# **INTRODUCTION**

The objective of this study was to independently validate the analytical method used in Smithers Viscient, Wareham Study No. 14113.6131 (Validation of an Environmental Chemistry Method for the Determination of Cycloate, Cycloate Sulfoxide, and N-ethylcyclohexylamine in Soil by LC-MS/MS). The analytical method was validated with regards to accuracy and precision, linearity, specificity and limit of quantification.

This study was conducted to support the registration of the test substances.

The method validation described in this report is designed to conform to SANCO 3029/99 rev 4 (2000) Guidance for generating and reporting methods of analysis in support of pre-registration data requirements for Annex II (part A, Section 4) and Annex III (part A, Section 5) of Directive 91/414, OCSPP 850.6100: Environmental Chemistry Methods and Associated Independent Laboratory Validation.

Analytical method 14113.6131 was supplied by Smithers Viscient, Wareham on behalf of the sponsor. The method was re-written in Smithers Viscient, Harrogate format as draft method SMV 3202259-01D, including the instrumentation available at Smithers Viscient, Harrogate. This was followed for method validation, and re-issued as SMV 3202259-01V when validation was complete.

The supplied method was followed with only minor changes (equipment and reagents were substituted for suitable equivalents, shaker speed was limited to 200 rpm rather than 300/250 rpm, Cycloate extracts were evaporated to 25-100  $\mu$ L rather than 25  $\mu$ L and Cycloate extracts were ultrasonicated to re-constitute in acetonitrile rather than vortex-mixed for 15 seconds).

Control samples of soil were separately fortified with Cycloate, Cycloate sulfoxide and ECHA at 10  $\mu$ g/kg in septuplicate and at 100  $\mu$ g/kg in quintuplicate and analysed. Cycloate samples were extracted with water and toluene, concentrated under nitrogen and reconstituted with acetonitrile. The extract was diluted into calibration range with acetonitrile: water (50:50 v/v).

Cycloate sulfoxide samples were extracted with saturated sodium chloride in methanol: water (50:50 v/v) and toluene. The extract was diluted into calibration range with acetonitrile: water (50:50 v/v).

ECHA samples were extracted with methanol, 3M sodium hydroxide in water and toluene. The extract was diluted into the calibration range with acetonitrile followed by water to a final ratio of acetonitrile: water (50:50 v/v).

To assess matrix effects, calibration standards were prepared in control soil final extract and in acetonitrile: water (50:50 v/v).

Samples were analysed using liquid chromatography with tandem mass spectrometry detection (LC-MS/MS).

Matrix effects, linearity and specificity of the method were determined. Precision and accuracy were calculated at each validation level for each compound.

One primary and one confirmatory LC-MS/MS transition were analysed for Cycloate, Cycloate sulfoxide and ECHA.

The study was initiated on 22 August 2018 (date the protocol was signed by the Study Director) and completed on the date the final report was signed by the Study Director. The practical phase of the study was conducted by Smithers Viscient (ESG) and was started on 22 August 2018 (stock preparation) and completed on 17 September 2018 (LC-MS/MS analysis).

### MATERIALS AND METHODS

The study was conducted in accordance with the protocol and one amendment. All deviations are given in Appendix 7.

**Test Substance** 

Test Substance Name:	Cycloate
CAS Number:	1134-23-2
Molecular Formula:	$C_{11}H_{21}NOS$
Molecular Weight:	215.39 g/mol
Sponsor Lot Number:	5608300
Structure:	CH



**Purity:** Storage Conditions: Recertification Date: 98.1% Room temperature (15-30°C) 18 January 2019

**Test Substance Name:** 

CAS Number: Molecular Formula: Molecular Weight: Sponsor Lot Number: Structure:

## Cycloate sulfoxide

N/A C<sub>11</sub>H<sub>21</sub>NO<sub>2</sub>S 231.35 g/mol ET18361-12

Purity: Storage Conditions: Recertification Date:

98.88% Room temperature (15-30°C) 28 November 2018 <sup>1</sup> The recertification date is 1 year from manufacture (28 November 2017) as confirmed by the sponsor.

