEL-TWA					$0.1 \text{ mg/m}^{3}$
(OSHA) <sup>a</sup> (fume)					$0.3 \text{ mg/m}^{3C}$
PEL-TWA					$0.2 \text{ mg/m}^3$
$(OSHA)^{b}(dust)$					$0.6 \text{ mg/m}^{3C}$
IDLH (NIOSH) <sup>c</sup>	$9 \text{ mg/m}^3$				
TLV-TWA					$0.01 \text{ mg/m}^{31}$
(ACGIH) <sup>d</sup>					$0.002 \text{ mg/m}^{3R}$
MAC-Peak					
Category (The					$0.005 \text{ mg/m}^3$
Netherlands) <sup>e</sup>					

<sup>b</sup> OSHA PEL-STEL (Permissible Exposure Limits - Short Term Exposure Limit) (OSHA, 2005) is defined analogous to the ACGIH-TLV-STEL. <sup>C</sup>-Acceptable ceiling concentration.

than 10 hours/day, 40 hours/week. <sup>C</sup>-Acceptable ceiling concentration.

<sup>a</sup> OSHA PEL-TWA (Occupational Health and Safety Administration, Permissible Exposure Limits - Time Weighted Average) (OSHA, 2005) is defined analogous to the ACGIH-TLV-TWA, but is for exposures of no more

- <sup>c</sup> IDLH (Immediately Dangerous to Life and Health, National Institute of Occupational Safety and Health) (NIOSH, 1996) represents the maximum concentration from which one could escape within 30 minutes without any escape-impairing symptoms, or any irreversible health effects.
- <sup>d</sup> ACGIH TLV-TWA (American Conference of Governmental Industrial Hygienists, Threshold Limit Value Time Weighted Average) (ACGIH, 2008) is the time-weighted average concentration for a normal 8-hour workday and a 40-hour work week, to which nearly all workers may be repeatedly exposed, day after day, without adverse effect. <sup>I</sup>- Inhalable particulate; <sup>R</sup>- Respirable fraction
- <sup>e</sup>MAC (Maximaal Aanvaaarde Concentratie [Maximal Accepted Concentration Peak Category]) (SDU Uitgevers [under the auspices of the Ministry of Social Affairs and Employment], The Hague, The Netherlands 2000) is defined analogous to the ACGIH-Ceiling.

### 8.3. Data Adequacy and Research

Toxicity data for cadmium are available for humans and animals. Most of the acute inhalation data for humans do not provide cadmium exposure concentrations, but report signs and symptoms of toxicity which are useful for noting effects of exposure. Short-term and chronic epidemiological studies in workers were available; however, the workers may have had concurrent exposures to other chemicals. Animal studies were conducted in at least three species and range from acute to chronic. The acute studies in rabbits and rats provide data suitable for deriving AEGL values and highlight the differences between the effects that occur following acute and chronic cadmium exposure.

1 2	9.	REFERENCES
3		
4 5 6 7	ACGIH	(American Conference of Governmental Industrial Hygienists). 1996. 1996 Supplement to the 6 <sup>th</sup> Edition. Documentation of the Threshold Limit Values and Biological Exposure Indices: Cadmium and Compounds. Seventh ed., ACGIH, Cincinnati, OH.
8 9 10	ACGIH	(American Conference of Governmental Industrial Hygienists). 2008. Threshold limit values (TLVs) for chemical and physical agents and biological exposure indices (BEIs). ACGIH, Cincinnati, OH.
12 13 14	ATSDR	R (Agency for Toxic Substances and Disease Registry). 2008. Draft toxicological profile for cadmium. U.S. Department of Health and Human Services, Atlanta, GA.
15 16 17	Barańsk	ki, B. and K. Sitarek . 1987. Effect of oral and inhalation exposure to cadmium on the oestrous cycle in rats. Toxicol. Lett. 36: 267-273.
18 19 20	Beton, I	D.C., G. S. Andrews, H.J. Davies, L. Howells and G.F. Smith. 1966. Acute cadmium fume poisoning. Five cases with one death from renal necrosis. Brit. J. Industr. Med 23: 292-301.
20 21 22 23	Buckley	y, B.J. and D.J.P. Bassett. 1987 Pulmonary cadmium oxide toxicity in the rat. J. Toxicol. Environ. Health. 21: 233-250.
23 24 25 26	Crump,	K.S. and R.B. Howe. 1984. The multistage model with a time-dependent dose pattern: Applications to carcinogenic risk assessment. Risk. Anal. 4: 163-176.
20 27 28 20	Fernánc	lez, M.A., P. Sanz, M. Palomar, J. Serra and E. Gadea. 1996. Fatal chemical pneumonitis due to cadmium fumes. Occup. Med. 46: 372-374.
29 30 31 32 33	Grose, I	E.C., J.H. Richards, R.H. Jaskot, M.G. Ménache, J.A. Graham and W.C. Dauterman. 1987. A comparative study of the effects of inhaled cadmium chloride and cadmium oxide: pulmonary response. J. Toxicol. Environ. Health, 21:219-232.
34 35 36 37	Haber, I	F.R. 1924. Zur geschichte des gaskrieges [On the history of the gas war]. In: Fuenf Vortraege aus den Jahren 1920-23 [Five lectures from the years 1920-1923]. Berlin, Germany: Verlag von Julius Springer; pp. 76-92.
38 39 40	Hadley,	J.G., A.W. Conklin, and C.L. Sanders. 1980. Rapid solubilization and translocation of <sup>109</sup> CdO following pulmonary deposition. Toxicol. Appl. Pharmacol. 54: 156-160.
41 42 43	HSDB	(Hazardous Substances Data Bank). 2005. Cadmium. TOXNET, Toxicology Data Network. US. Natl. Library of Medicine. Available Online. http://toxnet.nlm.nih.gov/.
44 45 46 47	IARC (	International Agency for Research on Cancer). 2003. IARC Monographs on the evaluation of the carcinogenic risk of chemicals to humans. Cadmium and Cadmium Compounds. Vol: 58: 119-237.
48 49 50 51	IPCS (I	nternational Programme on Chemical Safety). 1992. Environmental Health Criteria 134 Cadmium. Available online. http://www.intox.org/databank/documents/chemical/cadmium/ehc134.htm.

