

# Acute Exposure Guideline Levels for Selected Airborne Chemicals

## Volume 4

Subcommittee on Acute Exposure Guideline Levels

Committee on Toxicology

Board on Environmental Studies and Toxicology

**NATIONAL RESEARCH COUNCIL**  
*OF THE NATIONAL ACADEMIES*

THE NATIONAL ACADEMIES PRESS

Washington, D.C.

**[www.nap.edu](http://www.nap.edu)**

**THE NATIONAL ACADEMIES PRESS    500 Fifth Street, NW    Washington, DC 20001**

NOTICE: The project that is the subject of this report was approved by the Governing Board of the National Research Council, whose members are drawn from the councils of the National Academy of Sciences, the National Academy of Engineering, and the Institute of Medicine. The members of the committee responsible for the report were chosen for their special competences and with regard for appropriate balance.

This project was supported by Contract Nos. DAMD17-89-C-9086 and DAMD17-99-C-9049 between the National Academy of Sciences and the U.S. Army. Any opinions, findings, conclusions, or recommendations expressed in this publication are those of the author(s) and do not necessarily reflect the view of the organizations or agencies that provided support for this project.

International Standard Book Number 0-309-09147-0 (Book)

International Standard Book Number 0-309-53013-X (PDF)

Additional copies of this report are available from:

The National Academies Press  
500 Fifth Street., NW  
Box 285  
Washington, DC 20055

800-624-6242  
202-334-3313 (in the Washington metropolitan area)  
<http://www.nap.edu>

Copyright 2004 by the National Academy of Sciences. All rights reserved.

Printed in the United States of America

# THE NATIONAL ACADEMIES

## *Advisers to the Nation on Science, Engineering, and Medicine*

The **National Academy of Sciences** is a private, nonprofit, self-perpetuating society of distinguished scholars engaged in scientific and engineering research, dedicated to the furtherance of science and technology and to their use for the general welfare. Upon the authority of the charter granted to it by the Congress in 1863, the Academy has a mandate that requires it to advise the federal government on scientific and technical matters. Dr. Bruce M. Alberts is president of the National Academy of Sciences.

The **National Academy of Engineering** was established in 1964, under the charter of the National Academy of Sciences, as a parallel organization of outstanding engineers. It is autonomous in its administration and in the selection of its members, sharing with the National Academy of Sciences the responsibility for advising the federal government. The National Academy of Engineering also sponsors engineering programs aimed at meeting national needs, encourages education and research, and recognizes the superior achievements of engineers. Dr. Wm. A. Wulf is president of the National Academy of Engineering.

The **Institute of Medicine** was established in 1970 by the National Academy of Sciences to secure the services of eminent members of appropriate professions in the examination of policy matters pertaining to the health of the public. The Institute acts under the responsibility given to the National Academy of Sciences by its congressional charter to be an adviser to the federal government and, upon its own initiative, to identify issues of medical care, research, and education. Dr. Harvey V. Fineberg is president of the Institute of Medicine.

The **National Research Council** was organized by the National Academy of Sciences in 1916 to associate the broad community of science and technology with the Academy's purposes of furthering knowledge and advising the federal government. Functioning in accordance with general policies determined by the Academy, the Council has become the principal operating agency of both the National Academy of Sciences and the National Academy of Engineering in providing services to the government, the public, and the scientific and engineering communities. The Council is administered jointly by both Academies and the Institute of Medicine. Dr. Bruce M. Alberts and Dr. Wm. A. Wulf are chair and vice chair, respectively, of the National Research Council

**[www.national-academies.org](http://www.national-academies.org)**



## SUBCOMMITTEE ON ACUTE EXPOSURE GUIDELINE LEVELS

### *Members*

**DANIEL KREWSKI** (*Chair*), University of Ottawa, Ottawa, Ontario, Canada  
**EDWARD C. BISHOP**, Parsons Corporation, Pasadena, CA  
**JAMES V. BRUCKNER**, University of Georgia, Athens  
**DAVID P. KELLY**, Dupont Company, Newark, DE  
**KANNAN KRISHNAN**, University of Montreal, Montreal, Quebec, Canada  
**STEPHEN U. LESTER**, Center for Health, Environment and Justice, Falls Church,  
VA  
**JUDITH MACGREGOR**, Toxicology Consulting Services, Arnold, MD  
**PATRICIA MCGINNIS**, Syracuse Research Corporation, Ft. Washington, PA  
**FRANZ OESCH**, University of Mainz, Mainz, Germany  
**RICHARD B. SCHLESINGER**, Pace University, Pleasantville, NY  
**CALVIN C. WILLHITE**, Department of Toxic Substances, State of California,  
Berkeley  
**FREDERIK A. DE WOLFF**, Leiden University, Leiden, The Netherlands

### *Staff*

**KULBIR S. BAKSHI**, Program Director  
**KELLY CLARK**, Editor  
**AIDA C. NEEL**, Senior Project Assistant

## COMMITTEE ON TOXICOLOGY

### *Members*

**BAILUS WALKER, JR.** (*Chair*), Howard University Medical Center and American Public Health Association, Washington, DC  
**MELVIN E. ANDERSEN**, CIIT-Centers for Health Research, Research Triangle Park, NC  
**EDWARD C. BISHOP**, Parsons Corporation, Pasadena, CA  
**GARY P. CARLSON**, Purdue University, West Lafayette, IN  
**JANICE E. CHAMBERS**, Mississippi State University, Mississippi State  
**LEONARD CHIAZZE, JR.**, Georgetown University, Washington, DC  
**JUDITH A. GRAHAM**, American Chemistry Council, Arlington, VA  
**SIDNEY GREEN**, Howard University, Washington, DC  
**MERYL KAROL**, University of Pittsburgh, Pittsburgh, PA  
**STEPHEN U. LESTER**, Center for Health Environment and Justice, Falls Church, VA  
**DAVID H. MOORE**, Battelle Memorial Institute, Bel Air, MD  
**CALVIN C. WILLHITE**, Department of Toxic Substances, State of California, Berkeley  
**GERALD WOGAN**, Massachusetts Institute of Technology, Cambridge

### *Staff*

**KULBIR S. BAKSHI**, Program Director for Toxicology  
**ROBERTA M. WEDGE**, Program Director for Risk Analysis  
**SUSAN N.J. MARTEL**, Senior Staff Officer  
**ELLEN K. MANTUS**, Senior Staff Officer  
**KELLY CLARK**, Assistant Editor  
**AIDA C. NEEL**, Senior Project Assistant  
**TAMARA DAWSON**, Project Assistant

## BOARD ON ENVIRONMENTAL STUDIES AND TOXICOLOGY<sup>1</sup>

### *Members*

**JONATHAN SAMET** (*Chair*), Johns Hopkins University, Baltimore, MD  
**DAVID ALLEN**, University of Texas, Austin, Texas  
**THOMAS BURKE**, Johns Hopkins University, Baltimore, MD  
**JUDITH C. CHOW**, Desert Research Institute, Reno, NV  
**COSTEL D. DENSON**, University of Delaware, Newark  
**E. DONALD ELLIOTT**, Willkie, Farr & Gallagher, LLP, Washington, DC  
**CRISTOPHER B. FIELD**, Carnegie Institute of Washington, Stanford, CA  
**WILLIAM H. GLAZE**, Oregon Health and Sciences University, Beaverton  
**SHERRI W. GOODMAN**, Center for Naval Analyses Corporation, Alexandria, VA  
**DANIEL S. GREENBAUM**, Health Effects Institute, Cambridge, MA  
**ROGENE HENDERSON**, Lovelace Respiratory Research Institute, Albuquerque, NM  
**CAROL HENRY**, American Chemistry Council, Arlington, VA  
**ROBERT HUGGETT**, Michigan State University, East Lansing  
**BARRY L. JOHNSON**, Emory University, Atlanta, GA  
**JAMES H. JOHNSON**, Howard University, Washington, DC  
**JUDITH L. MEYER**, University of Georgia, Athens  
**PATRICK V. O'BRIEN**, Chevron Research and Technology, Richmond, CA  
**DOROTHY E. PATTON**, International Life Sciences Institute, Washington, DC  
**STEWART T.A. PICKETT**, Institute of Ecosystems Studies, Millbrook, NY  
**ARMISTEAD G. RUSSELL**, Georgia Institute of Technology, Atlanta  
**LOUISE M. RYAN**, Harvard University, Boston, MA  
**KIRK SMITH**, University of California, Berkeley  
**LISA SPEER**, Natural Resources Defense Council, New York, NY  
**G. DAVID TILMAN**, University of Minnesota, St. Paul  
**CHRIS G. WHIPPLE**, Environ, Inc., Emeryville, CA  
**LAUREEN A. ZEISE**, California Environmental Protection Agency, Oakland

### *Senior Staff*

**JAMES J. REISA**, Director  
**DAVID J. POLICANSKY**, Associate Director  
**RAYMOND A. WASSEL**, Senior Program Director for Environmental Sciences and Engineering  
**KULBIR BAKSHI**, Program Director for Toxicology  
**ROBERTA M. WEDGE**, Program Director for Risk Analysis  
**K. JOHN HOLMES**, Senior Staff Officer  
**SUSAN N. J. MARTEL**, Senior Staff Officer  
**SUZANNE VAN DRUNICK**, Senior Staff Officer  
**EILEEN N. ABT**, Senior Staff Officer  
**ELLEN K. MANTUS**, Senior Staff Officer  
**RUTH E. CROSSGROVE**, Managing Editor

---

<sup>1</sup>This study was planned, overseen, and supported by the Board on Environmental Studies and Toxicology.

**OTHER REPORTS OF THE  
BOARD ON ENVIRONMENTAL STUDIES AND TOXICOLOGY**

Air Quality Management in the United States (2004)  
Endangered and Threatened Fishes in the Klamath River Basin: Causes of Decline  
and Strategies for Recovery (2004)  
Cumulative Environmental Effects of Alaska North Slope Oil and Gas Development  
(2003)  
Estimating the Public Health Benefits of Proposed Air Pollution Regulations (2002)  
Biosolids Applied to Land: Advancing Standards and Practices (2002)  
The Airliner Cabin Environment and Health of Passengers and Crew (2002)  
Arsenic in Drinking Water: 2001 Update (2001)  
Evaluating Vehicle Emissions Inspection and Maintenance Programs (2001)  
Compensating for Wetland Losses Under the Clean Water Act (2001)  
A Risk-Management Strategy for PCB-Contaminated Sediments (2001)  
Acute Exposure Guideline Levels for Selected Airborne Chemicals (4 volumes,  
2000-2004)  
Toxicological Effects of Methylmercury (2000)  
Strengthening Science at the U.S. Environmental Protection Agency (2000)  
Scientific Frontiers in Developmental Toxicology and Risk Assessment (2000)  
Ecological Indicators for the Nation (2000)  
Waste Incineration and Public Health (1999)  
Hormonally Active Agents in the Environment (1999)  
Research Priorities for Airborne Particulate Matter (4 volumes, 1998-2003)  
Arsenic in Drinking Water (1999)  
The National Research Council's Committee on Toxicology: The First 50 Years  
(1997)  
Carcinogens and Anticarcinogens in the Human Diet (1996)  
Upstream: Salmon and Society in the Pacific Northwest (1996)  
Science and the Endangered Species Act (1995)  
Wetlands: Characteristics and Boundaries (1995)  
Biologic Markers (5 volumes, 1989-1995)  
Review of EPA's Environmental Monitoring and Assessment Program (3 volumes,  
1994-1995)  
Science and Judgment in Risk Assessment (1994)  
Pesticides in the Diets of Infants and Children (1993)  
Dolphins and the Tuna Industry (1992)  
Science and the National Parks (1992)  
Human Exposure Assessment for Airborne Pollutants (1991)  
Rethinking the Ozone Problem in Urban and Regional Air Pollution (1991)  
Decline of the Sea Turtles (1990)

*Copies of these reports may be ordered from the National Academies Press  
(800) 624-6242 or (202) 334-3313  
[www.nap.edu](http://www.nap.edu)*

## OTHER REPORTS OF THE COMMITTEE ON TOXICOLOGY

- Spacecraft Water Exposure Guidelines for Selected Contaminants, Volume 1 (2004)
- Toxicologic Assessment of Jet-Propulsion Fuel 8 (2003)
- Review of Submarine Escape Action Levels for Selected Chemicals (2002)
- Standing Operating Procedures for Developing Acute Exposure Guideline Levels for Hazardous Chemicals (2001)
- Evaluating Chemical and Other Agent Exposures for Reproductive and Developmental Toxicity (2001)
- Acute Exposure Guideline Levels for Selected Airborne Contaminants, Volume 1 (2000), Volume 2 (2002), Volume 3 (2003), Volume 4 (2004)
- Review of the US Navy's Human Health Risk Assessment of the Naval Air Facility at Atsugi, Japan (2000)
- Methods for Developing Spacecraft Water Exposure Guidelines (2000)
- Review of the U.S. Navy Environmental Health Center's Health-Hazard Assessment Process (2000)
- Review of the U.S. Navy's Exposure Standard for Manufactured Vitreous Fibers (2000)
- Re-Evaluation of Drinking-Water Guidelines for Diisopropyl Methylphosphonate (2000)
- Submarine Exposure Guidance Levels for Selected Hydrofluorocarbons: HFC-236fa, HFC-23, and HFC-404a (2000)
- Review of the U.S. Army's Health Risk Assessments for Oral Exposure to Six Chemical-Warfare Agents (1999)
- Toxicity of Military Smokes and Obscurants, Volume 1(1997), Volume 2 (1999), Volume 3 (1999)
- Assessment of Exposure-Response Functions for Rocket-Emission Toxicants (1998)
- Toxicity of Alternatives to Chlorofluorocarbons: HFC-134a and HCFC-123 (1996)
- Permissible Exposure Levels for Selected Military Fuel Vapors (1996)
- Spacecraft Maximum Allowable Concentrations for Selected Airborne Contaminants, Volume 1 (1994), Volume 2 (1996), Volume 3 (1996), Volume 4 (2000)



## Preface

Extremely hazardous substances (EHSs)<sup>1</sup> can be released accidentally as a result of chemical spills, industrial explosions, fires, or accidents involving railroad cars and trucks transporting EHSs. The people in communities surrounding industrial facilities where EHSs are manufactured, used, or stored and in communities along the nation's railways and highways potentially are at risk of being exposed to airborne EHSs during accidental releases. Pursuant to the Superfund Amendments and Reauthorization Act of 1986, the U.S. Environmental Protection Agency (EPA) has identified approximately 400 EHSs on the basis of acute lethality data in rodents.

As part of its efforts to develop acute exposure guideline levels for EHSs, EPA and the Agency for Toxic Substances and Disease Registry (ATSDR) in 1991 requested that the National Research Council (NRC) develop guidelines for establishing such levels. In response to that request, the NRC published *Guidelines for Developing Community Emergency Exposure Levels for Hazardous Substances* in 1993.

Using the 1993 NRC guidelines report, the National Advisory Committee (NAC) on Acute Exposure Guideline Levels for Hazardous Substances—consisting of members from EPA, the Department of Defense (DOD), the Department of Energy (DOE), the Department of Transportation, other federal and state governments, the chemical industry, academe, and other

---

<sup>1</sup>As defined pursuant to the Superfund Amendments and Reauthorization Act of 1986.

organizations from the private sector—has developed acute exposure guideline levels (AEGLs) for approximately 80 EHSs.

In 1998, EPA and DOD requested that the NRC independently review the AEGLs developed by NAC. In response to that request, the NRC organized within its Committee on Toxicology the Subcommittee on Acute Exposure Guideline Levels, which prepared this report. This report is the fourth volume in the series *Acute Exposure Guideline Levels for Selected Airborne Chemicals*. It reviews the AEGLs for chlorine, hydrogen chloride, hydrogen fluoride, toluene 2,4- and 2,6-diisocyanate, and uranium hexafluoride for scientific accuracy, completeness, and consistency with the NRC guideline reports.

This report was reviewed in draft by individuals selected for their diverse perspectives and technical expertise, in accordance with procedures approved by the NRC's Report Review Committee. The purpose of this independent review is to provide candid and critical comments that will assist the institution in making its published report as sound as possible and to ensure that the report meets institutional standards for objectivity, evidence, and responsiveness to the study charge. The review comments and draft manuscript remain confidential to protect the integrity of the deliberative process. We wish to thank the following individuals for their review of this report: David H. Moore of Battelle Memorial Institute; Sam Kacew of University of Ottawa; and Rakesh Dixit of Merck and Company, Inc.

Although the reviewers listed above have provided many constructive comments and suggestions, they were not asked to endorse the conclusions or recommendations nor did they see the final draft of the report before its release. The review of this report was overseen by Janice E. Chambers of Mississippi State University, appointed by the Division on Earth and Life Studies, who was responsible for making certain that an independent examination of this report was carried out in accordance with institutional procedures and that all review comments were carefully considered. Responsibility for the final content of this report rests entirely with the authoring committee and the institution.

The subcommittee gratefully acknowledges the valuable assistance provided by the following people: Ernest Falke and Paul Tobin, EPA; George Rusch, Honeywell, Inc.; Sylvia Talmage, Cheryl Bast, and Carol Wood, Oak Ridge National Laboratory; and Aida Neel, senior project assistant for the Board on Environmental Studies and Toxicology. Kelly Clark edited the report. We are grateful to James J. Reisa, director of the Board on Environmental Studies and Toxicology, for his helpful comments. The subcommittee particularly acknowledges Kulbir Bakshi, project director for

the subcommittee, for bringing the report to completion. Finally, we would like to thank all members of the subcommittee for their expertise and dedicated effort throughout the development of this report.

Daniel Krewski, *Chair*  
Subcommittee on Acute Exposure  
Guideline Levels

Bailus Walker, *Chair*  
Committee on Toxicology



# Contents

Introduction.....	1
Roster of the National Advisory Committee for Acute Exposure Guideline Levels for Hazardous Substances.....	7
Appendixes	
1 Chlorine: Acute Exposure Guideline Levels.....	13
2 Hydrogen Chloride: Acute Exposure Guideline Levels.....	77
3 Hydrogen Fluoride: Acute Exposure Guideline Levels.....	123
4 Toluene 2,4- and 2,6-Diisocyanate: Acute Exposure Guideline Levels.....	198
5 Uranium Hexafluoride: Acute Exposure Guideline Levels.....	250



Acute Exposure Guideline Levels  
for Selected Airborne Chemicals

Volume 4



# Introduction

This report is the fourth volume in the series *Acute Exposure Guideline Levels for Selected Airborne Chemicals*.

In the Bhopal disaster of 1984, approximately 2,000 residents living near a chemical plant were killed and 20,000 more suffered irreversible damage to their eyes and lungs following accidental release of methyl isocyanate. The toll was particularly high because the community had little idea what chemicals were being used at the plant, how dangerous they might be, and what steps to take in case of emergency. This tragedy served to focus international attention on the need for governments to identify hazardous substances and to assist local communities in planning how to deal with emergency exposures.

In the United States, the Superfund Amendments and Reauthorization Act (SARA) of 1986 required that the U.S. Environmental Protection Agency (EPA) identify extremely hazardous substances (EHSs) and, in cooperation with the Federal Emergency Management Agency and the Department of Transportation, assist Local Emergency Planning Committees (LEPCs) by providing guidance for conducting health-hazard assessments for the development of emergency-response plans for sites where EHSs are produced, stored, transported, or used. SARA also required that the Agency for Toxic Substances and Disease Registry (ATSDR) determine whether chemical substances identified at hazardous waste sites or in the environment present a public-health concern.

As a first step in assisting the LEPCs, EPA identified approximately 400 EHSs largely on the basis of their immediately dangerous to life and health (IDLH) values developed by the National Institute for Occupational Safety and Health (NIOSH) in experimental animals. Although several public and private groups, such as the Occupational Safety and Health Administration (OSHA) and the American Conference of Governmental Industrial Hygienists (ACGIH), have established exposure limits for some substances and some exposures (e.g., workplace or ambient air quality), these limits are not easily or directly translated into emergency exposure limits for exposures at high levels but of short duration, usually less than 1 h, and only once in a lifetime for the general population, which includes infants (from birth to 3 years of age), children, the elderly, and persons with diseases, such as asthma or heart disease.

The National Research Council (NRC) Committee on Toxicology (COT) has published many reports on emergency exposure guidance levels and spacecraft maximum allowable concentrations for chemicals used by the Department of Defense (DOD) and the National Aeronautics and Space Administration (NASA) (NRC 1968, 1972, 1984a,b,c,d, 1985a,b, 1986a,b, 1987, 1988, 1994, 1996a,b, 2000). COT has also published guidelines for developing emergency exposure guidance levels for military personnel and for astronauts (NRC 1986b, 1992). Because of COT's experience in recommending emergency exposure levels for short-term exposures, in 1991 EPA and ATSDR requested that COT develop criteria and methods for developing emergency exposure levels for EHSs for the general population. In response to that request, the NRC assigned this project to the COT Subcommittee on Guidelines for Developing Community Emergency Exposure Levels for Hazardous Substances. The report of that subcommittee, *Guidelines for Developing Community Emergency Exposure Levels for Hazardous Substances* (NRC 1993), provides step-by-step guidance for setting emergency exposure levels for EHSs. Guidance is given on what data are needed, what data are available, how to evaluate the data, and how to present the results.

In November 1995, the National Advisory Committee for Acute Exposure Guideline Levels for Hazardous Substances (NAC)<sup>1</sup> was established to identify, review, and interpret relevant toxicologic and other scientific data and to develop acute exposure guideline levels (AEGs) for high-priority, acutely toxic chemicals. The NRC's previous name for acute exposure levels—community emergency exposure levels (CEELs)—was re-

---

<sup>1</sup>NAC is composed of members from EPA, DOD, many other federal and state agencies, industry, academe, and other organizations. The roster of NAC is shown on page 8.

placed by “AEGLs” to reflect the broad application of these values to planning, response, and prevention in the community, the workplace, transportation, the military, and the remediation of Superfund sites.

