Reports:	ECM: EPA MRID No.: 49687805. Zha for a method for the determination of ca aquatic field dissipation studies. Wildli 118. Mana Study No.: R-35353. Report Evans Analytical Group, Easton, Mary Makhteshim Agan of North America, H Carolina; 44 pages. Final report issued issued June 9, 2015. ILV: EPA MRID No. 49687804. Arndt Laboratory Validation of Method for A PTRL West Project No: 2743W. Spons prepared by PTRL West (a division of sponsored and submitted by Makhteshi ADAMA), Raleigh, North Carolina; 56 2015.	aptan in freshwater for support of fe International Project No.: 234C- t prepared by Wildlife International, land, sponsored and submitted by nc. (d/b/a ADAMA), Raleigh, North May 8, 2015; Report Amendment t, T., L. Mannella. 2015. Independent nalysis of Captan in Surface Water. for Project No.: R-35996. Report EAG, Inc.), Hercules, California, m Agan of North America, Inc. (d/b/a
Document No.:	MRIDs 49687805 & 49687804	
Guideline:	850.6100	
Statements:	ECM: The study was conducted in com OECD Good Laboratory Practice (GLF test and reference substance characteriz testing facility (p. 3 of MRID 49687803 Confidentiality, GLP and Quality Assu 2-4). An Authenticity statement was no ILV: The study was conducted in comp with the exception of the sponsor-provi (p. 3 of MRID 49687804). Signed and and Quality Assurance statements were the authenticity of the study report was	P) standards, with the exception of the cations and storage stability at the 5). Signed and dated No Data rance statements were provided (pp. of provided. bliance with USEPA GLP standards, ided certification of captan standard dated No Data Confidentiality, GLP, e provided (pp. 2-4). A statement of
Classification:	This analytical method is classified as a is not sensitive enough for the lowest to a.i./L) for aquatic organisms. The meth based on acceptable procedures, as defi- was not satisfactory for the linear regre- calibration curves did not adequately be LOQ. A reagent blank was not included including the confirmation ion which we been provided.	bxicological level of concern (13.1 µg od used to calculate LOQ was not ined by 40 CFR Part 136. Linearity ssion in the ECM. The ECM and ILV racket the instrumental response at the d in the ECM. An updated ECM
Reviewer:	Faruque Khan Senior Scientist	Signature: Date: 12-18-2015

Executive Summary

This analytical method, Mana Study No. R-35353, is designed for the quantitative determination of captan in surface water using GC/MS. The method is quantitative for captan at the stated LOQ of 0.100 mg/L. The LOQ is greater than the lowest toxicological level of concern 13.1 μ g/L¹ (acute exposure to fish). The number of trials was not specified, but the reviewer assumed that the method was validated by the ILV after one trial. The ECM utilized lake surface water, while the ILV utilized creek surface water. Waters were sieved prior to processing and fortification. Captan was identified by one ion (*m*/*z* 79) in the ECM and by two ions (*m*/*z* 79 and *m*/*z* 149) in the ILV. Results were acceptable in the ECM and ILV; however, the calibration curves used for quantification were not fully optimized for accuracy. An updated ECM including the confirmation ion which was validated by the ILV could be recommended.

Table	1.	Anal	lytical	Method	Summary
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Analyte(s)	MR	D		PA view Matrix	Method Date (dd/mm/yyyy)	Registrant	Analysis	Limit of Quantitation (LOQ)
by Pesticide	Environmental Chemistry Method	Independent Laboratory Validation	EPA Review					
Captan	49687805	49687804		Water ^{1,2}	08/05/2015 (Original) 09/06/2015 (Amendment 1)	Makhteshim Agan of North America, Inc. (d/b/a ADAMA)	GC/MS	0.100 mg/L

1 In the ECM, the surface water was collected from Tuckahoe Lake, Ridgley, Maryland (p. 11 of MRID 49687805). The water was characterized (pH 7.56; Appendix 4, p. 42).

