

**ACUTE EXPOSURE GUIDELINE LEVELS (AEGLs)
FOR
THIONYL CHLORIDE
7719-09-7**



INTERIM

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5 **PREFACE**
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7 Under the authority of the Federal Advisory Committee Act (FACA) P. L. 92-463 of
8 1972, the National Advisory Committee for Acute Exposure Guideline Levels for Hazardous
9 Substances (NAC/AEGL Committee) has been established to identify, review and interpret
10 relevant toxicologic and other scientific data and develop AEGLs for high priority, acutely toxic
11 chemicals.
12

13 AEGLs represent threshold exposure limits for the general public and are applicable to
14 emergency exposure periods ranging from 10 minutes to 8 hours. Three levels — AEGL-1,
15 AEGL-2 and AEGL-3 — are developed for each of five exposure periods (10 and 30 minutes, 1
16 hour, 4 hours, and 8 hours) and are distinguished by varying degrees of severity of toxic effects.
17 The three AEGLs are defined as follows:
18

19 AEGL-1 is the airborne concentration (expressed as parts per million or milligrams per
20 cubic meter [ppm or mg/m³]) of a substance above which it is predicted that the general
21 population, including susceptible individuals, could experience notable discomfort, irritation, or
22 certain asymptomatic, non-sensory effects. However, the effects are not disabling and are
23 transient and reversible upon cessation of exposure.
24

25 AEGL-2 is the airborne concentration (expressed as ppm or mg/m³) of a substance above
26 which it is predicted that the general population, including susceptible individuals, could
27 experience irreversible or other serious, long-lasting adverse health effects or an impaired ability
28 to escape.
29

30 AEGL-3 is the airborne concentration (expressed as ppm or mg/m³) of a substance above
31 which it is predicted that the general population, including susceptible individuals, could
32 experience life-threatening health effects or death.
33

34 Airborne concentrations below the AEGL-1 represent exposure levels that could produce
35 mild and progressively increasing but transient and nondisabling odor, taste, and sensory
36 irritation or certain asymptomatic, non-sensory effects. With increasing airborne concentrations
37 above each AEGL, there is a progressive increase in the likelihood of occurrence and the severity
38 of effects described for each corresponding AEGL. Although the AEGL values represent
39 threshold levels for the general public, including susceptible subpopulations, such as infants,
40 children, the elderly, persons with asthma, and those with other illnesses, it is recognized that
41 individuals, subject to unique or idiosyncratic responses, could experience the effects described
42 at concentrations below the corresponding AEGL
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44

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EXECUTIVE SUMMARY

1
2 Thionyl chloride is a colorless to pale yellow liquid used in the production of batteries. It is
3 also used during synthesis of drugs, vitamins, and herbicides. It is commercially available and
4 used in industrial chemical reactions.
5

6 Thionyl chloride is a respiratory irritant causing bronchoconstriction, and can cause death
7 from pulmonary edema caused by the hydrolysis products, sulfur dioxide and hydrogen chloride.
8 Few human case reports are available, and they lack quantitative exposure data. The reports
9 confirm respiratory irritation in humans.
10

11 Exposure-response data from rat studies are used to derive acute exposure guideline level
12 (AEGL) values for thionyl chloride due to lack of quantitative data from human case reports.
13 The AEGL values for the exposure periods of concern were scaled from the experimental
14 exposure duration using exponential scaling ($C^n \times t = k$, where C = exposure concentration, t =
15 exposure duration, and k = a constant). Data are unavailable to empirically derive a scaling factor
16 (n) for thionyl chloride. The concentration-exposure time relationship for many irritant and
17 systemically acting vapors and gases may be described by $C^n \times t = k$, where the exponent n
18 ranges from 0.8 to 3.5 (ten Berge et al. 1986). Temporal scaling was performed using $n = 3$,
19 when extrapolating to shorter time points and $n = 1$ when extrapolating to longer time points for
20 AEGL values (NRC 2001).
21

22 Data are not available from human or animal studies to derive AEGL-1 values. Therefore,
23 AEGL-1 values are not recommended.
24

25 Rats exposed to 71 ppm thionyl chloride for one hour experienced swollen noses and
26 dyspnea (Pauluhn 1987). These are toxic responses but not irreversible or incapacitating effects
27 and will not impair ability to escape. The AEGL-2 values are derived from this data. A total
28 uncertainty factor of 30 was applied. A similar mechanism of action would be expected across
29 species, therefore, an uncertainty factor of 3 was used for interspecies variability while a factor
30 of 10 was used for intraspecies variability to account for sensitive populations. Thionyl chloride
31 hydrolyzes into sulfur dioxide and hydrogen chloride. Asthmatics are more sensitive than
32 healthy people to the effects of sulfur dioxide.
33

34 The AEGL-3 values were based upon the highest concentration causing no lethality in rats
35 exposed to thionyl chloride for one hour (Pauluhn 1987 and Nachreiner 1993). A one hour
36 exposure to 593 ppm produced 58% mortality (Nachreiner 1993). In Pauluhn (1987), the next
37 lowest experimental level, 407 ppm, did not cause lethality. This was used as the point of
38 departure. This concentration is only slightly higher than the lethality threshold (371 ppm)
39 reported in Nachreiner (1993). The same uncertainty factors and rationale used for AEGL-2
40 were applied to AEGL-3 calculations.
41

42 The calculated values are listed in Table 1.
43
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46

1 **TABLE 1. Summary of AEGL Values for Thionyl Chloride**
2

Classification	10-minute	30-minute	1-hour	4-hour	8-hour	Endpoint (Reference)
AEGL-1 (Notable discomfort)	NR	NR	NR	NR	NR	
AEGL-2 (Disabling)	4.3 ppm (21 mg/m ³)	3.0 ppm (15 mg/m ³)	2.4 ppm (12 mg/m ³)	0.59 ppm (2.9 mg/m ³)	0.30 ppm (1.5 mg/m ³)	Dyspnea (not incapacitating or irreversible) (Pauluhn 1987)
AEGL-3 (Lethal)	25 ppm (120 mg/m ³)	17 ppm (83 mg/m ³)	14 ppm (68 mg/m ³)	3.4 ppm (17 mg/m ³)	1.7 ppm (8.3 mg/m ³)	Threshold of lethality (Pauluhn 1987; Nachreiner 1993)

3
4 NR-Not recommended. Numeric values for AEGL-1 are not recommended because of insufficient data. Absence of
5 an AEGL-1 value does not imply that exposure below the AEGL-2 concentration is without adverse effects.
6
7
8

9 **1. INTRODUCTION**
10

11 Thionyl chloride is an industrially produced liquid used to make acyl chlorides and to
12 synthesize pharmaceuticals including drugs and vitamins. It is used to make dyes, to prepare
13 organic chlorine compounds, as a solvent in the production of herbicides, and as an electrolyte in
14 lithium batteries. Thionyl chloride is also used to prepare polyarylate-type engineering
15 thermoplastics made from iso- and tere-phthaloyl chlorides. The production is monitored and
16 controlled by the Chemical Weapons Convention under Schedule 3 because of the potential for it
17 to be used as a toxic chemical weapon and its widespread industrial use.
18

