#### Response to Federal Register Comments for DMF

Claudia M. Troxel Nancy Kim

Summary of proposed AEGL values for DMF					
Level	10-min	30-min	1-h	4-h	8-h
AEGL-1	NR	NR	NR	NR	NR
AEGL-2	160	110	90	55	38
AEGL-3	320	220	180	110	76

AEGL-1: Not recommended

**AEGL-2:** AEGL-3 ÷ 2 (because NAC felt the AEGL-3 values were protective base on monkey data)

**AEGL-3:** No mortality in rats exposed to 3700 ppm for 3 hours (Macdonald, 1982). Possible that proposed values are conservative. No effects observed in monkeys exposed to 500 ppm for 6 h/d, 5 d/wk for 2 or 13 wk, and rats exposed to 400 or 800 ppm DMF for 6 h/d, 5 d/wk for 13 wk exhibited only minimal to moderate necrosis of individual hepatocytes. Rats exposed to 200 ppm did not show evidence of any liver injury.

#### Total UF of 30

3 for interspecies: appears there are limited species differences regarding toxic response to DMF. Similar hepatic effects in humans as in animals. The mechanism of hepatotoxicity related to metabolism by CYP2E1 to reactive metabolite. Study demonstrates similar K<sub>m</sub> and V<sub>max</sub> between rat and human liver

10 for intraspecies:

- · CYP2E1 can be induced by alcohol, diabetes, and obesity
- Prior consumption of alcohol can exacerbate DMF toxicity
- Detoxification is partly dependent of glutathione conjugation; if GSH depleted, increased exposure to reactive metabolite
- DMF exposure can result in hepatotoxicity, so those with compromised liver function at increased risk

#### **Time scaling:**

Default value of n should be used in the temporal scaling of AEGL values across time. However, if one applies the default value of n = 1 for extrapolating from shorter to longer exposure periods, one obtains a 4 h value of 93 ppm and an 8 h value of 46 ppm. Using a default value results in AEGL values that are inconsistent with the available human data. Humans were exposed by inhalation to 87 ppm DMF for 4 h in a study designed to assess the metabolism of DMF (Kimmerle and Eben, 1975b). Although the study was not designed to assess the toxic effects resulting from DMF exposure, whatever effects may have been encountered were clearly not severe enough to be classified as AEGL-3 endpoints. Therefore, in the absence of any further data, an n of 2 was selected as a reasonable compromise between the possible values for n as reported by ten Berge et al. (1986).

AEGL-3 values are therefore derived using an n=3 for extrapolation to 10 and 30 min and 1 h, and an n=2 for extrapolation to 4 or 8 h.

#### Comments received from E.I. duPont Nemours, Inc., a producer of DMF

Overall, they think the numbers are too conservative, and that the AEGL values do not agree with the body of data on DMF.

- The 4-hour AEGL-2 is 55 ppm, while individuals were exposed to 87 ppm for 4 hours in a metabolism study.
- Data on repeated-exposure studies documented that no deaths occurred in monkeys exposed to 500 ppm or rats exposed to [800] ppm for 13 weeks, but the 10-minute AEGL-3 is 320 ppm.

Generally, NAC agrees with the comments from duPont, as there is already a statement in the AEGL-3 derivation section to that effect. However, there do not seem to be any viable alternatives at this point.

#### COMMENTS ON EPICHLOROHYDRIN TECHNICAL SUPPORT DOCUMENT

#### Derivation of AEGL-1 values.

The odor threshold should not be used as support. The values come from a secondary source. The cited values are in such a broad range (.08-12 ppm) as to be meaningless without a carefully controlled experiment with comparison to standards. If any good data exist, they should be applied to determination of an LOA, not the AEGL-1.

Is the Enterline reference a secondary source? If so it is not an appropriate basis for a derivation.

**Response**: AEGL-1 derivation needs to be revised because we no longer base AEGL-1 values on odor detection. Two proposals are presented for AEGL-1.

**Proposal No.1:** Use the UCC (1983) report that showed pharyngeal irritation in one of four subjects exposed to 68 ppm epichlorohydrin for 2 minutes. Exposure to 136 ppm resulted in irritation to the eyes and pharynx in two of four subjects.

Applying an uncertainty factor of 3 to the POD of 68 ppm and scaling based on n = 0.87 results in an AEGL-1 value for 10 minutes = **3.6 ppm**. This value should be maintained across all exposure durations. This chemical is an irritant.

**Proposal No. 2:** The NAC/AEGL Committee could recommend no AEGL-1 values because they would be below the level of odor detection. Shell Oil (1992) noted that epichlorohydrin is not detectable below about 10 ppm (OT<sub>50</sub> according to Shell Oil). There is no evidence of irritation occurring at  $\leq 10$  ppm. Therefore, any values derived would below the odor detection level as shown by the results in Proposal No. 1 above.

Calculation of LOA for Epichlorohydrin (see next page for derivation) The LOA = **46 ppm** 

#### Derivation of the Level of Distinct Odor Awareness (LOA) for Epichlorohydrin

#### Derivation of the Level of Distinct Odor Awareness (LOA)

The level of distinct odor awareness (LOA) represents the concentration above which it is predicted that more than one-half of the exposed population will experience at least a distinct odor intensity and about 10% of the population will experience a strong odor intensity. The LOA should help chemical emergency responders in assessing the public awareness of the exposure due to odor perception. The LOA derivation follows the guidance given by van Doorn et al., (2002).

The odor detection threshold  $(OT_{50})$  for epichlorohydrin is calculated from the odor threshold of 10 ppm (50% of unconditioned personnel) reported by Shell Oil (1992) and adjusted by Van Doorn (2002):

 $10 \text{ ppm} \times 40 \text{ ppm}/100 \text{ ppm} = 4.0 \text{ ppm}$ 

The concentration (C) leading to an odor intensity (I) of distinct odor detection (I=3) is derived using the Fechner function:

 $I = k_w \times \log(C/OT_{50}) + 0.5$ 

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For the Fechner coefficient, the default  $k_w = 2.33$  will be used because of the lack of chemical specific data.:

 $3 = 2.33 \times \log (C/4.0) + 0.5$ , which can be rearranged to log (C/4.0) = (3 - 0.5)/2.33 = 1.07, and results in  $C = (10^{1.07}) \times 4.0 = 34.4$  ppm

The resulting concentration is multiplied by an empirical field correction factor. It takes into account that in every day live factors, such as sex, age, sleep, smoking, upper airway infections, and allergy, as well as, distraction increase the odor detection threshold by a factor of 4. In addition, it takes into account that odor perception is very fast (about 5 seconds), which leads to the perception of concentration peaks. Based on the current knowledge, a factor of 1/3 is applied to adjust for peak exposure. Adjustments for distraction and peak exposure lead to a correction factor of 4/3 = 1.33.

 $LOA = C \times 1.33 = 34.4 \text{ ppm} \times 1.33 = 46 \text{ ppm}$  (van Doorn et al., 2002)

Therefore, the LOA for ethylene oxide is 46 ppm.

#### ATTACHMENT 8

#### Nitrogen Dioxide

#### Response to Federal Register Comments of October 7, 2004

One comment was received in response to the Federal Register notice for AEGL values for nitrogen dioxide. This comment was from Dr. George Alexeef regarding AEGL-1 values. Dr. Alexeef has requested that the NAC reconsider AEGL-1 values because effects were described at the concentration used as the basis for AEGL-1.

AEGL-1 values for nitrogen dioxide were set at 0.5 ppm for all time points. The basis for AEGL-1 was a study by Kerr et al. (1979, 1978) in which asthmatics were exposed to 0.5 ppm for 2 hours with exercise. At this concentration the odor was perceptible but the subjects became unaware of it after about 15 minutes. Seven of 13 asthmatics reported symptoms with exposure including two with slight burning of the eyes, one with a slight headache, three with chest tightness, and one with labored breathing during exercise. No changes in any pulmonary function tests were found immediately following the chamber exposure.

The NAC considered 0.5 ppm a NOAEL for objective tests of pulmonary function in **exercising asthmatics**. Dr. Alexeef considers the subjective symptoms an effect level. This issue was discussed by the NAC prior to voting overwhelmingly to adopt the AEGL-1 values. Neither new data nor alternatives to the standing NAC decision have been offered. Therefore, in the absence of specific recommendations or new information the NAC has no reason to reconsider AEGL-1 values for nitrogen dioxide. The values, as adopted, were based on the best data available and discussed openly by the full committee.

For comparison, the National Ambient Air Quality Standard is 0.053 ppm (annual average) with Significant Harm Levels of 2 ppm for a 1-hour average and 0.5 ppm for a 24-hour average; the Level of Concern (10% of the IDLH value) is 5 ppm<sup>1</sup>. The state of California has adopted 0.25 ppm as the standard for a 1-hour exposure to protect sensitive individuals.

<sup>&</sup>lt;sup>1</sup>The 5 ppm value for the Level of Concern was established prior to revision of the IDLH value from 50 ppm to 20 ppm in 1995.