2 In the ILV, the surface water (pH 7.0) was collected and fully characterized by Agvise Laboratories, Inc., Northwood, North Dakota (pp. 14-15; Appendix C, pp. 50-54 of MRID 49687804). The water was collected from Brandywine Creek, Chadds Ford, Pennsylvania.

I. Principle of the Method

Samples (unspecified volume) of freshwater were sieved (250 μ m) and added to Teflon® centrifuge tubes (tubes pre-rinsed with toluene; pp. 12-13; Figure 1, p. 19 of MRID 49687805). For fortification recovery samples, an aliquot of the water equivalent to the fortification stock volume was removed, then the water samples were fortified using gas-tight syringes or equivalent. Samples were immediately acidified by adding *ca*. one drop of 10% phosphoric acid in water per 10 mL of sample. The samples were extracted with 20.0 mL of toluene (added via volumetric pipette) via mixing on a vortex mixer for *ca*. 5 minutes. Separate toluene layer and discard aqueous layer. Pool the toluene extracts of the matrix blank samples. Recovery samples were diluted, if necessary, using the pooled toluene extracts of the matrix blanks so that instrument response was in the range of the calibration curve. Calibration standards were prepared using the pooled matrix blank toluene extracts and volumetric pipettes, volumetric flasks or equivalent. Final extracts and standards were transferred to autosampler vials for analysis by GC/MS.

¹ <u>http://www.epa.gov/pesticide-science-and-assessing-pesticide-risks/aquatic-life-benchmarks-pesticide-registration</u>

Samples were analyzed for captan by an Agilent Model 5890A GC system with an Agilent 5971A mass selective detector (p. 13; Table 1, p. 17 of MRID 49687805). The following instrumental conditions were employed: DB-5MS UI column (30 m x 250 μ m i.d., 0.25 μ m film thickness), injector temperature 250°C, helium carrier gas (0.7 mL/min.), temperature program 100°C (hold 1 min.) to 320°C (hold 1 min.; 15.0°C/min. rate), and selected ion monitoring (SIM) ionization mode. Injection volume was 2.00 μ L (splitless). Captan was identified and quantified with one ion, *m/z* 79.00. The retention time for captan was *ca*. 11.7 minutes. No further confirmation was performed.

In the ILV, the extraction was performed exactly as above, except that the samples were extracted with toluene using a Wrist-ActionTM shaker, instead of a vortex mixer (pp. 14, 16-18, 21; Figure 1, p. 30 of MRID 49687804). Additionally, the sample volume was specified as 10 mL of water, and the remaining extracts were transferred to amber bottles and stored frozen (<-4°C). The analytical procedure was performed using an Agilent Model 6890 GC system with an Agilent 5973 mass selective detector. All GC/MS conditions were the same as the ECM, except the following: the column was reported as Agilent DB-5 (30 m x 250 μ m i.d., 0.25 μ m film thickness); helium carrier gas flow (1.1 mL/min.); and the MS conditions specified electron impact (EI) in SIM mode. Captan was identified and quantified with two ions, *m*/*z* 79 (quantification) and *m*/*z* 149 (confirmation). The retention time for captan was not considered a significant modification of the ECM since a confirmatory method is not usually required when LC/MS and GC/MS is the primary method.

The LOQ for captan was 0.100 mg/L in the ECM and ILV (pp. 8, 13 of MRID 49687805; pp. 10, 20, 22-23 of MRID 49687804). In the ECM, the instrumental LOD was reported as 0.00989 mg/L. The LOD was not reported in the ILV.

II. Recovery Findings

ECM (MRID 49687805): Mean recoveries and relative standard deviations (RSDs) were within guideline requirements (mean 70-120%; RSD \leq 20%) for analysis of captan in surface water at fortification levels of 0.100 mg/L (LOQ) and 1.00 mg/L (10×LOQ; uncorrected recoveries; Table 2, p. 18). Captan was identified using one ion; no further confirmation was performed (Table 1, p. 17). The surface water matrix was collected from Tuckahoe Lake, Ridgley, Maryland (p. 11). The water was characterized (pH 7.56; Appendix 4, p. 42).