19 Thionyl chloride is a respiratory and skin irritant. Symptoms of exposure range from mild
20 irritation to death and include dyspnea, burns, pulmonary edema, and inflammation. When
21 exposed to water or water vapor, one mole of thionyl chloride produces one mole of sulfur
22 dioxide and two moles of hydrogen chloride very rapidly. Its hydrolysis occurs faster than that
23 of phosgene due to a pair of lone electrons in the S center of its pyramidal molecule.
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TABLE 2. Chemical and Physical Properties

Parameter	Value	References
Synonyms	Sulfurous oxychloride, sulfinyl chloride, sulfur chloride oxide, sulfurous dichloride, thionyl dichloride	O'Neil et al. 2001, RTECS 2006
Chemical formula	SOCl ₂	O'Neil et al. 2001
Molecular weight	118.98	O'Neil et al. 2001
CAS Reg. No.	7719-09-7	O'Neil et al. 2001
Physical state	Liquid- colorless, pale yellow, or red	O'Neil et al. 2001
Solubility in water	Reacts with H ₂ O to form SO ₂ and HCl, miscible with benzene, chloroform, and carbon tetrachloride	O'Neil et al. 2001
Vapor pressure	11.6 kPa at 20°C	AIHA 2002
Vapor density (air =1)	4.1 at 760 mmHg and 25°C	AIHA 2002
Melting point	104.5°C	O'Neil et al. 2001
Boiling point	75.6°C at 760 mmHg	O'Neil et al. 2001
Flammability limits	Nonflammable	O'Neil et al. 2001
Conversion factors	1 ppm= 4.87 mg/m ³ , 1 mg/m ³ = 0.206 ppm	AIHA 2002

2. HUMAN TOXICITY DATA

2.1. Acute Lethality

There are very few data on fatalities caused by acute thionyl chloride exposure. One case report is noted below.

2.1.1. Case Reports

Ducatmen et al. (1988) reported on a worker exposed to an unknown concentration of thionyl chloride for an estimated six minutes following a lithium thionyl chloride battery cell explosion. The worker developed severe respiratory distress and fell unconscious. He also suffered chemical burns and fractures from the exploding battery. Fatal pulmonary edema from inhalation of SO₂ and HCl mist, products of thionyl chloride and water, occurred three hours after exposure. Estimates of sulfur dioxide concentration in the room were calculated to be below 17,000 ppm.

2.2. Nonlethal Toxicity

Humans have been exposed to unknown amounts of thionyl chloride through their occupations. Most data on nonlethal effects of thionyl chloride in humans are case reports of accidental occupational exposures.

2.2.1. Odor Threshold/Odor Awareness

Thionyl chloride has a suffocating or pungent odor, but there is currently no odor threshold reported.

2.2.2. Case Reports

Few cases of thionyl chloride poisoning have been reported. These reports lack exposure and duration data and are described below. They do not provide quantitative data suitable for deriving AEGL values, but provide insight on human poisoning.

Grieco (1962) reported a case of a pharmaceutical production plant worker exposed for two minutes to an unknown amount of thionyl chloride vapor during repair of a vent pipe. The worker experienced burning of the eyes, upper airway, and chest followed by feelings of suffocation. He began coughing with expulsion of blood. He was hospitalized in June with cough, sero-mucosal excretion, pharyngeal and chest burning. His respiratory function test showed insufficient ventilation and examination revealed vesicular murmur with wheezing at end of expiration. He was diagnosed with acute asthmatic tracheal bronchitis, was treated with bronchodilators and antibiotics, and recovered.

Two cases were reported by Konichezky et al. (1993) of accidental inhalation in a battery plant by non-smoking employees. Exposure for one 30 year old employee occurred when a thionyl chloride tank burst in an open space. Immediately following exposure, he had no symptoms. Over a two week period, dyspnea developed and he was found to have mild hypocarbia (lower than normal levels of blood carbon dioxide) and moderate restrictive dysfunction in lung function tests. Treatment with salbutamol, oral aminophylline, and prednisone was successful after a six month period. The employee was accidentally exposed eight months later and suffered similar symptoms as before. Corticosteroid therapy over a six month period successfully returned the employee to normal health. The second employee was a 23 year old male exposed to thionyl chloride fumes through a similar accident with the exception that the tank burst occurred in a closed space. He was washed with tap water and admitted to the emergency department. He suffered chemical burns on various parts of his body including corneas, tongue, and nasal septum. He was treated with hydrocortisone and prednisone. He returned to the hospital 18 days later with shortness of breath and had hyperinflated and hyperlucent lungs. He was admitted to the intensive respiratory care unit and exhibited signs of dyspnea and sinus tachycardia. The employee continued to be ill and was diagnosed with end-stage lung disease 74 days post exposure. Eventually his respiratory condition improved somewhat, and he was able to perform minimal tasks for short periods of time without breathlessness. However, he was listed as permanently disabled (Konichezky et al. 1993).

A case report was submitted to the U.S. Environmental Protection Agency (USEPA/OPPT 2000). The employee suffered CNS effects after an acute exposure to an unknown concentration of what was believed to be thionyl chloride. The employee was not wearing any respiratory protection, inhaled 2-3 breaths of thionyl chloride vapor, and began suffering from rhinitis, lacrimation, sore throat, and coughing. The employee also exhibited signs of impaired hand-eye coordination, memory, judgment, and slurred speech. These clinical signs had improved six hours post exposure and no symptoms were present 72 hours post exposure.

2.2.3. Epidemiologic Studies

No epidemiological studies have been found in the available literature.

2.3. Neurotoxicity

No neurotoxicity studies were located; however neurotoxic effects were noted in the case report submitted to the U.S. Environmental Protection Agency (2000). A worker exposed to 2-3 breaths of an unknown concentration of thionyl chloride displayed impaired cerebellar function and ataxia. Normal cerebellar function returned twenty-four hours post exposure.

2.4. Developmental/Reproductive Toxicity

No data were available regarding the potential reproductive and developmental toxicity of thionyl chloride.

2.5. Genotoxicity

There are no data relevant to the derivation of AEGLs for thionyl chloride.

2.6. Carcinogenicity

No information was found regarding the carcinogenicity of thionyl chloride.

2.7. Summary

There is a lack of exposure and duration data regarding lethal and nonlethal human exposures to thionyl chloride. The case reports provide good qualitative data about health effects resulting from exposure. Signs of exposure include burning and watering of the eyes, difficulty breathing, coughing, and pulmonary edema.

3. ANIMAL TOXICITY DATA

3.1. Acute Lethality

Acute lethality data on thionyl chloride in laboratory animals are summarized in the following section and listed in Table 3. Benchmark concentrations (BMC) were calculated for mortality in the following studies.

3.1.1. Rats

Kinkead and Einhaus (1984) exposed male Fischer 344 rats to 99% pure thionyl chloride in a 60-liter whole-body inhalation chamber for one hour and observed the animals for at least two weeks post exposure. Analytical concentrations were measured by IR analysis. Moisture from the rats increased the humidity of the chamber to around 50% and caused almost complete hydrolysis of thionyl chloride to a mixture of sulfur dioxide and hydrogen chloride, and rats were exposed to this mixture more than to thionyl chloride. Exposure concentrations were reported as the mixture of sulfur dioxide and hydrogen chloride concentrations (906, 1080, 1239, 1509, or 1983 ppm). Five animals were exposed per group and no deaths were reported at 906 or 1080 ppm of the mixture. Two of five males died at 1239 ppm, three of five died at 1509 ppm, and four of five died at 1989 ppm of the mixture. All deaths occurred within 48 hours of exposure.