#### ATTACHMENT 9

October 6, 2004

Document Control Office (7407M) Office of Pollution Prevention and Toxics (OPPTS) EPA 1200 Pennsylvania Avenue Washington, DC 20460-0001

Docket control # OPPT-2004-0079

Peracetic Acid AEGL-2 values

I would like to state my concern regarding the explanation in the Peracetic Acid TSD of the exposure time periods that are associated with symptoms reported by investigators. Table 2 in the TSD accurately reports Fraser's exposures and the investigator's health effects. The question is how to interpret the time notations: is this an exposure 'for' or 'at' any stated time. This is a fogging simulation with the generation of peracetic acid exposure by a unit turned on at time 0 and then off after 5 minutes of generating the "fog" with exposures and investigator symptoms reported for 45 minutes. Rather than an exposure of "3.12 to 4.67 mg/m3 for 15 to 20 minutes" as reported in 5.1, Table 2 shows this exposure existed from 15 to 20 minutes after fogging began, a 5 minute period. Section 6.3 goes further by eliminating the lower end of this range and states that " a slightly lower concentration of 4.67 mg/m3 ..for exposure durations up to 20 minutes". Further the more serious health effects are omitted in Section 6.3. where 3.5 minutes exposure to 15.6 mg/m3 is reported to cause "lacrimation" while Table 2 states there was "extreme discomfort".

Although McDonagh conducted a study in a plant for up to 3 hours, it is unclear in the TSD summary what the exposure level was during the whole time period. Rather as in Fraser, there likely was variable exposure over the 3 hours in an industrial operation. According to section 2.2 the .53 mg/m3 exposure was for a 10 minute sample while 5.1 states this level existed for "up to 3 hours". The investigators also state that "the background level of peracetic acid was fluctuated significantly". Basically the exposures and associated symptoms in both studies are for short time periods leaving an open question what the resulting health effects would be for longer time periods, especially 1, 4 and 8 hours

In addition, the two reports (Fraser and McDonagh) relied on are not per reviewed studies and the subjects are few with no systematic interview. Although internal industrial hygiene company reports can often be used by the AEGL committee, there are significant questions if these reports do not include the practices normally found in a human exposure study, namely the systematic investigation of health effects that can be matched with exposure periods of known duration. In both studies the health effects reported are by an unknown number of investigators in Fraser and two investigators in McDonagh, not neutral or unbiased subjects. The subjects were not blind to the exposure levels and there are few subjects with little power to generalize to larger populations. For example, although Fraser reported extreme discomfort at 2 ppm after 5 minutes of exposure, it is unknown if there were more subjects, some might experience this effect at lower values or that some subjects might experience more serious symptoms.

Please note that section 2.6 should be labeled section 2.3. I urge the committee to lower the AEGL-2 values by time scaling for all values greater than 10 minutes. Although the resulting levels will probably be lower that the AEGL 1 values at longer time periods, the evidence is not available to state with certainty that these effects would not occur at 4 or 8 hours. Since the matched health effects, exposure levels and exposure periods in both studies used for AEGL 1 values are for short time periods, and few animal studies for extended time periods, the committee should consider lowering the AEGL 1 values for the longer time periods if they approach or exceed the AEGL 2 levels.

Response: I am not sure of issue being raised in this comment. Controlled human studies are not always available for deriving AEGL-values. However, it is my understanding that we use the available human data wherever possible. The AEGL-1 values are below any concentration shown to cause irritation and should be protective of the general public.

Sincerely,

John S. Morawetz

c: Larry Gregoire Eric Bray Michael Sprinker Bill Kojola, AFL-CIO George Rusch, AEGL Chairman Paul Tobin, EPA 3725 Andrew Ave. Cincinnati, OH 45203 October 1,2004

Document Control Office (7407M) Office of Pollution Prevention and Toxics (OPPTS) Environmental Protection Agency 1200 Pennsylvania Avenue Washington, DC 20460-0001

Docket control # OPPT-2004-0079

Proposed AEGLs for Peracetic acid

I write to seek clarity on a few points in the Technical Support document which you have written on peracetic acid, and specifically how you have made use of Fraser and Thorbison (1986) to estimate concentrations and the related acute health effects.

I have been looking through both documents and remain unclear about the concentration d t s used. In particular, you report that 15.6 milligrams per cubic meter of peracetic acid is equivalent to 5 ppm (on pg 5 line 14) - but I don't believe that 5 ppm of peracetic acid is necessarily equivalent to 5 ppm "Total as  $H_20_2$ )), the units used in Fraser and Thorbison (1986). Your document seems to be making that assumption since you are using 6.2 mg/m<sup>3</sup> peracetic acid to correspond to Fraser's use of 2 ppm as  $H_20_2$  - for ex on pg 19 of your TSD.

## Response: The rationale for converting ppm as $H_2O_2$ to mg/m<sup>3</sup> of peracetic is that one mole $H_2O_2$ is equivalent to one mole peracetic. Therefore, ppm of $H_2O_2$ can be converted to mg/m<sup>3</sup> peracetic acid.

Also, your discussion in section 6.3 (pg 19 line 20) may not clearly summarize the data in Tables 1 and 2 of Fraser. You say that exposure to 2 ppm "for 2 minutes was also considered tolerable"; but this seems to be a very selective reading of the last line of Fraser's Table 1, which is fairly confusing in that it presents variability due to location and time in the same table. Both tables together (including Table 2) could also be read to mean that this same concentration caused unbearable irritation (Table 1 middle) or extreme discomfort (Table 2 top) over brief periods of exposure.

# Response: This uncertainty regarding the effects at 2 ppm was taken into account when a lower concentration (1.5 ppm) was used as the point of departure for AEGL-2. Table 2 in the document and Fraser and Thorbinson's table showed fairly consistent responses at exposure concentrations $\leq 1.5$ ppm (4.7 mg/m<sup>3</sup>).

One difficulty in interpreting Fraser is estimating what the short-term average exposures might have been over the first 2,5 and 10 minutes of their fogging study, since the exposure seems to have 'ramped up quickly to (above?) 5 ppm in the first 3 minutes and

then gradually declined through 2 ppm at about 5 minutes. To come up with estimated short-term averages from this incomplete data seems to me to require some estimates of area under a plausible curve (then divided by duration), which it appears that you have not done.

It may be the case that your uncertainty factor of 3 takes care of these problems in the derivation of the proposed AEGL-2, but I believe additional attention to these issues is necessary before moving forward with the process.

Response: All studies of this type are accompanied by varying degrees of uncertainty. The exposure concentrations and effects presented by Fraser and Thorbinson were difficult to interpret. It may be possible to calculate the area-under-the-curve; however, this value is likely to be higher than that the point of departure reported in the document. Therefore, we have taken the more conservative approach. The uncertainty factor of 3 does provide an added level of protection.

Sincerely, Thurman B. Wenzl ScD CIH Adjunct Associate Professor University of Cincinnati Document Control Office (7407M) Office of Pollution Prevention and Toxics (OPPT) EPA 1200 Pennsylvania Avenue Washington, DC 20460-0001 October 6, 2004

Docket control # OPPT-2004-0079

Trichloroethylene values

I would like to raise concerns regarding the 10 and 30 minute AEGL-3 values recommended by the AEGL Committee for Trichloroethylene (TCE). The setting of environmental exposures above the anesthetic level, an AEGL-3 of 10,000 ppm for 10 minutes, would be potentially life threatening. The question is: If trichloroethylene was used safely as an anesthetic in an operating room, is it therefore safe for an environmental release? If the committee was setting safe levels for hospital use of TCE, the committee's values may stand. But this is neither the population nor the context of the AEGL's intended use.

Although there is evidence that TCE it is not immediately lethal to patients prepared for surgery, my concerns is that exposures to the general public without the medical support available in an operating room could be life threatening, the other AEGL-3 definition besides immediate lethality. The general population will not have medical support readily available. It is potentially dangerous to take safety data from the operating room and assume that it is transferable to the general public. Even healthy individuals who have just consumed a meal are at significant risk of aspiration of vomit and likely death. This danger is will recognized and leads to prohibitions from eating before surgery.

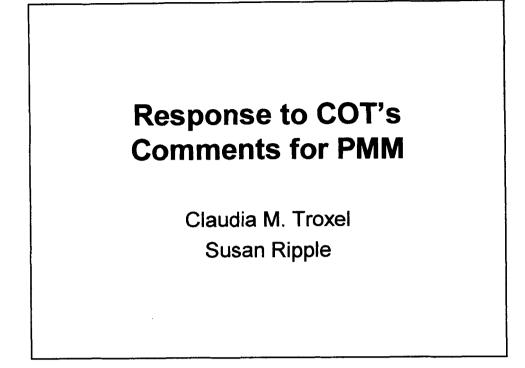
In hospitals, exposures at anesthetic levels occur to people who have been medically screened to be fit for an operation, they are monitored closely during the operation, the level of anesthetic dose is adjusted to provide a margin of safety, and they recover in a controlled hospital environment where they are regularly monitored by medical staff. What is "safe" under these conditions is not the same as during community exposures. People with pre-existing medical problems, such as cardiovascular disease, may be more susceptible to the lethal effects of these agents, there will be no trained medical staff observing for complications or reducing the exposure if problems do occur, and recovery from exposure likely will occur in an uncontrolled environment remote from expert care. Section 7.3 on the derivation of AEGL-3 states that "the obtained values are considered to be too low compared to the available human evidence". The studies that support this statement must be cited. The studies cited in the Technical Support document are mostly to exposures at 200 ppm or less. One study exposed people to 1,000 ppm for 2 hours but subjects began feeling lightheadedness, lethargy or dizziness, supportive of lower values.