AEGLs represent threshold exposure limits (exposure levels below which adverse health effects are not likely to occur) for the general public and are applicable to emergency exposures ranging from 10 min to 8 h. Three levels—AEGL-1, AEGL-2, and AEGL-3—are developed for each of five exposure periods (10 min, 30 min, 1 h, 4 h, and 8 h) and are distinguished by varying degrees of severity of toxic effects. The three AEGLs are defined as follows:

AEGL-1 is the airborne concentration (expressed as ppm [parts per million] or  $\text{mg}/\text{m}^3$  [milligrams per cubic meter]) of a substance above which it is predicted that the general population, including susceptible individuals, could experience notable discomfort, irritation, or certain asymptomatic nonsensory effects. However, the effects are not disabling and are transient and reversible upon cessation of exposure.

AEGL-2 is the airborne concentration (expressed as ppm or  $\text{mg}/\text{m}^3$ ) of a substance above which it is predicted that the general population, including susceptible individuals, could experience irreversible or other serious, long-lasting adverse health effects or an impaired ability to escape.

AEGL-3 is the airborne concentration (expressed as ppm or  $\text{mg}/\text{m}^3$ ) of a substance above which it is predicted that the general population, including susceptible individuals, could experience life-threatening adverse health effects or death.

Airborne concentrations below AEGL-1 represent exposure levels that can produce mild and progressively increasing but transient and nondisabling odor, taste, and sensory irritation or certain asymptomatic, nonsensory adverse effects. With increasing airborne concentrations above each AEGL, there is a progressive increase in the likelihood of occurrence and the severity of effects described for each corresponding AEGL. Although the AEGL values represent threshold levels for the general public, including susceptible subpopulations, such as infants, children, the elderly, persons with asthma, and those with other illnesses, it is recognized that individuals, subject to unique or idiosyncratic responses, could experience the effects described at concentrations below the corresponding AEGL.

## SUMMARY OF REPORT ON GUIDELINES FOR DEVELOPING AEGLS

As described in the *Guidelines for Developing Community Emergency Exposure Levels for Hazardous Substances* (NRC 1993) and the NAC guidelines report *Standing Operating Procedures on Acute Exposure Guideline Levels for Hazardous Substances* (NRC 2001), the first step in establishing AEGLs for a chemical is to collect and review all relevant published and unpublished information available on that chemical. Various types of evidence are assessed in establishing AEGL values for a chemical. They include information from (1) chemical-physical characterizations, (2) structure-activity relationships, (3) in vitro toxicity studies, (4) animal toxicity studies, (5) controlled human studies, (6) observations of humans involved in chemical accidents, and (7) epidemiologic studies. Toxicity data from human studies are most applicable and are used when available in preference to data from animal studies and in vitro studies. Toxicity data from inhalation exposures are most useful for setting AEGLs for airborne chemicals because inhalation is the most likely route of exposure and because extrapolation of data from other routes would lead to additional uncertainty in the AEGL estimate.

For most chemicals, actual human toxicity data are not available or critical information on exposure is lacking, so toxicity data from studies conducted in laboratory animals are extrapolated to estimate the potential toxicity in humans. Such extrapolation requires experienced scientific judgment. The toxicity data from animal species most representative of humans in terms of pharmacodynamic and pharmacokinetic properties are used for determining AEGLs. If data are not available on the species that best represents humans, the data from the most sensitive animal species are used to set AEGLs. Uncertainty factors are commonly used when animal data are used to estimate risk levels for humans. The magnitude of uncertainty factors depends on the quality of the animal data used to determine the no-observed-adverse-effect level (NOAEL) and the mode of action of the substance in question. When available, pharmacokinetic data on tissue doses are considered for interspecies extrapolation.

For substances that affect several organ systems or have multiple effects, all end points, including reproductive (in both genders), developmental, neurotoxic, respiratory, and other organ-related effects, are evaluated, the most important or most sensitive effect receiving the greatest attention. For carcinogenic chemicals, excess carcinogenic risk is estimated, and the AEGLs corresponding to carcinogenic risks of 1 in 10,000 ( $1 \times 10^{-4}$ ), 1 in

100,000 ( $1 \times 10^{-5}$ ), and 1 in 1,000,000 ( $1 \times 10^{-6}$ ) exposed persons are estimated.

## REVIEW OF AEGL REPORTS

As NAC began developing chemical-specific AEGL reports, EPA and DOD asked the NRC to review independently the NAC reports for their scientific validity, completeness, and consistency with the NRC guideline reports (NRC 1993, 2001). The NRC assigned this project to the COT Subcommittee on Acute Exposure Guideline Levels. The subcommittee has expertise in toxicology, epidemiology, pharmacology, medicine, industrial hygiene, biostatistics, risk assessment, and risk communication.

The AEGL draft reports are initially prepared by ad hoc AEGL Development Teams consisting of a chemical manager, two chemical reviewers, and a staff scientist of the NAC contractor—Oak Ridge National Laboratory. The draft documents are then reviewed by NAC and elevated from “draft” to “proposed” status. After the AEGL documents are approved by NAC, they are published in the *Federal Register* for public comment. The reports are then revised by NAC in response to the public comments, elevated from “proposed” to “interim” status, and sent to the NRC Subcommittee on Acute Exposure Guideline Levels for final evaluation.

The NRC subcommittee’s review of the AEGL reports prepared by NAC and its contractors involves oral and written presentations to the subcommittee by the authors of the reports. The NRC subcommittee provides advice and recommendations for revisions to ensure scientific validity and consistency with the NRC guideline reports (NRC 1993, 2001). The revised reports are presented at subsequent meetings until the subcommittee is satisfied with the reviews.

Because of the enormous amount of data presented in the AEGL reports, the NRC subcommittee cannot verify all the data used by NAC. The NRC subcommittee relies on NAC for the accuracy and completeness of the toxicity data cited in the AEGLs reports.

This report is the fourth volume in the series *Acute Exposure Guideline Levels for Selected Airborne Chemicals*. AEGL documents for chlorine, hydrogen chloride, hydrogen fluoride, toluene 2,4- and 2,6-diisocyanate, and uranium hexafluoride are published as an appendix to this report. The subcommittee concludes that the AEGLs developed in those documents are scientifically valid conclusions based on the data reviewed by NAC and are consistent with the NRC guideline reports. AEGL reports for additional chemicals will be presented in subsequent volumes.

**REFERENCES**

- NRC (National Research Council). 1968. Atmospheric Contaminants in Spacecraft. Washington, DC: National Academy of Sciences.
- NRC (National Research Council). 1972. Atmospheric Contaminants in Manned Spacecraft. Washington, DC: National Academy of Sciences.
- NRC (National Research Council). 1984a. Emergency and Continuous Exposure Limits for Selected Airborne Contaminants, Vol. 1. Washington, DC: National Academy Press.
- NRC (National Research Council). 1984b. Emergency and Continuous Exposure Limits for Selected Airborne Contaminants, Vol. 2. Washington, DC: National Academy Press.
- NRC (National Research Council). 1984c. Emergency and Continuous Exposure Limits for Selected Airborne Contaminants, Vol. 3. Washington, DC: National Academy Press.
- NRC (National Research Council). 1984d. Toxicity Testing: Strategies to Determine Needs and Priorities. Washington, DC: National Academy Press.
- NRC (National Research Council). 1985a. Emergency and Continuous Exposure Guidance Levels for Selected Airborne Contaminants, Vol. 4. Washington, DC: National Academy Press.
- NRC (National Research Council). 1985b. Emergency and Continuous Exposure Guidance Levels for Selected Airborne Contaminants, Vol. 5. Washington, DC: National Academy Press.
- NRC (National Research Council). 1986a. Emergency and Continuous Exposure Guidance Levels for Selected Airborne Contaminants, Vol. 6. Washington, DC: National Academy Press.
- NRC (National Research Council). 1986b. Criteria and Methods for Preparing Emergency Exposure Guidance Level (EEGL), Short-Term Public Emergency Guidance Level (SPEGL), and Continuous Exposure Guidance level (CEGL) Documents. Washington, DC: National Academy Press.
- NRC (National Research Council). 1987. Emergency and Continuous Exposure Guidance Levels for Selected Airborne Contaminants, Vol. 7. Washington, DC: National Academy Press.
- NRC (National Research Council). 1988. Emergency and Continuous Exposure Guidance Levels for Selected Airborne Contaminants, Vol. 8. Washington, DC: National Academy Press.
- NRC (National Research Council). 1992. Guidelines for Developing Spacecraft Maximum Allowable Concentrations for Space Station Contaminants. Washington, DC: National Academy Press.
- NRC (National Research Council). 1993. Guidelines for Developing Community Emergency Exposure Levels for Hazardous Substances. Washington, DC: National Academy Press.
- NRC (National Research Council). 1994. Spacecraft Maximum Allowable Concentrations for Selected Airborne Contaminants, Vol. 1. Washington, DC: National Academy Press.

- NRC (National Research Council). 1996a. Spacecraft Maximum Allowable Concentrations for Selected Airborne Contaminants, Vol. 2. Washington, DC: National Academy Press.
- NRC (National Research Council). 1996b. Spacecraft Maximum Allowable Concentrations for Selected Airborne Contaminants, Vol. 3. Washington, DC: National Academy Press.
- NRC (National Research Council). 2000. Spacecraft Maximum Allowable Concentrations for Selected Airborne Contaminants, Vol. 4. Washington, DC: National Academy Press.
- NRC (National Research Council) 2001. Acute Exposure Guideline Levels for Selected Airborne Chemicals. Volume 1. Washington, DC: National Academy Press.
- NRC (National Research Council). 2001. Standing Operating Procedures for Developing Acute Exposure Guideline Levels for Airborne Chemicals. Washington, DC: National Academy Press.
- NRC (National Research Council) 2002. Acute Exposure Guideline Levels for Selected Airborne Chemicals. Volume 2. Washington, DC: National Academy Press.
- NRC (National Research Council) 2003. Acute Exposure Guideline Levels for Selected Airborne Chemicals. Volume 3. Washington, DC: National Academy Press.

# Roster of the National Advisory Committee for Acute Exposure Guideline Levels for Hazardous Substances

## Committee Members

George Rusch  
Chair, NAC/AEGL Committee  
Department of Toxicology and  
Risk Assessment  
Honeywell, Inc.  
Morristown, NJ

Ernest Falke  
Chair, SOP Workgroup  
U.S. Environmental Protection  
Agency  
Washington, DC

George Alexeeff  
Office of Environmental Health  
Hazard Assessment  
California EPA  
Oakland, CA

Jonathan Borak  
Yale University  
New Haven, CT

Steven Barbee  
Arch Chemicals, Inc.  
Norwalk, CT

Lynn Beasley  
U.S. Environmental Protection  
Agency  
Washington, DC

Robert Benson  
U.S. Environmental Protection  
Agency  
Region VIII  
Denver, CO

William Bress  
Vermont Department of Health  
Burlington, VT

George Cushmac Office of Hazardous Materials Safety U.S. Department of Transportation Washington, DC	Larry Gephart Exxon Mobil Biomedical Sciences Annandale, NJ
John P. Hinz U.S. Air Force Brooks Air Force Base, TX	James Holler Agency for Toxic Substances and Disease Registry Atlanta, GA
Thomas C. Hornshaw Office of Chemical Safety Illinois Environmental Protection Agency Springfield, IL	Nancy K. Kim Division of Environmental Health Assessment New York State Department of Health Troy, NY
Loren Koller Loren Koller & Associates Corvallis, OR	Glenn Leach U.S. Army Center for Health Promotion and Preventive Medicine Aberdeen Proving Grounds, MD
John Morawetz International Chemical Workers Union Cincinnati, OH	Richard W. Niemeier National Institute for Occupational Safety and Health Cincinnati, OH
Marinelle Payton Department of Public Health Jackson State University Jackson, MS	George Rodgers Department of Pediatrics Division of Critical Care University of Louisville Louisville, KY
Robert Snyder Environmental and Occupational Health Sciences Institute Piscataway, NJ	Thomas J. Sobotka U.S. Food and Drug Administration Laurel, MD
Richard Thomas International Center for Environmental Technology McLean, VA	

**Oak Ridge National Laboratory Staff**

Cheryl Bast Oak Ridge National Laboratory Oak Ridge, TN	Sylvia Talmage Oak Ridge National Laboratory Oak Ridge, TN
Carol Wood Oak Ridge National Laboratory Oak Ridge, TN	

**National Advisory Committee Staff**

Paul S. Tobin Designated Federal Officer, AEGL Program U.S. Environmental Protection Agency Washington, DC	Marquea King Senior Scientist U.S. Environmental Protection Agency Washington, DC
---	---

# Appendix



# 1

## Chlorine<sup>1</sup>

### Acute Exposure Guideline Levels

#### SUMMARY

Chlorine is a greenish-yellow, highly reactive halogen gas that has a pungent, suffocating odor. The vapor is heavier than air and will form a cloud in the vicinity of a spill. Like other halogens, chlorine exists in the diatomic state in nature. Chlorine is extremely reactive and rapidly combines with both inorganic and organic substances. Chlorine is used in the manufacture of a wide variety of chemicals, as a bleaching agent in industry and household products, and as a biocide in water and waste treatment plants.

---

<sup>1</sup>This document was prepared by the AEGL Development Team comprising Sylvia Talmage (Oak Ridge National Laboratory) and members of the National Advisory Committee (NAC) on Acute Exposure Guideline Levels for Hazardous Substances, including Larry Gephart (Chemical Manager) and George Alexeeff and Kyle Blackman (Chemical Reviewers). The NAC reviewed and revised the document and AEGLs as deemed necessary. Both the document and the AEGL values were then reviewed by the National Research Council (NRC) Subcommittee on Acute Exposure Guideline Levels. The NRC subcommittee concludes that the AEGLs developed in this document are scientifically valid on the basis of the data reviewed by the NRC and are consistent with the NRC guidelines reports (NRC 1993, 2001).

Chlorine is an irritant to the eyes and respiratory tract; reaction with moist surfaces produces hydrochloric and hypochlorous acids. Its irritant properties have been studied in human volunteers, and its acute inhalation toxicity has been studied in several laboratory animal species. The data from the human and laboratory animal studies were sufficient for developing acute exposure guideline levels (AEGs) for the five exposure durations (i.e., 10 and 30 minutes [min] and 1, 4, and 8 hours [h]). Regression analysis of human data on nuisance irritation responses (itching or burning of the eyes, nose, or throat) for durations of 30-120 min and during exposures to chlorine at 0-2 parts per million (ppm) determined that the relationship between concentration and time is approximately  $C^2 \times t = k$  (where  $C$  = concentration,  $t$  = time, and  $k$  is a constant) (ten Berge and Vis van Heemst 1983).

The AEGL-1 was based on a combination of studies that tested healthy human subjects as well as atopic individuals (Rotman et al. 1983; Shusterman et al. 1998) and asthmatic patients (D'Alessandro et al. 1996). Atopic and asthmatic individuals have been identified as susceptible populations for irritant gases. The highest no-observed-adverse-effect level (NOAEL) for notable irritation and significant changes in pulmonary function parameters was 0.5 ppm in two studies. Eight atopic subjects were exposed for 15 min in one study (Shusterman et al. 1998), and eight healthy exercising individuals and an exercising atopic individual were exposed for two consecutive 4-h periods in the other (Rotman et al. 1983). The subjects in the Shusterman et al. (1998) study experienced nasal congestion, but irritation was described as none to slight. The exercising atopic individual in the Rotman et al. (1983) study experienced nondisabling, transient, asymptomatic changes in pulmonary function parameters. The selection of 0.5 ppm is supported by the lack of symptoms and lack of changes in pulmonary air flow and airway resistance in five asthmatic subjects inhaling 0.4 ppm for 1 h (D'Alessandro et al. 1996).

Because susceptible populations comprising atopic and asthmatic individuals were tested at similar concentrations, with incorporation of exercise into the protocol of one study, an intraspecies uncertainty factor (UF) of 1 was applied. The intraspecies UF of 1 is further supported by the fact that pediatric asthmatic subjects do not appear to be more responsive to irritants than adult asthmatic subjects (Avital et al. 1991). The AEGL-1 value was not time scaled for several reasons. First, the Rotman et al. (1983) study was for 8 h with a single 1-h break. Second, the response to chlorine appears to be concentration-dependent rather than time-dependent, as the pulmonary function parameters of individuals tested in this study, including

those for the atopic individual, did not increase between the 4- and 8-h measurements.

The AEGL-2 values were based on two of the studies used to derive the AEGL-1. Both healthy and susceptible human subjects inhaled chlorine at 1.0 ppm for 1 h (D'Alessandro et al. 1996) or 4 h (Rotman et al. 1983). Both healthy and susceptible subjects experienced some sensory irritation and transient changes in pulmonary function measurements. Greater changes were observed in pulmonary parameters among the susceptible subjects compared with the normal groups. In the latter study (Rotman et al. 1983), an atopic individual experienced no respiratory symptoms other than some sensory irritation during the 4-h exposure, but his airway resistance nearly tripled. He experienced shortness of breath and wheezing during a second 4-h exposure. Five individuals with nonspecific airway hyper-reactivity or asthma also experienced a statistically significant fall in pulmonary air flow and an increase in airway resistance during a 1-h exposure at 1.0 ppm (D'Alessandro et al. (1996). There were no respiratory symptoms during the exposure. The susceptible individual in the Rotman et al. (1983) study remained in the exposure chamber for the full 4 h without respiratory symptoms. Therefore, when considering the definition of the AEGL-2, the first 4 h of exposure was a no-effect level in a susceptible individual. Because the subjects were susceptible individuals, one of the subjects was undergoing light exercise during the exposures (making him more vulnerable to sensory effects), and an exercising susceptible individual exhibited effects that did not impede escape for the 4-h exposure duration (consistent with the definition of the AEGL-2), an intraspecies UF of 1 was applied.

Chlorine is a highly irritating and corrosive gas that reacts directly with the tissues of the respiratory tract with no pharmacokinetic component involved in toxicity; therefore, effects are not expected to vary greatly among other susceptible populations. Time-scaling was considered appropriate for the AEGL-2 because it is defined as the threshold for irreversible effects, which, in the case of irritants, generally involves tissue damage. Although the end point used in this case—a no-effect concentration for wheezing that was accompanied by a significant increase in airways resistance—has a different mechanism of action than that of direct tissue damage, it is assumed that some biomarkers of tissue irritation would be present in the airways and lungs. The 4-h 1-ppm concentration was scaled to the other time periods using the  $C^2 \times t = k$  relationship. The scaling factor was based on regression analyses of concentrations and exposure durations that attained nuisance levels of irritation in human subjects (ten Berge and Vis

van Heemst 1983). The 10-min value was set equal to the 30-min value in order to not exceed the highest exposure of 4.0 ppm in controlled human studies.

In the absence of human data, animal lethality data served as the basis for AEGL-3. The mouse was not chosen as an appropriate model for lethality because mice often showed delayed deaths, which several authors attributed to bronchopneumonia. Because the mouse was shown to be more sensitive to chlorine than the dog and rat, and because the mouse does not provide an appropriate basis for quantitatively predicting mortality in humans, a value below those resulting in no deaths in the rat (213 ppm and 322 ppm) and above that resulting in no deaths in the mouse (150 ppm) for a period of 1 h was chosen (MacEwen and Vernot 1972; Zwart and Woutersen 1988). The AEGL-3 values were derived from a 1-h concentration of 200 ppm. That value was calculated applying a total UF of 10—3 to extrapolate from rats to humans (interspecies values for the same end point differed by a factor of approximately 2 within each of several studies), and 3 to account for differences in human sensitivity. The susceptibility of asthmatic subjects relative to healthy subjects when considering lethality is unknown, but the data from two studies with human subjects showed that doubling a no-effect concentration for irritation and bronchial constriction resulted in potentially serious effects in asthmatic subjects but not in normal individuals. Time-scaling was considered appropriate for the AEGL-3, because tissue damage is involved. (Data in animal studies clearly indicate that time scaling is appropriate when lung damage is observed.) The AEGL-3 values for the other exposure times were calculated using the  $C^2 \times t = k$  relationship, which was derived based on the end point of irritation from a study with humans.

The calculated values are listed in Table 1-1.

## 1. INTRODUCTION

Chlorine is the most abundant naturally occurring halogen. Halogens do not occur in the elemental state in nature. When formed experimentally, chlorine is a greenish-yellow, diatomic gas ( $\text{Cl}_2$ ) with a pungent, suffocating odor. Chlorine is used in the manufacture of a variety of nonagricultural chemicals, such as vinyl chloride and ethylene dichloride; as a bleaching agent in the paper industry (along with chlorine dioxide [ $\text{ClO}_2$ ]); as commercial and household bleaching agents (in the form of chlorates [ $\text{ClO}_3^-$ ] and hypochlorites [ $\text{OCl}^-$ ]); and as a biocide in water purification and waste

**TABLE 1-1** Summary of AEGLs Values for Chlorine (ppm [mg/m<sup>3</sup>])

Classification	10 min	30 min	1 h	4 h	8 h	End Point (Reference)
AEGL-1 <sup>a</sup> (Nondisabling)	0.5 (1.5)	0.5 (1.5)	0.5 (1.5)	0.5 (1.5)	0.5 <sup>b</sup> (1.5)	No to slight changes in pulmonary function parameters in humans (Rotman et al. 1983; D'Alessandro et al. 1996; Shusterman et al. 1998)
AEGL-2 (Disabling)	2.8 (8.1)	2.8 (8.1)	2.0 (5.8)	1.0 (2.9)	0.7 (2.0)	1.0 ppm for 4 h was a NOAEL for an asthma-like attack in human subjects; the other values were time-scaled (Rotman et al. 1983; D'Alessandro et al. 1996)
AEGL-3 (Lethal)	50 (145)	28 (81)	20 (58)	10 (29)	7.1 (21)	Threshold for lethality in the rat (MacEwen and Vernot 1972; Zwart and Woutersen 1988)

<sup>a</sup>The distinctive, pungent odor of chlorine will be noticeable to most individuals at these concentrations.