- 1 Jakubowski, M., A. Abramowska-Guzik, W. Szymczak and M. Trzcinka-Ochocka. 2004. Influence of 2 3 4 5 long-term occupational exposure to cadmium on lung function tests results. Int. J. Occup. Med. Environ. Health 17: 361-368 Järup. L., A. Rogenfelt, C-G Elinder, K. Nogawa and T. Kjellström. 1983. Biological half-time of 6 7 cadmium in the blood of workers after cessation of exposure. Scand. J. Work Environ. Health 9:327-331. 8 9 Kjellström, T., L. Friberg and B. Rahnster. 1979. Mortality and cancer morbidity among cadmium-10 exposed workers. Environ. Health Perspect. 28:199-204. 11 12 Klimisch, H.J. 1993. Lung deposition, lung clearance and renal accumulation of inhaled cadmium 13 chloride and cadmium sulphide in rats. Toxicology 84:103-124. 14 15 Kutzman, R.S., R.T. Drew and R.N. Shiotsuka. 1986. Pulmonary changes resulting from subchronic 16 exposure to cadmium chloride aerosol. J. of Toxicol. Environ. Health 17: 175-189. 17 18 Lauwerys, R., H. Roels, M. Refniers, J.P. Buchett, A. Bernard and A. Goret. 1979. Significance of 19 cadmium concentration in blood and in urine in workers exposed to cadmium. Environ. Res. 20: 20 375-391. 21 22 McKenna, I.M., M.P. Waalkes, L.C. Chen and T. Gordon. 1997. Comparison of inflammatory lung 23 responses in Wistar rats and C57 and DBA mice following acute exposure to cadmium oxide 24 fumes. Toxicol. Appl. Pharmacol. 146:196-207. 25 26 Morrow, H. 2001. Cadmium and Cadmium Alloys. Kirk-Othmer Encyclopedia of Chemical 27 Technology, 4<sup>th</sup> Ed., M. Howe-Grant, ed. Available. Online. 28 http://mrw.interscience.wiley.com/emrw/9780471238966/kirk/article/cadmcarr.a01/current/abstra 29 ct?hd=All,cadmium&hd=All,cadmium&hd=All,alloys. 30 31 NIOSH (National Institute for Occupational Safety and Health). 1996. Documentation 32 for Immediately Dangerous to Life and Health Concentrations (IDLH). Available Online. 33 http://www.cdc.gov/niosh/idlh/7440439.html. 34 35 NIOSH (National Institute for Occupational Safety and Health). 2005. NIOSH Pocket 36 Guide to Chemical Hazards. NIOSH, Cincinnati, OH. Available Online. 37 http://www.cdc.gov/niosh/npgd0088.html. 38
- Nishijo, M., H. Nakagawa, R. Honda, K. Tanebe, S. Saito, H. Teranishi and K. Tawara. 2002. Effects of
   maternal exposure to cadmium on pregnancy outcome and breast milk. Occup. Environ. Med.
   59:394-397.
- 43 NRC (National Research Council). 2001. Standing Operating Procedures for Developing Acute
   44 Exposure Guideline Levels for Hazardous Chemicals. The National Academies Press,
   45 Washington, DC.
   46
- 47 NRC (National Research Council). 1985. Emergency and Continuous Exposure Guidance Levels for
   48 Selected Airborne Contaminants, Vol 5. The National Academies Press, Washington, DC.
   49

# NTP (National Toxicology Program). 1995. NTP technical report on toxicity studies of cadmium oxide administered by inhalation to F344/N rats and B6C3F<sub>1</sub> mice. NIH Publication 95-3388. U.S. Department of Health and Human Services, Public Health Service, National Institutes of Health.