The bottom line is that the one hour value of 3,800 ppm (.38%) should be used for the 10 and 30 minute values to provide a margin of safety for the threshold of potentially life threatening anesthetic effects. This value is consistent with the threshold of anesthesia found in the studies of Pembleton, 1974, Parfitt, 1999 and Langton-Hewer, 1975.

Sincerely,

John S. Morawetz

c: Larry Gregoire Secretary Treasurer's Office Eric Bray Michael Sprinker Bill Kojola, AFL-CIO George Rusch, AEGL Chairman Frank Mirer, UAW



#### **Properties:**

- · Oily yellow liquid
- Unbearable acrid odor
- Irritant, lacrimator

#### Human data:

- Lethal effects: case report of exposure to unknown concentration of vapor and liquid resulting in massive hemorrhaging lung edema with simultaneous heart, circulatory, and kidney failure from resultant hypoxia
- *Nonlethal effects:* Odor threshold (secondary sources)

0.001 ppm (Ruth); 0.24 ppm (Russian)

#### LETHAL ANIMAL DATA (only rats) Vernot et al., 1977; 1-hr LC<sub>50</sub> 11 ppm (males); 16 ppm (females); 13.5 ppm (combined) Stauffer Chemical Co., 1971; [nominal or measured conc. ?] 13 ppm (combined); 1-hr LC<sub>50</sub> 9 ppm no deaths; 18 ppm: 7/10 died Eye/mucosa irritation, dyspnea, acute depression Gage, 1970 (purity unknown; nominal conc.) 100 ppm for 1 hr – 4/4 M rats died (pulmonary edema) 10 ppm for 6 hr – 3/4 died; lethargy and respiratory difficulty

#### NONLETHAL ANIMAL DATA (only rats)

#### Knapp et al., 1987 (abstract)

15 M and Fe SD rats/group exposed to "cumulative" mean air concentrations 6 h/d, 5 d/wk, for 2 wks

Results:

- 0.02 ppm No effects
- 0.13 ppm Mild nasal epithelial changes
- 1.15 ppm Haircoat stains, labored breathing, tremor, ↓ bw, ↑lung wt, pulmonary edema, ↑ mucous secretions, alveolitis, interstitial fibroplasia, mild nasal epithelial changes

#### NONLETHAL ANIMAL DATA (only rats)

#### Knapp and Thomassen, 1987

18 SD rats/sex/group exposed 6 h/d, 5 d/wk, for 70 to 72 d

**Results:** 

- 0.014 ppm No effects
- 0.079 ppm 1 M, 1 Fe had residues of purulent or serum exudate
- 0.58 ppm Salivation (d 18) and sneezing (d 59); ↓ bw, ↑lung wt, mucous in trachea, respiratory nasal epithelium changes, residues of purulent or serum exudate, focal subacute interstitial pneumonia

#### NONLETHAL ANIMAL DATA (only rats)

*Gage, 1970* (purity unknown; nominal conc.; mixed with acetone)

Results

- 2 ppm (in acetone) for twenty, 6 hr exposures;
   4 M rats; initial respiratory difficulty, none died;
   gross necropsy pulmonary congestion
- 0.5 ppm (in acetone) for twenty, 6 hr exposures;
   4 M and 4 Fe rats; no signs of toxicity; no gross necropsy findings

Sur	nmary of	proposed	AEGL va	lues for P	MM
Level	10-min	30-min	1-h	4-h	8-h
AEGL-1	0.018	0.018	0.014	0.0090	0.0060
AEGL-2	0.044	0.044	0.035	0.022	0.015
AEGL-3	0.54	0.38	0.30	0.075	0.038

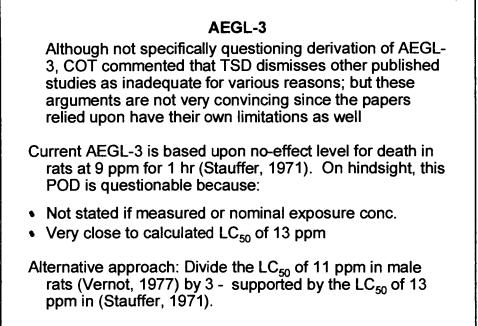
**AEGL-1:** Rat NOAEL of 0.079 ppm for 6 hr/d, 5 d/wk for 70-72 exposure days (Knapp and Thomassen, 1987)

**AEGL-2:** Mild/minimal focal subacute interstitial pneumonia and ↑ lung wt in rats exposed 0.58 ppm for 6 hr/d, 5 d/wk; 70-72 d (Knapp and Thomassen, 1987)

**AEGL-3:** No mortality in rats exposed to 9 ppm for 1 h (Stauffer Chemical Co., 1971)

#### COT COMMENTS 3 Main areas of concern:

- The NAC should consider including a MF to account for the poor data quality
- Concern that AEGL-1 and -2 appear to be based on systemic endpoint of pulmonary infection following a single exposure to an irritant
- The application of uncertainty factors (adjusted composite UF instead of individual components)



> Therefore, POD = 11 ppm ÷ 3 = 3.7 ppm

#### AEGL-3, con't

POD = 3.7 ppm

UF:

- 10 for interspecies ?
- 3 for intraspecies mechanism of toxicity direct contact effect; death by pulmonary edema
- n = default of 1; 3
- NOTE: Rats exposure to 0.58 ppm for 6 h/d, 5 d/wk, for 70 to 72 d: only effects related to repeated exposures

		AE	GL-3		
UF	10 min	30 min	1 hr	4 hr	8 hr
30	0.22	0.16	0.12	0.031	0.015
10	0.67	0.47	0.37	0.092	0.046

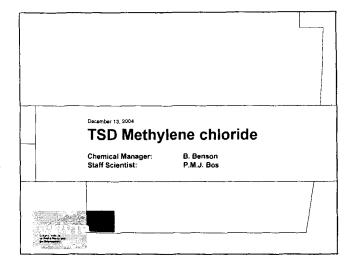
#### Comment:

AEGL-1 and -2 are based on the NOAEL or LOAEL for repeated exposure effects (6 hr/d, 5 d/wk for 70-72 d); mainly for prevention of pneumonia. NAC appears to be adopting the position that prevention of opportunistic pulmonary infection following a single exposure to an irritant is the proper public health end point for mercaptans and other irritants

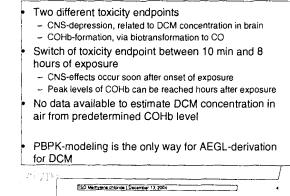
#### Alternatives:

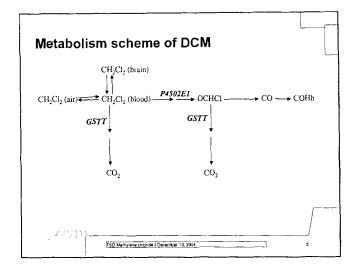
AEGL-2, could divide the AEGL-3 by 3 AEGL-1 – although from a secondary source, could use the odor threshold of 0.001 ppm (Ruth) on the basis that odor is "unbearable"

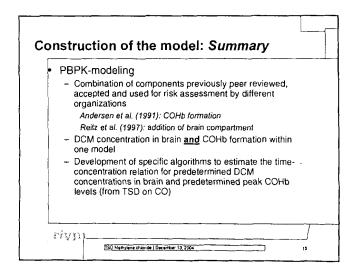
	Summ	ary of A	lternativ	e AEGLs	\$
AEGL	10 min	30 min	1 hr	4 hr	8 hr
Curren	t values	- <b>I</b>		.4	
1	0.018	0.018	0.014	0.0090	0.0060
2	0.044	0.044	0.035	0.022	0.015
3	0.54	0.38	0.30	0.075	0.038
AEGL-	1:odor; AEC	GL-2: AEG	L3÷3; AEC	GL-3: ⅓ LC	; <sub>50</sub> ; UF 10
1	0.001	0.001	0.001	0.001	0.001
2	0.22	0.16	0.12	0.031	0.015
3	0.67	0.47	0.37	0.092	0.046
Same a	as above, b	ut UF 30	<u> </u>		1
2	0.073	0.053	0.040	0.010	0.0050
3	0.22	0.16	0.12	0.031	0.015