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Test Substance Name:	N-Ethylcyclohexylamine (ECHA)
CAS Number:	5459-93-8
Sponsor Lot Number:	14128CO
Molecular Formula:	$C_6H_{11}NHC_2H_5$
Molecular Weight:	127.23 g/mol
Structure:	
Purity:	98.7%
Storage Conditions:	Room temperature (15-30°C)
Recertification Date:	28 August 2018 <sup>1</sup>

<sup>1</sup> The ECHA validation was performed on 24 August 2018, which was before the recertification date had passed.

The Certificates of Analysis for the test substances are presented in Appendix 1.

### **Test System**

A control sample of soil was provided by Smithers Viscient, Wareham (DU-L-PF soil ID# 10JAN18-SoilB). The soil was given the unique identification 18/0000000/17 and stored refrigerated.

Characterisation of the soil was performed by Agvise Laboratories, Northwood, North Dakota (not determined under this study). Characterisation data was taken from Smithers Viscient, Wareham study 14113.6131 and is presented in the following table:

Collection location:	Grand Forks, ND
% Organic carbon:	7.1%
USDA textural class:	Loam
Particle size distribution:	31% sand, 44% silt, 25% clay
pH (soil: water ratio 1:1):	6.7
% Water holding capacity (at 1/3 bar):	45.1%

The moisture content of the soil was determined at Smithers Viscient, Harrogate and was determined to be 29.4% of the dry soil weight.

Reagents	
Acetonitrile	HPLC grade, Honeywell
Methanol	HPLC grade, Honeywell
Toluene	HPLC grade, Honeywell
Water	Milli-Q with LCPAK polisher, In House
0.1% Formic acid in water	MS grade, Honeywell
0.1% Formic acid in acetonitrile	MS grade, Honeywell
Sodium chloride (NaCl)	Reagent grade, Fisher
Sodium hydroxide (NaOH)	Reagent grade, Fisher
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Equivalent or better reagents may have been used.

## Equipment

Shimadzu Nexera series HPLC system with AB Sciex API 5000 MS/MS detector Allegra X-15R centrifuge

# **Analytical Method**

The report for study 14113.6131 was supplied by Smithers Viscient, Wareham on behalf of the sponsor. The method was re-written in Smithers Viscient, Harrogate format as draft method SMV 3202259-01D, including the instrumentation available at Smithers Viscient, Harrogate. This was followed for method validation, and re-issued as SMV 3202259-01V when validation was complete. The method used LC-MS/MS analysis. The complete analytical procedure is presented in Appendix 5. One person can complete the extraction of 15 samples (for either Cycloate, Cycloate sulfoxide or ECHA) in 1 day (8 hour working period). Automated LC-MS/MS analysis of 15 sample extracts can be completed in a further 6 hours from submission. Schematic diagrams of the analytical method are presented in Appendix 6.

## **Preparation of Reagents**

Acetonitrile: water (50:50 v/v) 500 mL HPLC grade acetonitrile was mixed with 500 mL Milli-Q water.

## Saturated NaCl in methanol: water (50:50 v/v)

> 90 g NaCl was added to 250 mL Milli-Q water until no more dissolved, and mixed with 250 mL methanol.

3M NaOH in water

60 g NaOH was dissolved in 500 mL Milli-Q water.

## Preparation of Primary Stocks

Primary stock solutions of Cycloate, Cycloate sulfoxide and ECHA were prepared as described in the following table:

Stock ID	Test Substance	Amount Weighed (mg)	Purity (%)	Solvent	Final Volume (mL)	Concentration (µg/mL) <sup>1</sup>
Stock 1	0.000	10.60	98.1	-	10.398	1000
Stock 2	Cycloate	10.94	98.1		1	10.732
Stock 3	Cycloate	11.03	98.88		10.907	1000
Stock 4	sulfoxide	11.54	98.88	- Acetonitrile	11.411	1000
Stock 5		10.22	98.7		10.087	1000
Stock 6	ECHA	10.89	98.7		10.748	1000

<sup>1</sup>Corrected for Purity.

Duplicate stocks were prepared for correlation purposes.

Primary stocks were stored refrigerated in amber glass bottles and given a nominal expiry of three months.