1 Examination of the animals revealed mild to moderate multifocal congestion and alveolar
2 damage. Severe lung irritation and edema were reported as the cause of death. The BMCL₀₅ and
3 BMC₀₁ were 663 and 830 ppm, respectively.
4

5 Pauluhn (1987) exposed five male and five female Wistar rats in an OECD Guideline study
6 to 71, 407, 769, 2121, or 3441 ppm of thionyl chloride vapor for one hour and observed the
7 animals for 14 days post exposure. The thionyl chloride used for the study was reported to be
8 99.8% pure, and rats were exposed head/nose only in a chamber 30 X 28 cm that prevented
9 expired air from mixing with test atmosphere. Analytical concentration was determined using
10 spectrophotometry. The LC₅₀ was calculated to be 1277 ppm. Four of five males and four of
11 five females died at 2121 ppm, and five of five males and four of five females died at 3441 ppm
12 prior to sacrifice. Prior to death, labored breathing was observed in the animals. The author
13 noted opacity of cornea, lung distention, necrotic changes in the nose, and pulmonary edema in
14 the animals that died. The BMCL₀₅ and BMC₀₁ were 507 and 671 ppm, respectively.
15

16 Nachreiner (1993) exposed 6 male and 6 female Wistar rats to 196, 371, 593, 954, or 1241
17 ppm to 99.9% pure (a.i.) thionyl chloride vapor in a nose-only chamber for one hour and
18 observed the animals 14 days post exposure. Gas chromatography was used to analyze thionyl
19 chloride during exposure. The LC₅₀ was 736 ppm for males and females combined. No deaths
20 were recorded in the 196 or 371 ppm exposure groups. Five of six males and two of six females
21 died at 593 ppm. Five of six males and two of six females died at 954 ppm. Six of six males and
22 four of six females died at 1241 ppm. Prior to death, audible respiration was observed.
23 Hyperinflation of the lungs, perinasal and periocular encrustation were noted in some of the
24 animals that died in the 954 and 1241 ppm exposure groups. One control male rat died during
25 exposure and it was thought to be due to incorrect positioning in the chamber. The BMCL₀₅ and
26 BMC₀₁ were 193 and 222 ppm, respectively.
27

28 The variability of the lowest levels causing lethality among the three studies could possibly
29 be explained by the different exposure conditions of three studies. The rats in Kinkead and
30 Einhaus (1984) were exposed whole body to sulfur dioxide and hydrogen chloride produced by
31 thionyl chloride in 39% to 58% relative humidity. Pauluhn (1987) exposed rats head/nose only
32 and the input air had a relative humidity range from 29% to 51%. Although the chamber design
33 prevented moisture from rat expiration from mixing with thionyl chloride and forming
34 hydrolysates, it is possible that the humidity of the input air caused thionyl chloride hydrolysis
35 before inhalation by the rats. Nachreiner (1993) exposed rats using compressed air at 11%
36 relative humidity in a nose only chamber which greatly reduced the formation of thionyl chloride
37 hydrolysates in the chamber. The nose only route allowed for inhalation of thionyl chloride only
38 and the formation of hydrolysis products (sulfur dioxide and hydrogen chloride) within the
39 respiratory tract. Nachreiner (1993) estimated the half-life of thionyl chloride at 53% relative
40 humidity to be five minutes.

1 **TABLE 3. Inhalation Data on Rats Exposed to Thionyl Chloride**

2

Inhalation	Concentration (ppm)	Duration (min)	Effect	Reference
Head/Nose Only	71	60	Reddened, swollen noses, slight dyspnea, slight piloerection, 0% mortality	Pauluhn 1987
Nose Only	196	60	Color change in lungs, 0% mortality	Nachreiner, 1993
Nose Only	371	60	Color change in lungs, 0% mortality	Nachreiner, 1993
Head/Nose Only	407	60	Reddened, swollen noses, moderate dyspnea, moderate piloerection, 0% mortality	Pauluhn 1987
Nose Only	593	60	Color change in lungs, audible respiration, 58% mortality	Nachreiner, 1993
Head/Nose Only	769	60	Necrotic changes in nasal speculum, lung distention, wheezing sounds, 0% mortality	Pauluhn 1987
Whole Body	906	60	Eye irritation, shallow breathing and gasping, 0% mortality	Kinthead and Einhaus, 1984
Nose Only	954	60	Hyperinflation of the lungs, perinasal and periocular encrustation, 58% mortality	Nachreiner, 1993
Whole Body	1080	60	Eye irritation, shallow breathing and gasping, 0% mortality	Kinthead and Einhaus, 1984
Whole Body	1239	60	Pulmonary edema, 40% mortality	Kinthead and Einhaus, 1984
Nose Only	1241	60	Hyperinflation of the lungs, perinasal and periocular encrustation, 83% mortality	Nachreiner, 1993
Whole Body	1509	60	Severe lung irritation, pulmonary edema, 60% mortality	Kinthead and Einhaus, 1984
Whole Body	1983	60	Severe lung irritation, pulmonary edema, 80% mortality	Kinthead and Einhaus, 1984
Head/Nose Only	2121	60	Severe dyspnea, reduced motility, complete cornea opacity, pulmonary edema, 80% mortality	Pauluhn 1987
Head/Nose Only	3441	60	Pulmonary edema, cornea opacity, 90% mortality	Pauluhn 1987

3
4
5 **3.2. Nonlethal Toxicity**

6
7 The three studies reporting lethal toxicity effects of thionyl chloride also described nonlethal
8 toxicity and are listed in Table 3. A summary of each study is given below.

9
10 **3.2.1. Rats**

11 Kinthead and Einhaus (1984) exposed male Fischer 344 rats to thionyl chloride in an
12

1 inhalation chamber for one hour. Moisture from the rats caused decomposition of thionyl
2 chloride to a mixture of sulfur dioxide and hydrogen chloride (906, 1080, 1239, 1509, or 1983
3 ppm). Five animals were exposed per group and no deaths were reported at 906 or 1080 ppm of
4 the mixture. Animals were reported to have eye and respiratory irritation. Shallow breathing
5 and gasping were also observed in these animals during the exposure. Animals died at the higher
6 concentrations.

7
8 Pauluhn (1987) exposed male and female Wistar rats for one hour to 71, 407, 769, 2121, or
9 3441 ppm of thionyl chloride. Both genders exhibited swollen noses and dyspnea at 71 and 407
10 ppm. Symptoms were slight at 71 ppm and moderate at 407 ppm. Wheezing and necrotic
11 changes in the nose were observed at 769 ppm as well as moderate dyspnea and swollen noses.
12 One female rat from 769 ppm was observed to determine the duration of the symptoms. From
13 day 7 to 14, the rat was found to be apathetic. The dyspnea in the rat was non-reversible within
14 the post-treatment observation period. Some rats from all groups mentioned above had distended
15 lungs at necropsy. Mortality was noted at the higher concentrations.