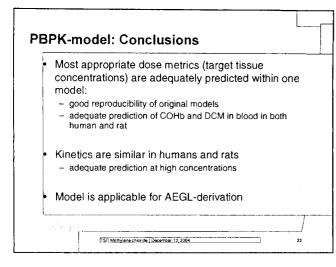


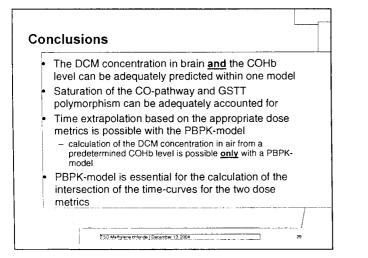
#### **Reasons for PBPK-modeling**

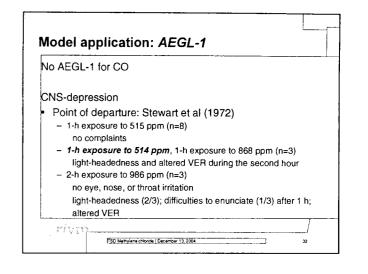


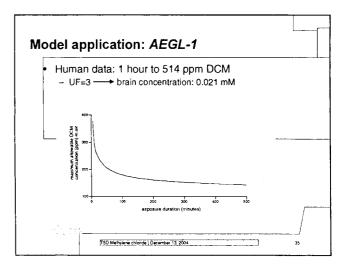


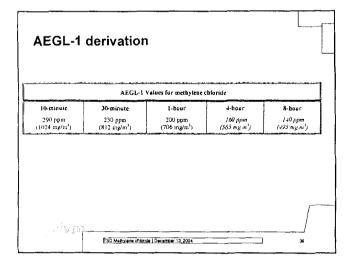


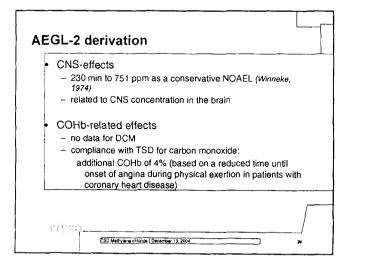


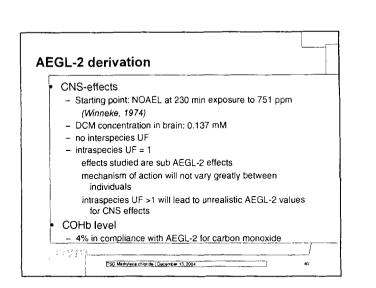


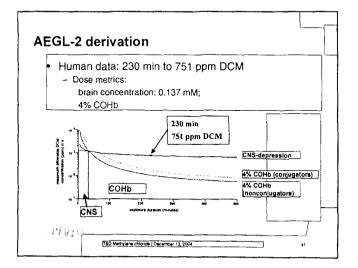


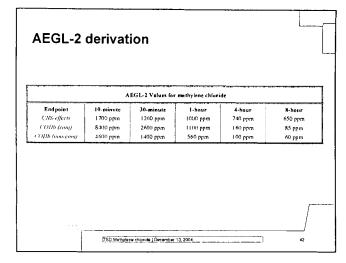


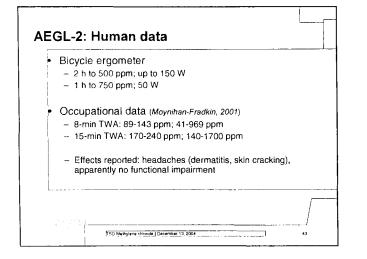


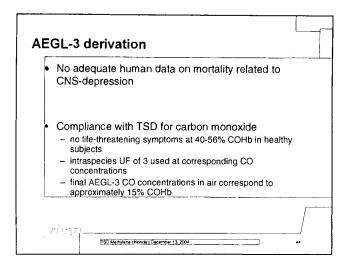


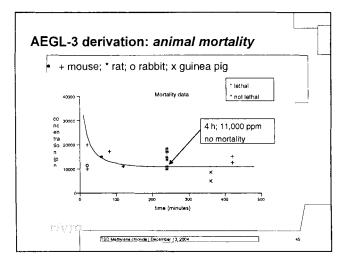


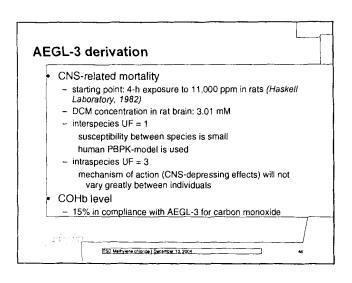


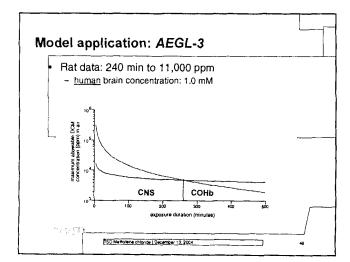












Endpoint 10-mia				
		l-bour	4-bour	8-hour
CNS-offects 12,000 p	ppm 8500 ppm	6900 ppm	4900 ppm	4200 ppni
OHb (cong)			 5200	
OHb (non-cong) 155,000	ppin 52,000 ppm	25,000 ppm	5300 ppm	2100 ppm

	Summary of AEGL Values								
Classification			Exposure Duratio	n					
	10-minute	30-minute	l-hour	4-hour	8-hour				
AEGL-I (Nondisabling)	290 ppm	230 թթո	200 pra	NR	NR				
AEGL-2	1700 ppm	1200 ppm							
(Disabling) Non-conjugators			560 ppm	100 ppm	60 ppu				
AEGL-3	12,100 ppm	8500 ppm	6900 ppm	4900 ppm					
(Lethal) Non-conjugators	1				2100 pp#				

#### **Update on Vinyl Acetate**

Claudia Troxel Richard Thomas

The AEGL-2 was based on a rat study (Bogdanffy et al. 1987) in which exposure for 6 h to 1000 ppm caused reversible nasal lesions, with the stipulation that the pathologist be contacted to confirm this.

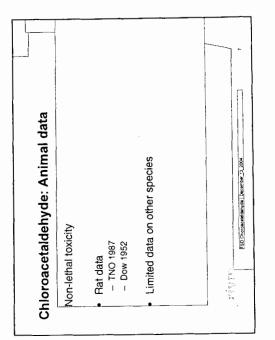
The report from the pathologist does indeed confirm that the lesions are reversible.

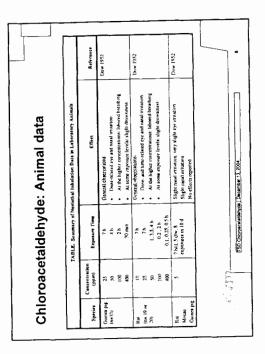
#### **ATTACHMENT 14** Chloroacetaldehyde: **Physical-chemical properties** Molecular weight: 78.5 Colorless liquid Water solubility: soluble December 13, 2004 Boiling point: 85°C (pure substance) **TSD** Chloroacetaldehyde Odor: acrid, penetrating Chemical Manager: M. Payton Flammability: not flammable Staff Scientist: M. Draaijer / P.M.J. Bos 1897 -TSD Chloroacetaldehyde i December 13, 2004 Chloroacetaldehyde: Uses Chloroacetaldehyde: Human data No human data available Chemical intermediate in manufacturing of chemicals Statement in a report on acute toxicity (Dow 1952) - "Every concentration employed including the lowest (10 ppm) produced lachrymation and nasal irritation in humans within a few minutes". 1

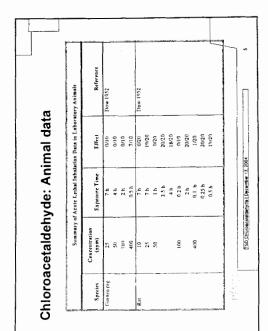
1

Chemical intermediate in manufacturing of chemicals
 Control of growth of algae, bacteria, and fungl in water

No human data available
Statement in a report on acute toxicity (Dow 1952)
- "Every concentration employed including the lowest (10
ppm) produced lachrymation and nasal irritation in humans
within a few minutes".

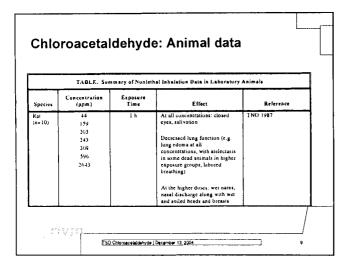


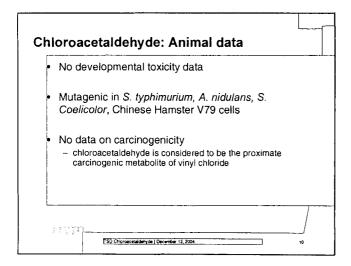


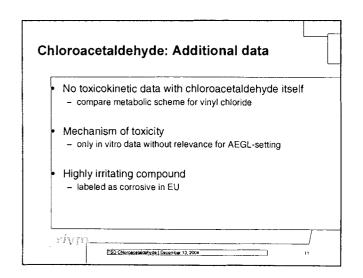


Species Concentration Exporter Tind (p)m) 44 In 10 159 201	NAMES OF TAXABLE PARTY OF TAXABLE PARTY OF TAXABLE PARTY.
	Effect Reference
202 596 2643 2643 16	0/10 TKO 1987 3/10 4/10 10/10 10/10 10/10 10/10 10/10

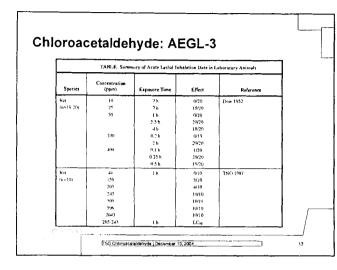
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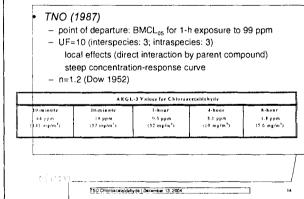