<sup>b</sup>Because effects were not increased following an interrupted 8-h exposure of an atopic individual to 0.5 ppm, the 8-h AEGL-1 was set equal to 0.5 ppm.

Abbreviations: mg/m<sup>3</sup>, milligrams per cubic meter; ppm, parts per million.

treatment systems (Perry et al. 1994). Chlorine gas was used as a chemical warfare agent during World War I (Withers and Lees 1987). The vapor is heavier than air and will form a cloud in the vicinity of a spill.

As of January 1999, world annual capacity for chlorine production was estimated at almost 50 million metric tons (CEH 2000). Chlorine is produced at chlor-alkali plants at over 650 sites worldwide, and North America accounts for 32% of capacity (operating rates are greater than 83% of capacity). In the early 1990s, chlorine was produced at 49 facilities, operated by 29 companies, in the United States (Perry et al. 1994). In 1993, U.S.

production was reported at 24 billion pounds (C&EN 1994). The major global market for chlorine is ethylene dichloride production (about 33%) (CEH 2000).

Chlorine is extremely reactive and enters into substitution or addition reactions with both inorganic and organic substances. Moist chlorine unites directly with most elements. Reaction with water produces hydrochloric (HCl) and hypochlorous acid (HClO) (Budavari et al. 1996; Perry et al. 1994). Other relevant chemical and physical properties are listed in Table 1-2. According to Amore and Hautala (1983), the odor threshold is 0.31 ppm, and a range of 0.2-0.4 ppm was reported in other studies. There is considerable variation in detecting the odor among subjects; for many individuals, the ability to perceive the odor decreases over exposure time (NIOSH 1976).

Chlorine is an eye and respiratory tract irritant and, at high doses, has direct toxic effects on the lungs. It reaches the lungs because it is only moderately soluble in water and it is not totally absorbed in the upper respiratory tract at high concentrations. The acute inhalation toxicity of chlorine has been studied in several laboratory animal species, and its irritant properties have been studied with human volunteers.

## 2. HUMAN TOXICITY DATA

### 2.1 Acute Lethality

For humans, a 5-min lethal concentration in 10% of subjects ( $LC_{10}$ ) of 500 ppm (NTIS 1996) and a possible 30-min lethal exposure of 872 ppm have been reported (Perry et al. 1995). Both of those secondary sources cited data from Prentiss (1937) as well as data from other early sources.

Although accidental releases have resulted in deaths (e.g., Jones et al. 1986), no studies were located in which acute lethal exposure concentrations were measured. Probit analysis of available information on the lethality of chlorine to animals and humans was used by Withers and Lees (1985b) to estimate a concentration lethal to 50% of the population ( $LC_{50}$ ). Their model incorporates the effects of physical activity, inhalation rate, the effectiveness of medical treatment, and the lethal toxic load function. The estimated 30-min  $LC_{50}$  at a standard level of activity (inhalation rate of 12 liters [L]/min) for the regular, vulnerable, and average (regular plus vulnerable) populations, as described by the authors, were 250, 100, and 210 ppm,

**TABLE 1-2** Chemical and Physical Properties of Chlorine

Parameter	Value	Reference
Synonyms	Bertholite; hypochlorite; hypochlorous acid	Budavari et al. 1996
Molecular formula	Cl <sub>2</sub>	Budavari et al. 1996
Molecular weight	70.9	Budavari et al. 1996
CAS registry no.	7782-50-5	Budavari et al. 1996
Physical state	Gas	Budavari et al. 1996
Color	Greenish-yellow	Budavari et al. 1996
Solubility in water	0.092 moles/L	Budavari et al. 1996
Vapor pressure	5,025 mm Hg at 20°C	Matheson Gas Co. 1980
Vapor density	1.4085 at 20°C	AIHA 1988
Density (water =1)	1.56 at boiling point	Perry et al. 1994
Melting point	-101°C	Budavari et al. 1996
Boiling point	-34.05°C	Budavari et al. 1996
Flammability	Nonflammable	Matheson Gas Co. 1980
Conversion factors in air	1 ppm = 2.9 mg/m <sup>3</sup> 1 mg/m <sup>3</sup> = 0.34 ppm	ACGIH 2001

respectively. The LC<sub>10</sub> for the three populations were 125, 50, and 80 ppm, respectively.

## 2.2. Nonlethal Toxicity

Exposures at 30 ppm and 40-60 ppm have been reported to cause intense coughing and serious damage, respectively (ILO 1998), but no documentation of those values was given.

### 2.2.1. Experimental Studies

Five well-conducted and well-documented studies using human volunteers were located. Those studies are summarized in Table 1-3.

**TABLE 1-3** Summary of Irritant Effects in Humans<sup>a</sup>

Concentration (ppm)	Exposure Time <sup>b</sup>	Effect	References
0.4	1 h	No pulmonary function changes in subjects with airway hyperreactivity/asthma	D'Alessandro et al. 1996
0.5	15 min	Change in nasal air resistance in rhinitic subjects (no change in nonrhinitic subjects); no effect on pulmonary peak flow, rhinorrhea, postnasal drip, or headache in either type of subject	Shusterman et al. 1998
0.5	8 h	Perception of odor, no discomfort, no effects, no changes in pulmonary function measurements for healthy individuals; some changes for atopic individual	Anglen 1981; Rotman et al. 1983
1.0	1 h	Statistically significant but modest changes in FEV <sub>1</sub> and R <sub>aw</sub> for normal and asthmatic subjects	D'Alessandro et al. 1996
1.0	2 h	No noticeable effects	Joosting and Verberk 1974
1.0	4 h	Irritation for some sensations; no changes in pulmonary function measurements	Anglen 1981
1.0	4 h	Transient changes in pulmonary function measurements (airway resistance)	Rotman et al. 1983
1.0	8 h	Irritation (itchy eyes, runny nose, mild burning in throat); transient changes in pulmonary function measurements; atopic subject could not complete full 8-h exposure because of wheezing and shortness of breath	Anglen 1981; Rotman et al. 1983

2.0	15 min	Perception of odor; no significant irritation effects	Anglen 1981
2.0	30 min	Not significantly different from control group for irritant effects, irritancy indices	Anglen 1981
2.0	1 h	Itching or burning of throat, urge to cough at nuisance level	Anglen 1981
2.0	2 h	Very slight irritation of eyes, nose, and throat; no changes in pulmonary function	Joosting and Verberk 1974
2.0	2 h	No significant changes in pulmonary function	Anglen 1981
2.0	4 h	50% response of subjects to sensations characterized as nuisance: itching or burning of nose or throat, urge to cough, runny nose, general discomfort; transient changes in pulmonary function	Anglen 1981
2.0	8 h	Not immediately irritating, objectionable after several hours; increased mucous; transient changes in pulmonary function	Anglen 1981
4.0	2 h	Nuisance level of throat irritation, perceptible to nuisance level of nose irritation and cough	Joosting and Verberk 1974

<sup>a</sup>The Anglen (1981) and Joosting and Verberk (1974) studies were performed with healthy adults. Atopic individuals were included in the Shusterman et al. (1998) and Rotman et al. (1983) studies, and healthy subjects as well as asthmatic subjects were included in the D'Alessandro (1996) study.

<sup>b</sup>8-h studies were composed of two segments with a 30-min or 1-h break after 4 h.

In the first part of a two-part experiment, 31 un-acclimated male and female subjects (age range 20-32 years [y]) were exposed to chlorine at 0.0, 0.5, 1.0, or 2.0 ppm for 4 h or at 0.5 ppm or 1.0 ppm for 8 h (Anglen 1981). Not all subjects were exposed to all concentrations. Exposure days were randomly assigned, and the subjects did not know the test concentration, although the investigator did. In part two, eight nonsmoking males, ages 23-33 y, were exposed to concentrations of chlorine at 0.0, 0.5, or 1.0 ppm for 8 h. The 8-h sessions were broken into two 4-h sessions with a 30- or 60-min lunch break. A 15-min exercise period during each hour of exposure was designed to increase the average heart rate to 100 beats per minute. During the exposures, the subjects filled out a subjective questionnaire concerning 14 sensations ranging from smell to shortness of breath and using a scale of 0 (no sensation) to 5 (unbearable). Eye irritation was documented photographically; other signs of irritation were documented with pre- and post-exposure examinations by a physician. Pulmonary function tests (forced vital capacity [FVC] and forced expiratory volume at 1 second [FEV<sub>1</sub>]) for each subject were measured before, during, and after exposures. The data were analyzed in terms of mean or median response, percent responding to greater than or equal to a set value, and responses to ranges or indices of irritation; the data were analyzed statistically where appropriate ( $p > 0.167$ ,  $p > 0.025$ ). Paired t-tests involving differences in values between pre-exposure and post-exposure times were used to analyze the pulmonary function measurements. Chlorine concentrations in the exposure chamber, measured by a variety of colorimetric and instrumental methods, were consistent.

At the tested concentrations, most of the subjects did not consider the sensations of smell or taste of chlorine unpleasant; therefore, those sensations were not included in the remainder of the analyses. In part one, the greatest number of subjects responded positively to the irritant sensation of itching or burning of the throat (described by the subjects as “feeling as if they had been talking for a long time”). Statistical differences were seen for that sensation at 1 and 2 ppm compared with controls; the level of response was  $\geq 3$  (nuisance or greater) for the 2-ppm concentration. A concentration of 0.5 ppm resulted in subjective irritation values between 1 (just perceptible) and 2 (distinctly perceptible) and produced no change in pulmonary functions. In part two, exposure at 1 ppm produced statistically significant changes in pulmonary function and increased subjective irritation at 8 h. No significant differences in pulmonary function measurements were seen at the end of a 4-h exposure at 1 ppm. Some differences in pulmonary function were seen at the end of 4 h at the 2-ppm concentration, but not

at the end of 2 h. Most of the exposed subjects did not report sensations of nausea, headache, dizziness, or drowsiness at any concentration during the exposures. Male subjects were more sensitive to the irritant effects of chlorine than female subjects. The author concluded that exposure at 2 ppm for up to 30 min produced no significant increase in subjective irritation (severity of response not stated) over that seen during control exposures, and 2-h exposures at 2 ppm or 4-h exposures at 1 ppm produced no significant changes in pulmonary function (Anglen 1981).

In a follow-up study, Rotman et al. (1983) reported pulmonary function tests of eight male subjects, ages 19-33 y, exposed at 0.0, 0.5, and 1.0 ppm, like in the Anglen (1981) study. Air samples were collected eight times daily, and analysis for chlorine was accomplished using a modified NIOSH-recommended methyl orange method. The subjects were blind to the exposure concentration, but the study investigators were not. While in the chamber, each subject exercised for 15 min of each hour on an inclined treadmill or a by a simple step test at a rate that produced a heart rate of 100 beats per minute. Subjects exited the exposure chamber after 4 h to undergo the pulmonary function tests and then reentered the chambers. Comparisons of pre- and post-exposure pulmonary functions were made by paired t-tests between the percent change from baseline values obtained at analogous times after a sham exposure. Insignificant differences were observed with the sham versus the 0.5-ppm exposure. Compared with the sham changes from baseline, the changes in the 1.0-ppm exposure group were small, but statistically significant ( $p < 0.05$ ); those changes were in FEV<sub>1</sub>, peak expiratory flow rate (PEFR), forced expiratory flow rate at 50% and 25% vital capacity (FEF<sub>50</sub> and FEF<sub>25</sub>), total lung capacity (TLC), airway resistance (R<sub>aw</sub>), and difference in nitrogen concentration between 750 milliliters (mL) and 1,250 mL of exhaled vital capacity ( $\Delta N_2$ ). At 8 h, changes were present in FVC, FEV<sub>1</sub>, forced expired volume in 1 second as %FEV (FEV<sub>1%</sub>), PEFR, FEF<sub>50</sub>, FEF<sub>25</sub>, and R<sub>aw</sub>. R<sub>aw</sub> had the greatest response to chlorine exposure with an increase of 31% after 4 h of exposure at 1.0 ppm compared with increases of up to 6% during sham exposures. Most of those parameters had returned to pre-exposure values by the following day. However, a ninth subject whose pre-exposure lung parameters indicated obstructive airway disease (as defined by DuBois et al. [1971]) did not complete the full 8 h of exposure at 1.0 ppm because of shortness of breath and wheezing; his values were not included in the statistical analysis. The atopic individual experienced changes in several pulmonary function parameters after exposure to chlorine at 0.5 ppm. The greatest change for this individual was in R<sub>aw</sub>, which increased by 40% over the pre-exposure value

after 4 h of exposure at 0.5 ppm and by 33% over the pre-exposure value after 8 h of exposure at 0.5 ppm. Changes in  $R_{aw}$  in the healthy subjects were 5% and 15% for the respective time periods. Following the 4-h exposure at 1.0 ppm,  $R_{aw}$  increased from a pre-exposure value of 3.3 centimeters (cm) of water per liter per second to 14.4 cm of water per liter per second in the atopic individual. The authors concluded that at the 1-ppm concentration, serious subjective symptoms of irritation were not produced in healthy adults, but transient altered pulmonary function was observed.

The authors discussed the implications of changing baselines and significant changes over time with sham exposures, the latter with respect to diurnal variation in pulmonary function. Baselines differed on different days of the tests (e.g., the pre-exposure baselines for total lung capacities [TLCs] were 7.09 L and 6.55 L on two different days, which is significantly different [ $p < 0.05$ ]; the TLC decreased after the sham exposure, but increased after the 0.5-ppm exposure). However, differences in parameters between baseline and post-exposure for sham exposures were fewer in number and generally smaller in amount than for the 1 ppm treatment. It should be noted that in normal subjects, several pulmonary function tests (FVC,  $FEV_1$ , and  $FEF_{25-75\%}$ ) may have daily changes of 5% to 13% and week to week changes of 11% to 21% (EPA 1994).

D'Alessandro et al. (1996) exposed 12 male and female volunteers, ages 18-50 y, to chlorine for 1 h. Five of the subjects were without airway hyper-reactivity (defined by baseline methacholine hyper-responsiveness) and seven were diagnosed with airway hyper-reactivity. Five of the seven with airway hyper-reactivity had clinical histories of asthma (one was being treated regularly with corticosteroids). Ten subjects, five normal and five hyper-reactive (three of which had asthma), were exposed to a concentration of chlorine at 1.0 ppm. Five of the subjects with asthma were exposed at 0.4 ppm. They were not blinded to the exposure status. The subjects were exposed to chlorine by mask while in the sitting position; there were no air exposures. Chlorine was measured with a chlorine analyzer. The following pulmonary function parameters were measured or calculated immediately following and 24 h after exposure:  $FEV_1$ , TLC, carbon monoxide diffusing capacity ( $D_{CO}$ ),  $R_{aw}$ , and  $FEF_{25-75\%}$ . After asthmatic subjects were exposed at 0.4 ppm, there were no statistically significant changes in any parameters, including  $FEV_1$  and  $R_{aw}$ , either immediately following or 24 h after exposure. Immediately following the exposure at 1.0 ppm, there were statistically significant changes in  $FEV_1$  and  $R_{aw}$  for both normal and hyper-reactive subjects compared with baseline values. Hyper-reactive subjects showed a greater relative decrease in  $FEV_1$  (16% compared with

4% for normal subjects) and a greater relative increase in  $R_{aw}$  (108% compared with 39% for the normal subjects). Although one hyper-responsive subjects' FEV<sub>1</sub> fell by 1,200 mL, and the  $R_{aw}$  more than tripled, the mean changes were considered modest by the authors. The hyper-responsive subject with the greatest increase in  $R_{aw}$  following exposure at 1.0 ppm showed virtually no change following exposure at 0.4 ppm. Two subjects characterized as hyper-responsive experienced undefined respiratory symptoms following exposure at 1.0 ppm. For all subjects, most values were close to baseline by 24 h post-exposure. The fact that none of the subjects found the odor of chlorine appreciable at either concentration is interesting.

In a single-blind crossover study, Shusterman et al. (1998) measured nasal air resistance via active posterior rhinomanometry in eight subjects with seasonal allergic rhinitis and eight nonrhinitic subjects. Measurements were made before, immediately after, and 15 min after a 15-min exposure to either filtered air or chlorine at 0.5 ppm in filtered air administered through a nasal mask in a climate-controlled chamber. Each subject served as his or her control, and subjects were free of medications for at least 24 h prior to testing. Subjects were between 18 and 40 y of age. The mean percent change in nasal air resistance from baseline to immediately after exposure was +24% in the subjects with allergic rhinitis and +3% in the nonrhinitic group. The mean percent change from baseline to 15 min after exposure was +21% in the subjects with allergic rhinitis and -1% in the nonrhinitic subjects. Differences between groups were significant ( $p < 0.05$ ) for both post-exposure times. Rhinitic subjects reported greater exposure-related increases in odor intensity, nasal irritation, and nasal congestion than did nonrhinitic subjects, but the relationship between subjective and objective nasal congestion was weak. No significant exposure-related changes were observed for rhinorrhea, postnasal drip, or headache. Pulmonary peak flow was also obtained before and after exposure and none of the subjects exhibited clinically significant changes in peak flow (decreases of  $\geq 10\%$  of baseline), nor did they complain of cough, wheezing, or chest tightness during chlorine exposure days.

Joosting and Verberk (1974) exposed eight subjects, ages 28-52, at 0.5-4 ppm for 2 h. The subjects were all members of a Dutch subcommittee on toxicology. Subjective reactions were noted every 15 min. Subjects exited the chambers every 15 min to perform spirometry tests. Concentrations of chlorine at 0.5 ppm and 1.0 ppm did not produce noticeable effects in 2 h; 2 ppm produced very slight eye, nose, and throat irritation; and 4 ppm resulted in a distinctly perceptible to offensive level of irritation of the nose and throat and desire to cough. The highest score was for irritation of the

throat, for which the average response was nuisance. No sensory irritation scores reached unbearable. No effects on lung function (vital capacity [VC], FEV, and forced inspiratory volume [FIV]) occurred at the lower concentrations; effects on lung function were not reported in the 4-ppm exposure group because only 2-3 subjects completed the exposure.

Older studies were located in the literature; those have been reviewed and critiqued by NIOSH (1976) and OSHA (1989). In several older studies in which measurement techniques and/or ranges of measured values were not given, concentrations as low as 0.027 ppm produced slight sensory effects in humans and concentrations at 0.5-4.0 ppm produced sensory irritation (NIOSH 1976). For example, Rupp and Henschler (1967) reported that human volunteers experienced burning of the eyes after exposure at 0.5 ppm for 15 min. In a separate test, the subjects reported respiratory irritation during exposure at 0.5 ppm and discomfort during exposure at 1 ppm. The study has been criticized for its lack of controls as well as the possible presence of confounding chemicals (OSHA 1989). However, the results of the Rupp and Henschler (1967) study that indicated some irritation at 0.5 ppm and 1 ppm are not all that different from the results of Anglen (1981), Rotman et al. (1983), and D'Alessandro et al. (1996).

### 2.2.2. Epidemiologic Studies

Few epidemiologic studies document average chlorine exposure concentrations over long periods of time; concentration variations over time; or exposure durations to various concentrations. Interpretation of results is often complicated by unknown previous exposures to chlorine, exposures to other chemicals, and smoking habits. NIOSH (1976) discussed available epidemiological studies conducted prior to 1976. In most of the cited studies, work-room concentrations averaged <1 ppm. The report noted the difficulty in correlating exposures to effects.

Patil et al. (1970) compared the health of 382 workers in 25 chlorine production plants in the United States and Canada with that of unexposed workers in the same plants. All subjects were male, between the ages of 19 and 69. Time-weighted average exposures to chlorine ranged from 0.006 ppm to 1.42 ppm, with a mean of 0.146 ppm; almost all workers were exposed to <1 ppm. The average number of exposure years was 10.9. There were no statistically significant ( $p < 0.05$ ) signs or symptoms on a dose-response relation basis in chest x-rays, electrocardiograms, or pulmonary function tests. Nor were there dose-response relationships with cough,

sputum production, frequency of colds, dyspnea, palpitation, chest pain, fatigue, tremors, gastrointestinal problems, dermatitis, or hematologic parameters. Subjective complaints of tooth decay were dose-related, but that complaint was not borne out by physical examination.

A study of respiratory effects in 52 Italian electrolytic cell workers with an average exposure to chlorine of  $0.298 \pm 0.181$  ppm was undertaken by Capodaglio et al. (1970, as cited in ACGIH 1995, 1996). Of five respiratory function tests ( $FEV_1$ , VC,  $D_{CO}$ , residual volume, and helium concentration gradient in a single breath during washout), only carbon monoxide diffusing capacity showed a slight but significant difference; however, cigarette smoking may have contributed to that observation.

Mortality and morbidity (respiratory symptoms, disease, and functions) of workers exposed to chlorine in the pulp and paper industry over a 10-y period (1963-1973) were similar to those of the general white male population (Ferris et al. 1979). Mean and maximum exposure concentrations in the pulp mill were to trace amounts ( $< 0.0005$  ppm). In an earlier study (Ferris et al. 1964), exposures in the pulp mill were more variable (mean, 7.4 ppm; range up to 64 ppm) and slight adverse effects on respiratory ailments were found. Exposures were also to chlorine dioxide.

Between 1984 and 1989, a prospective study was conducted on the effects of chlorine exposure on workers in a chlorine manufacturing plant (Kusch 1994). Chlorine exposures and pulmonary function tests (FVC,  $FEV_1$ , and  $FEF_{25-50\%}$ ), taken over a period of 5 y were compared with a control group. The average exposure in the control group was 0.058 ppm, and the average exposure in the workers was 0.092 ppm. There were no measurable effects on pulmonary function related to chlorine exposure.