16
17 Nachreiner (1993) exposed male and female Wistar rats to 196, 371, 593, 954, or 1241 ppm
18 to thionyl chloride vapor in a nose-only chamber for one hour and observed the animals 14 days
19 post exposure. All animals exposed to thionyl chloride were observed using mouth breathing.
20 No other outward clinical signs attributed to treatment were observed in the 196 and 371 ppm
21 exposed animals. Histopathology revealed color changes in the lungs of males and females
22 starting at the lowest dose. Higher concentrations led to death.

23 24 **3.3. Developmental/Reproductive Toxicity**

25
26 There are currently no studies on the developmental/reproductive toxicity of thionyl chloride.

27 28 **3.4. Genotoxicity**

29
30 There are no genotoxicity studies with thionyl chloride.

31 32 **3.5. Carcinogenicity**

33
34 There are no data to suggest that thionyl chloride is a carcinogen.

35 36 **3.6. Summary**

37
38 The effects of acute inhalation exposure of rats to thionyl chloride are noted in the above
39 section. Effects consisted of ocular and respiratory irritation, lung discoloration, and/or death.
40 Death appeared to be caused by lung irritation and pulmonary edema. LC_{50} values for the two of
41 the three rat studies were 1277 ppm (Pauluhn 1987), and 736 ppm (Nachreiner 1993). Data from
42 the Kinkead and Einhaus study (1984) suggested lethality values much higher than Nachreiner
43 (1993) and Pauluhn (1987), but due to the study design, were not considered useful.

4. SPECIAL CONSIDERATIONS

4.1. Metabolism and Disposition

Thionyl chloride is rapidly hydrolyzed to sulfur dioxide and hydrogen chloride and the metabolism and disposition follow those two compounds. Following inhalation, sulfur dioxide is distributed throughout the body after dissolving into surface fluid. Some remains in the respiratory system for a week or more following exposure. Urinary excretion clears it from the body (Costa 2001). Hydrogen chloride dissolves in the nasal passages. Hydrogen chloride is not metabolized; however, hydrogen and chloride ions resulting from adsorption in the respiratory tract may be distributed throughout the body (NRC, 2000).

4.2. Mechanism of Toxicity

Sulfur dioxide acts on the respiratory system via stimulation of bronchoconstriction and mucus secretion in the upper airways. It injures cells lining the airway passages and causes mucus-secreting goblet cells to proliferate. These two events result in airway narrowing and increased airflow resistance (Costa 2001). Inhaled hydrogen chloride irritates the respiratory tract following a latency period of several hours. Following exposure, the epithelial barrier in the alveolar zone breaks down and begins to leak, flooding the alveoli and causing pulmonary edema (Witschi and Last 2001).

4.3. Structure Activity Relationships

There are no structure-activity relationships applicable for estimating acute exposure limits for thionyl chloride.

4.4. Other Relevant Information

4.4.1. Susceptible Populations

Asthmatics have been shown to be sensitive to respiratory effects caused by sulfur dioxide exposure. As noted above, sulfur dioxide stimulates bronchoconstriction and increases airflow resistance. Response to inhalation of sulfur dioxide is enhanced with moderate exertion and/or mouth breathing and can lead to decreased forced expiratory volume (Koren 1995). Exercise exacerbates the respiratory effects of sulfur dioxide in both healthy and asthmatic subjects (Kulle et al. 1984; Horstman et al. 1988). Exercising is also known to increase hydrogen chloride intake and exacerbate respiratory effects.

5. DATA ANALYSIS FOR AEGL-1

5.1. Summary of Human Data Relevant to AEGL-1

Human data were not available for deriving an AEGL.

5.2. Summary of Animal Data Relevant to AEGL-1

Nonlethal toxicity data in animals consistent with AEGL-1 effects were not available.

5.3. Derivation of AEGL-1 Values

No AEGL-1 values were derived because of insufficient data.

TABLE 4. AEGL-1 Values for Thionyl Chloride

10-minute	30-minute	1-hour	4-hour	8-hour
NR	NR	NR	NR	NR

NR-Not recommended. Numeric values for AEGL-1 are not recommended because of insufficient data. Absence of an AEGL-1 value does not imply that exposure below the AEGL-2 concentration is without adverse effects.

6. DATA ANALYSIS FOR AEGL-2

6.1. Summary of Human Data Relevant to AEGL-2

Human data were not available for deriving an AEGL based upon non-lethal, irreversible effects of thionyl chloride exposure.

6.2. Summary of Animal Data Relevant to AEGL-2

Exposure to 71 and 407 ppm produced dyspnea in male and female rats (Pauluhn 1987). Nachreiner (1993) noted lung discoloration at 196 and 371 ppm. It is unknown if discoloration of the organs was reversible.

6.3. Derivation of AEGL-2

The one hour exposure data (71 ppm) from the study by Pauluhn 1987 were selected to derive AEGL-2 values as there was an effect level for a toxic response that was not incapacitating or irreversible. One hour exposures to 71 ppm produced slight dyspnea and nasal irritation in rats. The next higher concentration (407 ppm) caused moderate dyspnea and nasal irritation. A total uncertainty factor of 30 was applied to account for interspecies extrapolation (3) and intraspecies variability (10). An appropriate animal model was used and, the mechanism of action (bronchoconstriction, irritation of the epithelial lining, and increased mucus production via goblet cell proliferation) would not differ across species. Although human data did not provide quantitative information, there was qualitative data that showed similarities to rat thionyl chloride effects; irritation and hyperinflation of the lungs and pulmonary edema. An uncertainty factor of 10 was used for intraspecies variability due to the wide variability in response to sulfur dioxide exposure between healthy and asthmatic humans. Although data on a sensitive subpopulation are lacking for thionyl chloride, it is known that asthmatics are more sensitive than healthy humans to sulfur dioxide, one of thionyl chloride's hydrolysis products. It is believed that the intraspecies uncertainty factor of 10 is protective of asthmatics.

The concentration exposure time relationship for many irritant and systemically acting vapors and gases may be described by $C^n \times t = k$, where the exponent ranges from 0.8 to 3.5 (ten Berge et al. 1986). In lieu of a definitive data set allowing empirical derivation of the exponent n , temporal scaling was performed, using $n = 3$ when extrapolating to shorter time points and $n =$

1 when extrapolating to longer time points using the $C^n \times t = k$ equation (NRC 2001).

TABLE 5. AEGL-2 Values for Thionyl Chloride

10-minute	30-minute	1-hour	4-hour	8-hour
4.3 ppm (21 mg/m ³)	3.0 ppm (15 mg/m ³)	2.4 ppm (12 mg/m ³)	0.59 ppm (2.9 mg/m ³)	0.30 ppm (1.5 mg/m ³)

7. DATA ANALYSIS FOR AEGL-3

7.1. Summary of Human Data Relevant to AEGL-3

Human data were not available for deriving thionyl chloride AEGL-3 values based upon lethality.

7.2. Summary of Animal Data Relevant to AEGL-3

Lethality data were available for rats. Exposures producing lethality range from 593 to 3441 ppm, with all values representing a one hour exposure. Rats exposed to a mixture of the hydrolysis products of thionyl chloride at concentrations at or lower than 1080 ppm survived. A $BMCL_{05}$ of 663 ppm and BMC_{01} of 830 were calculated from this study (Kinkead and Einhaus 1984). No deaths occurred in rats exposed to 769 ppm thionyl chloride; an LC_{50} of 1277 ppm; and $BMCL_{05}$ of 507 ppm and BMC_{01} of 671 ppm were also calculated from the same study (Pauluhn 1987). In the report from Nachreiner (1993) no animals died at or below exposures of 371 ppm; the LC_{50} was 736 ppm, and the $BMCL_{05}$ and BMC_{01} were calculated at 193 and 222 ppm, respectively.