1

8

9 10

11

12 13

14

15 16

17

18

20 21

22

23 24

25

26

27 28

29

30

31 32

33

- Oberdörster, G., M.G. Cherian and R.B. Baggs. 1994. Importance of species difference in experimental pulmonary carcinogenicity of inhaled cadmium for extrapolation to humans. Toxicol. Lett. 72:339-343.
- Oberdörster, G., D. Hochrainer and C. Cox. 1987. Acute pulmonary toxicity of cadmium compounds: dependence on physico-chemical form. Toxicology of Metals: Clinical and Experimental Research. S.S. Brown and Y. Kodama., eds. West Sussex, England: Ellis Horwood Limited. pp. 319-320.
- Oldiges, H., D. Hochrainer and U. Glaser. 1989. Long-term inhalation study with Wistar rats and four cadmium compounds. Toxicol. Environ. Chem. 19:217-222.
- OSHA (Occupational Safety and Health Administration). 2005. Code of Federal Regulations, CFR 29, Part 1910, Table Z-2.
- Panchal, L. and P. Vaideeswar. 2006. Acute lung injury due to cadmium inhalation- a case report. Indian J. Pathol. Microbiol. 49: 265-266.
- 19 Prigge, E. 1978. Early signs of oral and inhalative cadmium uptake in rats. Arch. Toxicol. 40: 231-247.
  - Rinehart, W. E., Hatch, T. 1964. Concentration-time product (CT) as an expression of dose in sublethal exposures to phosgene. Ind. Hyg. J. 25: 545-553.
  - Rusch, G.M., J.S. O'Grodnick and W.E. Rinehart. 1986. Acute inhalation study in the rat of comparative uptake, distribution and excretion for different cadmium containing materials. Am. Ind. Hyg. Assoc. J. 47(12): 754-763.
  - Schoeters, G., E. Den Hond, M. Zuurbier, R. Naginiene, P. Van Den Hazel, N. Stilianakis, R. Ronchetti and J.G. Koppe. 2006. Cadmium and children: exposure and health effects. Acta. Pædiatrica. 95:50-54.
  - SDU Uitgevers 2000. MAC Ministry of Social Affairs and Employment. Nationale MAC (Maximum Allowable Concentration) List, 2000. The Hague, The Netherlands.
- Sorahan, T. and N.A. Esmen. 2004. Lung cancer mortality in UK nickel-cadmium battery workers, 1947 2000. Occup. Environ. Med. 61:108-116.
- Takenaka, S., E. Karg, W.G. Kreyling, B. Lentner, H. Schultz, A. Ziesenis, P. Schramel and J. Heyder.
   2004. Fate and toxic effects of inhaled ultrafine cadmium oxide particles in the rat lung. Inhal.
   Toxicol. 16 (suppl.1): 83-92.
- Takenaka, S., H. Oldiges, H. König, D. Hochrainer and G. Oberdöster. 1983. Carcinogenicity of cadmium
   chloride aerosols in W rats. J. Natl. Cancer Inst. 70: 367-373.
- 45 ten Berge, W.F., A. Zwart, and L.M. Appelman. 1986. Concentration-time mortality response
  46 relationship of irritant and systemically acting vapours and gases. J. Hazard. Materials: 13:
  47 301-309.
  48
- Thun, M.J., T.M. Schnorr, A.B. Smith, W.E. Halperin and R.A. Lemen. 1985. Mortality among a cohort
  of US cadmium production workers- an update. J. Natl. Cancer Inst. 74: 325-333.
- Trottier, B., J. Athot, A.C. Ricard and J. Lafond. 2002. Maternal-fetal distribution of cadmium in the
   guinea pig following a low dose inhalation exposure. Toxicol. Lett. 129:189-197.

1	
2	U.S. EPA (United States Environmental Protection Agency). 1994. Integrated Risk Information System
3	Summary of Cadmium (CASRN 7440-43-9). United States Environmental Protection Agency,
4	Washington, D.C. Available online. http://www.epa.gov/ncea/iris/subst/0141.htm.
5	
6	U.S. EPA (United States Environmental Protection Agency). 1986. Final draft for the drinking water
7	criteria document on cadmium. EPA PB89-192140. Office of Drinking Water, Washington, D.C.
8	
9	Wittman, R. and H. Hu. 2002. Cadmium exposure and nephropathy in a 28-year old female metals
10	worker. Environ. Health Perspecti. 110: 1261-1266.
11	
12	Yamamoto, K., M. Ueda, H. Kikuchi, H. Hattori and Y. Hiraoka. 1983. An acute fatal occupational
13	cadmium poisoning by inhalation. Z. Rechtsmed. 91: 139-143.

1	<b>APPENDIX A: Derivation of AEGL Values</b>				
2 3	Derivation of AEGL-1				
4 5 6 7 8 9	Key Study:	Takenaka, S., E. Karg, W.G. Kreyling, B. Lentner, H. Schultz, A. Ziesenis, P. Schramel and J. Heyder. 2004. Fate and toxic effects of inhaled ultrafine cadmium oxide particles in the rat lung. Inhal. Toxicol. 16 (suppl.1): 83-92.			
10 11 12	Toxicity endpoint:	The experimental concentration, $0.55 \text{ mg Cd/m}^3$ caused slight respiratory irritation in female rats exposed for 6 hours.			
12 13 14	Time scaling:	$C^n \ge t = k$ ; n=3 when extrapolating to shorter time points and n = 1 when extrapolating to longer time points.			
15 16 17 18 19 20 21 22 23 24 25	Uncertainty factors: Interspecies:	3, Cadmium is a direct-acting respiratory irritant and it is not expected that toxicity would differ among species. Rabbits and rats exposed for 2 hours to 0.25-4.5 mg/m <sup>3</sup> displayed similar histological and biochemical pulmonary effects including pneumonitis, increased lung weight, pulmonary inflammatory cell influx, and decreased glutathione peroxidase activity (Grose et al. 1987). Rats exposed to cadmium (0.00169-5.3 mg/m <sup>3</sup> ) from 1-6 hours (Buckley and Bassett 1987; Oberdörster et al. 1987; Takenaka et al. 2004) exhibited the same effects as those observed in the Grose et al. (1987) study.			
26 27 28 29 30 31	Intraspecies:	3, Cadmium is a direct-acting respiratory irritant, and respiratory effects due to irritation are not expected to differ greatly among individuals. After a five hour exposure to cadmium, workers experienced cough, throat irritation, dyspnea, and pulmonary edema (Beton et al. 1966) which are signs of respiratory irritation.			
32 33	Modifying factor:	None applied.			
34 35 36 37 38 39	Calculations:	$\begin{array}{l} 0.55 \ mg/m^3 /  10 = 0.055 \ mg/m^3 \\ C^3 \ x \ t = k \\ (0.055 \ mg/m^3)^3 \ x \ 360 \ min = 0.059895 \ mg/m^3 \text{-min} \\ \end{array}$			
40 41 42	10-min AEGL-1	30-min value adopted as 10-min value = $0.13 \text{ mg/m}^3$			
42 43 44 45	30-min AEGL-1	$C^3 x 30 min= 0.059895 mg/m^3-min$ C = 0.13 mg/m <sup>3</sup>			
45 46 47	1-hr AEGL-1	$C^3 x 60 min= 0.059895 mg/m^3-min$ C = 0.10 mg/m <sup>3</sup>			
48 49	4-hr AEGL-1	$C^3 \ge 240 \text{ min} = 0.059895 \text{ mg/m}^3 \text{-min}$			