### **Preparation of Secondary Stocks**

Secondary stock solutions were separately prepared as described in the following table:

Test Substance	Stock Concentration (µg/mL)	Volume Taken (mL)	Solvent	Final Volume (mL)	Concentration (µg/mL)
Cycloate	1000	0.1		10	10
Cycloate sulfoxide	1000	0.1	Acetonitrile	10	10
ECHA	1000	0.1		10	10

Secondary stock solutions were stored refrigerated in amber glass bottles and given a nominal expiry of one month.

### **Preparation of Sub-Stocks**

Sub-stock solutions were prepared as described in the following table:

Test Substance	Stock Concentration (µg/mL)	Volume Taken (mL)	Solvent	Final Volume (mL)	Concentration (µg/mL)
	10	1		10	1
Cycloate	1	1		10	0.1
	0.1	1		10	0.01
	10	1	] Г	10	1
Cycloate sulfoxide	1	1	Acetonitrile	10	0.1
	0.1	1		10	0.01
ECHA	10	1	1 Г	10	1
	1	1	1 F	10	0.1
	0.1	1	1 [	10	0.01

<sup>1</sup> Equivalent to 10 µg/L.

Sub-stock solutions were prepared on the day of use and stored refrigerated until the corresponding analysis was complete.

### **Preparation of Matrix Matched Standards for Matrix Assessment**

Matrix matched standards of Cycloate, Cycloate sulfoxide and ECHA were prepared in disposable glass vials as described in the following tables:

### Cycloate

Stock Concentration (µg/L)	Volume Taken (mL)	Solvent	Final Volume (mL)	Concentration (µg/L)
10	0.2	Control soil final	10	0.2
10	0.2		10	0.2
10	0.2	extract	10	0.2

# Cycloate sulfoxide

Stock Concentration (µg/L)	Volume Taken (mL)	Solvent	Final Volume (mL)	Concentration (µg/L)
10	0.02	0	10	0.02
10	0.02	Control soil final	10	0.02
10	0.02	extract	10	0.02

## ECHA

Stock Concentration (µg/L)	Volume Taken (mL)	Solvent	Final Volume (mL)	Concentration (µg/L)
10	0.04	0	10	0.04
10	0.04	Control soil final	10	0.04
10	0.04	extract	10	0.04

## **Preparation of Non-Matrix Matched Standards for Matrix Assessment** Non-matrix standards of Cycloate, Cycloate sulfoxide and ECHA were prepared in

Non-matrix standards of Cycloate, Cycloate sulfoxide and ECHA were prepared in disposable glass vials as described in the following tables:

### Cycloate

Stock Concentration (µg/L)	Volume Taken (mL)	Solvent	Final Volume (mL)	Concentration (µg/L)
10	0.2		10	0.2
10	0.2	Acetonitrile:	10	0.2
10	0.2	water (50:50 v/v)	10	0.2

# Cycloate sulfoxide

Stock Concentration (µg/L)	Volume Taken (mL)	Solvent	Final Volume (mL)	Concentration (µg/L)
10	0.02	4	10	0.02
10	0.02	Acetonitrile:	10	0.02
10	0.02	water (50:50 v/v)	10	0.02

## ECHA

Stock Concentration (µg/L)	Volume Taken (mL)	Solvent	Final Volume (mL)	Concentration (µg/L)
10	0.04		10	0.04
10	0.04	Acetonitrile: water (50:50 v/v)	10	0.04
10	0.04		10	0.04

The matrix matched standards were analysed alternately with the non-matrix standards and the mean peak areas compared.

## **Preparation of Calibration Standards**

Non-Matrix matched calibration standards of Cycloate, Cycloate sulfoxide and ECHA were prepared as described in the following tables:

Cycloate

Stock Concentration (µg/L)	Volume Taken (mL)	Solvent	Final Volume (mL)	Concentration (µg/L)	
10	0.5		10	0.5	
0.5	0.8		1	0.4	
0.5	0.6	Acetonitrile:	1	0.3	
0.5	0.4	water (50:50 v/v)	1	0.2	
0.5	0.2		1	0.1	
0.5	0.1		1	0.05	