### 2.2.3. Accidents

An accident at a chemical plant in India resulted in the exposure of 88 workers as well as police and fire-fighting personnel to a measured concentration of 66 ppm for an unspecified amount of time (Shroff et al. 1988). No further details on the exposure concentration or duration were given. The workers, ages 21-60 y, were admitted to the hospital within an hour of exposure with symptoms of dyspnea, coughing, irritation of the throat and eyes, headache, giddiness, chest pain, and abdominal discomfort. Examinations revealed hilar congestion, bronchial vasculature, respiratory incapacitation at the PFT (undefined); bronchoscopy revealed tracheobronchial congestion, chronic bronchitis, scattered hemorrhages, and bronchial ero-

sion. Bronchial smears of 28 patients on day 5 after the accident showed basal-cell and goblet-cell hyperplasia; acute inflammation; and chromatolysis of columnar epithelial cells, multinucleated syncytial respiratory epithelial cells with degenerating cilia, and nonpigmented alveolar macrophages. In some patients these effects progressed to bronchopneumonia, epithelial regeneration, and repair by fibrosis by day 25 post-exposure.

During an industrial accident, a group of workers was presumably exposed to concentrations up to 30 ppm (based on symptoms—actual exposure concentration and duration not known) (Abhyankar et al. 1989). Initial symptoms included watering eyes, sneezing, cough, sputum, retrosternal burning, dyspnea, apprehension, and vomiting. All patients were asymptomatic by 2 weeks (wk) post-exposure; at 6 months (mo) post-exposure, all spirometry tests (FVC, FEV<sub>1</sub>) were within the normal range. Those patients known to have a pre-existing lung condition did not show any additional evidence of lung damage.

Incidents of acute poisonings at indoor swimming pools have been reported (Decker 1988), but air concentrations during those incidents are unknown. In a study in Spain, the mean air concentration measured during five nonconsecutive days in four enclosed swimming pools was  $0.42 \pm 0.24$  milligrams per cubic meter (mg/m<sup>3</sup>) ( $0.14 \pm 0.08$  ppm) (Drobic et al. 1996). The samples were taken at <10 cm above the water, the breathing zone of swimmers.

Acute exposure to chlorine/chloramine gas occurs often among the general public through the mixing of domestic home cleaners (Mrvos et al. 1993); swimming pool chlorinator tablets (Wood et al. 1987); and intentional self administration (Rafferty 1980). In the case of home exposures, a review of 216 cases reported to a Regional Poison Information Center showed that symptoms, primarily cough with resulting shortness of breath, resolved within 1 to 6 h without medical intervention. There was no information on exposure concentrations (Mrvos et al. 1993).

### **2.3. Developmental and Reproductive Effects**

No studies on developmental and reproductive effects in humans were located.

### **2.4. Genotoxicity**

No data concerning the genotoxicity of chlorine in humans via inhala-

tion exposures were identified in the available literature. When chlorine (sodium hypochlorite) at concentrations of 20 ppm and above was added to cultures of human lymphocytes, chromosomal aberrations (breaks and rearrangements) and endomitotic figures were observed (Mickey and Holden 1971).

### 2.5. Carcinogenicity

No increase in neoplasms was reported in an epidemiology study of workers engaged in the production of chlorine (Patil et al. 1970). The range of time-weighted exposures to chlorine was 0.006-1.42 ppm.

### 2.6. Summary

Anglen (1981) conducted a study on 31 male and female subjects in which slight but statistically significant changes in pulmonary function and subjective irritation resulted from exposure to chlorine at 1 ppm for 8 h (two 4-h sessions). Subjective sensory irritation (itching or burning of the throat) was described as "just perceptible" or "distinctly perceptible." A 30-min exposure at 2 ppm produced no increase in subjective irritation. An 8-h exposure at 0.5 ppm produced no changes in lung function and no significant sensory irritation. A 4-h exposure at 1 ppm produced no changes in pulmonary function tests, but an 8-h exposure at 1 ppm produced slight declines in some pulmonary function tests. Most of these findings were confirmed in a study by Rotman et al. (1983) in which eight healthy volunteers were exposed at 0.5 ppm or 1.0 ppm for an interrupted 8 h. Transient but statistically significant declines in six of 15 pulmonary function tests were associated with exposure at 1 ppm for 4 or 8 h but not with exposure at 0.5 ppm for 4 or 8 h. These studies reported that there was no effect of chlorine exposure on carbon monoxide diffusing capacity, thus indicating there is no significant pulmonary edema from the exposures. However, an atopic subject in the Rotman et al. study (1983) suffered an asthma-like attack resulting from exposure to chlorine at 1 ppm; that subject withstood exposure, testing, and exercise for the first 4 h, but exited the exposure chamber "before the full 8-h exposure to 1 ppm." Changes in his pulmonary function measurements were greater than those of the other test subjects. The atopic subject completed the interrupted 8-h exposure at 0.5 ppm.

The Anglen and Rotman studies were supported by two additional studies. In the first study (Angelen 1981), subjects with airway hyper-reactivity, including asthmatic subjects, showed no significant changes in FEV<sub>1</sub> or R<sub>aw</sub> following a 1-h exposure at 0.4 ppm. Exposure of both normal and hyper-reactive subjects at 1.0 ppm significantly decreased FEV<sub>1</sub> and significantly increased R<sub>aw</sub> in both sets of subjects; the hyper-reactive subjects showed a significantly greater response than normal subjects. However, the mean changes were considered modest by the authors. In the second study, subjective sensory irritation of healthy individuals reached “nuisance” level at an exposure concentration of 4 ppm for 2 h (Joosting and Verberk 1974). This study showed that time, as well as concentration, was a factor in subjective response to chlorine inhalation.

No useful exposure data could be derived from epidemiology studies or human exposures to accidental releases of chlorine. No studies were located on developmental and reproductive effects. In an *in vitro* study, chlorine was genotoxic at 20 ppm.

### 3. ANIMAL TOXICITY DATA

#### 3.1 Acute Lethality

A summary of the acute lethality data is presented in Table 1-4. Some of those studies were reviewed by Withers and Lees (1985a). According to Withers and Lees (1985a), many of the older studies had deficiencies in gas analysis methods and exposure conditions. More recent studies contradict the results in the older studies (e.g., the 3-h LC<sub>50</sub> of 10 ppm in mice reported by Schlagbauer and Henschler [1967, as cited in AIHA 1988] is contradicted by the nonlethal 6-h exposure at 9.3 ppm for 5 d reported in the more recent study by Buckley et al. [1984]). Nevertheless, some of the older studies (Lipton and Rotariu 1941, as cited in Withers and Lees 1985a; Silver et al. 1942, as cited in Withers and Lees 1985a; Schlagbauer and Henschler 1967, as cited in AIHA 1988) are cited in Table 1-4 for comparison purposes. More recent studies are discussed below.

##### 3.1.1. Dogs

In an early study, Underhill (1920) reported on mortality in dogs following 30-min exposures to a range of concentrations (50-2,000 ppm). A total of 112 male and female dogs of several breeds were used. Acute mor-

**TABLE 1-4** Summary of Acute Lethal Inhalation Data in Animals

Species	Concentration (ppm)	Exposure Time	Effect <sup>a</sup>	Reference
Dog	650	30 min	LC <sub>50</sub>	Underhill 1920; Withers and Lees 1985a
Rat	5,500	5 min	LC <sub>50</sub>	Zwart and Woutersen 1988
	2,841	5 min	No deaths	
Rat	1,946	10 min	LC <sub>50</sub>	Zwart and Woutersen 1988
Rat	700	30 min	LC <sub>50</sub>	Zwart and Woutersen 1988
	547	30 min	No deaths	
Rat	1,000	53 min	LC <sub>50</sub>	Weedon et al. 1940
Rat	455	1 h	LC <sub>50</sub>	Zwart and Woutersen 1988
Rat	288 <sup>b</sup>	1 h	LC <sub>01</sub>	Zwart and Woutersen 1988
	322	1 h	No deaths	
Rat	293 <sup>c</sup>	1 h	LC <sub>50</sub>	Back et al. 1972; MacEwen and Vernot 1972; Vernot et al. 1977
	213	1 h	No deaths	
Rat	250	7.3 h	LC <sub>50</sub>	Weedon et al. 1940
Rat	63	>16 h	LC <sub>50</sub>	Weedon et al. 1940
Mouse	290	6 min	No deaths	Bitron and Aharonson 1978
Mouse	1,057	10 min	LC <sub>50</sub>	Zwart and Woutersen 1988
	754	10 min	No deaths	
Mouse	676	10 min	LC <sub>50</sub>	Silver et al. 1942 <sup>d</sup>
Mouse	628	10 min	LC <sub>50</sub>	Lipton and Rotariu 1941 <sup>d</sup>
Mouse	549	10 min	25-45% mortality	Silver et al. 1942 <sup>d</sup>
Mouse	380	10 min	10% mortality	Silver et al. 1942 <sup>d</sup>
Mouse	302	10 min	LC <sub>50</sub>	Alarie 1980
Mouse	290	11 min	LC <sub>50</sub>	Bitron and Aharonson 1978

*(Continued)*

TABLE 1-4 Continued

Species	Concentration (ppm)	Exposure Time	Effect <sup>a</sup>	Reference
Mouse	290	15 min	80% mortality	Bitron and Aharonson 1978
Mouse	290	25 min	100% mortality	Bitron and Aharonson 1978
Mouse	1,000	28 min	LC <sub>50</sub>	Weedon et al. 1940
Mouse	504	30 min	LC <sub>50</sub>	Zwart and Woutersen 1988
Mouse	127	30 min	LC <sub>50</sub>	Schlagbauer and Henschler 1967 <sup>e</sup>
	55	30 min	No deaths	
Mouse	170	55 min	LC <sub>50</sub>	Bitron and Aharonson 1978
Mouse	137 <sup>c</sup>	1 h	LC <sub>50</sub>	Back et al. 1972; MacEwen and Vernot 1972; Vernot et al. 1977
Mouse	250	1 h	LC <sub>80</sub>	O'Neil 1991
	200	1 h	LC <sub>01</sub>	
	150	1 h	No deaths	
Mouse	170	2 h	80% mortality	Bitron and Aharonson 1978
Mouse	10	3 h	80% mortality <sup>f</sup>	Schlagbauer and Henschler 1967 <sup>e</sup>
Mouse	250	7.3 h	LC <sub>50</sub>	Weedon et al. 1940
Mouse	63	>16 h	LC <sub>50</sub>	Weedon et al. 1940
Rabbit	500	30 min	100% mortality	Barrow and Smith 1975

<sup>a</sup>LC<sub>50</sub> and LC<sub>100</sub> values were obtained immediately after exposure (Weedon et al. 1940), 3 h post-exposure (Alarie 1980), 10 d post-exposure (Silver et al. 1942), 14 d post-exposure (Back et al. 1972; MacEwen and Vernot 1972; Vernot et al. 1977; Zwart and Woutersen 1988), and 30 d post-exposure (Bitron and Aharonson 1978).

<sup>b</sup>Calculated by Zwart and Woutersen (1988) using probit analysis; note this value is lower than the concentration resulting in no deaths.

<sup>c</sup>Authors report a 20-30% loss of chlorine in the exposure chambers; the concentrations given are measured concentrations.

<sup>d</sup>As cited in Withers and Lees 1985a.

<sup>e</sup>As cited in AIHA 1988.

<sup>f</sup>These results conflict with results of other studies.

talities, defined as deaths within 3 d of the exposure, were 0%, 6%, 20%, 43%, 50%, 87%, and 92% at concentration ranges of 50-250 ppm, 400-500 ppm, 500-600 ppm, 600-700 ppm, 700-800 ppm, 800-900 ppm, and 900-2,000 ppm, respectively. However, some delayed deaths, occurring as the result of bronchopneumonia following subsidence of acute pulmonary edema, resulted in all groups; one of nine suffered a delayed death (time not given) at the 50-250 ppm range. Withers and Lees (1985a) analyzed these data using the method of Litchfield and Wilcoxon (1949) and calculated an  $LC_{50}$  of 650 ppm. That value is based on the concentration in ppm at 25°C. Underhill (1920) initially made the conversion from  $mg/m^3$  to ppm based on a temperature of 0°C. Dogs exposed to chlorine became excited and displayed signs of respiratory irritation; those signs were followed by labored breathing (Underhill 1920).

### 3.1.2. Rats

Back et al. (1972), MacEwen and Vernot (1972), and Vernot et al. (1977) reported the same 1-h  $LC_{50}$  of 293 ppm (95% confidence limits, 260-329 ppm) for Sprague-Dawley rats. MacEwen and Vernot (1972) noted a 20-30% loss of chlorine in the exposure chambers, probably due to condensation on the walls. Therefore, the concentrations given are measured concentrations. Rats experienced immediate eye and nose irritation followed by lacrimation, rhinorrhea, and gasping after 1 h of exposure at all tested concentrations (213, 268, 338, and 427 ppm). Rats surviving the 213 ppm and 268-ppm exposures gained less weight than the control group during the 14-d post-exposure period. No deaths occurred in rats exposed at 213 ppm for 1 h. Weedon et al. (1940) exposed groups of eight rats to 63, 240, or 1,000 ppm for 16 h or until death. Times to 50% mortality were >16 h, 7.3 h, and 53 min, respectively.

Zwart and Woutersen (1988) exposed specific pathogen free (SPF) Wistar-derived rats to chlorine at 322-5,793 ppm for exposure durations of 5 min to 1 h to calculate  $LC_{50}$  values. Observations were made over a 14-d period after which the animals were sacrificed and histologic examinations were made; additional groups of rats were exposed and examined 2 d after exposure. Rats exposed at the highest concentrations during the 30- and 60-min exposures showed signs of restlessness, eye and nasal irritation, labored breathing, and reduced respiratory rate. Mortalities occurred during exposure as well as within the first week of the observation period. Increased lung weights were positively correlated with higher concentrations and

longer exposure durations. At the high concentrations, 5,793 ppm for 5 min and 2,248 ppm for 10 min, effects were observed in the nose, larynx, and trachea; at those and the lower concentrations, lung lesions, including focal aggregates of mononuclear inflammatory cells, increased septal cellularity, squamous metaplasia of bronchiolar epithelium, and edema, were observed in one or more animals. Hyperplasia of the larynx and trachea observed at 2 d post-exposure was resolved by 14 d post-exposure in surviving rats. In rats in which minute-volume was measured, death occurred in several animals following a reduction in minute-volume to  $\leq 39\%$  of the pre-exposure level. According to the authors, the 1-h  $LC_{01}$  of 288 ppm (95% confidence interval, 222-345 ppm), estimated by probit analysis, appeared to correspond with the onset of irreversible lung damage. The breathing pattern during the exposures changed from regular inspiration directly followed by a regular expiration to rapid shallow breathing that lasted less than a minute. That was followed by maximal inhalation directly after expiration and a long post-inspiratory pause. No deaths occurred at 2,841 ppm for 5 min, 547 ppm for 30 min, or 322 ppm for 60 min.

In a subchronic study, three of 10 female F-344 rats exposed at 9 ppm for 6 h/d, 5 d/wk died by the 30th exposure (Barrow et al. 1979). In another study, groups of eight 30-wk-old male and female SPF rats were exposed to chlorine at approximately 117 ppm for 3 h/d, 7 d/wk until half of the animals of each gender died (Bell and Elmes 1965). Total exposure time to 50% mortality was 29 h for males and 32 h for females. When the response of conventional rats was compared with that of SPF rats, the response was more severe in the conventional rats, who exhibited proliferation of goblet cells and increased mucus, emphysema, and polymorphonuclear cells in the lungs.

### 3.1.3. Mice

Back et al. (1972), MacEwen and Vernot (1972), and Vernot et al. (1977) reported a 1-h  $LC_{50}$  of 137 (95% confidence limits, 119-159 ppm). Weedon et al. (1940) exposed groups of four mice to 63, 240, or 1,000 ppm for 16 h or until death. Times to 50% mortality were >16 h, 7.3 h, and 28 min, respectively.

Bitron and Aharonson (1978) exposed male albino mice to 170 ppm and 290 ppm for several exposure times and calculated 50% mortality as a function of exposure time ( $Lt_{50}$ ). Mice were restrained during the exposures. Observations were made over a 30-d period.  $Lt_{50}$  for the 170-ppm

and 290-ppm exposures were 55 min and 11 min, respectively. The results of this work were unusual in that many of the deaths were delayed, occurring during the second week of the observation period, rather than during and immediately following exposure. No mice died within a 30-d observation period following exposure at 290 ppm for 6 min. No deaths occurred in BALB/c mice exposed at 50, 100, or 150 ppm for 1 h (O'Neil 1991). Mice were observed for at least 5 d post-exposure.

Zwart and Woutersen (1988) exposed Swiss mice to chlorine at 579-1,654 ppm for 10 min and 458-645 ppm for 30 min to calculate LC<sub>50</sub> values. Mortality observations were made over a 14-d period. Nearly one-third of the mice died during the second week post-exposure, indicating to the authors that the deaths may have been due to secondary infection. Increased lung weights were positively correlated with higher concentrations and longer exposure durations. No deaths occurred at a concentration of chlorine at 754 ppm for 10 min.

Alarie (1980) reported that the 10-min LC<sub>50</sub> of male Swiss-Webster mice decreased from 302 ppm in uncannulated mice to 131 ppm when chlorine was delivered directly to the trachea via cannulation.

As part of an experiment on immune response, groups of 10 BALB/c mice were exposed at 50, 100, 150, 200, or 250 ppm for 1 h (O'Neil 1991). Mortality occurred at 200 ppm (two mice, 4-5 d post-exposure) and at 250 ppm (8/10 mice).

### 3.2. Nonlethal Toxicity

Data on effects following exposures to nonlethal concentrations of chlorine are available for the monkey, rat, mouse, guinea pig, and rabbit. Studies utilizing acute exposure durations are summarized in Table 1-5.

#### 3.2.1. Nonhuman Primates

No studies on single acute exposures were located. Klonne et al. (1987) exposed Rhesus monkeys to chlorine at 0, 0.1, 0.5, or 2.3 ppm for 6 h/d, 5 d/wk for 1 y. In the group exposed at 2.3 ppm, ocular irritation as well as treatment-related histopathologic changes limited to the nasal passages and trachea were observed at 1 y. Those lesions, consisting of focal, epithelial hyperplasia with loss of cilia and decreased numbers of goblet cells, were considered mild in the group exposed at 2.3 ppm and were not present in all

animals. Lesions in the lower exposure groups were minimal. No statistically significant differences were observed between control and exposure groups for pulmonary diffusing capacity of carbon monoxide or distribution of ventilation values (number of breaths to 1% N<sub>2</sub>).

### 3.2.2. Rats

Demnati et al. (1995) exposed groups of four male Sprague-Dawley rats (nose-only) to chlorine at 0, 50, 100, 200, 500, or 1,500 ppm for 2-10 min in order to study effects on airway mucosa and lung parenchyma. Histologic examinations were performed at 1, 3, 6, 12, 24, and 72 h after exposure. Exposures to concentrations of  $\leq 500$  ppm did not induce significant histologic changes. Lungs from control rats and from rats exposed at 50-100 ppm for 2 min were normal within 72 h; at concentrations of 200 ppm and 500 ppm for 2-5 min, there was only slight perivascular edema in all exposed rats. Exposure at 1,500 ppm for 2 min produced only slight effects, including mild perivascular edema and occasional small clusters of polymorphonuclear leukocytes in the mucosa of large airway. The 10-min exposure at 1,500 ppm caused significant changes that varied with time after exposure—airspace and interstitial edema associated with bronchial epithelial sloughing at 1 h, decreased edema and the appearance of mucosal polymorphonuclear leucocytes at 6-24 h, and epithelial regeneration as evidenced by hyperplasia and goblet cell metaplasia at 72 h. No deaths were reported.

Exposure at 25 ppm lowered the respiratory rate by 50% (RD<sub>50</sub>) in F-344 rats, presumably during a 10-min test (Barrow and Steinhagen 1982); during exposure for 6 h, the RD<sub>50</sub> was 10.9 ppm (Chang and Barrow 1984). Groups of 9-10 male F-344 rats were exposed to a concentration of chlorine at 9.1 ppm (the RD<sub>50</sub> of mice) for 1, 3, or 5 d (6 h/d) and examined for respiratory tract pathology (Jiang et al. 1983). Sacrifice took place immediately after exposure. In all animals, lesions were present in the nasal passages with less severe changes in the nasopharynx, larynx, trachea, and lungs. Lesions in the nasal passages involved epithelial degeneration with epithelial cell exfoliation, erosion, and ulceration (respiratory epithelium) and extensive epithelial erosion and ulceration (olfactory epithelium of the dorsal meatus). Electron microscopy examination revealed loss of respiratory and olfactory cilia and cellular exfoliation of the naso- and maxillo-turbinates.

**TABLE 1-5** Summary of Sublethal Effects in Laboratory Animals

Species	Concentration (ppm)	Exposure Time	Effect <sup>a,b</sup>	Reference
Rat	1,500	2 min	Mild perivascular edema of lung, leucocytic infiltration	Dennati et al. 1995
Rat	200, 500	2, 5 min	Slight perivascular edema of lung	Dennati et al. 1995
Rat	50, 100	2 min	No effect	Dennati et al. 1995
Rat	1,500	10 min	Epithelial hyperplasia, goblet cell metaplasia of lung	Dennati et al. 1995
Rat	25	10 min	RD <sub>50</sub>	Barrow and Steinhagen 1982
Rat	10.9	6 h	RD <sub>50</sub>	Chang and Barrow 1984
Rat	9.1	6 h	Lesions in nasal passages; less severe changes in nasopharynx, larynx, trachea, and lungs	Jiang et al. 1983
Mouse	9.3	10 min	RD <sub>50</sub>	Barrow et al. 1977
Mouse	3.5	1 h	RD <sub>50</sub>	Gagnaire et al. 1994
Mouse	9.1	6 h	Lesions in nasal passages; less severe changes in nasopharynx, larynx, trachea, and lungs	Jiang et al. 1983
Rabbit	50	30 min	No gross or microscopic lung changes	Barrow and Smith 1975
Rabbit	100, 200	30 min	Initial changes in lung function; hemorrhage, pneumonitis, bronchitis; recovery at 60 d except pulmonary compliance	Barrow and Smith 1975

<sup>a</sup>Observed immediately after exposure (Jiang et al. 1983), 72 h post-exposure (Dennati et al. 1995), 5 d post-exposure (O'Neil 1991), 14 d post-exposure (MacEwen and Vernot 1972; Barrow and Smith 1975; Zwart and Woutersen 1988), 30 d post-exposure (Bitron and Aharanson 1978).