1		$C = 0.063 \text{ mg/m}^3$
2 3	8-hr AEGL-1	$C^1 \ge 480 \text{ min} = 19.8 \text{ mg/m}^3\text{-min}$
4		$C = 0.041 \text{ mg/m}^3$

1	Derivation of AEGL-2			
2 3 4	Key Studies:	Buckley, B.J. and D.J.P. Bassett. 1987. Pulmonary cadmium oxide toxicity in the rat. J. Toxicol. Environ. Health. 21: 233-250.		
5 6 7	Toxicity endpoints:	The experimental concentration, $5.3 \text{ mg Cd/m}^3$ caused overt respiratory irritation and pathology in exposed for 3 hours.		
8 9 10	Time scaling:	$C^n$ x t = k; n=3 when extrapolating to shorter time points and n = 1 when extrapolating to longer time points.		
11 12 13 14 15 16 17 18 19 20	Uncertainty factors: Interspecies:	3, Cadmium is a direct-acting respiratory irritant and it is not expected that toxicity would differ among species. Rabbits and rats exposed for 2 hours to 0.25-4.5 mg/m <sup>3</sup> displayed similar histological and biochemical pulmonary effects including pneumonitis, increased lung weight, pulmonary inflammatory cell influx, and decreased glutathione peroxidase activity (Grose et al. 1987). Rats exposed to cadmium (0.00169-5.3 mg/m <sup>3</sup> ) from 1-6 hours (Buckley and Bassett 1987; Oberdörster et al. 1987; Takenaka et al. 2004) exhibited the same effects as those observed		
21 22 23 24 25 26	Intraspecies:	in the Grose et al. (1987) study. 3, Cadmium is a direct-acting respiratory irritant, and respiratory effects due to irritation are not expected to differ greatly among individuals. After a five hour exposure to cadmium, workers experienced cough, throat irritation, dyspnea, and pulmonary edema (Beton et al. 1966) which are signs of respiratory irritation.		
27 28 20	Modifying factor:	None applied.		
29 30 31 32 33 24	Calculations:	5.3 mg/m <sup>3</sup> / 10 = 0.53 mg/m <sup>3</sup> $C^{3} x t = k$ (0.53 mg/m <sup>3</sup> ) <sup>3</sup> x 180 min = 26.79786 mg/m <sup>3</sup> -min		
34 35 26		C x t = k 0.53 mg/m <sup>3</sup> x 180 min = 95.4 mg/m <sup>3</sup> -min		
30 37 38 39	10-min AEGL-2	$C^3 x 10 min= 26.79786 mg/m^3-min$ C = 1.4 mg/m <sup>3</sup>		
40 41 42	30-min AEGL-2	$C^3 x 30 min= 26.79786 mg/m^3-min$ C = 0.96 mg/m <sup>3</sup>		
43 44 45	1-hr AEGL-2	$C^3 \ge 60 \text{ min} = 26.79786 \text{ mg/m}^3\text{-min}$ $C = 0.76 \text{ mg/m}^3$		
46 47 48	4-hr AEGL-2	$C^{1} x 240 min = 95.4 mg/m^{3}$ -min C = 0.40 mg/m <sup>3</sup>		
40 49	8-hr AEGL-2	$C^1 \ge 480 \min = 95.4 \text{ mg/m}^3 \text{-min}$		