## Cycloate sulfoxide

Stock Concentration (µg/L)	Volume Taken (mL)	Solvent	Final Volume (mL)	Concentration (µg/L)	
10	0.05		10	0.05	
0.05	0.8		1	0.04	
0.05	0.6	Acetonitrile:	1	0.03	
0.05	0.4	water (50:50 v/v)	1	0.02	
0.05	0,2		1	0.01	
0.05	0.1		1	0.005	

## ECHA

Stock Concentration (µg/L)	Volume Taken (mL)	Solvent	Final Volume (mL)	Concentration (µg/L)
10	0.1		10	0.1
0,1	0.8		1	0.08
0.1	0.6	Acetonitrile:	1	0.06
0.1	0.4	water (50:50 v/v)	1	0.04
0.1	0.2		1	0.02
0.1	0.1		1	0.01

Calibration standards were prepared on the day of use, transferred into disposable glass vials and stored refrigerated until the analysis was complete.

A single set of calibration standards was prepared for each validation batch, which was analysed twice during the batch, in random order interspersed with the samples.

## Sample Fortification

5 g dry weight of soil was weighed into a Nalgene centrifuge tube. Soil samples were fortified with Cycloate, Cycloate sulfoxide or ECHA standard in acetonitrile at 10  $\mu$ g/kg in septuplicate and at 100  $\mu$ g/kg in quintuplicate. Duplicate control water samples and a reagent blank were also prepared, as described in the following tables:

#### Cycloate

Sample ID	Stock Concentration (µg/mL)	Volume Added (mL)	Sample Weight (g)	Fortified Concentration (µg/kg)
Reagent blank A & D	N/A	N/A	N/A	N/A
Control A, D-E & J-K	N/A	N/A	5	N/A
F10 A-G & V-AB	0,1	0.5	5	10
F100 A-E	1	0.5	5	100

N/A = Not Applicable.

Con A was used for matrix assessment.

#### Cycloate sulfoxide

Sample ID	Stock Concentration (µg/mL)	Volume Added (mL)	Sample Weight (g)	Fortified Concentration (µg/kg)
Reagent blank B	N/A	N/A	N/A	N/A
Control B & F-G	N/A	N/A	5	N/A
F10 H-N	0,1	0.5	5	10
F100 F-J	1	0.5	5	100

N/A = Not Applicable.

Con B was used for matrix assessment.

#### ECHA

Sample ID	Stock Concentration (µg/mL)	Volume Added (mL)	Sample Weight (g)	Fortified Concentration (µg/kg)
Reagent blank C	N/A	N/A	N/A	N/A
Control C & H-I	N/A	N/A	5	N/A
F10 O-U	0.1	0.5	5	10
F100 K-O	1	0.5	5	100

N/A = Not Applicable.

Con C was used for matrix assessment.

### Sample Extraction

#### Cycloate

10 mL water was added to each sample directly after fortification and vortex mixed for approximately 5 seconds. 15 mL toluene was added and vortex mixed for approximately 15 seconds, then placed on a shaker set to 200 rpm for 1 hour and centrifuged at 3000 rpm for 10 minutes. Approximately 10 mL of toluene extract was removed from the top layer of the sample and transferred into a glass jar. The sample extraction was repeated with 10 mL toluene, and the extracts combined. The toluene extract was made to 50 mL volume in a measuring cylinder. 1 mL extract was transferred to a borosilicate glass tube and evaporated to 25-100  $\mu$ L under a gentle stream of nitrogen (no heating). It was critical that the extract was not reduced to dryness (resulting in low recovery), or that too much toluene remained (resulting in partition of the extract). The sample was reconstituted with 1 mL acetonitrile and ultrasonicated for 5 minutes. The sample was diluted into calibration range with acetonitrile: water (50:50 v/v) and transferred into an HPLC vial for analysis.

Sample ID	Fortified Concentration (µg/kg)	Sample Weight (g)	Extract Volume (mL)	Dilution (mL-mL)	Dilution Factor
Reagent blank A & D	N/A	N/A	50	1-10	100
Control A, D-E & J-K	N/A	5	50	1-10	100
F10 A-G & V-AB	10	5	50	1-10	100
F100 A-E	100	5	50	1-10 then 0,3-1	333

The dilution procedure is given in the following table:

N/A = Not Applicable.