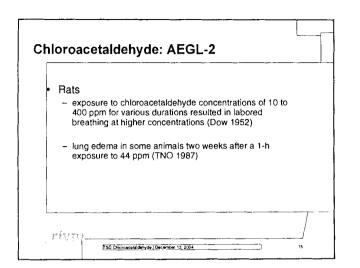


D	ata availability
•	No adequate human data
•	Animal data indicate that irritation/corrosivity is the major effect of chloroacetaldehyde – increasing severity with increasing exposure – death due to severe lung damage – steep dose-response curve

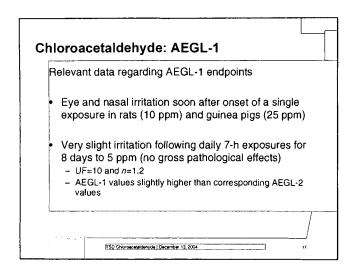


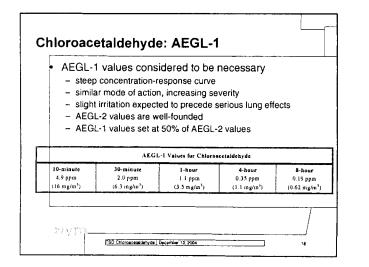






ppm – modi – UF=	edema in rats t fying factor of 2 10 (interspecie 2 (from mortali	(LOAEL) s: 3; intraspecie ty data; similar	es:3) mode of action	
10-minute 9.8 ppm (31 mg/m <sup>2</sup> )	AEG1 30-minute 3.9 ppm (13 mg/m <sup>2</sup> )	L-2 Values for Chloros L-hour 2.2 ppm (7.1 mg/m <sup>3</sup> )	4-hour 0.69 ppm (2.2 mg/m <sup>3</sup> )	8-hour 0.39 ppm (1.5 mg/m²

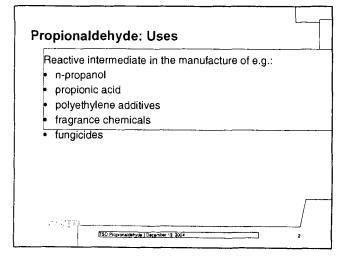




#### Chloroacetaldehyde: Summary of AEGL-values

		Exposure Duration				
Classification	10-minute	30-minute	l-bour	4-hour	8-bour	
AEGL-1 (Nondisabling)	4.9 ppm	2.0 ppm	1.1 թթու	0.35 ppm	0.19 ррп.	
AEGL-2 (Disabling)	9.8 ppm	4.0 ppm	2.2 ppm	0.69 ppm	0.39 ppm	
AEGL-3 (Lethal)	44 ppm	18 ppin	9,9 ppm	3.1 ppm	1.8 ppm	
					[	
rivm					1	

### December 13, 2004 December 13, 2004 TSD Propionaldehyde Chemical Manager: M. Payton Staff Scientist: A. Muller / P. M.J. Bos



#### Propionaldehyde: Physical-chemical properties Molecular weight: 58.08 Colorless liquid Water solubility: soluble in 5 vol. water Boiling point: 49°C

#10

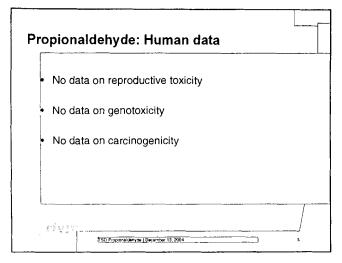
- Odor: suffocating odor
- Flammability: flash point between -18 and -40°C

TSD Propionaldehyde | December 13, 2004

• LEL: 2.3 - 2.9%

rivyy

# Propionaldehyde: Human data No relevant case reports available Experiments with volunteers — Mild irritation in 12 males exposed to 134 ppm for 30 min f<sup>i</sup>/VJT

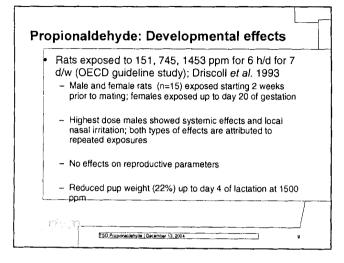


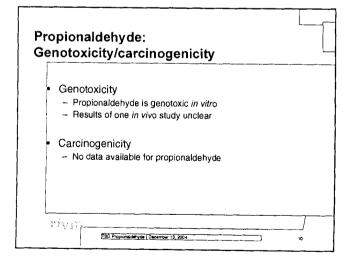
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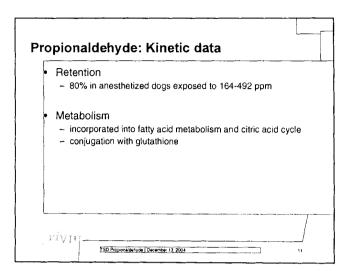
	Summery	of Acute Lethal Inhala	tion Data in Labor	utory Animals
Species	Concentration (ppm)	Exposure Time	Effect	Reference
Rat	1930 ppm	4 h	No lethality	Eschbach 1981
Rai	8000 ppm	4 h	5/6 deaths	Smyth et al. 1951
Rat	25,420 ypm	30 min	LC12	Skog 1950
Rat	Saturated vapor pressure	30 min	4/4 deaths	Gage 1910
Mouse	2868 mg/m <sup>3</sup> (acrosol)	5 h on average	50/50 deaths	Salem and Cullumbine 1960
Rabbit	2868 mg/m <sup>3</sup> (aerosol)	4 h on average	3/5 deaths	Salein and Cullumbine 1960
Gumea pig	2868 mg/m <sup>3</sup> (acrosol)	10.6	3/20 deaths	Salem and Cullumbine 1960

<ul> <li>RD<sub>50</sub> values</li> <li>– rat: 6789 ppm</li> <li>– mouse: 2052 ppm</li> </ul>	
2078 ppm	
3703 ppm	

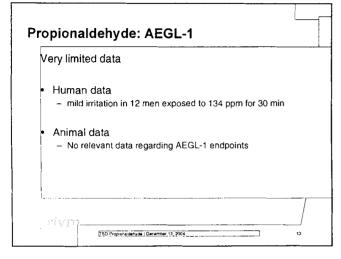
Summary of NonLethal Inhalation Data in Laboratory Animals					
Species	Concentration (ppm)	Exposure Time	Effect	Reference	
<b>₹</b> ⊒1	1226 ppm	) min	No effect on blood pressure	Egle 1972	
Rat	4100 ppm	1 min	Increase in blood pressure	Egle 1972	
Rat	1930 ppm	4 h	Lachrymation	Eschbach 1981	
Mouse	5230 ppm	5 min	Anesthesia	Axelsson et al. 195	
se					

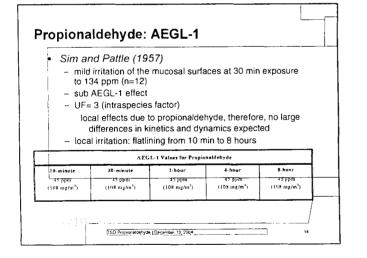




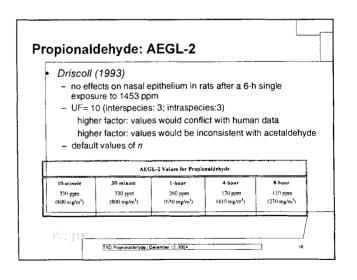


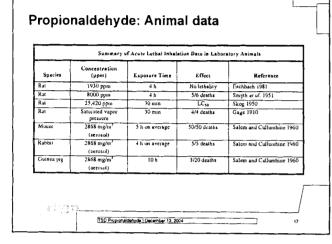
Comp	trison of effects of propional	dehyde and acetaldehyde	
Effect	Acetaldehvde	Fropionaldehvde	Keference
Increase us blood prossure in rat			Fale et al 1973
5 mg/kg (V (%)	10.7	10.5	
10 mg/kg IV (%)	6.7	12.4	1
NOE2, in ppm for			Egle 1972a
Increased heart rate	5560	4220	1-0
Increased blood pressure	556	1266	
Deposition in dog upper resputtory tract (%)	30-55	59-63	Egle 1972b
Aldrhyde dehydrogenase, partially purified			Petersen et al 1977
from mover liver cytospi		1	
Km (uM)	0.59	0.36	1
Vmax	4.40	3.30	1
RD50 values (ppm)	1		
B6C3F1 mice	2932	2078	Steinhagen and Harrow 1984
Swina Websier mice	2845	2052	Stewlingen and Barrow 1984
Swigs Webster mule	1900	2750	Alune 1981
F-344 rat	2991	6900	Babuk ct of 1985
Mortalay	8000 ppm. 8 hour. 0/6 16000 ppm, 8 hour. 0/6	8000 ppm, 4 hour. 5/6	Smyth et al. 1951
1.C50 in 30 manates (ppm)	20720	25420	Skog 1950

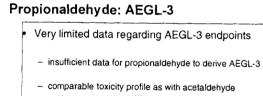




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•	Rats
	<ul> <li>lachrymation in rats exposed for 4 h to 1930 ppm (starting after 15 min)</li> </ul>
	<ul> <li>effects on nasal epithelium upon repeated exposure; considered not relevant for single exposure (highest concentration: 1453 ppm for 6 h/d)</li> </ul>
•	Mice
	<ul> <li>anesthesia in mice exposed for 5 min to 5230 ppm</li> </ul>