$\frac{1}{2}$	Derivation of AEGL-3			
2 3 4 5 6 7	Key Studies:	Rusch, G.M., J.S. O'Grodnick and W.E. Rinehart. 1986. Acute inhalation study in the rat of comparative uptake, distribution and excretion for different cadmium containing materials. Am. Ind. Hyg. Assoc. J. 47: 754-763.		
8 9 10	Toxicity endpoint:	The threshold of lethality was calculated from the 2-hr $LC_{50}$ for cadmium fume in rats, 112 mg/m <sup>3</sup> .		
11 12 13	Time scaling:	$C^n x t = k$ ; n=3 when extrapolating to shorter time points, and n = 1 when extrapolating to longer time points.		
14	Uncertainty factors:			
14 15 16 17 18 19 20 21 22 23	Interspecies:	3, Cadmium is a direct-acting respiratory irritant and it is not expected that toxicity would differ among species. Rabbits and rats exposed for 2 hours to 0.25-4.5 mg/m <sup>3</sup> displayed similar histological and biochemical pulmonary effects including pneumonitis, increased lung weight, pulmonary inflammatory cell influx, and decreased glutathione peroxidase activity (Grose et al. 1987). Rats exposed to cadmium (0.00169-5.3 mg/m <sup>3</sup> ) from 1-6 hours (Buckley and Bassett 1987; Oberdörster et al. 1987; Takenaka et al. 2004) exhibited the same effects as those observed in the Grose et al. (1987) study.		
24 25 26 27 28 29	Intraspecies:	3, Cadmium is a direct-acting respiratory irritant, and respiratory effects due to irritation are not expected to differ greatly among individuals. After a five hour exposure to cadmium, workers experienced cough, throat irritation, dyspnea, and pulmonary edema (Beton et al. 1966) which are signs of respiratory irritation.		
30 31	Modifying factor:	None applied.		
32 33 34 35 36 37	Calculations	112 mg/m <sup>3</sup> /3/10= 3.733 mg/m <sup>3</sup> $C^3 x t = k$ (3.733 mg/m <sup>3</sup> ) <sup>3</sup> x 120 min = 6242.45206 mg/m <sup>3</sup> -min $C^1 x t = k$ 3.733 mg/m <sup>3</sup> x 120 min = 447.96 mg/m <sup>3</sup> -min		
39 40	10-minute AEGL-3	$C^3 \ge 10 \text{ min} = 6242.45206 \text{ mg/m}^3\text{-min}$ $C = 8.5 \text{ mg/m}^3$		
41 42 43	30-minute AEGL-3	$C^3 \ge 30 \text{ min} = 6242.45206 \text{ mg/m}^3\text{-min}$ $C = 5.9 \text{ mg/m}^3$		
44 45 46	1-hr AEGL-3	$C^{3} x 60 min = 6242.45206 mg/m^{3}$ -min C = 4.7 mg/m <sup>3</sup>		
47 48 49	4-hr AEGL-3	$C^{1} x 240 min = 44.796 mg/m^{3}$ -min C = 1.9 mg/m <sup>3</sup>		

1		
2	8-hr AEGL-3	$C^{T} x 480 min = 44.796 mg/m^{3}-min$
3		$C = 0.93 \text{ mg/m}^3$
4		C .

1 2

### **APPENDIX B: Time-Scaling Calculations**

3 The relationship between dose and time for any given chemical is a function of the 4 physical and chemical properties of the substance and the unique toxicological and 5 pharmacological properties of the individual substance. Historically, the relationship according 6 to Haber (1924), commonly called Haber's Law or Haber's Rule (i.e., C x t = k, where C =7 exposure concentration, t =exposure duration, and k =a constant) has been used to relate 8 exposure concentration and duration to effect (Rinehart and Hatch 1964). This concept states 9 that exposure concentration and exposure duration may be reciprocally adjusted to maintain a 10 cumulative exposure constant (k) and that this cumulative exposure constant will always reflect a 11 specific quantitative and qualitative response. This inverse relationship of concentration and 12 time may be valid when the toxic response to a chemical is equally dependent upon the 13 concentration and the exposure duration. However, an assessment by ten Berge et al. (1986) of LC<sub>50</sub> data for certain chemicals revealed chemical-specific relationships between exposure 14 concentration and exposure duration that were often exponential. This relationship can be 15 expressed by the equation  $C^n x t = k$ , where *n* represents a chemical specific, and even a toxic 16 endpoint specific, exponent. The relationship described by this equation is basically in the form 17 18 of a linear regression analysis of the log-log transformation of a plot of C vs. ten Berge et al. 19 (1986) examined the airborne concentration (C) and short-term exposure duration (t) relationship 20 relative to death for approximately 20 chemicals and found that the empirically derived value of 21 *n* ranged from 0.8 to 3.5 among this group of chemicals. Hence, the value of the exponent (n) in 22 the equation  $C^n x t = k$  quantitatively defines the relationship between exposure concentration 23 and exposure duration for a given chemical and for a specific health effect endpoint. Haber's 24 Rule is the special case where n = 1. As the value of *n* increases, the plot of concentration vs. 25 time yields a progressive decrease in the slope of the curve.

26

27 The available data do not allow for empirical derivation of a temporal scaling factor (*n*) for

28 cadmium. The exposure concentration-exposure duration relationship for many irritant and

systemically acting vapors and gases may be described by  $C^n x t = k$ , where the exponent, *n*,

ranges from 0.8 to 3.5 (ten Berge et al., 1986). In the absence of an empirically derived

exponent (*n*), temporal scaling from the experimental durations of the respective PODs to AEGL-specific durations was performed using n = 3 when extrapolating to shorter time points

and n = 1 when extrapolating to longer time points using the  $C^n x t = k$  equation.