<sup>1</sup>3 aliquots of Control A extract were evaporated and diluted for matrix assessment.

#### Cycloate sulfoxide

10 mL saturated NaCl in methanol: water (50:50 v/v) was added to each sample directly after fortification and vortex mixed for approximately 5 seconds. 15 mL toluene was added and vortex mixed for approximately 15 seconds, then placed on a shaker set to 200 rpm for 30 minutes and centrifuged at 3000 rpm for 5 minutes. Exactly 10 mL toluene was removed from the top layer of the sample and transferred into a glass jar. The sample extraction was repeated with 10 mL toluene three more times, and the extracts combined to give a total of 40 mL toluene extract. An aliquot of the toluene extract was diluted into calibration range with acetonitrile: water (50:50 v/v) and transferred into an HPLC vial for analysis.

The dilution procedure is given in the following table:

Sample ID	Fortified Concentration (µg/kg)	Sample Weight (g)	Extract Volume (mL)	Dilution (mL-mL)	Dilution Factor
Reagent blank B	N/A	N/A	40	0.08-10	1000
Control B & F-G	N/A	5	40	0.08-10 <sup>1</sup>	1000
F10 H-N	10	5	40	0.08-10	1000
F100 F-J	100	5	40	0.025-10	3200

N/A = Not Applicable.

<sup>1</sup> 3 aliquots of Control B extract were diluted for matrix assessment.

#### ECHA

5 mL methanol was added to each sample directly after fortification and vortex mixed for approximately 5 seconds. 5 mL 3M NaOH was added and vortex mixed for approximately 5 seconds. 10 mL toluene was added and vortex mixed for approximately 15 seconds, then placed on a shaker set to 200 rpm for 10 minutes and centrifuged at 3000 rpm for 5 minutes. Exactly 5 mL toluene was removed from the top layer of the sample and transferred into a 20 mL glass vial. The sample extraction was repeated with 5 mL toluene three more times, and the extracts combined to give a total of 20 mL toluene extract. A portion of extract was diluted with 5 mL acetonitrile and vortex mixed for approximately 30 seconds. The sample was left for at least 10 minutes to allow the analyte to partition out of the toluene. 5 mL water was added to the acetonitrile and mixed, to give a ratio of acetonitrile: water (50:50 v/v). A portion of the final extract was transferred to an HPLC vial for analysis.

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Sample ID	Fortified Concentration (µg/kg)	Sample Weight (g)	Extract Volume (mL)	Dilution (mL-mL)	Dilution Factor
Reagent blank C	N/A	N/A	20	0.06-10	667
Control C & H-I	N/A	5	20	0.06-101	667
F10 O-U	10	5	20	0.06-10	667
F100 K-O	100	5	20	0.025-10	1600

N/A = Not Applicable. <sup>1</sup> 3 aliquots of Control C extract were diluted for matrix assessment.

### Instrument Conditions

LC-MS/MS analysis was performed using the following instrument conditions:

### Cycloate HPLC Parameters:

Instrument	Shimadzu Nexera series HPLC system						
Column#	Waters Atlantis T3, 3 µm, 4.6 × 100 mm						
Mobile Phase A#	0.1% Formic acid in water						
Mobile Phase B#	0.1% Formic acid in acctonitrile						
Flow Rate	0.8 mL/min						
Gradient	Time (min)	Mobile Phase A (%)	Mobile Phase B (%)				
	0.0	98	2				
	0.5	98	2				
	0.6	50	50				
	6.0	0	100				
	7.0	0	100				
	7.1	98	2				
	8.5	98	2				
Run Time	8.5 minutes						
Column Temperature	40°C						
Autosampler Temperature	10°C						
Injection Volume	50 µL						
Retention Time	Approx. 6.2 minutes	s (Cycloate)					
Valco Valve Diverter	Time (mi	n)	Position				
	0		A (to waste)				
	1		B (to MS)				
	7.5		A (to waste)				
100 (C D							