<sup>b</sup>The RD<sub>50</sub> test is usually a 10-min test.

When male and female F-344 rats were exposed to chlorine at 0, 1, 3, or 9 ppm for 6 h/d, 5 d/wk for 6 wk, effects were observed in the upper and lower respiratory tract (Barrow et al. 1979). Lesions in male and female rats exposed at 9 ppm included widespread inflammation throughout the respiratory tract with hyperplasia and hypertrophy of epithelial cells of the respiratory bronchioles, alveolar ducts, and alveoli. Hepatocellular cytoplasmic vacuolation was observed in both genders exposed at 3 ppm or 9 ppm, and renal tubule effects were observed in male rats exposed at 9 ppm. Effects observed in animals exposed at 1 ppm or 3 ppm were much less severe than those observed at 9 ppm. In addition, decreased body weight gains were observed in females at all exposure concentrations and in males at 3 ppm and 9 ppm.

Groups of 14-wk-old male and female SPF rats were exposed to chlorine at approximately 40 ppm for 3 h/d for a total of 42 h (Bell and Elmes 1965). No deaths occurred. Signs and symptoms of exposure included coughing, sneezing, and runny and blood-stained noses after 3 h. Histologic examinations of the lungs revealed recovery by 14 d post-exposure. Exposure of conventional rats for 1 h daily to 14-18 ppm, for a total of 24 exposure hours in 4 wk, also resulted in no mortality (Elmes and Bell 1963). The authors considered the chlorine concentrations overestimates, because the exposed rats huddled together in the exposure cage.

Groups of 70 male and 70 female F-344 rats were exposed to chlorine gas at 0, 0.4, 1.0, or 2.5 ppm for 6 h/d, 5 d/wk (males) or 3 alternate d/wk (females) for 2 y (CIIT 1993; Wolf et al. 1995). Concentration-dependent lesions confined to the nasal passages were observed in all animals. These lesions were most severe in the anterior nasal cavity and included respiratory and olfactory epithelial degeneration, septal fenestration, mucosal inflammation, respiratory epithelial hyperplasia, squamous metaplasia and goblet cell hypertrophy and hyperplasia, and secretory metaplasia of the transitional epithelium of the lateral meatus. Body weights were depressed compared with controls, but no early deaths occurred.

### 3.2.3. Mice

Groups of four male Swiss-Webster mice were exposed to chlorine concentrations at 0.7-38.4 ppm for 10 min (Barrow et al. 1977). The  $RD_{50}$  was 9.3 ppm. The  $RD_{50}$  of male  $OF_1$  mice was calculated at 3.5 ppm by Gagnaire et al. (1994). Although that exposure was for 60 min, the decrease occurred by 10 min. The protocol of ASTM (1991) was followed by Gagnaire et al., but the  $OF$  strain is not the strain of mice suggested for use

in measuring sensory irritation. The ASTM (1991) RD<sub>50</sub> test calls for male Swiss-Webster mice and a 10-min exposure period.

In a follow-up study to that of Barrow et al. (1977), Buckley et al. (1984) exposed male Swiss-Webster mice to the RD<sub>50</sub> (9.3 ppm) for 6 h/d for 5 d. Half of each group was necropsied immediately after the last exposure and the other half was necropsied 72 h post-exposure. No deaths were reported. Lesions in both the anterior respiratory epithelium adjacent to the dorsal meatus and in the respiratory epithelium included exfoliation, inflammation erosion, ulceration, and necrosis. Chlorine reached the lower respiratory tract, as indicated by tracheal lesions and terminal bronchiolitis, with occlusion of the affected bronchioles by serocellular exudate. Recovery was minimal to moderate after 72 h.

Groups of 9-10 male Swiss-Webster mice were exposed at 9.1 ppm, the approximate RD<sub>50</sub> for 1, 3, or 5 d (6 h/d) and examined for respiratory tract pathology (Jiang et al. 1983). Sacrifice took place immediately after exposure. Respiratory tract lesions were similar to those of the rat described above (lesions in the nasal passages with less severe changes in the nasopharynx, larynx, trachea, and lungs).

Groups of 70 male and 70 female B6C3F<sub>1</sub> mice were exposed to chlorine gas at 0, 0.4, 1.0, or 2.5 ppm for 6 h/d, 5 d/wk for 2 y (CIIT 1993; Wolf et al. 1995). Concentration-dependent lesions confined to the nasal passages were observed in all animals. These lesions were most severe in the anterior nasal cavity and included respiratory and olfactory epithelial degeneration, septal fenestration, mucosal inflammation, respiratory epithelial hyperplasia, squamous metaplasia and goblet cell hypertrophy and hyperplasia, and secretory metaplasia of the transitional epithelium of the lateral meatus. Body weights were depressed (males, all exposures; females, 2.5 ppm) compared with controls, but no early deaths occurred.

#### **3.2.4. Guinea pigs**

Arlong et al. (1940, as cited in NIOSH 1976) exposed guinea pigs to 1.7 ppm for 5 h daily over 87 d. No deaths were reported during the 300-d observation period. No other details of the study were available.

#### **3.2.5. Rabbits**

Groups of two male and two female rabbits were exposed at 0, 50, 100, or 200 ppm for 30 min and tested for lung changes as measured by volume-

pressure relationships and inspiratory-expiratory flow rate at times from 30 min to 60 d post-exposure (Barrow and Smith 1975). Rabbits exposed at 50 ppm showed no changes at any time periods. Recovery of flow rate ratios occurred by 14 d post-exposure in the 100-ppm groups and by 60 d post-exposure in the 200-ppm group. Pulmonary compliance in the 100-ppm and 200-ppm groups did not return to control levels within 60 d. Examinations of the lungs revealed hemorrhages, pneumonitis and anatomic emphysema in the 100-ppm and 200-ppm groups at 3 and 14 d post-exposure; those changes were not present at 60 d post-exposure.

### 3.3. Developmental and Reproductive Effects

No studies addressing developmental or reproductive effects following inhalation exposure to chlorine were located. However, because effects on development and reproduction would be systemic, due to circulating chlorine, the effects of oral administration of chlorine may have bearing on the chlorine hazard assessment. Those data, reviewed by EPA (1996) and AIHA (1988), demonstrated no or insufficient evidence of reproductive or developmental toxicity.

Groups of 10 male B6C3F<sub>1</sub> mice were dosed by oral gavage with 1 mL of test solution containing OCl<sup>-</sup> or HOCl at 40, 100, or 200 mg/L/d for 5 d (Meier et al. 1985). Animals were sacrificed at 1, 3, or 5 wk after the last treatment, and the caudae epididymides were examined for sperm head abnormalities. No abnormalities were observed at 1 and 5 wk post-treatment in the groups treated with OCl<sup>-</sup> or at any time in the groups treated with HOCl. A small but statistically significance difference compared with controls was observed in the groups administered OCl<sup>-</sup> at all dose levels at 3 wk. These results do not clearly indicate an effect on fertility.

Druckrey (1968) administered highly chlorinated drinking water (100 mg/L) to seven consecutive generations of BD II rats. The average daily dose was estimated at 10 mg/kg/d. No treatment-related effects were observed on any generation. Carlton et al. (1986) administered chlorine in deionized water by gavage male and female Long Evans rats at doses of 1.0, 2.0, or 5.0 mg/kg/d to for 66-76 d. Dosing began prior to mating and continued during gestation and lactation. Groups of offspring were necropsied at 21 d after birth or at 40 d of age; the latter group was dosed following weaning. No statistically significant differences were observed between the control and treated rats in litter survival, litter size, or pup weight.

HOCl, formed by bubbling chlorine gas through water, was administered in the drinking water to Sprague-Dawley rats for 2.5 mo, prior to and

throughout gestation (Abdel-Rahman 1982). Concentrations were 0, 1, 10, or 100 mg/L. Rats were sacrificed on day 20 of gestation and fetuses were examined for bone and soft-tissue defects. No increase in resorptions were found in any treatment group. A significant increase in skeletal anomalies at the 100 mg/L concentration (incompletely ossified or missing sternbrae and rudimentary ribs) was interpreted as a nonspecific retardation in growth. The increase in skeletal defects was not significant in the 10 mg/L and 100 mg/L groups compared with the controls. Total defects—skeletal and soft tissue—were increased significantly over the control group in the 100 mg/L group. Maternal toxicity—body weight, food consumption, clinical signs—was not described.

### 3.4. Genotoxicity

No data on inhalation exposures were located in the available literature. Genotoxicity studies were conducted via oral dosing of groups of 10 male B6C3F<sub>1</sub> mice with 1 mL of test solution containing OCl<sup>-</sup> or HOCl at concentrations of 40, 100, or 200 mg/L for 5 d (Meier et al. 1985). Chlorine was not mutagenic in the bone marrow micronucleus and cytogenetic assays. Sodium hypochlorite produced chromosomal aberrations in several mammalian cell tests (NTP 1992).

### 3.5. Chronic Toxicity and Carcinogenicity

Groups of male and female F-344 rats were exposed by inhalation to chlorine concentrations at 0, 0.4, 1.0, or 2.5 ppm for 2 y (CIIT 1993; Wolf et al. 1995). Histologic examinations of the nose and major organs revealed no increase in the incidence of neoplasia over that of control groups. F-344 rats and B6C3F<sub>1</sub> mice of both genders administered chlorinated or chloraminated drinking water for 2 y showed no increased incidences of neoplasms (NTP 1992).

### 3.6. Summary

Few animal studies addressed no- or mild-effect levels at exposure times of 10 min to 8 h. No gross or microscopic lung changes occurred in rabbits following a 30-min exposure at 50 ppm (Barrow and Smith 1975). The highest 30-min values resulting in no deaths (LC<sub>0</sub>) for the rat and rabbit

were 547 ppm (Zwart and Woutersen 1988) and 200 ppm (Barrow and Smith 1975), respectively. The 60-min concentrations resulting in no deaths in the rat and mouse were 322 (Zwart and Woutersen 1988) and 150 ppm (O'Neil 1991), respectively. No deaths, but moderate to severe lesions of the respiratory tract and peribronchiolitis, occurred in rats following a 6-h exposure at 9.1 ppm (Jiang et al. 1983).

Thirty-minute LC<sub>50</sub> values ranged from 137 ppm in the mouse (Back et al. 1972) to 700 ppm in the rat (Zwart and Woutersen 1988). The 60-min LC<sub>50</sub> and LC<sub>01</sub> values for the rat were 455 ppm and 288 ppm (Zwart and Woutersen 1988).

Chlorine administered in the drinking water or by gavage to rats or mice did not cause reproductive or developmental problems (Druckrey 1968; Abdel-Rahman 1982; Meier et al. 1985; Carlton et al. 1986). A 2-y inhalation study with rats showed no evidence of carcinogenicity (CIIT 1993; Wolf et al. 1995). Mutagenicity tests were generally negative (Meier et al. 1985).

## 4. SPECIAL CONSIDERATIONS

### 4.1. Metabolism and Disposition

Pharmacokinetic data following acute exposures were not available. Metabolic and kinetic considerations are not relevant regarding the determination of AEGL values because animals die of acute respiratory failure. Chlorine gas reacts at the site of contact, and very little of the chemical is absorbed into the bloodstream (Eaton and Klaassen 1996).

### 4.2. Mechanism of Toxicity

Although of moderate solubility, chlorine is categorized as a Category I gas because it is so rapidly irreversibly reactive in the surface liquid and tissue of the respiratory tract (EPA 1994). Studies with repeated exposures of laboratory animals indicate that at moderate to high concentrations, chlorine is not effectively scrubbed in the upper respiratory tract and is therefore capable of exerting its effects over the entire respiratory tract (Barrow et al. 1979); however, at low concentrations ( $\leq 2.5$  ppm for up to 2 y), chlorine is effectively scrubbed in the anterior nasal passages as indicated by the absence of lesions in the lower respiratory tract of rats, mice (Wolf et al. 1995), and monkeys (Klonne et al. 1987).

Chlorine gas combines with tissue water to form hydrochloric and hypochlorous acids (HCl and HClO); the latter spontaneously breaks down into HCl and free O<sup>•</sup>, which combines with water, releasing oxygen radicals (O<sup>-</sup>). The oxygen radical produces major tissue damage, which is enhanced by the presence of HCl (Perry et al. 1994; Wolf et al. 1995).

The response to inhalation of chlorine can range from sensory irritation and reflex bronchoconstriction to death, the latter due to pulmonary edema. The sensory irritation response to chlorine is due to stimulation of the trigeminal nerve endings in the respiratory mucosa, which results in a decrease in respiratory rate (Alarie 1981). Reflex bronchoconstriction is a local reaction in which cholinergic-like agents bind to respiratory tract cell surface receptors and trigger an increase in the intracellular concentration of cyclic guanosine monophosphate. That facilitates contraction of the smooth muscles that surround the trachea and bronchi, causing a decrease in airway diameter and a corresponding increase in resistance to airflow that may result in wheezing, coughing, a sensation of chest tightness, and dyspnea (Witschi and Last 1996). Death can occur from lack of oxygen during an asthmatic attack or if chlorine reaches the lungs and causes pulmonary edema; delayed deaths that occur starting 3 d after exposure might be due to bronchial infection (Underhill 1920; Withers and Lees 1985a; Bitron and Aharonson 1978), which may be treatable in humans.

### 4.3. Structure-Activity Relationships

The combined human and animal data on chlorine are sufficient for derivation of inhalation exposure guidelines and the use of structure-activity comparisons is not necessary. Like hydrogen chloride (HCl) and fluorine (F<sub>2</sub>), Cl<sub>2</sub> is an irritant to the eyes, skin, and respiratory tract. When compared with mortality data for HCl and F<sub>2</sub> (Wohlslagel et al. 1976; ATSDR 1993; Perry et al. 1994; NAC 1996), chlorine is more toxic than HCl but slightly less toxic than F<sub>2</sub> to laboratory rodents.

### 4.4. Concentration-Exposure Duration Relationship

When considering low concentrations of chlorine that do not result in tissue damage and the response of atopic and asthmatic individuals, time-scaling might not be relevant. Several studies, such as those of Shusterman et al. (1998) and D'Alessandro et al. (1996), were conducted for short periods of time—15 min and 1 h, respectively—because responses were ex-

pected within those time periods. Furthermore, in the Rotman et al. (1983) study, pulmonary function parameters of the tested individuals, including those for the atopic individual, did not change between the 4- and 8-h measurements, indicating a sustained effect during exposure.

Time-scaling is relevant at the AEGL-2 and AEGL-3 levels where tissue damage is involved. When data are lacking for desired exposure times, scaling across time might be based on the relationship between concentration and exposure duration when a common end point is used. The relationship between concentration and time is described by  $C^n \times t = k$ , in which the exponent  $n$  may be different from 1. ten Berge and Vis van Heemst (1983) analyzed the data of Anglen (1981) on irritation response in humans. Irritation consisted of itching or burning of the eyes, nose, or throat and scores ranged from 0 (no sensation) to 5 (unbearable), with a score of 3 representing a nuisance level or irritation. Regression analysis of the percent of subjects reporting a nuisance irritation response to concentrations at 1 ppm and 2 ppm over exposure durations of 30 min and 120 min resulted in an  $n$  value of 1.9.

Several sets of mortality data from animal studies were available for calculating the relationship between concentration and exposure time. Using the probit analysis method of ten Berge et al. (1986) and/or regression analysis, the data sets and their values for  $n$  are Bitron and Aharonson (1978),  $n = 3.5$ ; Zwart and Woutersen (1988),  $n = 1.0$ ; and Weedon et al. (1940),  $n = 1.1$ . The probit analysis method applied to the data for 11 irritant gases from ten Berge et al. (1986) also results in a range of values for  $n$  of 1.0-3.5. In the Bitron and Aharonson (1978) study, the mice were restrained (which usually results in lower lethality values than for unrestrained animals) and deaths were delayed, occurring during the second week of the observation period rather than during and immediately following exposure. Because of the questionable methodology and the delayed deaths (possibly due to bacterial infection) in the Bitron and Aharonson (1978) study, that study will not be considered for calculating a time-scaling relationship for chlorine. Respiratory irritation is an initial step in the progression of irritation that leads to pulmonary edema and death. Based on the evidence for a similar mechanism of action for irritation and death, an  $n$  value of 2 will be used for time-scaling for chlorine. An  $n$  value of 2 for time-scaling the threshold for lethality for chlorine is supported by data for another halogen. Using the end point of lethality, the concentration-exposure duration relationship for fluorine for several mammalian species was  $C^n \times t = k$ , where  $n$  was approximately slightly less than 2 (NAC 1996). Based on relative toxicity (Section 4.3), the  $n$  value for chlorine would be

higher than that of fluorine when extrapolating from longer to shorter exposure durations.

## 4.5. Other Relevant Information

### 4.5.1. Susceptible Populations

Chlorine is highly irritating and corrosive to the tissues of the respiratory tract. At low concentrations, the direct action of chlorine on the respiratory tract is not expected to vary greatly among most healthy individuals, including infants, children, and the elderly. For example, at the low concentration of 0.5 ppm, neither healthy nor atopic subjects exhibited clinically significant changes (decreases of  $\geq 10\%$ ) in peak air flow (Shusterman et al. 1998). At 0.4 ppm, there were no statistically significant changes in several respiratory parameters in either healthy or asthmatic subjects (D'Alessandro et al. 1996). In the Rotman et al. (1983) study where numerous pulmonary parameters were measured, healthy subjects responded in a similar manner at 1 ppm.

Data from the Rotman et al. (1983) and D'Alessandro et al. (1996) studies on chlorine exposures and individuals with airway hyper-reactivity or asthma indicate that, compared with the general population, the respiratory tracts of those individuals may be very reactive to the presence of chlorine, as reported in Section 2.2.1. Responsiveness to inhaled agents varies among individuals with airway hyper-reactivity and asthma. In children, asthma may be defined as mild, moderate, or severe depending on the response to an inhaled agonist (Larsen 1992). With mild asthma, symptoms are infrequent and brief, and treatment is with inhaled  $\beta$ -agonists as needed. Moderate asthma is defined by symptoms occurring twice weekly, and treatment is with cromolyn sodium or slow-release theophylline. Severe asthmatic patients have daily symptoms, and daily treatment with oral or inhaled steroids is required. One individual in the D'Alessandro et al. (1996) study with airway hyper-reactivity had a clinical history of asthma and "was being treated regularly with inhaled or systemic corticosteroids."

There is a concern that children with airway hyper-reactivity and asthma may be more sensitive to inhaled irritants and allergens than adult asthmatic subjects. Avital et al. (1991) studied the response to methacholine challenge in 182 asthmatic children (132 males and 50 females) of various ages. All therapy except corticosteroid therapy and slow-release theophylline was discontinued prior to the study. The children were divided

into three age groups—1-6 y, 7-11 y, and 12-17 y—and into three clinical groups according to their minimal therapeutic requirements—mild, moderate, and severe asthma. In older children, responsiveness was measured by the methacholine provocation concentration that produced a 20% fall in FEV<sub>1</sub>. In younger children, responsiveness was determined by wheezing, persistent cough, or tachypnea following a methacholine challenge. Bronchial reactivity correlated inversely with the severity of bronchial asthma according to minimal drug requirements and was similar over the age range for each severity group. That is, in each severity category (mild, moderate, or severe asthma), the mean concentration of methacholine that evoked the designated response was the same regardless of the age group. Avital et al. (1991) compared their results with the responses of asthmatic adults challenged with histamine from a study by Cockcroft et al. (1977). Mean ages of the adults were between 30 y and 40 y. Although the classes of asthmatic severity in Cockcroft et al. (1977) were only broadly comparable to those in Avital et al. (1991), there was a “striking similarity” in the results. That is, the concentration at which adults in each class of severity responded to histamine was similar to the methacholine concentration at which the children reacted in the respective severity classes.

Adults were tested in the D’Alessandro et al. (1996) study, but the ages of the individuals in the lower range of age (18 y) were close to those of the older children in the Avital et al. (1991) study. The range of provocative concentrations of methacholine in the D’Alessandro et al. (1996) study overlapped the range of the severe asthmatic subjects in the Avital et al. (1991) study, indicating that at least one adult in the former study was as responsive to methacholine as the children with severe asthma. Furthermore, during the exposure to chlorine at 1 ppm, the atopic individual in the Rotman et al. (1983) study, who was not on medication, responded to a greater degree, as measured by a fall in FEV<sub>1</sub>, than any of the subjects with airway reactivity or asthma in the D’Alessandro et al. (1996) study. Following a 4-h exposure to chlorine at 1 ppm, the FEV<sub>1</sub> of the atopic individual was 45% of the pre-exposure value (Rotman et al. 1983), whereas following the 1-h exposure at 1 ppm, the FEV<sub>1</sub> of the most sensitive individual with airway hyper-reactivity or asthma was 61% of the pre-exposure value (D’Alessandro et al. 1996). The final FEV<sub>1</sub> for both subjects was the same, 1,900 mL.