7.3. Derivation of AEGL-3

The data sets for deriving AEGL-3 values are that of Pauluhn (1987) and Nachreiner (1993). Both provide exposure response data for rats exposed to thionyl chloride for one hour at concentrations of 71, 407, 769, 2121, or 3441 ppm (Pauluhn 1987) and 196, 371, 593, 954, and 1241 ppm (Nachreiner 1993). There was 80% mortality at 2121 and 91% mortality at 3441 ppm. No mortality was observed in the 769 ppm exposure group, but wheezing and necrotic changes in the nasal speculum were noted in these animals. At 407 ppm, the animals displayed moderate dyspnea and reddened and swollen noses, but there was no mortality (Pauluhn 1984). There was 58% mortality at 593 and 954 ppm and 83% mortality at 1241 ppm (Nachreiner 1987). No mortality was observed in the 371 ppm exposure group, but lung discoloration was noted in these animals. These data provide a basis for a lethality threshold. A benchmark dose concentration was calculated from the combined data of the Pauluhn (1984) and Nachreiner (1993) studies. However, the p value of the fitted model for the benchmark analysis was too low ($p = 0.0022$), indicating that the model did not fit the combined data set, and the combined benchmark value could not be used. (A p value greater than 0.1 indicates that the model fits the data.) The calculated benchmark dose concentration for either study was not used to derive AEGL-3 values because of the high variability between the two. In the absence of knowledge of which value is “more accurate”, 407 ppm was chosen as the point of departure.

1 The one hour inhalation study by Pauluhn (1987) supported by Nachreiner (1993) provided
 2 the most sound basis and were selected to derive AEGL-3 values as it had the highest
 3 experimental concentration at which no mortality was observed (407 ppm). A total uncertainty
 4 factor of 30 was applied to account for interspecies extrapolation (3) and intraspecies variability
 5 (10). A factor of 3 was applied for interspecies variability since an appropriate animal model
 6 was used and, the mechanism of action (bronchoconstriction, irritation of the epithelial lining,
 7 and increased mucus production via goblet cell proliferation) would not differ across species.
 8 Although human data did not provide quantitative exposure information, they described effects
 9 similar to those seen in rats; irritation and hyperinflation of the lungs and pulmonary edema. An
 10 uncertainty factor of 10 was used for intraspecies variability due to the wide variability in
 11 response to sulfur dioxide exposure between healthy and asthmatic humans. Although data on
 12 sensitive subpopulations are lacking for thionyl chloride, it is known that asthmatics are more
 13 sensitive than healthy humans to sulfur dioxide, one of thionyl chloride's hydrolysis products. It
 14 is believed that the intraspecies uncertainty factor of 10 is protective of asthmatics.

15
 16 The concentration exposure time relationship for many irritant and systemically acting
 17 vapors and gases may be described by $C^n \times t = k$, where the exponent ranges from 0.8 to 3.5 (ten
 18 Berge et al. 1986). In lieu of a definitive data set allowing empirical derivation of the exponent
 19 n, temporal scaling was performed, using $n = 3$ when extrapolating to shorter time points and $n =$
 20 1 when extrapolating to longer time points using the $C^n \times t = k$ equation (NRC 2001).

21
 22 **TABLE 6. AEGL-3 Values for Thionyl Chloride**

10-minute	30-minute	1-hour	4-hour	8-hour
25 ppm (120 mg/m ³)	17 ppm (83 mg/m ³)	14 ppm (68 mg/m ³)	3.4 ppm (17 mg/m ³)	1.7 ppm (8.3 mg/m ³)

24
 25
 26 **8. SUMMARY OF AEGLS**

27
 28 **8.1. AEGL Values and Toxicity Endpoints**

29
 30 Data consistent with the AEGL-1 effects are not available. Therefore, AEGL-1 values are
 31 not recommended. AEGL-2 values are based on experimental concentrations that were neither
 32 incapacitating nor irreversible in rats (71 ppm). AEGL-3 values are based on the highest
 33 experimental concentration that did not produce lethality. AEGL values for thionyl chloride are
 34 listed in Table 7.
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1 **TABLE 7. Summary of AEGL Values**
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Classification	Exposure Duration				
	10-minute	30-minute	1-hour	4-hour	8-hour
AEGL-1 (Notable discomfort)	NR	NR	NR	NR	NR
AEGL-2 (Disabling)	4.3 ppm	3.0 ppm	2.4 ppm	0.59 ppm	0.30 ppm
AEGL-3 (Lethal)	25 ppm	17 ppm	14 ppm	3.4 ppm	1.7 ppm

3
4 NR-Not recommended. Numeric values for AEGL-1 are not recommended because of insufficient data. Absence of
5 an AEGL-1 value does not imply that exposure below the AEGL-2 concentration is without adverse effects.
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8 **8.2. Comparison with Other Standards and Guidelines**

9
10 All currently available standards and guidelines are shown in Table 8. Emergency response
11 planning guideline (ERPG) values and Dutch maximum allowable concentration (MAC) values
12 have been published. The National Institute of Occupational Safety and Health (NIOSH)
13 recommended exposure limits-short term exposure (REL-STEL) and American Conference of
14 Governmental Industrial Hygienists (ACGIH) threshold limit value-short term exposure limits
15 (TLV-STEL) both have ceiling values. The derived AEGL values are consistent with current
16 standards and guidelines.
17

18 Thionyl chloride hydrolyzes into sulfur dioxide and hydrogen chloride. Kinkead and
19 Einhaus (1984) suggest that the toxicity of thionyl chloride is greater than the simple additivity
20 of its hydrolysis products, and this concept was incorporated into ERPG recommendations
21 (AIHA 2002). AEGL values have been recommended for sulfur dioxide (USEPA 2006) and
22 hydrogen chloride (NRC 2004) and are listed in Table 8. A weight of evidence approach using
23 data from exercising asthmatics was used to derive AEGL-2 values for sulfur dioxide. Sulfur
24 dioxide AEGL3 values were derived from the calculated BMCL₀₅ in rats following a 4 hour
25 exposure. Compared to sulfur dioxide, derived AEGL-2 thionyl chloride values are based on
26 animal studies and are greater at timepoints of 1 hour or less. The AEGL-3 values for thionyl
27 chloride are lower than the same values for sulfur dioxide and are expected to be protective.
28 Hydrogen chloride AEGL-2 values were based on histopathology and the mouse RD₅₀, and
29 AEGL-3 values were derived from the estimated NOEL for death from the one hour rat LC₅₀.
30 Hydrogen chloride is well scrubbed by the upper respiratory tract and would require higher
31 concentrations to cause adverse effects. The derived thionyl chloride AEGL values are lower
32 than both of the hydrogen chloride values and would be expected to be protective of individuals
33 exposed. No other standards and guidelines are available for thionyl chloride.
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TABLE 8. Extant Standards and Guidelines for Thionyl Chloride