1 **APPENDIX C: Carcinogenicity Assessment** 2 3 The carcinogenicity data are summarized in Section 3.6 and Table 9. The U.S. EPA (1994) 4 concluded that cadmium is a "probable human carcinogen" based on evidence of carcinogenicity in animals and limited evidence of carcinogenicity in an exposed human population. The 5 inhalation unit risk calculated is  $1.8 \times 10^{-3} \,\mu g/m^3$ , and the concentration associated with a risk 6 level of 1 in 10.000 is 6 x  $10^{-2}$  µg/m<sup>3</sup>. 7 8 9 To convert a 70-year (25,600 days) exposure to a 24-hr exposure: 10 24-hr exposure =  $d \ge 25,600$ ; where  $d = 6 \ge 10^{-2} \ \mu g/m^3$ 11  $= (6 \times 10^{-2} \,\mu g/m^3) \times 25,600 \,days$ 12 13  $= 1536 \,\mu g/m^3 = 1.54 \,m g/m^3$ 14 To account for uncertainty regarding the variability in the stage of the cancer process at which 15 cadmium may act, a multi-stage factor of 6 is applied (Crump and Howe, 1984): 16 17  $1.54 \text{ mg/m}^3/6 = 0.26 \text{ mg/m}^3$ 18 19 20 Therefore, a single exposure to cadmium at  $0.26 \text{ mg/m}^3$  for 24 hrs would represent a cancer risk of  $10^{-4}$ . If the exposure is limited to a fraction (f) of a 24-hr period, the fractional exposure 21 22 becomes 1/f x 24 hr (NRC 1985). 23  $= 0.26 \text{ mg/m}^3$ 24 24 hr exposure 25 8 hr exposure =  $0.78 \text{ mg/m}^3$ 4 hr exposure =  $1.56 \text{ mg/m}^3$ 26  $= 6.24 \text{ mg/m}^3$ 1 hr exposure 27  $30 \text{ min exposure} = 12.48 \text{ mg/m}^3$ 28 29  $10 \text{ min exposure} = 36.66 \text{ mg/m}^3$ 30 31 32 The AEGL values for 10 minutes, 30 minutes, 1, 4, and 8 hours are presented below for

33 34 risks of 10<sup>-4</sup>, 10<sup>-5</sup>, and 10<sup>-6</sup>

Time (h)	10 <sup>-4</sup>	10 <sup>-5</sup>	10 <sup>-6</sup>
0.17	36.7	3.67	0.367
0.5	12.5	1.25	0.125
1	6.24	0.624	0.0624
4	1.56	0.156	0.0156
8	0.78	0.078	0.0078

35

36 These values based on carcinogenicity are not proposed for AEGL values.

## 6

AEGL-1 VALUES					
10-min	30-min	1-hr	4-hr	8-hr	
0.13 mg/m <sup>3</sup>	0.13 mg/m <sup>3</sup>	0.10 mg/m <sup>3</sup>	$0.063 \text{ mg/m}^3$	0.041 mg/m <sup>3</sup>	
Key Reference:					
Takenaka, S., E. Karg,	W.G. Kreyling, B. Lent	ner, H. Schultz, A. Zies	enis, P. Schramel and J.	Heyder. 2004. Fate	
and toxic effects of inl	naled ultrafine cadmium	oxide particles in the rat	t lung. Inhal. Toxicol. (si	uppl.1): 83-92.	
Test Species/Strain/N	umber: Rat/Fischer 34	4/24/group			
Exposure Route/Con	centrations/Duration: 1	Inhalation/0.07, 0.550 m	$g/m^3/6$ hr		
Effects:					
$0.07 \text{ mg/m}^3$ No	o morphological changes	s or inflammatory respon	nse		
$0.550 \text{ mg/m}^3$ Ir	creased neutrophils and	multifocal alveolar infla	ammation		
Endpoint/Concentrat	tion/Rationale: Slight re	espiratory irritation /0.55	mg Cd/m <sup>3</sup> administered	l as CdO	
Uncertainty Factors/	Rationale:				
Total uncertainty fac	tors: 10				
Interspecies:	3, Cadmium is a dir	ect-acting respiratory in	ritant and it is not expect	ed that toxicity would	
	differ among species. Rabbits and rats exposed for 2 hours to 0.25-4.5 mg/m <sup>3</sup> displayed				
	similar histological and biochemical pulmonary effects including pneumonitis, increased				
	lung weight, pulmo	nary inflammatory cell i	nflux, and decreased glu	tathione peroxidase	
	activity (Grose et al. 1987). Rats exposed to cadmium (0.00169-5.3 mg/m <sup>3</sup> ) from 1-6				
	hours (Buckley and	Bassett 1987; Oberdörs	ter et al. 1987; Takenaka	1 et al. 2004)	
	exhibited the same	effects as those observed	1 in the Grose et al. (198	7) study.	
Intraspecies:	3, Cadmium is a dir	ect-acting respiratory irr	ritant, and respiratory eff	ects due to irritation	
	are not expected to	differ greatly among ind	lividuals. After a five ho	our exposure to	
	cadmium, workers	experienced cough, throa	at irritation, dyspnea, and	d pulmonary edema	
	(Beton et al. 1966)	which are signs of respir	atory irritation.		
Modifying Factor: None					
Animal to Human Dosimetric Adjustment: None					
Time Scaling: C <sup>n</sup> x t =	= k; n=3 when extrapolat	ing to shorter time point	ts (10-, 30-, and 60- min	, and 4-hr), and $n = 1$	
when extrapolating to longer time points (8 hr). The 30-minute AEGL-1 value was adopted as the 10-minute					
value due to the added uncertainty of extrapolating from a 6-hr time point to 10 min.					
Data Adequacy: Data were available and adequate for deriving AEGL-1 values.					