ILV (MRID 49687804): Mean recoveries and RSDs were within guideline requirements for analysis of captan in surface water at fortification levels of 0.1 mg/L (LOQ) and 1 mg/L (10×LOQ; uncorrected recoveries; Table I, p. 27; Appendix D, pp. 55-56). Captan was identified using two ions; recoveries of the quantification ion were higher than those of the confirmation ion at both fortification levels (pp. 10-11). The number of trials was not specified, but the reviewer assumed that the method was validated after one trial. The surface water (pH 7.0) was collected and fully characterized by Agvise Laboratories, Inc., Northwood, North Dakota (pp. 14-15; Appendix C, pp. 50-54). The water was collected from Brandywine Creek, Chadds Ford, Pennsylvania.

Analyte	Fortification Level (mg/L)	Number of Tests	v	Mean Recovery (%)	Standard Deviation (%)	Relative Standard Deviation (%)
Primary ion $(m/z 79)$						
Conton	0.100 (LOQ)	5	85.4-91.0	88.7	2.26	2.55
Captan	1.00	5	87.6-98.7	92.6	4.15	4.48

Table 2. Initial Validation Method Recoveries for Captan in Surface Water (Lake)^{1,2}

Data (uncorrected recovery results; Table 2, p. 18) were obtained from Table 2, p. 18 of MRID 49687805.

1 The surface water was collected from Tuckahoe Lake, Ridgley, Maryland (p. 11). The water was characterized (pH 7.56; Appendix 4, p. 42).

2 Captan was identified and quantified with one ion, *m/z* 79.00 (Table 1, p. 17). No further confirmation was performed.

Table 3. Independent Validation Method Recoveries for Captan in Surface Water (Creek)	$()^{1,2}$
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Analyte	Fortification Level (mg/L)	Number of Tests	Recovery Range (%)	Mean Recovery (%)	Standard Deviation (%)	Relative Standard Deviation (%)	
Primary ion $(m/z 79)$							
Conton	0.100 (LOQ)	5	84-103	95	7	7	
Captan	1.00	5	82-94	88	5	6	
	Secondary ion $(m/z \ 149)$						
Conton	0.100 (LOQ)	5	77-100	89	9	10	
Captan	1.00	5	79-83	81	1	1	

Data (uncorrected recovery results; p. 20; Appendix D, pp. 55-56) were obtained from Table I, p. 27 of MRID 49687804.

1 The surface water (pH 7.0) was collected and fully characterized by Agvise Laboratories, Inc., Northwood, North Dakota (pp. 14-15; Appendix C, pp. 50-54). The water was collected from Brandywine Creek, Chadds Ford, Pennsylvania.

2 Captan was identified and quantified with two ions, m/z 79 (quantification) and m/z 149 (confirmation; p. 18).

III. Method Characteristics

The method LOQ for captan was 0.100 mg/L in the ECM and ILV (pp. 8, 13 of MRID 49687805; pp. 10, 20, 22-23 of MRID 49687804). In the ECM, the LOQ was defined as the lowest fortification level fortified and analyzed during the verification set. Other criteria included the confirmation that interferences in the reagent blank and matrix blank were <30% of the LOQ. The theoretical LOQ was calculated to be 0.0422 mg/L in the ECM; the actual LOQ was calculated to be 0.0659 mg/L. Both values supported the method LOQ. The calculations for each were as follows:

Theoretical LOQ = (lowest calibration standard) \times (dilution factor of the matrix blank samples) \div (purity of test substance);

Actual LOQ = (lowest calibration standard) \div (average signal to noise ratio) \times 10 \times (dilution factor of the matrix blank samples) \div (purity of test substance).

In the ECM, the instrumental LOD was reported as 0.00989 mg/L. The calculation was as follows:

Instrumental LOD = (lowest calibration standard) \div (average signal to noise ratio) $\times 3 \times$ (dilution factor of the matrix blank samples) \div (purity of test substance).