TSD Propionaldehyde | December 13, 2004

equate data on AEGL-3 endpoints (BMDL <sub>os</sub> ) for acetaldehyde
GL-3 values for acetaldehyde adopted for propionaldehyde

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#### Propionaldehyde: AEGL-3 Appelman et al. (1982) - point of departure: BMDL<sub>os</sub> of 5295 ppm (4-h exposure) - UF=10 (interspecies:3; intraspecies: 3) - default values for *n* (flatline from 30- to 10-min) AEGL-3 Values for Propionaldehyde 30-minute 10-minute 1-huur 4-bour 8-bour 530 ppm 1100 ppm 1100 ppm 840 ppm 260 թրա (1300 mg/m<sup>3</sup>) (2700 mg/m<sup>3</sup>) (2700 mg/m<sup>3</sup>) (2000 mg/m<sup>2</sup>) (630 mg/m<sup>3</sup>) Alternative: POD: 30-min LC<sub>30</sub> of 25,420 ppm; MF=3; UF=10; n=1 30-min: 850 ppm; 1-h: 420 ppm; 4-h: 110 ppm; 8-h: 53 ppm river TSD Propionaldehyde | December 13, 2004

		Summary of Al	GL Values		
	Exposure Duration				
Classification	10-minute	30-minute	l-kour	4-hour	8-bour
AEGL-1 (Nondisabling)	45 ppin	45 ppin	45 ppm	45 ppm	45 ppm
AEGL-2 (Disabling)	330 ppm	3.30 ppm	260 ppm	170 ppm	110 pp.u
AEGL-3 (Lethal)	1100 ppm	1100 ppm	840 ррни	530 ppm	260 ppm

#### ACUTE EXPOSURE GUIDELINE LEVELS (AEGLs) FOR BIPHENYL

NAC/AEGL-35 December 13-15, 2004 Washington D.C.

ORNL Staff Scientist: Dana F. Glass Chemical Manager: Richard Thomas Chemical Reviewers: Susan Ripple and Bob Benson

#### **Biphenyl- Background**

• Currently used as heat-transfer agent and fungistat for citrus

- Production greatly decreased due to PCB restrictions
- Colorless to pale yellow/white solid at room temperature
- Distinct, pleasant odor with low odor threshold
- Limited data available on inhalation studies

### **Exposure Symptoms**

### • Exposures from inhalation or dermal contact

- Eye and throat irritation
- Headaches
- Nausea

### **AEGL-1 Values**

- AEGL-1 values not recommended
- Lack of data available

### **AEGL-2 Values**

AEGL-2 Values										
10 min	30 min	1 hr	4 hr	8 hr						
2.9 ppm	2.9 ppm	2.3 ppm	1.4 ppm	0.73 ppm						

- Key Reference:
  - Cannon Laboratories, Inc. 1977
  - National Research Council. 2001

• Endpoint/Concentration/Rationale: Three-fold reduction of AEGL-3 values. Estimated threshold for impaired ability to escape.

### **AEGL-3 Values**

AEGL-3 Values								
10 min	30 min	1 hr	4 hr	8 hr				
8.6 ppm	8.6 ppm	6.8 ppm	4.3 ppm	2.2 ppm				

• Key Reference:

- Cannon Laboratories, Inc. 1977

- Test Species: 10 male/10 female mice
- Exposure: Inhalation: 14, 38 or 43 ppm, 4 hrs
- Effect:
  - 14 ppm: hyperactivity and shallow respiration
  - 38 and 43 ppm:

 hyperactivity, nasal discharge & rapid respiration. Moderate weight loss (day 1)

• 1/10 death- 2 hours at 43 ppm (not compound-related)

• slight lung congestion on gross pathological examination

• Endpoint/Concentration/Rationale: 43 ppm- highest concentration used in acute inhalation studies resulting in clinical signs without death

### AEGL-3 Values (cont'd.)

### • Uncertainty Factors/Rationale: 10

- Interspecies: 3, clinical signs similar among different species

- Intraspecies: 3, using intraspecies UF of 10 creates levels unrealistically low compared to occupational levels

Time-scaling: Extrapolation to time-points was done
n = 3, for 30 min, 1 hr and 4 hr

-n = 1, for 8 hr

- 30-minute AEGL-3 value also adopted as the 10minute value

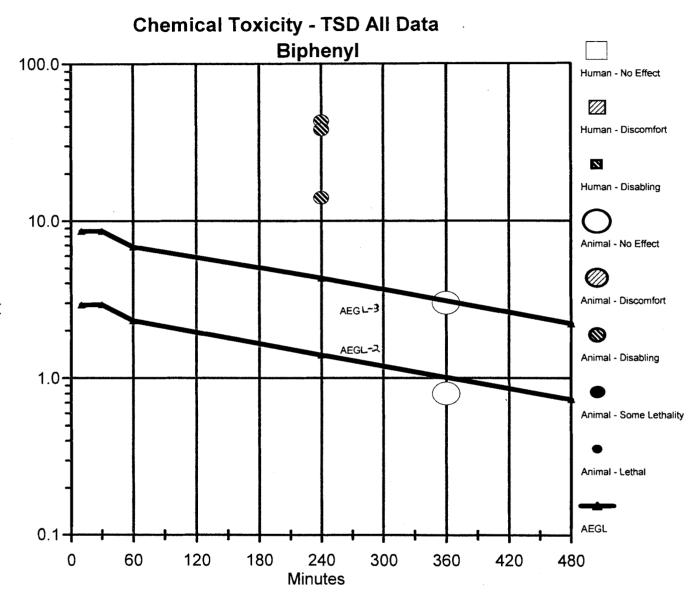
### Exposure Guidelines (expressed as ppm)

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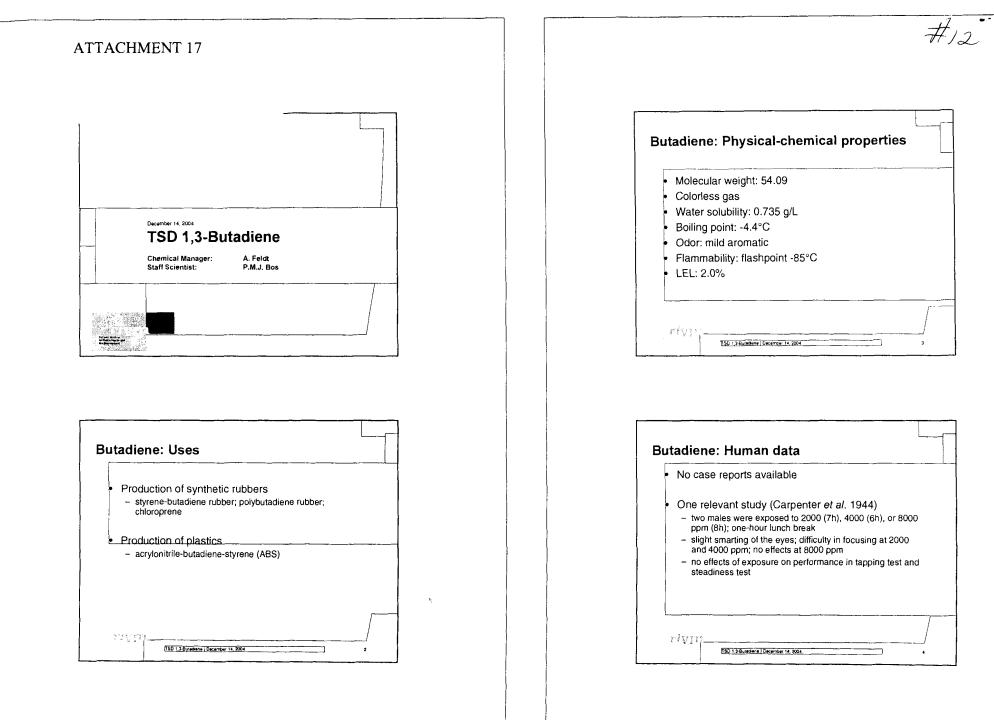
Extant	Standards	and Guid	lelines fo	r Chemi	cal				
C	Exposure Duration								
Guideline	10 min	30 min	1 hr	4 hr	8 hr				
AEGL-1	NR	NR	NR	NR	NR				
AEGL-2	2.9	2.9	2.3	1.4	0.73				
AEGL-3	8.6	8.6	6.8	4.3	2.2				
PEL-TWA (OSHA)					0.2				
IDLH (NIOSH)		16							
REL-TWA (NIOSH)					0.2 (10 hr)				
TLV-TWA (ACGIH)					0.2 (lung)				
MAK (Germany)					0.2				
MAC (Dutch)					0.2				
STV/LLV (Sweden)	0.4 (15 min)				0.2				