Guideline	Exposure Duration				
	10 minute	30 minute	1 hour	4 hour	8 hour
AEGL-1	NR	NR	NR	NR	NR
AEGL-2	4.3 ppm	3.0 ppm	2.4 ppm	0.59 ppm	0.30 ppm
AEGL-3	25 ppm	17 ppm	14 ppm	3.4 ppm	1.7 ppm
AEGL-1 SO ₂	0.20 ppm	0.20 ppm	0.20 ppm	0.20 ppm	0.20 ppm
AEGL-2 SO ₂	0.75 ppm	0.75 ppm	0.75 ppm	0.75 ppm	0.75 ppm
AEGL-3 SO ₂ *	30 ppm	30 ppm	30 ppm	19 ppm	9.6 ppm
AEGL-1 HCl	1.8 ppm	1.8 ppm	1.8 ppm	1.8 ppm	1.8 ppm
AEGL-2 HCl	100 ppm	43 ppm	22 ppm	11 ppm	11 ppm
AEGL-3 HCl	620 ppm	210 ppm	100 ppm	26 ppm	26 ppm
ERPG-1 (AIHA) ^a			0.2 ppm		
ERPG-2 (AIHA) ^a			2 ppm		
ERPG-3 (AIHA) ^a			10 ppm		
REL-STEL (NIOSH) ^b					1 ppm ceiling
TLV-STEL (ACGIH) ^c					1 ppm ceiling
MAC (The Netherlands) ^d					1 ppm

NR-Not recommended. Numeric values for AEGL-1 are not recommended because of insufficient data. Absence of an AEGL-1 value does not imply that exposure below the AEGL-2 concentration is without adverse effects.

*Proposed AEGL-3 values adopted at the NAC-41 meeting.

^aERPG (Emergency Response Planning Guidelines, American Industrial Hygiene Association (AIHA 2006)

- The ERPG-1 is the maximum airborne concentration below which it is believed nearly all individuals could be exposed for up to one hour without experiencing other than mild, transient adverse health effects or without perceiving a clearly defined objectionable odor. The ERPG-1 is based on increased airway resistance in exercising asthmatics exposed to sulfur dioxide.
- The ERPG-2 is the maximum airborne concentration below which it is believed nearly all individuals could be exposed for up to one hour without experiencing or developing irreversible or other serious health effects or symptoms that could impair an individual's ability to take protective action. The ERPG-2 is based on bronchoconstriction in resting asthmatics exposed to 5 ppm of sulfur dioxide.
- The ERPG-3 is the maximum airborne concentration below which it is believed nearly all individuals could be exposed for up to one hour without experiencing or developing life-threatening health effects. The ERPG-3 is based on the toxicity of sulfur dioxide with an additional safety factor because the mixture of sulfur dioxide and hydrogen chloride is more toxic than simple additivation of the two compounds would indicate.

^bNIOSH REL-STEL (Recommended Exposure Limits - Short Term Exposure Limit) (NIOSH 2005) is defined analogous to the ACGIH TLV-STEL.

^cACGIH TLV-STEL (Threshold Limit Value - Short Term Exposure Limit) (ACGIH 2006)

1 is defined as a 15-minute TWA exposure which should not be exceeded at any time during the
2 workday even if the 8-hour TWA is within the TLV-TWA. Exposures above the TLV-TWA up
3 to the STEL should not be longer than 15 minutes and should not occur more than 4 times per
4 day. There should be at least 60 minutes between successive exposures in this range.
5

6 ^dMAC Ministry of Social Affairs and Employment (SDU Uitgevers (Maximaal Aanvaarde
7 Concentratie [Maximal Accepted Concentration]), The Hague, The Netherlands 2000) is defined
8 analogous to the ACGIH-TLV-TWA.
9

10 **8.3. Data Adequacy and Research Needs**

11

12 The data reported for human exposure to thionyl chloride are qualitative only and do not
13 contain actual concentration and /or duration parameters. The absence of exposure
14 concentrations for humans is one key area where data are lacking. However, the data confirm
15 the respiratory effects of thionyl chloride exposure, and add support to the toxicological
16 endpoints of the animal data. Quantitative animal data are available from three studies that
17 demonstrate a respiratory response similar to that observed in humans. The animal data are
18 sufficient for showing lethality and non-incapacitating exposures. Additional data providing
19 information at concentrations below 71ppm (dyspnea in rats) would be useful to determine a
20 NOAEL or reversible mild irritation level in rats to derive AEGL-1 values. There are no data for
21 addressing exposure concentration-duration relationships as all available studies exposed the
22 animals for only one hour. Data on more exposure durations would be useful in the development
23 of a more precise temporal extrapolation for the development of AEGL values of varying
24 exposure time durations.

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APPENDIX A: Derivation of AEGL Values

Thionyl Chloride

INTERIM: 05/2008

- 1 Derivation of AEGL-1
- 2 No AEGL-1 values were derived due to insufficient data.
- 3
- 4 Key Study:
- 5
- 6 Toxicity endpoint:
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- 8 Time scaling:
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- 10 Uncertainty factors:
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- 12 Modifying factor:
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- 14 Calculations:
- 15
- 16 10-minute AEGL-1
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- 18 30-minute AEGL-1
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- 20 1-hour AEGL-1
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- 22 4-hour AEGL-1
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- 24 8-hour AEGL-1
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1 Derivation of AEGL-2 Values

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3
4 Key Study: Pauluhn (1987), based upon slight dyspnea and reddened and swollen noses at 71
5 ppm for one hour.

6
7 Toxicity endpoint: Slight dyspnea.

8
9 Time scaling: The concentration exposure time relationship for many irritant and systemically
10 acting vapors and gases may be described by $C^n \times t = k$, where the exponent ranges from 0.8 to
11 3.5 (ten Berge et al. 1986). In lieu of a definitive data set allowing empirical derivation of the
12 exponent n, temporal scaling was performed, using n = 3 when extrapolating to shorter time
13 points and n = 1 when extrapolating to longer time points using the $C^n \times t = k$ equation (NRC
14 2001).

15
16 Uncertainty factors: A total uncertainty factor of 30 was applied. A factor of 3 was applied for
17 interspecies variability. An appropriate animal model was used and, the mechanism of action
18 (bronchoconstriction, irritation of the epithelial lining, and increased mucus production via
19 goblet cell proliferation) would not differ across species. Although human data did not provide
20 quantitative exposure data, they did show similar effects; irritation and hyperinflation of the
21 lungs and pulmonary edema. An uncertainty factor of 10 was used for intraspecies variability.
22 Although data on a sensitive subpopulation are lacking for thionyl chloride, it is known that
23 asthmatics are more sensitive than healthy humans to sulfur dioxide, one of thionyl chloride's
24 hydrolysis products. It is believed that the intraspecies uncertainty factor of 10 is protective of
25 asthmatics.