**APPENDIX D: Derivation Summary for Cadmium AEGLs** 

ACUTE EXPOSURE GUIDELINE LEVELS FOR

CADMIUM (CAS Reg. No. 7440-43-9)

**DERIVATION SUMMARY** 

AEGL-2 VALUES								
10-min	30-min	1-hr	4-hr	8-hr				
1.4 mg/m <sup>3</sup>	0.96 mg/m <sup>3</sup>	0.76 mg/m <sup>3</sup>	$0.40 \text{ mg/m}^3$	$0.20 \text{ mg/m}^3$				
Key Reference:								
Buckley, B.J. and D	J.P. Bassett. 1987. Pulm	onary cadmium oxide t	oxicity in the rat. J. Tox	kicol. Environ.				
Health. 21: 233-250								
Test Species/Strain	Number: Rats/Wistar/	16/group						
Exposure Route/Co	oncentrations/Duration	: Inhalation/0.0.5, 5.3 n	ng/m <sup>3</sup> CdO/ 3 hr					
Effects:								
0. 5 mg/m <sup>3</sup>	Transient mild hypercellu	larity at bronchoalveol	ar junctions and adjacer	nt alveoli,				
	inflammatory cell influx							
$5.3 \text{ mg/m}^3$	Interstitial thickening, $\uparrow$ c	cuboidal alveolar cells,	↑inflammatory cells, ↑c	lry lung weight,				
	↑protein content, ↑DNA o	content, ↑ GP, GR, G6P	PD, 6PGD activity					
Endpoint/Concent	ration/Rationale: Overt	respiratory tract irritation	on and pathology /5.3 m	ng/m <sup>3</sup> administered as				
CdO								
<b>Uncertainty Factor</b>	s/Rationale:							
Total uncertainty fa	actors: 10							
Interspecies:	3, Cadmium is a d	irect-acting respiratory	irritant and it is not exp	ected that toxicity				
	would differ among species. Rabbits and rats exposed for 2 hours to $0.25-4.5 \text{ mg/m}^3$							
	displayed similar h	nistological and biocher	nical pulmonary effects	sincluding				
	pneumonitis, incre	ased lung weight, pulm	ionary inflammatory ce	ll influx, and				
	decreased glutathi	one peroxidase activity	(Grose et al. 1987). Ra	ats exposed to				
	cadmium (0.00169	9-5.3 mg/m <sup>3</sup> ) from 1-6 l	nours (Buckley and Bas	sett 1987;				
	Oberdörster et al.	1987; Takenaka et al. 2	004) exhibited the same	e effects as those				
	observed in the Gr	ose et al. (1987) study.						
Intraspecies:	3, Cadmium is a d	irect-acting respiratory	irritant, and respiratory	effects due to				
	irritation are not ex	spected to differ greatly	among individuals. A	fter a five hour				
	exposure to cadmi	um, workers experience	ed cough, throat irritation	on, dyspnea, and				
	pulmonary edema (Beton et al. 1966) which are signs of respiratory irritation.							
Modifying Factor:	Modifying Factor: None							
Animal to Human	Dosimetric Adjustment	: None						
<b>Time Scaling:</b> C <sup>n</sup> x	t = k; $n=3$ when extrapol	ating to shorter time po	oints (10, 30, and 60 min	n), and $n = 1$ when				
extrapolating to longer time points (4 hr, 8 hr).								
Data Adequacy: Da	ata were available and ad	equate for deriving AE	GL-2 values.	Data Adequacy: Data were available and adequate for deriving AEGL-2 values.				

AEGL-3 VALUES					
10-min	30-min	1-hr	4-hr	8-hr	
8.5 mg/m <sup>3</sup>	5.9 mg/m <sup>3</sup>	4.7 mg/m <sup>3</sup>	1.9 mg/m <sup>3</sup>	$0.93 \text{ mg/m}^3$	
Key Reference:					
Rusch, G.M., J.S. O'C	Frodnick, and W.E. Rineh	art. 1986. Acute inhala	tion study in the rat	t of comparative	
uptake, distribution an	d excretion for different c	admium containing m	aterials. Am. Ind. H	lyg. Assoc. J 47: 754-	
763.					
Test Species/Strain/N	umber: Rat/Sprague-Da	wley/26/group	1	11 100 / 3 0 1	
Exposure Route/Con	centrations/Duration: In $(3, 3, 2, 1, 5, 5)$	nhalation/97 mg/m <sup>3</sup> Cd	red, 99 mg/m <sup>3</sup> Cd y	vellow, 132 mg/m <sup>3</sup> Cd	
carbonate, 112 mg	g/m <sup>*</sup> Cd fume/2 hr				
Effects: $0.7 \text{ m} \text{ s/m}^3 \text{ Cd} \text{ rad}$	I comingation and	nol dissolvation			
$97 \text{ mg/m}^3 \text{ Cd yell}$	Lacrimation, lei				
$132 \text{ mg/m}^3 \text{ Cd ca}$	rhonate 5.8% mortality	dry rales nulmonary e	edema		
$132 \text{ mg/m}^2 \text{ Cd cu}$ 112 mg/m <sup>3</sup> Cd fu	me 48% mortality.	hypoactivity, closed ex	ves, dry and moist r	ales, LC <sub>50</sub>	
Endpoint/Concentra	tion/Rationale: Threshold	d of lethality based on	$\frac{2}{2-\text{hr}}$ LC <sub>50</sub> 112 mg/r	$n^3$ for cadmium fumes	
Uncertainty Factors/	Rationale:	,	30 0		
Total uncertainty fac	ctors: 10				
Interspecies:	3, Cadmium is a dire	ct-acting respiratory ir	ritant and it is not e	xpected that toxicity	
	would differ among species. Rabbits and rats exposed for 2 hours to 0.25-4.5 mg/m <sup>3</sup>				
	displayed similar his	tological and biochemi	ical pulmonary effe	cts including	
	pneumonitis, increas	ed lung weight, pulmo	nary inflammatory	cell influx, and	
	decreased glutathion	e peroxidase activity (	Grose et al. 1987).	Rats exposed to	
	cadmium (0.00169-5	$1.3 \text{ mg/m}^3$ ) from 1-6 hc	burs (Buckley and B	assett 1987;	
	Oberdorster et al. 19	8/; Takenaka et al. 200	04) exhibited the sa	me effects as those	
Intraspacias	3 Cadmium is a dire	e et al. (1987) study.	ritant and respirato	ry affacts due to	
mu aspecies.	irritation are not exp	ected to differ greatly a	among individuals	After a five hour	
	exposure to cadmiun	1. workers experienced	l cough, throat irrita	tion, dyspnea, and	
	pulmonary edema (B	seton et al. 1966) which	h are signs of respir	atory irritation.	
Modifying Factor: None					
Animal to Human Dos	simetric Adjustment:				
Time Scaling: C <sup>n</sup> x t	<b>Time Scaling:</b> $C^n \ge t = k$ ; n=3 when extrapolating to shorter time points (10-, 30-, and 60- min), and n = 1				
when extrapolatin	ng to longer time points (4	-hr, 8 hr).			
Data Adequacy: Data	a were available and adequ	uate for deriving AEG	L-3 values.		