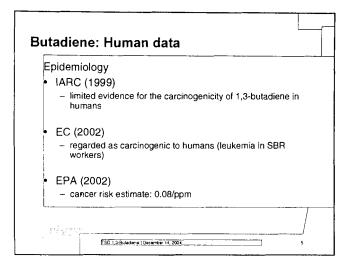




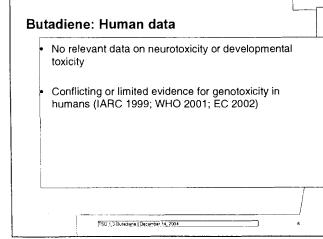
No effect= No effect or mild discomfort Discomfort= Notable transient discomfort/irritation Disabling= Irreversible/long lasting effects or impaired ability to escape Some lethality= Some, but not all, exposed animals died Lethal= All exposed animals died

bpm





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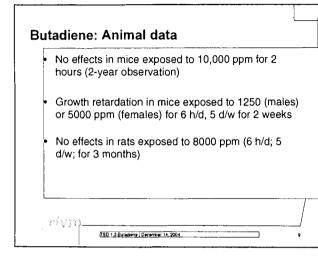
	Summer	of Acute Lethal (pho	lation Data in Labor	ntery Animals	٦
Species	Concentration (ppni)	Esposure Time	Effect"	Rekrence	1
Rabbiz	150,000 250,000	25 min unknowa	No monality Mortality	Larienov w al. (1934)	1
lõvinak pig	50,000 69,000 89,000 200,000 200,000	12 h 2 b 10 h 31 min 1 h	3/5 deaths 195% environ 100% environ 100% environ 100% environ 1/5 deathe	GR193 (1943)	
Rai	50,000 39,000 89,000 209,000	Հ4 հ հե 19 հ 30 թ.թ.	100% surviva) 100% survival 5/7 deaths 2/5 deaths	BRPG (1997)	
Rat	79.009 124.000 207.090	46	ենդ ենտ ենտ	Shuges = /1969)	1
Rai	2760-4000	ISh	(V) denthe	Kreiting et al. (1987)	
Mouse	10,000	2 h	100% pervexi	Buches er at (1993)	
Moure	91,000 122,0%0 169,0%6	22	ենը։ ԼՀա ԵՇա	Shugar v (1969)	
Moure	2000-4000	13 h	<b>Hithaliry</b>	Kreibug et al. (1987)	

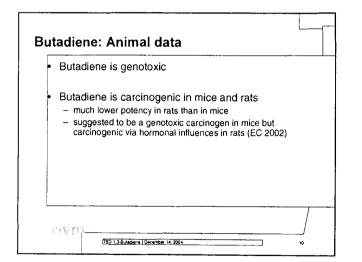
Nor	eye effects in rabbits and dogs exposed to 6700
	7.5 h/d for 8 months
No r	arcosis in rabbits exposed to 150,000 ppm for
25 n	nin (250,000 ppm was lethal)
	espiratory stress in pregnant female rats
expo	osed to 7647 ppm (6 h/d)
	pnea during the first 30 min in mice exposed to
498	) ppm (6 h/d for 5 d)

3

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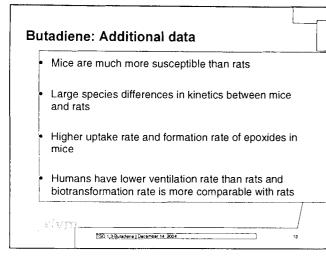
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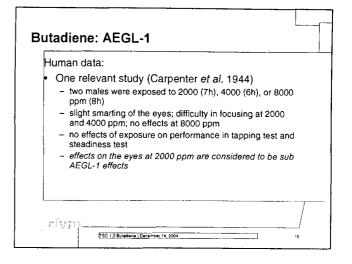
But	adie	ne: Developmental toxicity	
•	Rats (Irish – no – sm	exposed up to 7647 ppm for 6 h/d on day 6-7 et al. 1981) effects on pregnancy or implantation naller fetuses with skeletal effects (wavy ribs) due to aternal growth retardation (predominantly at 7647 ppm	
•	day 6 rat	and mice exposed up to 1000 ppm for 6 h/d 6-15 (Hackett <i>et al.</i> 1987) Is: maternal growth retardation; no exposure-related fe fects	
	an	ce: maternal growth retardation; reduced fetal weight d minor skeletal abnormalities at 200 and 1000 ppm b	out
	no rivn	1 at 40 ppm	U

ur in the presence
due to single
single exposures)

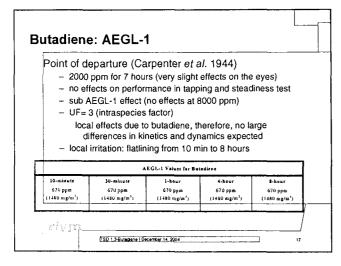


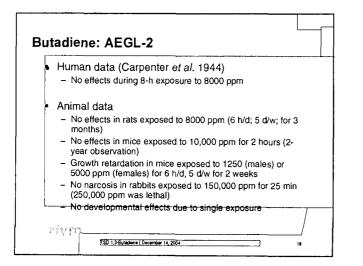
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C	ata availability
	Acute toxicity of 1,3-butadiene is low, even in mice
	Limited human data available
	TSU 1.3 Dutadiene   Gecember 14, 2004



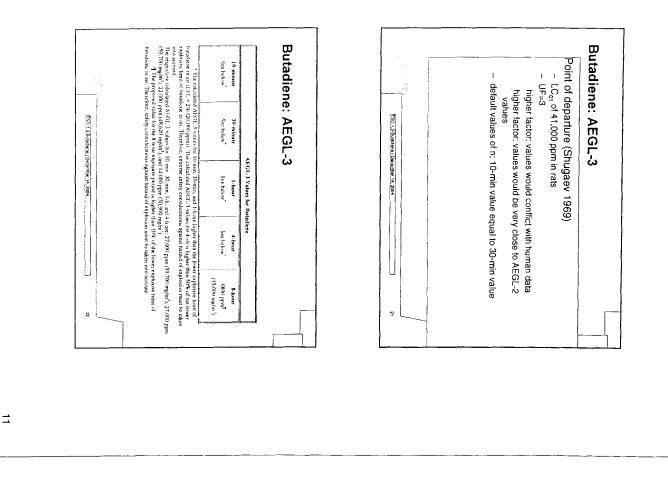
4	nimal data
	No eye effects in rabbits and dogs exposed to 6700 ppm 7.5 h/d for 8 months
•	No narcosis in rabbits exposed to 150,000 ppm for 25 min
•	No respiratory stress in pregnant female rats exposed to 7647 ppm (6 h/d)
•	Dyspnea during the first 30 min in mice exposed to 4980 ppm (6 h/d for 5 d)
•	No effects in mice exposed to 10,000 ppm for 2 hours (2-year observation)





int or depar	ture (Croud	ch et al. 19	79)	
<ul> <li>No effects months)</li> </ul>	in rats expos	ed to 8000 p	pm (6 h/d; 5	d/w; for 3
– UF=3				
highest o	concentration	tested		
repeated	l exposure			
higher fa	ictor: values v	would conflict	with humar	data
	lues of n: 10-			
		AEGL-2 Values for But	adiene	
	30-minute	1-hour	4-hour	6-hour

Butadiene: AEGL-3	
No adequate human data	
<ul> <li>Animal data: Shugaev (1969)         <ul> <li>mice: 2-h LC<sub>50</sub> of 122,000 ppm</li> <li>rats: 4-h LC<sub>50</sub> of 128,000 ppm (LC<sub>01</sub>: 41,000 ppm)</li> </ul> </li> </ul>	
riym	
[SD 1.3Builadens   December 14, 2004	20



8-hour 670 ppm 2000 ppm 3-2000 ppm 6800 ppm 7800	STATUTE STATUTE CONTRACT OF STATE	The respective calculated AEGL 3 values for 10-min, 30-min, 1-b, and 4-h are: 27.000 ppm (59.700 mg/m <sup>2</sup> ) 27.000 ppm (59.700 mg/m <sup>2</sup> ), 22.000 ppm (48.620 mg/m <sup>2</sup> ), and 1-0.000 ppm (59.500 mg/m <sup>2</sup> ). 97 the proposed value is righter than (10-46 to lower explores time of hundeter in an: Therefore, safety considerations Against hazved of Explorion must be taken non-account.	<ul> <li>The calculated AFGL's values for thoma 30 own and 1-have higher than the hover explosive that of bundlers an util LL = 3% (2000 ppm)). The calculated AFGL's values for 4-1 is higher than 5% of the lower explosive than of bundlers in an Therefore, carrente and/sy considerations spanint hard of explosion must be here non-account.</li> </ul>	AEGL.3 See below See below See below See below See below	AEGL-2 6100 ppm <sup>3</sup> 6100 ppm <sup>3</sup> 4900 ppm <sup>3</sup> 3100 ppm <sup>3</sup>	AEGL.1 670 ppm 670 ppm 670 ppm 670 ppm 670 ppm	Classification 10-minute 30-minute 1-hour 4-hour	Exposure Duration	Summary of AECI, Values	Butadiene: Summary of AEGL-values
		n (59,700 mg/m <sup>4</sup> ). ). 3n ant. Therefore, safe	ower explosive firmit o or than 50% of the low of explosion must he		րա <sup>1</sup> 2000 չիա					lues

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## Table 1 - Chemical and Physical Properties #//