In the ILV, the LOQ was defined as the lowest fortification level yielding acceptable recovery. The LOD was mentioned in the ILV, but no value was reported. No calculations for the LOQ or LOD were reported in the ILV.

		Captan			
Limit of Quantitation (LOQ)		0.100 mg/L			
Limit of Detection (LOD)	ECM	0.00989 mg/L (instrumental)			
	ILV	Not reported			
Linearity (calibration curve r^2 and concentration range)	ECM	$r^2 = 0.993 (m/z 79)$			
and concentration range)	ILV	$r^2 = 0.9992 (m/z 79)$ $r^2 = 0.9964 (m/z 149)$			
	Concentration range	0.02 to 0.2 µg/mL (solvent-based calibrants)			
Repeatable	ECM ¹	Yes at LOQ and $10 \times \text{LOQ}$ (<i>m</i> / <i>z</i> 79). No confirmation ion was monitored. ²			
	ILV ³	Yes at LOQ and $10 \times LOQ$ (<i>m/z</i> 79 and <i>m/z</i> 149).			
Reproducible		Yes at LOQ and 10×LOQ.			
Specific	ECM	Only chromatograms for m/z 79 provided. Matrix interferences were not observed.			
	ILV	Chromatograms for <i>m</i> / <i>z</i> 79 and <i>m</i> / <i>z</i> 149 provided. Matrix interferences were not observed. Multiple non-interfering peaks were observed in the confirmation ion chromatograms (peak heights were >3xs the peak height of the analyte).			

Table 4. Method Characteristics

Data were obtained from pp. 8, 12-13; Table 2, p. 18; Figure 2, p. 20; Figures 5-7, pp. 23-25 of MRID 49687805; pp. 10-11, 20, 22-23; Tables I-II, pp. 27-28; Figures 4-9, pp. 34-39; Appendix 4, pp. 55-56 of MRID 49687804.

1 The surface water was collected from Tuckahoe Lake, Ridgley, Maryland (p. 11 of MRID 49687805). The water was characterized (pH 7.56; Appendix 4, p. 42).

2 A confirmatory method is not usually required when LC/MS and GC/MS is the primary method.

3 The surface water (pH 7.0) was collected and fully characterized by Agvise Laboratories, Inc., Northwood, North Dakota (pp. 14-15; Appendix C, pp. 50-54 of MRID 49687804). The water was collected from Brandywine Creek, Chadds Ford, Pennsylvania.

Linearity is satisfactory when $r^2 \ge 0.995$

IV. Method Deficiencies and Reviewer's Comments

1. The ECM study authors supported their determinations of the LOQ and LOD with calculations; however, these determinations of the LOQ and LOD were not based on scientifically acceptable procedures, as defined by 40 CFR Part 136 (pp. 8, 13 of MRID 49687805). The theoretical LOQ equaled the product of the lowest calibration standard and dilution factor of the matrix blank samples, divided by the purity of test substance. The actual LOQ equaled the product of the lowest calibration standard, which was divided by the average signal to noise ratio, and 10 and the dilution factor of the matrix blank samples, divided LOQ was higher than both the theoretical and actual LOQs. In the ECM, the instrumental LOD was calculated as the lowest calibration standard, divided by the average signal to noise ratio, and rof the matrix blank samples, divided by the purity of test substance. The reported LOQ was higher than both the theoretical and actual LOQs. In the ECM, the instrumental LOD was calculated as the lowest calibration standard, divided by the average signal to noise ratio, multiplied by the product of 3 and the dilution factor of the matrix blank samples, divided by the purity of test substance. In the ILV, the LOQ was reported from the ECM, and the LOD was not reported.

It is preferred that detection limits are not be based on the arbitrarily selected lowest concentration in the spiked samples.