Another consideration when evaluating the response of individuals with airway hyper-reactivity and asthma to irritants and allergens is the time of response following challenge. Individuals in the D’Alessandro et al. (1996) study were exposed to chlorine for only 1 h; the response of the atopic individual in the Rotman et al. (1983) study may have occurred early, al-

though wheezing did not occur until after 4 h of exposure. According to the literature, asthmatic reactions may be immediate (within minutes of exposure) or delayed (hours after exposure) (Larsen 1992). In the Avital et al. (1991) study, response to provocative concentrations of methacholine occurred within 2 min. In both studies with chlorine (Rotman et al. 1983; D'Alessandro et al. 1996), individuals had responded by the time they were tested for pulmonary function changes (after 1 h and after 4 h of exposure), and there were no reported delayed or greater effects within the 24-h post-exposure period. Although not conclusive, the data appear to indicate that the response to chlorine is concentration-dependent rather than time-dependent. Thus, a 1-h exposure to chlorine is sufficient to elicit a response in susceptible individuals; if a response is not present at 1 h, it is unlikely to occur with continued exposure.

#### **4.5.2. Reactive Airways Dysfunction Syndrome**

In humans, the reported long-term effects of accidental exposures at high concentrations of chlorine (evidenced by the presence of a yellow-green cloud) are conflicting, some authors noting residual pulmonary abnormalities (Kowitz et al. 1967; Alberts and do Pico 1996), and others either reporting no significant permanent damage (Weill et al. 1969; Kaufman and Burkons 1971; Jones et al. 1986) or reporting that the presence of permanent damage was questionable (Charan et al. 1985). Several case reports described respiratory hyper-responsiveness following acute exposures to chlorine at high concentrations. This syndrome, called reactive airways disease or reactive airways dysfunction syndrome (RADS), is initiated by one or several exposures to high concentrations of an irritating gas. Case studies were reviewed by Alberts and do Pico (1996) and Lemiere et al. (1996). In several of the studies, a clear interpretation of the results was complicated by the lower values in pulmonary function tests of smokers.

#### **4.5.3. Gender and Species Variability**

Several studies indicated gender differences in responses to chlorine exposure, females rats (Barrow et al. 1979; Wolf et al. 1995) and monkeys (Klonne et al. 1987) developing lesions at lower concentrations than males. However, male mice were slightly more sensitive to chlorine than female mice (Wolf et al. 1995).

The data also allow an examination of interspecies differences. In those studies in which investigators tested the lethality of chlorine to two species (rat and mouse), the  $LC_{50}$  values were within a factor of approximately 2 of each other (Weedon et al. 1940; Back et al. 1972; Zwart and Woutersen 1988).

#### 4.5.4. Tolerance to Repeated Exposures

Following repeated exposures, rats developed sensory irritation tolerance to chlorine, as indicated by an increase in the  $RD_{50}$  following pretreatment (Barrow and Steinhagen 1982). The  $RD_{50}$  was 25 ppm in non-pretreated rats, whereas  $RD_{50}$  values increased to 90, 71, and 454 ppm when rats were pretreated at 1, 5, or 10 ppm, respectively, for 6 h/d, 5 d/wk for 2 wk and exposed 16-24 h following the last day of pretreatment.

## 5. DATA ANALYSIS FOR AEGL-1

### 5.1. Summary of Human Data Relevant to AEGL-1

NIOSH (1976) discussed available epidemiology studies conducted prior to 1976. In those studies, and in a more recent study (Ferris et al. 1979), work room concentrations averaged <1 ppm, and no effects could be clearly documented. However, those studies generally included a "healthy worker" population.

The studies by Anglen (1981) and Rotman et al. (1983) indicate that there are no significant sensory irritation and no serious pulmonary function changes associated with 4- or 8-h exposures in healthy human subjects of both genders at 0.5 ppm or 1.0 ppm. However, an atopic individual, whose baseline pulmonary parameters were outside of the normal range, as defined by DuBois et al. (1971), experienced changes in several pulmonary function parameters after exposure to chlorine at 0.5 ppm (Rotman et al. 1983). The greatest change for that individual was in  $R_{aw}$ , which increased 40% over the pre-exposure value after 4 h of exposure at 0.5 ppm and increased 33% over the pre-exposure value after 8 h of exposure at 0.5 ppm. Changes in  $R_{aw}$  in the healthy subjects were 5% and 15% for the respective time periods. Transient changes in pulmonary function (specifically  $R_{aw}$ ) were seen at 1.0 ppm for 4 h in the Rotman et al. study (1983) and at 2 ppm for 4 h in the Anglen (1981) study; however, sensory irritation reached nuisance

levels at the latter concentration-exposure time combination. In healthy subjects, no lung function measurements were changed at 2.0 ppm for 2 h (Anglen 1981). The FEV<sub>1</sub>, a particularly reproducible and sensitive measure of obstructive or restrictive flows in the lung (Witschi and Last 1996), changed by less than 10% following exposure at 1 ppm for 4 h. The atopic individual did not tolerate the 1-ppm exposure during a second 4-h exposure period because of serious respiratory effects.

In the study by D'Alessandro et al. (1996), subjects with a clinical history of asthma were tested. In the study by Shusterman et al. (1998), subjects with seasonal allergic rhinitis were tested. A concentration at 0.4 ppm for 1 h elicited no statistically significant response in airflow or resistance in asthmatic subjects (D'Alessandro et al. 1996). In the same study, a concentration at 1.0 ppm for 1 h elicited significant changes in several pulmonary function parameters for both normal subjects and subjects with asthma, but the mean changes were considered modest by the study authors. However, the R<sub>aw</sub> of one subject with asthma more than tripled during the exposure at 1.0 ppm.

## 5.2. Summary of Animal Data Relevant to AEGL-1

All short-term animal studies were conducted at concentrations that produced effects greater than those defined by the AEGL-1.

## 5.3. Derivation of AEGL-1

The studies by Anglen (1981), Rotman et al. (1983), D'Alessandro et al. (1996), and Shusterman et al. (1998) are all relevant to the development of AEGL-1 values. Those studies addressed sensory irritation as well as symptomatic and asymptomatic changes in several pulmonary function parameters. In addition, the studies used a diverse population, including healthy, atopic, and asthmatic subjects. Exercise was incorporated into the protocol of the Rotman et al. (1983) study, simulating conditions of stress. The studies indicate that 0.5 ppm, for 15 min (Shusterman et al. 1998) or for an interrupted 8 h with incorporation of exercise into the protocol (Rotman et al. 1983), is the highest NOAEL consistent with the definition of an AEGL-1 (i.e., a NOAEL for notable discomfort and irritation accompanied by non-disabling, transient, asymptomatic effects). The next highest concentration tested, 1.0 ppm for more than 4 h, resulted in effects, such as shortness of breath and wheezing, greater than those defined by an AEGL-1. The NOAEL of 0.5 ppm was identified from the study of Rotman et al.

subjects in the study of D'Alessandro et al. (1996). That NOAEL supports the choice of 0.5 ppm in the study of Rotman et al. (1983). In fact, the single exercising atopic subject proved to be more sensitive at 0.5 ppm, as indicated by asymptomatic changes in pulmonary function parameters, than were the non-exercising asthmatics exposed at 0.4 ppm.

Because subjects identified as most susceptible to the irritant effects of chlorine were tested (atopic and asthmatic individuals), an intraspecies uncertainty factor (UF) of 1 was applied. The intraspecies UF of 1 is also supported by the fact that bronchoconstriction is induced in pediatric and adult asthmatic subjects at similar levels of challenge (Avital et al. 1991). Therefore, an additional UF to protect pediatric asthmatic subjects is not necessary. Time-scaling was not applied to the AEGL-1 for several reasons. The study conducted by Rotman et al. (1983) actually lasted more than 8 h (two 4-h sessions with a 1-h break between). That reduces the uncertainty usually associated with scaling from shorter to longer time periods. Because effects were not increased following an interrupted 8 h of exposure at 0.5 ppm in the susceptible individual, the 8-h AEGL-1 was also set at 0.5 ppm. The use of the same value across all exposure durations is supported by the fact that the response to the irritant effects of chlorine appears to be concentration-dependent rather than time-dependent. Calculations are presented in Appendix A; results are presented in Table 1-6. Figure 1-1 is a plot of the derived AEGL values and all of the human and animal data on chlorine.

## 6. DATA ANALYSIS FOR AEGL-2

### 6.1. Summary of Human Data Relevant to AEGL-2

The studies by Joosting and Verberk (1974), Anglen (1981), Rotman et al. (1983), and D'Alessandro et al. (1996) address sensory irritant effects in humans as well as differences in pulmonary function tests during exposures at 1.0 ppm for up to 8 h and 2.0 ppm and 4.0 ppm for 4 h. As previously noted, Rotman et al. (1983) used an exercising atopic individual, and D'Alessandro et al. (1996) used five subjects with nonspecific airway hyper-reactivity, three of which had clinical histories of asthma. For healthy individuals, the concentrations tested by Rotman et al. (1983) and D'Alessandro et al. (1996) resulted in effects below the definition of the AEGL-2. However, in one atopic subject, the 1-ppm concentration for

**TABLE 1-6** AEGL-1 Values for Chlorine (ppm [mg/m<sup>3</sup>])

10 min	30 min	1 h	4 h	8 h
0.5 (1.5)	0.5 (1.5)	0.5 (1.5)	0.5 (1.5)	0.5 (1.5)

more than 4 h resulted in shortness of breath and wheezing as well as serious pulmonary function changes (Rotman et al. 1983). In the D'Alessandro et al. (1996) study, there was a significant fall in FEV<sub>1</sub> and airway resistance immediately after exposure at 1.0 ppm in both healthy and asthmatic subjects, but the fall among asthmatic subjects was greater. Two of the subjects with airway hyper-reactivity experienced undefined respiratory symptoms following the exposure. Those subjects were not specifically identified as asthmatics. The 2-ppm concentration tested by Anglen (1981) reached nuisance levels by 4 h and was accompanied by transient changes in pulmonary functions. Joosting and Verberk (1974) did not find differences in pulmonary function after exposure at 2 ppm for 2 h; concentrations at 4 ppm for 2 h were irritating, but pulmonary function measurements were not made. Subjective sensory irritation of the throat reached an average level of "nuisance" (distinctly perceptible to offensive) at 4 ppm for 2 h; irritation of the nose and the desire to cough averaged "distinctly perceptible" (range, up to nuisance). No sensory response was reported as unbearable during these exposures. Although airflow resistance, the major pulmonary effect found in the Rotman et al. (1983) study, was not measured by Joosting and Verberk (1974), Anglen (1981) reported that the changes in other pulmonary parameters at 2 ppm for 4 or 8 h were completely reversible by the next day.

## 6.2. Summary of Animal Data Relevant to AEGL-2

Alarie (1981), the author of the ASTM (1981) RD<sub>50</sub> test, considered the RD<sub>50</sub> for male Swiss-Webster mice to be intolerable to humans but non-lethal over a period of hours. The RD<sub>50</sub> of male Swiss-Webster mice was 9.3 ppm, as reported by Barrow et al. (1977). Lesions were present in the nasal passages, and some changes were present in the lower respiratory tract of rats exposed at 9.1 ppm for 6 h (Jiang et al. 1983). Mice "tolerated" that concentration 6 h/d for 5 d without deaths (Buckley et al. 1984). No gross or microscopic lung changes occurred in rabbits following a 30-min exposure at 50 ppm (Barrow and Smith 1975).



**TABLE 1-7** AEGL-2 Values for Chlorine (ppm [mg/m<sup>3</sup>])

10 min	30 min	1 h	4 h	8 h
2.8 (8.1)	2.8 (8.1)	2.0 (5.8)	1.0 (2.9)	0.71 (2.0)

### 6.3. Derivation of AEGL-2

Because human data are available, they should be used to calculate the AEGL-2. In the Rotman et al. (1983) study, the exposure of a susceptible subject to chlorine at 1 ppm for 4 h did not produce respiratory symptoms, but an exposure for more than 4 h resulted in serious asthma-like symptoms and serious pulmonary function changes. The pulmonary function changes were reversible, and the ability to escape was not impaired, but an asthma-like attack that occurred during the longer exposure must be considered a serious health effect. Therefore, the first 4-h exposure, which was without symptoms, can be considered a NOAEL for the symptoms that define AEGL-2. In the D'Alessandro et al. study (1996), two subjects with airway hyper-reactivity and three subjects with asthma also experienced significant changes in several pulmonary function parameters following a 1-h exposure at 1.0 ppm. No respiratory symptoms were experienced during the exposure, although undefined symptoms were experienced later. Because individuals representative of the susceptible population were tested and their reactions met the definition of the AEGL-2, no UF for differences in human sensitivity was applied. Time-scaling was applied to the AEGL-2 values because the 1.0-ppm concentration meets the definition of nuisance irritation, as described by ten Berge and Vis van Heemst (1983), and evacuation or sheltering should take place at the high short-term exposure concentrations that might cause an asthma attack. The 4-h 1-ppm concentration was scaled to the other time periods using the  $C^2 \times t = k$  relationship, derived by ten Berge and Vis van Heemst (1983). The 10-min value was set equal to the 30-min value so that the highest exposure of 4.0 ppm in the controlled human studies was not exceeded. Results are presented in Table 1-7 (above), and the calculations are presented in Appendix A.

The 1-y study in which exposure of Rhesus monkeys to a concentration of Cl<sub>2</sub> at 2.3 ppm resulted in mild histopathologic changes in the nasal passages and trachea (Klonne et al. 1987) supports the safety of this short-term exposure for humans. Regarding size, anatomy, and tissue distribution, the respiratory tract of the monkey is an appropriate model for humans.

## 7. DATA ANALYSIS FOR AEGL-3

### 7.1. Summary of Human Data Relevant to AEGL-3

Although exposure concentrations as high as 30 ppm (actual exposure concentration and duration not known) and 66 ppm for unspecified amounts of time have been reported by Abhyankar et al. (1989) and Shroff et al. (1988), respectively, measurement methods and/or exposure times were not well-documented. No deaths were reported at those exposures. No reliable data on concentrations that caused death in humans were located.

### 7.2. Summary of Animal Data Relevant to AEGL-3

The most recent available data for relevant exposure times show that 1-h concentrations that resulted in no deaths range from 150 ppm in the mouse (O'Neil 1991) to 322 ppm in the rat (Zwart and Woutersen 1988). On the basis of the rat data, Zwart and Woutersen (1988) calculated an  $LC_{01}$  of 288 ppm. Thirty-minute concentrations that resulted in no deaths were 547 ppm in the rat (Zwart and Woutersen 1988) and 200 ppm in the rabbit (Barrow and Smith 1975). The mouse data of Zwart and Woutersen (1988) were insufficient for calculating an  $LC_{01}$ . In addition, lethality measurements in many of the mouse studies were complicated by delayed deaths (Bitron and Aharonson 1978; O'Neil 1991), which were attributed to pneumonia. Mice might not be an appropriate model for extrapolation to humans, as stated by ten Berge et al. (1986) and the NRC (1991) for HCl. The NRC (1991) states that when considering lethal concentrations of respiratory irritants (such as HCl), the mouse "may not be an appropriate model for extrapolation to humans," because "mice appear to be much more susceptible to the lethal effects of HCl than other rodents or baboons. To some extent, this increased susceptibility may be due to less effective scrubbing of HCl in the upper respiratory tract."

### 7.3. Derivation of AEGL-3

Because the experimental data in mice appeared to provide an overly conservative estimate of lethality that was not consistent with the overall preponderance of the data, a value less than the concentration that resulted in no deaths in rats but greater than the value that resulted in no deaths in

mice was chosen as the basis for the AEGL-3 values. The 200-ppm value is below the 1-h highest nonlethal concentrations (213 ppm and 322 ppm) and the  $LC_{01}$  (288 ppm) in two well-conducted studies with rats (MacEwen and Vernot 1972; Zwart and Woutersen 1988) and above the 1-h highest nonlethal concentration in mice, 150 ppm (O'Neil 1991). The 200-ppm concentration is an  $LC_{20}$  for the mouse. A UF of 3 was used to extrapolate from rats to humans (the data show that interspecies differences were within a factor of approximately 2 for lethality). In addition, chlorine is a contact-site, direct-acting toxicant; there is no metabolic or pharmacokinetic component to chlorine-induced effects, and there is likely to be little difference between species in the response of biologic tissues to chlorine exposure. Also, for intraspecies differences, corrosive gases acting at the point of contact would predict low variability in a population; thus a UF of 3 is applied to protect susceptible individuals. The relative sensitivity of asthmatic individuals when considering lethality is unknown, but the data from the Rotman et al. (1983) and D'Alessandro et al. (1996) studies show that doubling a no-effect concentration for irritation and bronchial constriction resulted in potentially serious effects. The data were divided by a combined UF of 10 and were scaled to the 30-min and 4- and 8-h exposure durations using the  $C^2 \times t = k$  relationship, which was based on a nuisance level of human irritation (ten Berge and Vis van Heemst 1983). Application of a larger intraspecies UF, such as 10 (for a total of 30), would result in an 8-h AEGL-3 value of 2.3 ppm. It is unlikely that asthmatic subjects would suffer life-threatening symptoms at 2.3 ppm. The values appear in Table 1-8 and the calculations are presented in Appendix A.

It should be noted that the 1-h AEGL-3 of 20 ppm is below the AEGL-3 derived from the 1-h rat  $LC_{01}$  (the threshold for lethality) of 28.8 ppm (288/10).

## 8. SUMMARY OF AEGLs

### 8.1. AEGL Values and Toxicity End Points

In summary, the AEGL values for various levels of effects and various exposure periods were derived using the following methods. The AEGL-1 was based on a study with human volunteers, including a susceptible individual, in which a concentration of chlorine at 0.5 ppm for 4 h produced no sensory irritation and resulted in only mild transient effects on pulmonary parameters in the healthy individuals. Pulmonary changes in the susceptible

**TABLE 1-8** AEGL-3 Values for Chlorine (ppm [mg/m<sup>3</sup>])

10 min	30 min	1 h	4 h	8 h
50 (145)	28 (81)	20 (58)	10 (29)	7.1 (21)

individual were greater than those in healthy subjects, but did not result in symptoms above the definition of the AEGL-1. Because both genders were tested in one of the studies, a susceptible individual was observed in the other study, and subjects were undergoing light exercise (making them more vulnerable to sensory irritation), no UF to account for differences in human sensitivity was applied. The 0.5-ppm no-effect concentration for susceptible individuals is supported by a 1-h 0.4-ppm no-effect concentration for individuals with airway hyper-reactivity or asthma. The 0.5-ppm exposure was considered a threshold for more severe effects, regardless of exposure duration.

The AEGL-2 values were based on the same studies used to derive the AEGL-1s. In those studies healthy human volunteers experienced transient changes in pulmonary function measurements and a susceptible individual experienced an asthma-like attack (shortness of breath and wheezing) following a more than 4-h exposure to chlorine at 1 ppm. The susceptible individual remained in the exposure chamber for the full 4 h before the symptoms occurred. Because both genders were tested, subjects were undergoing light exercise (making them more vulnerable to sensory irritation), and a susceptible individual was tested, no UF was applied to account for differences in human sensitivity. Similar effects and symptoms in individuals with airway hyper-reactivity or asthma exposed at 1.0 ppm for 1 h supports the application of no intraspecies UF for the 4-h concentration. The 4-h 1-ppm concentration was scaled to the other time periods using the  $C^2 \times t = k$  relationship.

In the absence of human data, the AEGL-3 values were based on a concentration lower than the highest value that resulted in no deaths for the rat but equal to the LC<sub>20</sub> for the mouse. The data resulting from exposure of mice to chlorine were either incomplete or complicated by delayed deaths. The 1-h concentration of 200 ppm was divided by a UF of 3 to extrapolate from animals to humans (interspecies values for the same end point differed by a factor of approximately 2 within each of several studies) and by a UF of 3 to account for differences in human sensitivity (the toxic effect is the result of a chemical reaction with biologic tissue of the respiratory tract, which is unlikely to differ among individuals). The AEGL-3

values for the other exposure times were calculated based on the  $C^2 \times t = k$  relationship.

The three AEGL levels for the four exposure times are listed in Table 1-9.

## 8.2. Extant Standards and Guidelines for Chlorine

Standards and guidance levels for workplace and community exposures are listed in Table 1-10. The 8-h AEGL-1 and the ACGIH TLV-TWA values are the same, and, although not specifically stated, the TLV-TWA is based on the absence of irritation during the 8-h exposure at 0.5 ppm in the Anglen (1981) and Rotman et al. (1983) studies. Those and another, more recent study are also the basis for the AEGL-1 and AEGL-2 values. The AEGL-1 is specifically protective of asthmatic subjects.

The NIOSH IDLH is based on acute inhalation toxicity data in humans, Freitag (1941, as cited in NIOSH 1994a), and several secondary sources, such as ILO (1998). Exposure at 30 ppm is stated to cause intense coughing fits, and exposure at 40-60 ppm for 30-60 min might cause serious damage (ILO 1998). The IDLH is higher than the AEGL-2, but less than the AEGL-3. The AEGL-2 is lower because it is specifically protective of asthmatic patients.

The ERPG-1 is based on an 8-h human exposure at 1 ppm that produced slight transient pulmonary effects in healthy subjects (Anglen 1981; Rotman et al. 1983). The 1-h AEGL-1 is lower than the ERPG-1 because

**TABLE 1-9** Summary and Relationship of AEGL Values

Classification	Exposure Duration				
	10 min	30 min	1 h	4 h	8 h
AEGL-1 (Nondisabling)	0.5 ppm (1.5 mg/m <sup>3</sup> )				
AEGL-2 (Disabling)	2.8 ppm (8.1 mg/m <sup>3</sup> )	2.8 ppm (8.1 mg/m <sup>3</sup> )	2.0 ppm (5.8 mg/m <sup>3</sup> )	1.0 ppm (2.9 mg/m <sup>3</sup> )	0.71 ppm (2.0 mg/m <sup>3</sup> )
AEGL-3 (Lethal)	50 ppm (145 mg/m <sup>3</sup> )	28 ppm (81 mg/m <sup>3</sup> )	20 ppm (58 mg/m <sup>3</sup> )	10 ppm (29 mg/m <sup>3</sup> )	7.1 ppm (21 mg/m <sup>3</sup> )

Abbreviations: mg/m<sup>3</sup>, milligrams per cubic meter; ppm, parts per million.