### **APPENDIX E: Category Plot for Cadmium**

**Chemical Toxicity - TSD Data Cadmium** 



# 45 67 89

The values for AEGL-2 and AEGL-3 are above a concentration, 0.45 mg Cd/m<sup>3</sup>, at which 2 animals died. Based on the information provided in the study (Grose et al. 1987), it is possible that the animal deaths were not the result of exposure to cadmium but were the result of exposure apparatus difficulties (reversal of animal body and resulting asphyxiation). The mortality at this concentration was inconsistent with the other animal data resulting from exposures at similar concentrations. There was also a lack of dose response as no mortality occurred following

10 exposure to a higher dose of cadmium.

1	
T	

Category Plot Data							
For Category 0 = No effect, 1 = Discomfort, 2 = Disabling, SL = Some Lethality, 3 = Lethal							
Source	Species	Sex	# Exposures	mg/m <sup>3</sup>	Min	Category	Comments
NAC/AEGL-1				0.13	10	AEGL	
NAC/AEGL-1				0.13	30	AEGL	
NAC/AEGL-1				0.10	60	AEGL	
NAC/AEGL-1				0.063	240	AEGL	
NAC/AEGL-1				0.041	480	AEGL	
NAC/AEGL-2				1.4	10	AEGL	
NAC/AEGL-2				0.96	30	AEGL	
NAC/AEGL-2				0.76	60	AEGL	
NAC/AEGL-2				0.40	240	AEGL	
NAC/AEGL-2				0.20	480	AEGL	
NAC/AEGL-3				8.5	10	AEGL	
NAC/AEGL-3				5.9	30	AEGL	
NAC/AEGL-3				4.7	60	AEGL	
NAC/AEGL-3				1.9	240	AEGL	
NAC/AEGL-3				0.93	480	AEGL	
Beton et al. 1966	Human	m	1	8.6	300	SL	20% mortality, pulmonary edema.
							dyspnea in others
Grose et al. 1987	Rabbit	m	1	0.25	120	1	Decreased GSH peroxidase activity
Grose et al. 1987	Rabbit	m	1	0.45	120	1	Increased GSH transferase activity
Grose et al. 1987	Rabbit	m	1	4.5	120	2	Pneumonitis
Grose et al. 1987	Rabbit	m	1	0.25	120	0	No effects
Grose et al. 1987	Rabbit	m	1	0.45	120	1	Increase alveolar macrophages
Grose et al. 1987	Rabbit	m	1	4.5	120	2	Pneumonitis
Grose et al. 1987	Rat	m	1	0.25	120	1	Decreased GSH peroxidase activity
Grose et al. 1987	Rat	m	1	0.45	120	1	Decreased GSH peroxidase activity;
							body weight
Grose et al. 1987	Rat	m	1	4.5	120	2	Pneumonitis
Grose et al. 1987	Rat	m	1	0.25	120	0	No effects
Grose et al. 1987	Rat	m	1	0.45	120	SL	Pulmonary congestion
Grose et al. 1987	Rat	m	1	4.5	120	2	Pneumonitis
Takenaka et al 2004	Rat	f	1	0.07	360	0	No effects
Takenaka et al 2004	Rat	f	1	0.55	360	1	Alveolar inflammation, neutrophils
Oberdorster et al. 1987	Rat	m	1	0.00195	60	1	Decreased alveolar macrophages,
Oberdorster et al. 1987	Rat	m	1	0.00169	60	1	Decreased alveolar macrophages,
Oberdorster et al. 1987	Rat	m	1	0.00182	60	0	No effects
Rusch et al. 1986	Rat	b	1	97	120	1	Lacrimation; renal discoloration
Rusch et al. 1986	Rat	b	1	99	120	1	Lacrimation
Rusch et al. 1986	Rat	b	1	132	120	SL	5.8% mortality, dry rales; pulmonary
							edema
Rusch et al. 1986	Rat	b	1	112	120	SL	48% mortality; hypoactivity, LC50