- 2. In the ECM, linearity was not satisfactory for the linear regression of the quantification ion of captan (see above). Linearity is satisfactory when $r^2 \ge 0.995$. Linear regressions were matrix-matched. A confirmation ion was not monitored in the ECM.
- 3. The calibration curves did not adequately bracket the instrumental response at the LOQ. In the ILV, the two lowest standards, 0.02 and 0.05 mg/L generated instrument responses (peak areas) of 3398 (Q) and 517 (C) and 10176 (Q) and 1467 (C), respectively (Figure 3, pp. 32-33 of MRID 49687804). The LOQ peak area was 8596 (Q) and 1116 (C; Figure 8, p. 38). In the ECM, the bracketing of the LOQ by the calibration standards could not be evaluated since the individual peak response data was not provided of the calibration curves or the fortified samples and representative chromatograms were only provided for the highest and lowest calibration standard. For accuracy of calculations at the LOQ based on the calibration curve, at least two calibration standards should generate instrumental responses below and above the instrument response at the LOQ.
- 4. A reagent blank was not included in the ECM.
- 5. No confirmation method was included in the ECM. A confirmation ion was not monitored, and a second confirmatory analytical method was not presented. A confirmatory method is not usually required when LC/MS and GC/MS is the primary method.

In the ILV, captan was identified and quantified with two ions, m/z 79 (quantification) and m/z 149 (confirmation; pp. 17-18, 21 of MRID 49687804). The addition of a second ion for confirmation of the identification of captan was not considered a significant modification of the ECM since a confirmatory method is not usually required when LC/MS and GC/MS is the primary method; however, an updated ECM including the confirmation ion which was validated by the ILV should have been provided.

- 6. The ILV study authors noted that recoveries of the quantification ion were higher than those of the confirmation ion at both fortification levels (pp. 10-11 of MRID 49687804). No reason was proposed to explain this observation. The smaller peak response of the confirmation ion versus the quantification ion was noted in the ILV representative chromatograms (Figures 8-9, pp. 38-39). The GC/MS full scan of captan was included in the ILV (Figure 2, p. 31)
- 7. The ECM reported the reason for the amendment of the original study report: to correct the Sponsor Study number from R-35553 to R-35353 (Appendix 6, p. 44 of MRID 49687805).
- 8. The ILV study author reported that communications between the independent laboratory and the originating laboratory, Wildlife International, were not necessary (p. 25 of MRID 49687804).
- 9. In the ILV, no significant matrix effects were found (p. 23; Table III, p. 29 of MRID 49687804). Matrix effects were determined to be 9-13% for both ions monitored. Non-matrix matched standards were used in the ECM and ILV (p. 12 of MRID 49687805; pp. 15-16; Figures 5-6, pp. 35-36 of MRID 49687804).

10. It was reported in the ILV that the method could be performed within 24 hours or 3 eighthour working days (p. 25 of MRID 49687805).

V. References

- U.S. Environmental Protection Agency. 2012. Ecological Effects Test Guidelines, OCSPP 850.6100, Environmental Chemistry Methods and Associated Independent Laboratory Validation. Office of Chemical Safety and Pollution Prevention, Washington, DC. EPA 712-C-001.
- 40 CFR Part 136. Appendix B. Definition and Procedure for the Determination of the Method Detection Limit-Revision 1.11, pp. 317-319.

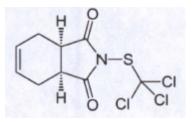
Attachment 1: Chemical Names and Structures

Captan (Captan Tec)

IUPAC Name:

SMILES String:

CAS Name: CAS Number: (3aR,7aS)-2-[(trichloromethyl)sulfanyl]-3a,4,7,7a-tetrahydro-1H-isoindole-1,3(2H)-dione Not reported 133-06-2 Not found



Information related to Primary Review

For CDM Smith

Primary Reviewer: Lisa Muto

Signature:

Lesa Muto

Date: 11/10/15

Secondary Reviewer: Kathleen Ferguson

Signature:

Date: 11/10/15

QC/QA Manager: Joan Gaidos

Signature:

Date: 11/10/15