**TABLE 1-10** Extant Standards and Guidelines for Chlorine (ppm)

Guideline	Exposure Duration				
	10 min	30 min	1 h	4 h	8 h
AEGL-1	0.5	0.5	0.5	0.5	0.5
AEGL-2	2.8	2.8	2.0	1.0	0.71
AEGL-3	50	28	20	10	7.1
ERPG-1 (AIHA) <sup>a</sup>			1		
ERPG-2 (AIHA)			3		
ERPG-3 (AIHA)			20		
EEGL (NRC) <sup>b</sup>					3
PEL-Ceiling (OSHA) <sup>c</sup>					1
IDLH (NIOSH) <sup>d</sup>		10			
REL-Ceiling (NIOSH) <sup>e</sup>					0.5
TLV-TWA (ACGIH) <sup>f</sup>					0.5
TLV-STEL (ACGIH) <sup>g</sup>					1.0
MAK (Germany) <sup>h</sup>					0.5
MAK - Peak Limit (Germany) <sup>i</sup>					1.0
MAC - Ceiling (The Netherlands) <sup>j</sup>					1.0

<sup>a</sup>ERPG (emergency response planning guidelines) of the American Industrial Hygiene Association (AIHA 2001). ERPG-1 is the maximum airborne concentration below which it is believed nearly all individuals could be exposed for up to 1 h without experiencing symptoms other than mild, transient adverse health effects or without perceiving a clearly defined objectionable odor. The ERPG-1 for chlorine is based on the transient effects on pulmonary function parameters observed during exposure at 1.0 ppm in the studies of Anglen (1981) and Rotman et al. (1983). The ERPG-2 is the maximum airborne concentration below which it is believed nearly all individuals could be exposed for up to 1 h without experiencing or developing irreversible or other serious health effects or symptoms that could impair an individual's ability to take protection action. The ERPG-2 for chlorine is based on the

human data and the subchronic and chronic animal studies. The ERPG-3 is the maximum airborne concentration below which it is believed nearly all individuals could be exposed for up to 1 h without experiencing or developing life-threatening health effects. The ERPG-3 for chlorine is based on scientific judgment; the studies of Schlagbauer and Henschler (1967, as cited in AIHA 1988) and Withers and Lees (1985) are mentioned.

<sup>b</sup>EEGL (emergency exposure guidance level) (NRC 1985). The EEGL is the concentration of a contaminant that can cause discomfort or other evidence of irritation or intoxication in or around the workplace, but avoids death, other severe acute effects, and long-term or chronic injury. EEGLs were developed for healthy military personnel. The EEL (emergency exposure limit) for chlorine was set in 1966 and 1971 on the basis of nose and eye irritation. After review of the Rotman et al. (1983) data, the 8-h EEL was kept at 3 ppm, but the 24-h EEGL was lowered to 0.5 ppm.

<sup>c</sup>OSHA PEL–ceiling (permissible exposure limit–ceiling of the Occupational Safety and Health Administration) (OSHA 1997). The PEL–ceiling should not be exceeded at any time during a work day (instantaneous monitoring or 15-min TWA).

<sup>d</sup>IDLH (immediately dangerous to life and health standard of the National Institute of Occupational Safety and Health) (NIOSH 1994). The IDLH represents the maximum concentration from which one could escape within 30 min without any escape-impairing symptoms or irreversible health effects. The IDLH for chlorine is based on acute inhalation toxicity data in humans, specifically Freitag (1941, as cited in NIOSH 1994a) and several secondary sources.

<sup>e</sup>NIOSH REL–ceiling (recommended exposure limits–ceiling) (NIOSH 1997). The REL–ceiling should not be exceeded at any time during a work day.

<sup>f</sup>ACGIH TLV–TWA (Threshold Limit Value–time-weighted average of the American Conference of Governmental Industrial Hygienists) (ACGIH 2000). The time-weighted average concentration for an 8-h work day and a 40-h work week to which nearly all workers may be repeatedly exposed, day after day, without adverse effects. The TWA is based on the absence of pulmonary effects at a concentration of chlorine at 0.5 ppm (from Rotman et al. 1983).

<sup>g</sup>ACGIH TLV–STEL (Threshold Limit Value–short-term exposure limit) (ACGIH 2001). A 15-min TWA exposure that should not be exceeded at any time during the work day even if the 8-h TWA is within the TLV–TWA. Exposures above the TLV–TWA up to the STEL should not be longer than 15 min and should not occur more than 4 times per day. There should be at least 60 min between successive exposures in this range. The STEL is based on significant changes in pulmonary function parameters during exposure at 1.0 ppm observed in the Rotman et al. (1983) study.

<sup>h</sup>MAK (Maximale Arbeitsplatzkonzentration [Maximum Workplace Concentration]) (Deutsche Forschungsgemeinschaft [German Research Association] 2000). The MAK is defined analogous to the ACGIH TLV–TWA.

<sup>i</sup>MAK Spitzenbegrenzung (peak limit) (German Research Association 2001). The peak limit constitutes the maximum average concentration to which workers can be

exposed for a period up to 30 min with no more than two exposure periods per work shift; total exposure may not exceed the 8-h MAK.

<sup>4</sup>MAC (Maximaal Aanvaarde Concentratie [Maximal Accepted Concentration]) (SDU Uitgevers [under the auspices of the Ministry of Social Affairs and Employment], The Hague, The Netherlands 2000). The MAC is defined analogous to the ACGIH TLV-TWA.

it is specifically protective of the asthmatic population. The ERPG-2 is based on human data and subchronic and chronic animal data (unreferenced) and is only slightly higher than the 1-h AEGL-2. The ERPG-3 is based on a preponderance of the animal data (unreferenced) and is the same as the 1-h AEGL-3.

### 8.3. Data Adequacy and Research Needs

The human database is extensive (Joosting and Verberk 1974; Anglen 1981; Rotman et al. 1983; D'Alessandro et al. 1996; Shusterman et al. 1998) and addresses the healthy population as well as asthmatic and atopic individuals. Both genders were tested during exercise in the Anglen (1981) study, and both sensory irritation and pulmonary function parameters were measured. Rotman et al. (1983) measured a range of pulmonary function parameters and included a susceptible individual. D'Alessandro et al. (1996) tested subjects with airway hyper-reactivity or asthma. In addition, in the study by Anglen (1981), exposure concentrations were measured by several different methods, all of which gave similar results. Exposure concentrations were similarly measured in the Rotman et al. (1983) study. Animal studies involved single and multiple species acute tests with multiple dosing regimens and indicated a clear dose-response relationship. Longer-term animal studies that can be used to support the safety of acute exposures were also available. At the higher concentrations, tissue and organ pathology indicated the same toxic mechanism across species.

The mammalian lethality database is extensive and includes four species and exposure times of 5 min to 16 h, as well as chronic studies. However, some of the studies were judged to suffer from methodology and concentration-analysis shortcomings. The mouse appeared to be more sensitive to chlorine than the rat, and in several studies, lethality may have been due to bronchopneumonia, a treatable effect in humans. When developing emergency exposure guidance levels (EEGLs) for hydrogen chloride,

also an irritant chemical, the NRC did not consider the mouse an appropriate model for extrapolations to humans. Therefore, basing the AEGL-3 levels on a concentration higher than the highest nonlethal level in mice but lower than the highest nonlethal concentration in rats is considered appropriate.

## 9. REFERENCES

- Abdel-Rahman, M.S., M.R. Berardi, and R.J. Bull. 1982. Effect of chlorine and monochloramine in drinking water on the developing rat fetus. *J. Appl. Toxicol.* 2:156-159.
- Abhyankar, A., N. Bhambure, N.N. Kamath, S.P. Pajankar, S.T. Nabar, A. Shrenivas, A.C. Shah, and S.N. Deshmukh. 1989. Six month follow-up of fourteen victims with short-term exposure to chlorine gas. *J. Soc. Occup. Med.* 39:131-132.
- ACGIH (American Conference of Governmental Industrial Hygienists). 2001. Documentation of the Threshold Limit Values and Biological Exposure Indices: Chlorine, 6th Ed. Cincinnati, OH: ACGIH.
- ACGIH (American Conference of Governmental Industrial Hygienists). 2001. Threshold Limit Values (TLVs) for Chemical and Physical Agents and Biological Exposure Indices (BEIs). Cincinnati, OH: ACGIH.
- AIHA (American Industrial Hygiene Association). 1988. Emergency Response Planning Guidelines, Chlorine. Akron, OH: AIHA.
- Alarie, Y. 1980. Toxicological evaluation of airborne chemical irritants and allergens using respiratory reflex reactions. Pp. 207-231 in *Proceedings, Symposium on Inhalation Toxicology and Technology*. Ann Arbor, MI: Ann Arbor Science.
- Alarie, Y. 1981. Dose-response analysis in animal studies: Prediction of human responses. *Environ. Health Perspect.* 42:9-13.
- Alberts, W.M., and G.A. do Pico. 1996. Reactive airways dysfunction syndrome. *Chest* 109:1618-1626.
- Amoore, J.E., and E. Hautala. 1983. Odor as an aid to chemical safety: Odor thresholds compared with Threshold Limit Values and volatilities for 214 industrial chemicals in air and water dilution. *J. Appl. Toxicol.* 3:272-290.
- Anglen, D.M. 1981. Sensory response of human subjects to chlorine in air. Ph.D. Dissertation, University of Michigan.
- Arlong, F., E. Berthet, and J. Viallier. 1940. Action of chronic intoxication by low concentration chlorine fumes on experimental guinea pigs. *Presse Med.* 48:361.
- ASTM (American Society for Testing and Materials). 1991. Standard test method for estimating sensory irritancy of airborne chemicals. In *Annual Book of ASTM Standards*, Vol. 11.04. Philadelphia, PA: ASTM.

- ATSDR (Agency for Toxic Substances and Disease Registry). 1993. Fluorides, Hydrogen Fluoride, and Fluorine (F). U.S. Department of Health and Human Services, Public Health Service, Atlanta, GA.
- Avital, A., N. Noviski, E. Bar-Yishay, C. Springer, M. Levy, and S. Godfrey. 1991. Nonspecific bronchial reactivity in asthmatic children depends on severity but not on age. *Am. Rev. Respir. Dis.* 144:36-38.
- Back, K.C., A.A. Thomas, and J.D. MacEwen. 1972. Reclassification of Materials Listed as Transportation Health Hazards. Report No. TSA-20-72-3. Aerospace Medical Research Laboratory, Wright-Patterson AFB, Dayton, OH.
- Barrow, C.S., and W.H. Steinhagen. 1982. Sensory irritation tolerance development to chlorine in F-344 rats following repeated inhalation. *Toxicol. Appl. Pharmacol.* 65:383-9.
- Barrow, C.S., Y. Alarie, J.C. Warrick, and M.F. Stock. 1977. Comparison of the sensory irritation response in mice to chlorine and hydrogen chloride. *Arch. Environ. Health* 32:68-76.
- Barrow, C.S., R.J. Kociba, L.W. Rampy, D.G. Keyes, and R.R. Albee. 1979. An inhalation toxicity study of chlorine in Fischer-344 rats following 30 days of exposure. *Toxicol. Appl. Pharmacol.* 49:77-88.
- Barrow, R.E., and R.G. Smith. 1975. Chlorine induced pulmonary function changes in rabbits. *Am. Ind. Hyg. Assoc. J.* 36:398-403.
- Bell, D.P., and P.C. Elmes. 1965. The effects of chlorine gas on the lungs of rats without spontaneous pulmonary disease. *J. Pathol. Bacteriol.* 89:307-317.
- Bitron, M.D., and E.F. Aharonson. 1978. Delayed mortality of mice following inhalation of acute doses of formaldehyde, sulfur dioxide, chlorine and bromine. *Am. Ind. Hyg. Assoc. J.* 39:129-138.
- Buckley, L.A., X.Z. Jiang, R.A. James, K.T. Morgan, and C.S. Barrow. 1984. Respiratory tract lesions induced by sensory irritants and the RD<sub>50</sub> concentration. *Toxicol. Appl. Pharmacol.* 74:417-429.
- Budavari, S., M.J. O'Neil, A. Smith, P.E. Heckelman, and J.F. Kinneary, eds. 1996. *The Merck Index*, 12th Ed. Rahway, NJ: Merck & Co., Inc.
- C&EN (Chemical & Engineering News). 1994. Chlorine industry running flat out despite persistent health fears. 72:13.
- Capodaglio, E., G. Pezzagno, J.C. Bobbio, and F. Cazzoli. 1970. Indagine sulla funzionalità respiratoria di lavoratori addetti a produzione elettrolitica di cloro e soda [in Italian]. *Med. Lav.* 60:192-202.
- Carlton, B.D., P. Bartlett, A. Basaran, K. Colling, I. Osis, and M.K. Smith. 1986. Reproductive effects of alternative disinfectants. *Environ. Health Perspect.* 69:237-241.
- CEH (Chemical Economics Handbook). 2000. Chlorine/sodium hydroxide [Online]. Available: <http://ceh.sric.sri.com/Public/Reports/733.1000/> [October 2001].
- Chang, J.C.F., and C.S. Barrow. 1984. Sensory tolerance and cross-tolerance in F-344 rats exposed to chlorine or formaldehyde gas. *Toxicol. Appl. Pharmacol.* 76:319-327.

- Charan, N.B., S. Lakshminarayan, G.C. Myers, and D.D. Smith. 1985. Effects of accidental chlorine inhalation on pulmonary function. *West. J. Med.* 143:333-336.
- CIIT (Chemical Industry Institute of Toxicology). 1993. A Chronic Inhalation Toxicity Study of Chlorine in Female and Male B6C3F<sub>1</sub> Mice and Fischer 344 Rats. Research Triangle Park, NC: CIIT.
- Cockcroft, D.W., D.N. Killian, J.J.A. Mellon, and F.E. Hargreave. 1977. Bronchial reactivity to inhaled histamine: A method and clinical survey. *Clin. Allergy* 7:235-243.
- D'Alessandro, A., W. Kuschner, H. Wong, H.A. Boushey, and P.D. Blanc. 1996. Exaggerated responses to chlorine inhalation among persons with nonspecific airway hyperreactivity. *Chest* 109:331-337.
- Decker, W.J. 1998. Chlorine poisoning at the swimming pool revisited: Anatomy of two minidisasters. *Vet. Hum. Toxicol.* 30:584-585.
- Demnati, R., R. Fraser, G. Plaa, and J.L. Malo. 1995. Histopathological effects of acute exposure to chlorine gas on Sprague-Dawley rat lungs. *J. Environ. Pathol. Toxicol. Oncol.* 14:15-19.
- Drobnic, F., A. Freixa, P. Casan, J. Sanchis, and X. Guardino. 1996. Assessment of chlorine exposure in swimmers during training. *Med. Sci. Sports Exerc.* 28:271-274.
- Druckrey, H. 1968. Chlorinated drinking water toxicity tests involving seven generations of rats. *Food Cosmet. Toxicol.* 6:147-154.
- DuBois, A.B., S.Y. Botelho, and J.H. Comroe. 1971. A new method for measuring airway resistance in man using a body plethysmograph: Values in normal subjects and in patients with respiratory disease. *J. Clin. Invest.* 35:327-335.
- Eaton, D.L., and C.D. Klaassen. 1996. Principles of toxicology. In Casarett and Doull's *Toxicology: The Basic Science of Poisons*. New York, NY: McGraw Hill.
- Elmes, P.C., and D.P. Bell. 1963. The effects of chlorine gas on the lungs of rats with spontaneous pulmonary disease. *J. Pathol. Bacteriol.* 86:317-326.
- EPA/FEMA/DOT (U.S. Environmental Protection Agency/Federal Emergency Management Association/U.S. Department of Transportation). 1987. Technical Guidance for Hazards Analysis: Emergency Planning for Extremely Hazardous Substances. EPA-OSWER-88-0001. U.S. Environmental Protection Agency, Washington, DC.
- EPA (U.S. Environmental Protection Agency). 1994. Methods for Derivation of Inhalation Reference Concentrations and Application of Inhalation Dosimetry. EPA/600/8-90/066F. Environmental Criteria and Assessment Office, Washington, DC.
- EPA (U.S. Environmental Protection Agency). 1996. Chlorine. Integrated Risk Information System (IRIS) [Online]. Available: <http://www.epa.gov/iris/subst/0405.htm> [September 2, 1996].

- Ferris, B.G., W.A. Burgess, and J. Worchester. 1964. Prevalence of chronic respiratory disease in a pulp mill and a paper mill in the United States. *Br. J. Ind. Med.* 24:26-37.
- Ferris, B.G., S. Puleo, and H.Y. Chen. 1979. Mortality and morbidity in a pulp and a paper mill in the United States: A ten-year follow-up. *Br. J. Ind. Med.* 36:127-134.
- Freitag. 1941. Dangers of chlorine gas [in German]. *Z. Gesamte Schiess-Sprengstoffwesen* 35:159.
- Gagnaire, F., S. Azim, P. Bonnet, G. Hecht, and M. Hery. 1994. Comparison of the sensory irritation response in mice to chlorine and nitrogen trichloride. *J. Appl. Toxicol.* 14:405-409.
- ILO (International Labour Office). 1998. Chlorine and Compounds. In *Encyclopaedia of Occupational Health and Safety*, 4th Ed., Vol. 1 (A-K). Geneva, Switzerland: ILO.
- Jiang, X.Z., L.A. Buckley, and K.T. Morgan. 1983. Pathology of toxic responses to the RD<sub>50</sub> concentration of chlorine gas in the nasal passages of rats and mice. *Toxicol. Appl. Pharmacol.* 71:225-236.
- Jones, R.N., J.M. Hughes, H. Glindmeyer, and H. Weill. 1986. Lung function after acute chlorine exposure. *Am. Rev. Resp. Dis.* 134:1190-1195.
- Joosting, P., and M. Verberk. 1974. Emergency population exposure: A methodological approach. *Recent Adv. Assess. Health Eff. Environ. Pollut.* 4:2005-2029.
- Kaufman, J., and B. Burkons. 1971. Clinical, roentgenologic, and physiologic effects of acute chlorine exposure. *Arch. Environ. Health* 23:29-34.
- Klonne, D.R., C.E. Ulrich, M.G. Riley, T.E. Hamm Jr., K.T. Morgan, and C.S. Barrow. 1987. One-year inhalation toxicity study of chlorine in rhesus monkeys (*Macaca mulatta*). *Fundam. Appl. Toxicol.* 9:557-572.
- Kowitz, T.A., R.C. Reba, R.T. Parker, and W.S. Spicer. 1967. Effects of chlorine gas upon respiratory function. *Arch. Environ. Health* 14:545-558.
- Kusch G.D. 1994. Prospective study of the effects of chronic chlorine exposure in manufacturing on respiratory health. In *Chlorine Plant Operations Seminar and Workshop Proceedings*. Washington, DC: The Chlorine Institute, Inc.
- Larsen, G.L. 1992. Asthma in children. *N. Engl. J. Med.* 326:1540-1545.
- Lemiere, C., J.L. Malo, and D. Gautrin. 1996. Nonsensitizing causes of occupational asthma. *Obstruct. Lung Dis.* 80:749-774.
- Lipton, M.A., and G.J. Rotariu. 1941. In Progress report on toxicity of chlorine gas for mice, E.M.K. Gelling and F.C. McLean, eds. Report No. 286. U.S. National Defense Committee, Office of Science Research and Development.
- Litchfield, J.T., and F. Wilcoxon. 1949. A simplified method of evaluating dose-effect experiments. *J. Pharmacol. Exp. Ther.* 96:99.
- MacEwen, J.D., and E.H. Vernot. 1972. Toxic Hazards Research Unit Annual Technical Report: 1972. AMRL-TR-72-62. Aerospace Medical Research Laboratory, Wright-Patterson Air Force Base, OH; National Technical Information Service, Springfield, VA.

- Matheson Gas Co. 1980. Matheson Gas Data Book, 6th Ed. Lyndhurst, NJ: Division Searle Medical Products USA, Inc.
- Meier, J.R., R.J. Bull, J.A. Stober, and M.C. Cimino. 1985. Evaluation of chemicals used for drinking water disinfection for production of chromosomal damage and sperm-head abnormalities in mice. *Environ. Mut.* 7:201-211.
- Mickey, G., and H. Holden. 1971. Effects of chlorine on mammalian cells in vitro. *EMS Newsletter* 4:39-41.
- Mrvos, R., B.S. Dean, and E.P. Krenzelok. 1993. Home Exposures to chlorine/chloramine gas: Review of 216 cases. *South. Med. J.* 86:654-657.
- NAC (National Advisory Committee). 1996. Acute Exposure Guideline Levels (AEGs) for Fluorine. Interim Report.
- NIOSH (National Institute for Occupational Safety and Health). 1976. Criteria for a recommended standard: Occupational exposure to chlorine. NIOSH Publication 76-170. U.S. Department of Health and Human Services, Washington, DC.
- NIOSH (National Institute for Occupational Safety and Health). 1994a. Documentation for Immediately Dangerous to Life or Health Concentrations (IDLHs). PB94-195047. National Institute for Occupational Safety and Health, Cincinnati, OH; National Technical Information Service, Springfield, VA.
- NIOSH (National Institute for Occupational Safety and Health). 1994b. NIOSH Pocket Guide to Chemical Hazards. NIOSH Publication 94-116. U.S. Department of Health and Human Services, U.S. Government Printing Office, Washington, DC.
- NRC (National Research Council). 1984. Emergency and Continuous Exposure Limits for Selected Airborne Contaminants, Vol. 2. Washington, DC: National Academy Press.
- NRC (National Research Council). 1991. Permissible Exposure Levels and Emergency Exposure Guidance Levels for Selected Airborne Contaminants. Washington, DC: National Academy Press.
- NRC (National Research Council). 1993. Guidelines for Developing Community Emergency Exposure Levels for Hazardous Substances. Washington, DC: National Academy Press.
- NRC (National Research Council). 2001. Standing Operating Procedures for Developing Acute Exposure Guideline Levels for Hazardous Chemicals. Washington, DC: National Academy Press.
- NTIS (National Technical Information Service). 1996. Chlorine. Registry of Toxic Effects of Chemical Substances (RTECS) [Online]. Available: Available: <http://www.ntis.gov/search/product.asp?ABBR=SUB5363&starDB=GRAHIST> [December 1996].
- NTP (National Toxicology Program). 1992. Toxicology and Carcinogenesis of Chlorinated Water (CAS Nos. 7782-50-5 and 7681-52-9) and Chloraminated Water (CAS No. 10599-90-3) (Deionized and Charcoal-Filtered) in F344/N Rats and B6C3F<sub>1</sub> Mice (Drinking Water Studies). NTP TR 392. National Institutes of Health, Bethesda, MD.