#### MS/MS Parameters:

Instrument	AB Sciex API 5000 Triple Quadrupole Mass Spectrometer					
Ionisation Type#	Electrospray (ESI)					
Polarity#	Positive					
Scan Type#	Multiple reaction monitoring (MRM)					
Ion Spray Voltage	5500 V					
Collision Gas (CAD)	5					
Curtain Gas (CUR)	25					
Gas Flow 1 (GS1)	40					
Gas Flow 2 (GS2)	40					
Vaporiser Temperature (TEM)	550°C					
Interface Heater (ihe)	On					
Entrance Potential (EP)	10					
Collision Exit Potential (CXP)	13					
Compound Name	MRM Transition Ions Monitored	Declustering Potential (DP)	Collision Energy (CE)	Dwell Time (ms)		
Cycloate (Primary)	215.9/83.0	120	22.26	100		
Cycloate (Confirmatory)	215.9/154.1	120	18.06	100		

Parameters marked # may not be modified. Minor adjustments to the remaining parameters may be required in order to fully optimise the system.

Dwell Time (ms)

200

200

Energy

(CE)

18.8

18.0

## Cycloate Sulfoxide HPLC Parameters:

Curtain Gas (CUR) Gas Flow 1 (GS1)

Gas Flow 2 (GS2)

Compound Name

Interface Heater (ihe)

Entrance Potential (EP)

Vaporiser Temperature (TEM)

Collision Exit Potential (CXP)

Cycloate sulfoxide (Primary)

Cycloate sulfoxide (Confirmatory)

Instrument Column#	Shimadzu Nexera series HPLC system Waters Atlantis T3, 3 µm, 4.6 × 100 mm					
Mobile Phase A#	0.1% Formic acid in water					
Mobile Phase B#	0.1% Formic acid in acetonitrile					
Flow Rate	1.2 mL/min					
Gradient	Time (min)	Mobile Phase A (%)	Mobile Phase B (%)			
	0.0	95	5			
	1.0	95	5			
	4.0	0	100			
	5.0	0	100			
	5.1	95	5			
	6.0	95	5			
Run Time	6 minutes					
Column Temperature	35°C					
Autosampler Temperature	10°C					
Injection Volume	25 μL					
Retention Time	Approx. 3.8 minutes (Cycloate sulfoxide)					
Valco Valve Diverter	Time (min)		Position			
	0		A (to waste)			
	1	1				
	5		A (to waste)			
MS/MS Parameters:						
Instrument	AB Sciex API 5000 Triple Quadrupole Mass Spectrometer					
Ionisation Type#	Electrospray (ESI)					
Polarity#	Positive					
Scan Type#	Multiple reaction monitoring (MRM)					
Ion Spray Voltage	5500 V					
Collision Gas (CAD)	5					
	2.2					

Parameters marked # may not be modified. Minor adjustments to the remaining parameters may be required in order to fully optimise the system.

Ions Monitored

253.9/226.1

253.9/177.1

MRM Transition Declustering Collision

Potential (DP)

135

135

25

40

40 650°C

On

10

13

# ECHA HPLC Parameters:

Instrument Column# Mobile Phase A# Mobile Phase B# Flow Rate	Shimadzu Nexera series HPLC system Waters Atlantis T3, 3 μm, 4.6 × 100 mm 0.1% Formic acid in water 0.1% Formic acid in acetonitrile 1.2 mL/min						
Gradient	Time (min)	Mobile Pha	se A (%)	Mobile Phase B (%)			
	0.0	98	(S) (S)	2			
	0.5	98		2			
	3.0	0		100			
	4.0	0		100			
	4.1	98		2			
	5.0	98		2			
Run Time Column Temperature Autosampler Temperature	5 minutes 40°C 15°C						
Injection Volume	50 µL						
Retention Time							
Valco Valve Diverter	Tíme (mi			Position			
	0	0		A (to waste)			
	1		B (to MS)				
	4		1	A (to waste)			
MS/MS Parameters:							
Instrument	AB Sciex API 5000 Triple Quadrupole Mass Spectrometer						
Ionisation Type#	Electrospray (ESI)						
Polarity#	Positive						
Scan Type#	Multiple reaction mo	onitoring (MRM	I)				
Ion Spray Voltage	5500 V						
Collision Gas (CAD)	5						
Curtain Gas (CUR)	25						
Gas Flow 1 (GS1)	40						
Gas Flow 2 (GS2)	40						
Vaporiser Temperature (TEM)	650°C						
Interface Heater (ihe)	On						
Entrance Potential (EP)	10						
Collision Exit Potential (CXP)	13						
Compound Name	MRM Transition Ions Monitored	Declustering Potential (DP)	Collision Energy (CE)	Dwell Time (ms)			
ECHA (Primary)	128.2/83.0	124	20.6	200			
ECHA (Confirmatory)	128.2/55.1	124	27.1	200			
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Parameters marked # may not be modified. Minor adjustments to the remaining parameters may be required in order to fully optimise the system.