26
27
28 Modifying factor: None

29
30 Calculations: $71 \text{ ppm}/30 = 2.37 \text{ ppm}$
31 $C^3 \times t = k$
32 $(2.37 \text{ ppm})^3 \times 60 \text{ min} = 798.7 \text{ ppm}^3 \cdot \text{min}$

33
34 $C^1 \times t = k$
35 $2.37 \text{ ppm} \times 60 \text{ min} = 142.2 \text{ ppm} \cdot \text{min}$

36
37 10-minute AEGL-2 $C^3 \times 10 \text{ min} = 798.7 \text{ ppm}^3 \cdot \text{min}$
38 $C = 4.3 \text{ ppm}$

39 30-minute AEGL-2 $C^3 \times 30 \text{ min} = 798.7 \text{ ppm}^3 \cdot \text{min}$
40 $C = 3.0 \text{ ppm}$

41 1-hour AEGL-2 $C^3 \times 60 \text{ min} = 798.7 \text{ ppm}^3 \cdot \text{min}$
42 $C = 2.4 \text{ ppm}$

43 4-hour AEGL-2 $C^1 \times 240 \text{ min} = 142.2 \text{ ppm} \cdot \text{min}$
44 $C = 0.59 \text{ ppm}$

45 8-hour AEGL-2 $C^1 \times 480 \text{ min} = 142.2 \text{ ppm} \cdot \text{min}$
46 $C = 0.30 \text{ ppm}$

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1 Derivation of AEGL-3 Values

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Key Studies: Pauluhn (1987) and Nachreiner (1993), based upon an estimate of a lethality threshold experimental concentration of 407 ppm. Experimental concentrations greater than 407 ppm produced mortality.

Toxicity endpoint: Threshold for lethality (407 ppm)

10 Time scaling: $C^n \times t = k$, temporal scaling, using $n = 3$ when extrapolation to shorter time points
11 and $n = 1$ when extrapolating to longer time points due to lack of data to derive the value of n
12 (NRC 2001).

13
14 Uncertainty factors: A total uncertainty factor of 30 was applied to account for interspecies
15 extrapolation. A factor of 3 was applied for interspecies variability. An appropriate animal
16 model was used and, the mechanism of action (bronchoconstriction, irritation of the epithelial
17 lining, and increased mucus production via goblet cell proliferation) would not differ across
18 species. Although human data did show similar effects; irritation and hyperinflation of the lungs
19 and pulmonary edema. An uncertainty factor of 10 was used for intraspecies variability.
20 Although data on a sensitive subpopulation are lacking for thionyl chloride, it is known that
21 asthmatics are more sensitive than healthy humans to sulfur dioxide, one of thionyl chloride's
22 hydrolysis products. It is believed that the intraspecies uncertainty factor of 10 is protective of
23 asthmatics.

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25

Modifying factor: None

26
27

Calculations: $407 \text{ ppm}/30 = 13.57 \text{ ppm}$

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29

$$C^3 \times t = k$$

$$(13.57 \text{ ppm})^3 \times 60 \text{ min} = 149930 \text{ ppm}^3 \cdot \text{min}$$

30
31

$$C^1 \times t = k$$

$$13.57 \text{ ppm} \times 60 \text{ min} = 814.2 \text{ ppm min}$$

32
33

34 10-minute AEGL-3 $C^3 \times 10 \text{ min} = 149930 \text{ ppm}^3 \cdot \text{min}$

35

$$C = 25 \text{ ppm}$$

36

36 30-minute AEGL-3 $C^3 \times 30 \text{ min} = 149930 \text{ ppm}^3 \cdot \text{min}$

37

$$C = 17 \text{ ppm}$$

38

38 1-hour AEGL-3 $C^3 \times 60 \text{ min} = 149930 \text{ ppm}^3 \cdot \text{min}$

39

$$C = 14 \text{ ppm}$$

40

40 4-hour AEGL-3 $C^1 \times 240 \text{ min} = 814.2 \text{ ppm} \cdot \text{min}$

41

$$C = 3.4 \text{ ppm}$$

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42 8-hour AEGL-3 $C^1 \times 480 \text{ min} = 814.2 \text{ ppm} \cdot \text{min}$

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$$C = 1.7 \text{ ppm}$$

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APPENDIX B: Time-Scaling Calculations

Data were unavailable to empirically derive a scaling factor (n) for thionyl chloride. The concentration exposure time relationship for many irritant and systemically acting vapors and gases may be described by $C^n \times t = k$, where the exponent ranges from 0.8 to 3.5 (ten Berge et al. 1986). In the absence of an empirically derived exponent (n), and to obtain AEGL values, temporal scaling was performed, using $n = 3$ when extrapolating to shorter time points and $n = 1$ when extrapolating to longer time points using the $C^n \times t = k$ equation (NRC 2001).

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APPENDIX C: Derivation Summary for Thionyl Chloride

1 ACUTE EXPOSURE GUIDELINE LEVELS FOR
 2 THIONYL CHLORIDE (CAS Reg. No. 7719-09-7)
 3 DERIVATION SUMMARY

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 5
 6

AEGL-1 VALUES

10-minute	30-minute	1-hour	4-hour	8-hour
NR	NR	NR	NR	NR
Key Reference: There were insufficient data to derive AEGL-1 values for thionyl chloride.				
Test Species/Strain/Number: Not Applicable				
Exposure Route/Concentrations/Durations: Not Applicable				
Effects: Not Applicable				
Endpoint/Concentration/Rationale: Not Applicable				
Uncertainty Factors/Rationale: Not Applicable				
Modifying Factor: Not Applicable				
Animal to Human Dosimetric Adjustment: Not Applicable				
Time Scaling: Not Applicable				
Data Adequacy: NR-Not recommended. Numeric values for AEGL-1 are not recommended because of insufficient data. Absence of an AEGL-1 value does not imply that exposure below the AEGL-2 concentration is without adverse effects.				

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AEGL-2 VALUES

10-minute	30-minute	1-hour	4-hour	8-hour
4.3 ppm	3.0 ppm	2.4 ppm	0.59 ppm	0.30 ppm
Key Reference: Pauluhn, J. 1987. Study for acute inhalation toxicity in rats in accordance with OECG Guideline No. 403 (Exposure: 1 x 1 Hour). Report No. 15403. Leverkusen, Germany: Bayer AG.				
Test Species/Strain/Number: Male and female Wistar rats, 5/gender/group				
Exposure Route/Concentrations/Durations: Inhalation (Head/Nose Only): 0, 71, 407, 769, 2121, and 3441 ppm for 1 hour				
Effects: dyspnea, reddened and swollen noses 71 ppm 0% mortality, reddened and swollen noses, slight dyspnea (determinant for AEGL-2) 407 ppm 0% mortality, reddened and swollen noses, moderate dyspnea 769 ppm 0% mortality, reddened and swollen noses, moderate dyspnea, wheezing, necrotic nasal changes, lung distention 2121 ppm 80% mortality (80% male: 4 post exposure, 80% female: 4 post exposure) 3441 ppm 90% mortality (100% male: 5 post exposure, 80% female: 4 post exposure)				
Endpoint/Concentration/Rationale: 71 ppm for 1 hour considered for basis for AEGL-2 values				
Uncertainty Factors/Rationale: Total uncertainty factor: 30 Interspecies: 3- An appropriate animal model was used and, the mechanism of action (bronchoconstriction, irritation of the epithelial lining, and increased mucus production via goblet cell proliferation) would not differ across species. Although human data did not provide quantitative exposure data, they did provide show similar effects; irritation and hyperinflation of the lungs and pulmonary edema. Intraspecies: 10- Although data on a sensitive subpopulation are lacking for thionyl chloride, it is known that asthmatics are sensitive to sulfur dioxide, one of thionyl chloride's hydrolysis products. It is believed that the intraspecies uncertainty factor of 10 is protective of asthmatics and other sensitive populations.				
Modifying Factor: Not Applicable				
Animal to Human Dosimetric Adjustment: None applied, insufficient data.				
Time Scaling: In the absence of an empirically derived exponent (n), and to obtain AEGL values, $C^n \times t = k$, temporal scaling, using $n = 3$ when extrapolating to shorter time points and $n = 1$ when extrapolating to longer time points. The concentration exposure time relationship for many irritant and systemically acting vapors and gases may be described by $C^n \times t = k$, where the exponent ranges from 0.8 to 3.5 (ten Berge et al. 1986).				
Data Adequacy: The study was considered adequate for AEGL-2 derivation. It was a well-designed and performed OECD Guideline study, adequate numbers of animals were used, and an endpoint consistent with AEGL-2 definition and toxicity of thionyl chloride was observed.				