- O'Neil, C.E. 1991. Immune responsiveness in chlorine exposed rats. PB92-124478. National Institute for Occupational Safety and Health, Cincinnati, OH.
- OSHA (Occupational Safety and Health Administration). 1989. 29 CFR Part 1910, Air Contaminants: Final Rule. Fed. Regist. 54(12):2455-2456 (Thursday, January 19, 1989).
- Patil, L.R.S., R.G. Smith, A.J. Vorwald, and T.F. Mooney. 1970. The health of diaphragm cell workers exposed to chlorine. *Am. Ind. Hyg. Assoc. J.* 31:678-686.
- Perry, W.G., F.A. Smith, and M.B. Kent. 1994. The halogens. Pp. 4482-4505 in *Patty's Industrial Hygiene and Toxicology*, Vol. 2, Part F, G.F. Clayton and F.E. Clayton, eds. New York, NY: John Wiley & Sons, Inc.
- Prentiss, A.M. 1937. *Chemicals in War; a Treatise on Chemical Warfare*. New York: McGraw-Hill Book Company.
- Rafferty, P. 1980. Voluntary chlorine inhalation: A new form of self-abuse? *Br. Med. J.* 281:1178-1179.
- Rothery, S.P. 1991. Hazards of chlorine to asthmatic patients. *Br. J. Gen. Pract.* 41:39.
- Rotman, H.H., M.J. Fliegelman, T. Moore, R.G. Smith, D.M. Anglen, C.J. Kowalski, and J.G. Weg. 1983. Effects of low concentration of chlorine on pulmonary function in humans. *J. Appl. Physiol.* 54:1120-1124.
- Rupp, H., and D. Henschler. 1967. Effects of low chlorine and bromine concentrations on man. *Int. Arch. Gewerbepathol.* 23:79-90.
- Schlagbauer, M., and D. Henschler. 1967. Toxicity of chlorine and bromine with single and repeated inhalation. *Int. Arch. Gewerbepath. Gewerbehyg.* 23:91.
- Shroff, C.P., M.V. Khade, and M. Srinivasan. 1988. Respiratory cytopathology in chlorine gas toxicity: A study in 28 subjects. *Diagn. Cytopathol.* 4:28-32.
- Shusterman, D.J., M.A. Murphy, and J.R. Balmes. 1998. Subjects with seasonal allergic rhinitis and nonrhinitic subjects react differentially to nasal provocation with chlorine gas. *J. Allergy Clin. Immunol.* 101:732-740.
- Silver, S.D., F.P. McGrath, and R.L. Ferguson. 1942. Chlorine median lethal concentration data for mice. DATR 373. Edgewood Arsenal, MD.
- ten Berge, W.F., and M. Vis van Heemst. 1983. Validity and accuracy of a commonly used toxicity-assessment model in risk analysis. *ICHEME Symposium Series No. 80*:I1-I12.
- ten Berge, W.F., A. Zwart, and L.M. Appleman. 1986. Concentration-time mortality response relationship of irritant and systemically acting vapors and gases. *J. Hazard. Mater.* 13:301-310.
- Underhill, F.P. 1920. *The lethal War Gases: Physiology and Experimental Treatment*. New Haven, CT: Yale University Press. Pp. 20.
- Vernot, E.H., J.D. MacEwen, C.C. Haun, and E.R. Kinkead. 1977. Acute toxicity and skin corrosion data for some organic and inorganic compounds and aqueous solutions. *Toxicol. Appl. Pharmacol.* 42:417-423.

- Weedon, F.R., A. Hartzell, and C. Setterstrom. 1940. Toxicity of ammonia, chlorine, hydrogen cyanide, hydrogen sulfide and sulfur dioxide gases. V. Animals. *Contrib. Boyce Thompson Inst.* 11:365-385.
- Weill, H., R. George, M. Schwarz, and M. Ziskind. 1969. Late evaluation of pulmonary function after acute exposure to chlorine gas. *Amer. Rev. Resp. Dis.* 99:374-379.
- Withers, R.M.J., and F.P. Lees. 1985a. The assessment of major hazards: The lethal toxicity of chlorine, Part 1: Review of information on toxicity. *J. Hazard. Mater.* 12:231-282.
- Withers, R.M.J., and F.P. Lees. 1985b. The assessment of major hazards: The lethal toxicity of chlorine., Part 2: model of toxicity to man. *J. Hazard. Mater.* 12:283-302.
- Withers, R.M.J., and F.P. Lees. 1987. The assessment of major hazards: The lethal toxicity of chlorine, Part 3: Crosschecks from gas warfare. *J. Hazard. Mater.* 15:301-342.
- Witschi, H.R., and J.A. Last. 1996. Toxic responses of the respiratory system. In *Casarett and Doull's Toxicology: The Basic Science of Poisons*. New York, NY: McGraw-Hill.
- Wohlslagel, J., L.C. DiPasquale, and E.H. Vernot. 1976. Toxicity of solid rocket motor exhaust: Effects of HCl, HF, and alumina on rodents. *J. Combust. Toxicol.* 3:61-69.
- Wolf, D.C., K.T. Morgan, E.A. Gross, C. Barrow, O.R. Moss, R.A. James, and J.A. Popp. 1995. Two-year inhalation exposure of female and male B6C3F1 mice and F344 rats to chlorine gas induces lesions confined to the nose. *Fundam. Appl. Toxicol.* 24:111-131.
- Wood, B.R., J.L. Colombo, and B.E. Benson. 1987. Chlorine inhalation toxicity from vapors generated by swimming pool chlorinator tablets. *Pediatrics* 79:427-430.
- Zwart, A., and R.A. Woutersen. 1988. Acute inhalation toxicity of chlorine in rats and mice: Time-concentration-mortality relationships and effects on respiration. *J. Hazard. Mater.* 19:195-208.

**APPENDIX A****Derivation of Chlorine AEGLs****Derivation of AEGL-1**

Key studies:	Rotman et al. 1983; D'Alessandro et al. 1996; Shusterman et al. 1998
Toxicity end point:	Transient pulmonary function changes in atopic individual exposed at 0.5 ppm for an interrupted 8 h; non-significant changes in pulmonary peak air flow in eight atopic individuals exposed at 0.5 ppm for 15 min; no statistically significant pulmonary parameter changes in asthmatic subjects exposed at 0.4 ppm for 1 h
Time-scaling:	No time scaling; because there is adaptation to the slight irritation that defines the AEGL-1 end point, the same value (0.5 ppm) was used across all time points
Uncertainty factors:	1, because susceptible individuals were tested and one of the susceptible individuals was exercising, making him more susceptible to sensory irritation (no-effect level in healthy exercising individuals of both genders)
Calculations:	Because the 0.5 ppm concentration was indicative of a NOAEL for more serious pulmonary changes, the 0.5 ppm concentration was used for all exposure durations. The susceptible individual underwent an interrupted 8-h exposure at 0.5 ppm without increased symptoms, so that concentration was also used for the 8-h AEGL-1

**Derivation of AEGL-2**

Key studies:	Rotman et al. 1983; D'Alessandro et al. 1996
Toxicity end point:	No-effect concentration for serious health effect (asthma-like attack) in a sensitive, exercising individual exposed at 1 ppm for 4 h and in individuals with airway hyper-reactivity (including 3 asthmatic individuals) exposed at 1 ppm for 1 h
Time-scaling:	$C^2 \times t = k$ (ten Berge and Vis van Heemst 1983) $(1 \text{ ppm})^2 \times 240 \text{ min} = 240 \text{ ppm}^2 \cdot \text{min}$
Uncertainty factors:	1. The value was based on effects consistent with the AEGL-2 definition in a susceptible, exercising individual and in asthmatics subjects
<i>30-min AEGL-2:</i>	$C^2 \times 30 \text{ min} = 240 \text{ ppm}^2 \cdot \text{min}$ $C = 2.8 \text{ ppm}$
<i>1-h AEGL-2:</i>	$C^2 \times 60 \text{ minutes} = 240 \text{ ppm}^2 \cdot \text{min}$ $C = 2 \text{ ppm}$
<i>4-h AEGL-2:</i>	1 ppm for 4 h; basis for derivation of other exposure durations
<i>8-h AEGL-2:</i>	$C^2 \times 480 \text{ min} = 240 \text{ ppm}^2 \cdot \text{min}$ $C = 0.71 \text{ ppm}$

The 10-min AEGL-2 was set equal to the 30-min AEGL-2 so that the highest human test concentration of 4.0 ppm was not exceeded.

**Derivation of AEGL-3**

Key studies:	Zwart and Woutersen 1988; MacEwen and Vernot 1972
--------------	---

Toxicity end point:	1-h lethality value; an end point below the highest concentration resulting in no deaths in the rat and above the highest concentration resulting in no deaths in the mouse was chosen because the mouse was shown to be more sensitive than other mammals to irritant gases, including chlorine, and does not provide an appropriate basis for quantitatively predicting mortality in humans
Time-scaling:	$C^2 \times t = k$ (ten Berge and Vis van Heemst 1983) $(200 \text{ ppm}/10)^2 \times 60 \text{ min} = 24,000 \text{ ppm}^2 \cdot \text{min}$
Uncertainty factors:	Combined uncertainty factor of 10  3 for interspecies variability (interspecies values for the same end point differed by a factor of approximately 2 in several studies)  3 for differences in human sensitivity (the toxic effect is the result of a chemical reaction with biologic tissue of the respiratory tract, which is unlikely to differ among individuals)
<i>10-min AEGL-3:</i>	$C^2 \times 10 \text{ min} = 24,000 \text{ ppm}^2 \cdot \text{min}$ $C = 50 \text{ ppm}$
<i>30-min AEGL-3:</i>	$C^2 \times 30 \text{ min} = 24,000 \text{ ppm}^2 \cdot \text{min}$ $C = 28.3 \text{ ppm}$
<i>1-h AEGL-3:</i>	$200 \text{ ppm}/10 = 20 \text{ ppm}$
<i>4-h AEGL-3:</i>	$C^2 \times 240 \text{ min} = 24,000 \text{ ppm}^2 \cdot \text{min}$ $C = 10 \text{ ppm}$
<i>8-h AEGL-3:</i>	$C^2 \times 480 \text{ min} = 24,000 \text{ ppm}^2 \cdot \text{min}$ $C = 7.1 \text{ ppm}$

## APPENDIX B

ACUTE EXPOSURE GUIDELINE LEVELS FOR  
CHLORINE (CAS No. 7782-50-5)

## DERIVATION SUMMARY

AEGL-1				
10 min	30 min	1 h	4 h	8 h
0.5 ppm	0.5 ppm	0.5 ppm	0.50 ppm	0.50 ppm
<p>Key references: (1) Rotman, H.H., M.J. Fliegelman, T. Moore, R.G. Smith, D.M. Anglen, C.J. Kowalski, and J.G. Weg. 1983. Effects of low concentrations of chlorine on pulmonary function in humans. <i>J. Appl. Physiol.</i> 54:1120-1124.</p> <p>(2) Shusterman, D.J., M.A. Murphy, and J.R. Balmes. 1998. Subjects with seasonal allergic rhinitis and nonrhinitic subjects react differentially to nasal provocation with chlorine gas. <i>J. Allergy Clin. Immunol.</i> 101:732-740.</p> <p>(3) D'Alessandro, A., W. Kuschner, H. Wong, H.A. Boushey, and P.D. Blanc. 1996. Exaggerated responses to chlorine inhalation among persons with nonspecific airway hyperreactivity. <i>Chest</i> 109:331-337.</p>				
<p>Test species/strain/number: Eight male subjects, one atopic subject (Rotman et al. 1983); eight atopic subjects and eight nonatopic subjects (Shusterman et al. 1998); five asthmatic subjects and five nonasthmatic subjects (D'Alessandro et al. 1996).</p>				
<p>Exposure route/concentrations/durations:</p> <p>Inhalation; 0.0, 0.5, 1.0 ppm for 8 h; break at 4 h for an unreported period of time to undergo pulmonary function tests followed by chamber reentry; subjects exercised for 15 min of every hour during exposures; sham exposures were included (Rotman et al. 1983)</p> <p>Inhalation; 0.0 ppm or 0.5 ppm for 15 min (Shusterman et al. 1998)</p> <p>Inhalation; 0.4 ppm or 1.0 ppm for 1 h (D'Alessandro et al. 1996)</p>				
<p>Effects:</p> <p>0.5 ppm for 4 h—no effects in eight of nine subjects; transient changes in pulmonary functions in one of nine subjects.</p> <p>1.0 ppm for 4 h—some irritation, transient changes in pulmonary functions in nine subjects including an atopic individual; asthma-like episode in one of nine subjects when exposure duration extended to more than 4 h (Rotman et al. 1983).</p>				

<b>AEGL-1</b> <i>Continued</i>
0.5 ppm for 15 min—nasal congestion; nonsignificant changes in pulmonary peak flow (Shusterman et al. 1998). 0.4 ppm for 1 h—no statistically significant pulmonary function effect in asthmatic individuals (D’Alessandro et al. 1996).
End point/concentration/rationale: 0.5 ppm for 4 h resulted in no effects in healthy human subjects and transient changes in pulmonary functions for a susceptible individual who had obstructive airways disease prior to the exposure. The 0.5-ppm concentration was chosen as the basis for the AEGL-1 because the next highest concentration produced effects consistent with an AEGL-2 (coughing, wheezing, and a considerable increase in airways resistance) in a susceptible individual. Supported by studies of Shusterman et al. (1998) and D’Alessandro et al. (1996).
Uncertainty factors/rationale: Total uncertainty factor: 1 Interspecies: Not applicable; human subjects tested. Intraspecies: 1. An atopic individual who had obstructive airways disease prior to the exposure and was considered characteristic of the “susceptible” population was tested. This individual was did not exhibit adverse effects. The choice of an intraspecies uncertainty factor of 1 is supported by another study in which a concentration of 0.4 ppm for 1 h was a no-effect concentration for changes in pulmonary function parameters in individuals with airway hyper-reactivity/asthma and by a study in asthmatic subjects exposed at 0.4 ppm
Modifying factor: Not applicable
Animal to human dosimetric adjustment: Not applicable; human data used
Time-scaling: Not applied; because 0.5 ppm appeared to be the threshold for more severe changes in pulmonary parameters in the atopic individual regardless of exposure duration, the 0.5 ppm was used for all AEGL-1 exposure durations.
Data adequacy: The Angelen (1981) study was well conducted and documented and reinforces a study conducted earlier at the same facilities in which 31 male and female subjects were tested for sensory irritation. The Rotman et al. (1983) study went into greater detail than the earlier study, measuring 15 pulmonary function parameters before, during, and after exposures. Subjects were exercising during exposures and this study included a susceptible individual. The choice of intraspecies uncertainty factor was supported by a study of shorter duration with asthmatics.

AEGL-2				
10 min	30 min	1 h	4 h	8 h
2.8 ppm	2.8 ppm	2.0 ppm	1.0 ppm	0.71 ppm
<p>Key references: (1) Rotman, H.H., M.J. Fliegelman, T. Moore, R.G. Smith, D.M. Anglen, C.J. Kowalski, and J.G. Weg. 1983. Effects of low concentrations of chlorine on pulmonary function in humans. <i>J. Appl. Physiol.</i> 54:1120-1124.</p> <p>(2) D'Alessandro, A., W. Kuschner, H. Wong, H.A. Boushey, and P.D. Blanc. 1996. Exaggerated responses to chlorine inhalation among persons with nonspecific airway hyperreactivity. <i>Chest</i> 109:331-337.</p>				
<p>Test species/strain/gender/number: Nine human male subjects, including atopic individual (Rotman et al. 1983); 10 human subjects of which five had airway reactivity/asthma (D'Alessandro et al. 1996)</p>				
<p>Exposure route/concentration/duration:            Inhalation; 0.0, 0.5, 1.0 ppm for 8 h; break at 4 h for an unreported time period to undergo pulmonary function tests followed by chamber reentry; subjects exercised for 15 min of every hour during exposures; sham exposures were included (Rotman et al. 1983)            Inhalation; 0.4 or 1.0 ppm for 1 h (D'Alessandro et al. 1996)</p>				
<p>Effects:            0.5 ppm for 4 h—no effects in eight healthy subjects; transient changes in pulmonary functions in one of nine subjects.            1.0 ppm for 4 h—some irritation, transient changes in pulmonary functions nine subjects including an atopic individual; asthma-like episode in one of nine subjects when exposure duration extended beyond 4 h.            1.0 ppm for 1 h—increased airway resistance in asthmatic individuals (D'Alessandro et al. 1996).</p>				
<p>End point/concentration/rationale: 1 ppm for 4 h was a no-effect exposure for serious health symptoms in an atopic exercising individual, and 1 ppm for 1 h was a symptomless effect on airway resistance in asthmatic individuals. However, the increase in airways resistance was considered the NOAEL for an AEGL-2 effect.</p>				
<p>Uncertainty factors/rationale:            Total uncertainty factor: 1            Interspecies: Not applicable; human subjects tested.            Intraspecies: 1. A susceptible exercising individual who had obstructive airways disease prior to the exposure and was considered characteristic of the “susceptible” population was tested. The application of an intraspecies uncertainty factor of 1 is supported by another study in which individuals with airway hyperreactivity/asthma</p>				

<b>AEGL-2 Continued</b>
showed similar pulmonary function changes and some clinical symptoms but no asthma-like attack following exposure at 1.0 ppm for 1 h.
Modifying factor: Not applicable
Animal to human dosimetric adjustment: Not applicable; human data used
Time-scaling: $C^n \times t = k$ where $n = 2$ . This value describes the concentration-exposure duration relationship for the end point of nuisance irritation (ten Berge and Vis van Heemst 1983, IChemE Symposium Series No. 80:17-21).
Data adequacy: The Angelen (1981) study was well conducted and documented and reinforces a study conducted earlier at the same facilities in which 31 male and female subjects were tested for sensory irritation. The Rotman et al. (1983) study went into greater detail than the earlier study, measuring 15 pulmonary function parameters before, during, and after exposures. Subjects were exercising during exposures, and a susceptible individual was included. The choice of intraspecies uncertainty factor was supported by a study of shorter duration with asthmatic subjects (D'Alessandro et al. 1996).

AEGL-3				
10 min	30 min	1 h	4 h	8 h
50 ppm	28 ppm	20 ppm	10 ppm	7.1 ppm
Key references: (1) MacEwen, J.D. and E.H. Vernot. 1972, Toxic Hazards Research Unit Annual Technical Report. 1972. Wright-Patterson Air Force Base, Dayton, OH; (2) Zwart, A. and Woutersen. 1988. Acute inhalation toxicity of chlorine in rats and mice: time-concentration mortality relationships and effects on respiration. J. Hazard. Mater. 19:195-208; (3) O'Neil, C.E. 1991. Immune responsiveness in chlorine exposed mice. PB92-124478, Prepared for NIOSH, Cincinnati, OH.				
Test species/strain/gender/number: (1) Sprague-Dawley rats, 10/exposure group; (2) Wistar-derived rats, 10/exposure group; (3) BALB/c mice, 10/exposure group				
Exposure route/concentrations/durations: Inhalation; (1) 213-427 ppm for 1 h, (2) 322-595 ppm for 1 h, (3) 50-250 ppm for 1 h				
Effects: (1) no deaths at 213 ppm for 1 h (Sprague-Dawley rat); (2) no deaths at 322 ppm for 1 h (Wistar-derived rat); (3) no deaths at 150 ppm for 1 h (BALB/c mouse)				
End point/concentration/rationale: 200 ppm for 1 h (the estimated mean of highest experimental nonlethal values for the rat and mouse) was chosen as the basis for the 1-h AEGL-3. Mice appeared to be unusually sensitive to chlorine, and in some studies, delayed deaths were attributed to bronchopneumonia rather than direct effects of chlorine.				
Uncertainty factors/rationale: Total uncertainty factor: 10 Interspecies: 3. The mouse and rat LC <sub>50</sub> values did not differ by more than a factor of 2 to 3, and the mouse was consistently more sensitive. In some mouse studies delayed deaths were attributed to bronchopneumonia rather than direct effects of chlorine exposure. Intraspecies: 3. Chlorine is a highly reactive, irritating, and corrosive gas whose effect on respiratory tissues is not expected to differ greatly among individuals.				
Modifying factor: Not applicable				
Animal to human dosimetric adjustment: Not applied				
Time-scaling: $C^n \times t = k$ where $n = 2$ . This value describes the concentration-exposure duration relationship for the end point of nuisance irritation (ten Berge and Vis van Heemst 1983, IChemE Symposium Series				

**AEGL-3** *Continued*

No. 80:17-21). The irritation mechanism of action leads to pulmonary edema and potential lethality. An *n* of 2 is also relevant to animal lethality studies.

Data adequacy: The database for chlorine is extensive with multiple studies of lethality conducted at several exposure durations and involving several species. Studies with multiple dosing regimens showed a clear dose-response relationship. Longer-term studies that support the safety of the values were also available. Tissue and organ pathology indicated that the toxic mechanism was the same across species.