LC-MS/MS data were collected using Analyst 1.6.2.

### **Calculation of Results**

LC-MS/MS data were calculated using Analyst 1.6.2. When the calibration fit is linear as in this study, Analyst uses the following formula to calculate the concentration of test substance present in the sample extract:

$$x=\frac{(y-c)}{m}\times DF$$

Where:

x = concentration of test substance in sample extract ( $\mu g/kg$ ) y = peak area due to test substance c = y intercept on calibration graph m = gradient of the calibration graph DF = sample dilution factor

The sample dilution factor is calculated as follows:

DF = Final extract volume (mL) / Amount of soil in final extract (g)

Where:

Final extract volume = volume of the final diluted sample Amount of soil in final extract = equivalent amount of soil in the extract after final dilution

Procedural recovery from fortified samples is calculated as follows:

Recovery (%) = Sample concentration / Fortified concentration × 100

The Limit of Detection (LOD) in  $\mu$ g/kg was calculated according to the following equation (U.S. EPA, 2016, 1994, Definition and Procedure for the Determination of the Method Detection Limit, Revision 1.11 and Revision 2):

 $LOD = (t 0.99 \times SD) + mean apparent residue in the control samples$ 

Where SD = standard deviation for 7 replicate samples at the LOQ t 0.99 = one-tailed statistic at the 99% confidence interval for 7 replicates (3.143)

The Method Detection Limit (MDL) in  $\mu g/kg$  was calculated by multiplying the lowest calibration standard (in  $\mu g/L$ ) by the dilution factor for the control.

### Validation Pass Criteria

The validation was deemed acceptable if the following criteria were met for the primary and confirmatory transitions monitored for each compound.

### Mean Recovery and Precision

Recovery and precision were acceptable if each fortification level had a mean recovery between 70 and 110% and a %RSD (relative standard deviation)  $\leq 20\%$ .

# Specificity/Selectivity

Specificity was acceptable if the amounts found in blank samples were  $\leq$  30% of the LOQ.

# Linearity

Linearity was acceptable if the lowest calibration standard concentration was  $\leq 80\%$  of the equivalent LOQ final extract concentration. The highest calibration standard concentration was  $\geq 120\%$  of the 10 × LOQ extract concentration (after dilution if applicable). The correlation coefficient (r) was acceptable if it was  $\geq 0.995$ .

## LOD (Limit of Detection) Assessment

The LOD was calculated according to the U.S. EPA, 2016, 1994, Definition and Procedure for the Determination of the Method Detection Limit, Revision 1.11 and Revision 2.

# MDL (Method Detection Limit)

The MDL was calculated as the sample concentration equivalent to the lowest calibration standard.

# Matrix Assessment

An assessment of matrix effects was made by comparison of the peak areas from triplicate standards prepared in blank solvent and in each control matrix final extract. This was assessed for the primary and confirmatory transitions.

Results were presented as a % difference from the mean non-matrix standard value.

A difference of  $\geq 20\%$  was considered significant.

Cycloate, Cycloate sulfoxide and ECHA were analysed using non-matrix matched calibration standards because matrix effects were insignificant.

Appendix 6 Schematic Diagram of the Analytical Method

Cycloate:



Cycloate sulfoxide:



### ECHA



### Appendix 7 Deviations

The protocol specified the MRM transitions to two decimal places, as given in the primary analytical method. It was not the intention to use these exact masses, as the actual mass measured is dependent on the tuning and optimisation of the mass spectrometer, however no such statement was included in the protocol. The MRM transitions used in this study were only given to one decimal place, as this is the level of precision displayed when tuning. This protocol deviation did not affect the integrity of the study.