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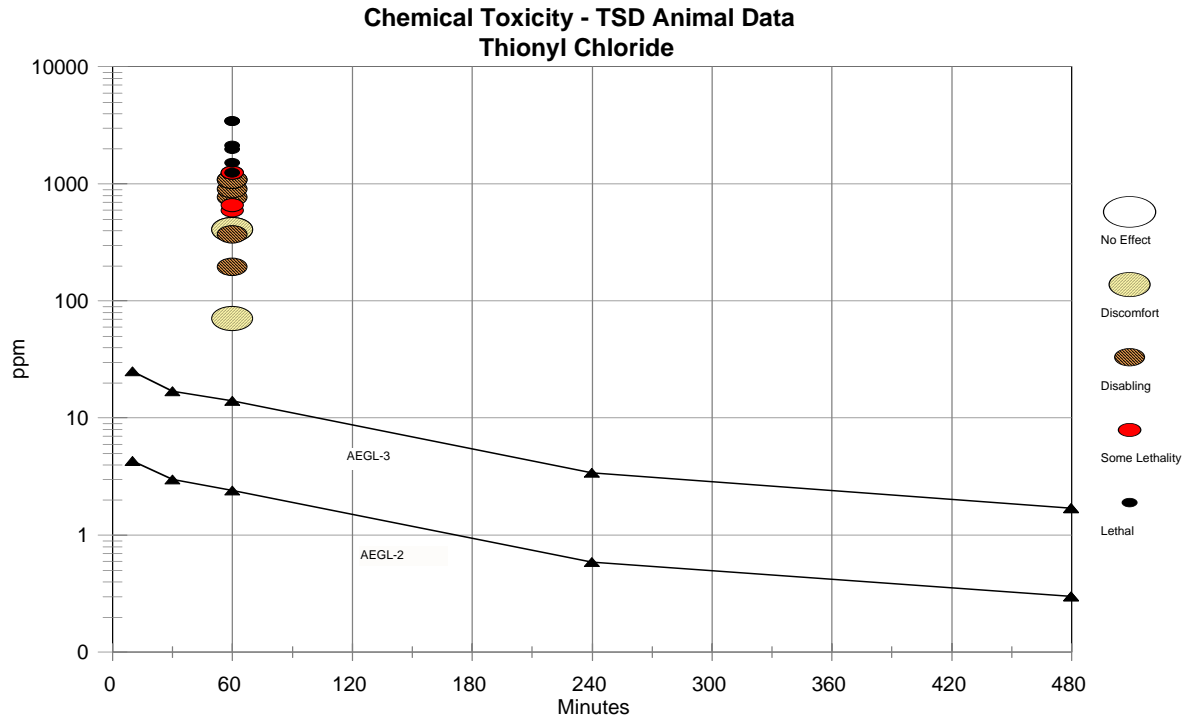
AEGL-3 VALUES

10-minute	30-minute	1-hour	4-hour	8-hour
25 ppm	17 ppm	14 ppm	3.4 ppm	1.7 ppm
Key Reference: Pauluhn, J. 1987. Study for acute inhalation toxicity in rats in accordance with OECG Guideline No. 403 (Exposure: 1 x 1 Hour). Report No. 15403. Leverkusen, Germany: Bayer AG. Nachreiner, D.J. 1993. Thionyl chloride: Acute vapor inhalation toxicity study in rats. Union Carbide Chemicals and Plastics Company, Inc. Export, PA				
Test Species/Strain/Number: Male and female Wistar rats, 5/gender/group; Male and female Wistar rats, 6/gender/group				
Exposure Route/Concentrations/Durations: Inhalation (Head/Nose Only): 0, 71, 407, 769, 2121, and 3441 ppm for 1 hour; (Nose Only): 0, 196, 371, 593, 954, and 1241 ppm for 1 hour. (The concentration where no deaths were observed, 407 ppm in Pauluhn (1987), was determinant for AEGL-3.)				
Effects: Lethality Pauluhn (1987) 71 ppm 0% mortality, reddened and swollen noses, slight dyspnea (determinant for AEGL-2) 407 ppm 0% mortality, reddened and swollen noses, moderate dyspnea 769 ppm 0% mortality, reddened and swollen noses, moderate dyspnea, wheezing, necrotic nasal changes, lung distention 2121 ppm 80% mortality (80% male: 4 post exposure, 80% female: 4 post exposure) 3441 ppm 90% mortality (100% male: 5 post exposure, 80% female: 4 post exposure) Nachreiner (1993) 196 ppm 0% mortality, lung discoloration 371 ppm 0% mortality, lung discoloration 593 ppm 58% mortality (83% male: 5 during exposure, 33% female: 2 during exposure) 954 ppm 58% mortality (83% male: 5 during exposure, 33% female: 1 during exposure and 1 post exposure) 1241 ppm 83% mortality (100% male: 3 during exposure and 3 post exposure, 67% female: 2 during exposure and 2 post exposure)				
Endpoint/Concentration/Rationale: 407 ppm for 1 hour considered for basis for AEGL-3 values				
Uncertainty Factors/Rationale: Total uncertainty factor: 30 Interspecies: 3- An appropriate animal model was used and, the mechanism of action (bronchoconstriction, irritation of the epithelial lining, and increased mucus production via goblet cell proliferation) would not differ across species. Although human data did not provide quantitative exposure data, they did show similar effects; irritation and hyperinflation of the lungs and pulmonary edema. Intraspecies: 10- Although data on a sensitive subpopulation are lacking for thionyl chloride, it is known that asthmatics are sensitive to sulfur dioxide, one of thionyl chloride's hydrolysis products. It is believed that the intraspecies uncertainty factor of 10 is protective of asthmatics and other sensitive populations.				
Modifying Factor: None				
Animal to Human Dosimetric Adjustment: None applied, insufficient data				
Time Scaling: In the absence of an empirically derived exponent (n), and to obtain AEGL values, $C^n \times t = k$, temporal scaling, using $n = 3$ when extrapolating to shorter time points and $n = 1$ when extrapolating to longer time points. The concentration exposure time relationship for many irritant and systemically acting vapors and gases may be described by $C^n \times t = k$, where the exponent ranges from 0.8 to 3.5 (ten Berge et al. 1986).				
Data Adequacy: The study was considered adequate for AEGL-3 derivation. It was a well-designed and performed study, adequate numbers of animals were used, and an endpoint consistent with the AEGL-3 definition and toxicity of thionyl chloride was observed.				

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APPENDIX D: Category Plot for Thionyl Chloride



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