Parameter	Value	Reference
Synonyms	DMA;	Bingham et al.,
~ j	<i>N</i> -methylmethanamine;	2001
	methanamine N-methyl	
Chemical	C <sub>2</sub> H <sub>7</sub> N	Bingham et al.,
formula		2001
Molecular	45.08	Bingham et al.,
weight		2001
CAS Reg. No.	124-40-3	Bingham et al.,
		2001
Physical state	Colorless gas	Steinhagen et al.,
		1982
Solubility in	Very soluble	Bingham et al.,
water		2001
Vapor pressure	2 atm	Bingham et
		al.,2001
Vapor density	1.55	Bingham et
(air =1)	*	al.,2001
Liquid density	0.6804 g/mL	Bingham et
(water =1)		al.,2001
Melting point	-93 °C	Bingham et al.,
		2001
Boiling point	7.4 °C	Bingham et al.,
		2001
Odor threshold	1.6 ppm; smell of rotting	Bingham et al.,
	fish	2001
Flammability	2.8 through 14.4%	www.east-
limits	2	westglobal.com
Conversion	$1 \text{ ppm} = 1.84 \text{ mg/m}^3$	Bingham et al.,
factors		2001

# Table 2 - Main Parameters of Acute Inhalation Toxicity ofDimethylamine

Species	Concentration (ppm)	Exposure Time (hours)	Effect	References
Rats	4,700	4	LC <sub>50</sub>	Koch et al., 1980
Rats	4,540	6	LC <sub>50</sub>	Steinhagen et al., 1982
Mice	4,725	2	LC <sub>50</sub>	Mezentseva, 1956

# Table 3 - Summery of Nonlethal Inhalation Data onLaboratory Animals

Species	Concentration (ppm)	Exposure Time	Effect	References
Rats	573	10- min	RD <sub>50</sub>	Steinhagen et al., 1982
Mice	511	10-min	RD <sub>50</sub>	Steinhagen et al., 1982
Rats	175	6-hour	Discharge of modified mucus	Gross et al., 1987
Rats / Mice	100	10-min	Mild levels of discomfort	Steinhagen et al., 1982

to « lek let stampare sitte late • Laser e copiateri questo late « let d'enere inpressian antes laser e Laser e Copieur sur cette face » lek det stampare sitte late • Laser e Table 4 - AEGL-1 Values for Dimethylamine

10-minute	30-minute	1-hour	4-hour	8-hour
3.3 ppm				
$6.1 (mg/m^3)$				

### **Derivation of AEGL-1**

Key Study: Toxicity Endpoint:	Steinhagen et al. (1982). Single Intake of DMA in at Concentrations of 100 ppm and Higher for 10 minutes in Rats and Mice caused minimal reversible changes determined as "MILD" LEVELS OF DISCOMFORT (hyperemia in the mucous and discharge from nasal routs) in the absence of any pathomorphological modifications. Increase in level of single exposure to the substance at concentrations of 175 – 200 ppm caused a wide spectrum fo disorders from epithelial vacuolization to reversible ulceration or chronic inflammation. Secretion of modified mucus was observed (Gross et al., 1987; Steinhagen et al.,
Uncertainty Factors:	<ul> <li>1982).</li> <li>Lowering the levels of chronic DMA inhalation exposure of rats to (10 and 30 ppm) did not lead to any histopathological changes (CIIT, 1982).</li> <li>In order to account for interspecies variability of DMA induced rhinitis an uncertainty factor of 10 was used. UF 3 was used to account for intraspecies variability. Based on the fact that in deriving AEGL-1 the starting point was reversible rhinitis in rats in rats exposed to DMA at 100 ppm concentration for 10 minutes, further lowering the values seems to be unjustified.</li> </ul>
Scaling Process:	was not done
Time Scaling:	was not done
Calculations:	
<u>10-min AEGL-1</u>	100  ppm/30 = 3.3  ppm
<u>30-min AEGL-1</u>	100  ppm/30 = 3.3  ppm
<u>1-hour AEGL-1</u>	100  ppm/30 = 3.3  ppm
4-hour AEGL-1	100  ppm/30 = 3.3  ppm 100  prm/20 = 2.2  prm
8-hour AEGL-1	100  ppm/30 = 3.3  ppm

### Table 5 - AEGL-2 Values for Dimethylamine

10-minute	30-minute	1-hour	4-hour	8-hour
19.3 ppm	13.4 ppm	10.6 ppm	6.7 ppm	4.4 ppm
$35.6(mg/m^3)$	$24.7 \text{mg/m}^3$ )	$19.5(mg/m^3)$	$12.4(mg/m^3)$	$8.1(mg/m^3)$

### **Derivation of AEGL-2**

Key Study: Toxicity	Gross et al. (1987). Single 6-hour exposure of male rats to dimethylamine at the concentration of 175 ppm caused a wide spectrum of disorders from epithelial vacuolization to ulceration and acute or chronic inflammation. No irreversible histological disorders were observed. Increase of DMA chronic inhalation exposure level up to 185
Endpoint:	ppm concentration caused clinical signs in the form of central lobular degeneration (Hollingsworth, R. L. and Rowe, V. K., 1964). Lowering the levels of chronic DMA inhalation exposure of
	rats to (10, 30, and 100 ppm) did not lead to any histopathological changes (CIIT, 1982).
Uncertainty	In order to account for interspecies variability of DMA
Factors:	induced rhinitis, erosion of anterior edges, and fenestration of limiting layer an uncertainty factor of 10 was used. UF 3 was used to account for intraspecies variability.
Scaling process:	$C^{1} \times t = k$ (ten Berge et al., 1986); 175 ppm/30 = 5,833 ppm $C^{3} * t = k$ ; (5,833 ppm) <sup>3</sup> * 360 min = 71,458.333 ppm <sup>3</sup> * min $C^{1} * t = k$ ; 5,833 ppm * 360 min = 2,099.88 ppm * min
Time scaling:	The relation of exposure concentration and exposure time for most irritants and for vapors and gases of systemic action can be described as follows: $C^{n} * t = k$ , where the exponent n varies from 0.8 to 3.5 (ten Berge et al, 1986). Using n=3 for cases of extrapolation on a shorter exposure time and n=1 is used for extrapolation on a longer exposure.
<b>Calculations:</b>	
<u>10-min AEGL-1</u>	$C_{2}^{3} * 10 \text{ min} = 71458.333 \text{ ppm}_{2}^{3} * \text{ min}; C = 19.3 \text{ ppm}_{2}^{3}$
<u>30-min AEGL-1</u>	$C_{2}^{3} * 30 \text{ min} = 71458.333 \text{ ppm}_{2}^{3} * \text{ min}; C = 13.4 \text{ ppm}_{2}^{3}$
<u>1-hour AEGL-1</u>	$C_{2}^{3} * 60 \text{ min} = 71458.333 \text{ ppm}^{3} * \text{ min}; C = 10.6 \text{ ppm}$
4-hour AEGL-1	$C^{3} * 240 \text{ min} = 71458.333 \text{ ppm}^{3} * \text{min}; C = 6.7 \text{ ppm}$
<u>8-hour AEGL-1</u>	$C^{1} * 480 \text{ min} = 2099.88 \text{ ppm} * \text{min}; C = 4.4 \text{ ppm}$

· eet a certie impression cente tere - Laser of Lopinst sur cette lace o the det structure tere - Laser e ceptatori questo late o det a'onstie impres Table 6 - AEGL-3 Values for Dimethylamine

10-minute	30-minute	1-hour	4-hour	8-hour
275.2 ppm	190.8 ppm	151.4  ppm	95.4 ppm	62.5 ppm
$507.4(mg/m^3)$	$351.8(mg/m^3)$	$279.1(mg/m^3)$	$175.9(mg/m^3)$	115.2(mg/r

### **Derivation of AEGL-3**

Key Study:	Steinhagen et al., 1982 was based on determining the threshold by the lethal effect of 2,500 ppm followed by severe injuries in respiratory tract and a number of other internal organs.			
Toxicity Endpoint:	Increase of exposure level to the substance up to 2,800 ppm causes minimal lethality effect; while DMA exposure at 4,540 ppm concentration is the mean lethal level (Steinhagen et al., 1982).			
Uncertainty	Based on $LC_{50}$ values an uncertainty factor of 10 was used to			
Factors:	account for interspecies variability of DMA induced toxicity. Intraspecies variability was limited by a factor of 3 since in majority of individuals exposure of respiratory tract and a number of internal organs is manifested in similar boundaries and with similar sensitivity. Based on the fact that in deriving AEGL values non-lethal toxic effect followed by severe injuries in respiratory tract and number of internal organs was the			
Scaling process:	starting point further decrease of the values was not justified. $C^{1} x t = k$ (ten Berge et al., 1986); 2,500 ppm/30 = 83.33 ppm $C^{3} * t = k$ ; ( $83.33$ ppm) <sup>3</sup> * 360 min = 208308334.3 ppm <sup>3</sup> min $C^{1} * t = k$ ; 83.33 ppm * 360 min = 29998.8 ppm * min			
Time scaling:	Relation of concentration and exposure time for most irritants and for vapors and gases with systemic action can be described as follows: $C^{n} * t = k$ , where the exponent n varies from 0.8 to 3.5 (ten Berge et al., 1986). Due to absence of specific data intermediary scaling was done by formula $C^{n} * t = k$ with use of n=3 for extrapolation cases on shorter exposures; and n=1 was used for extrapolation on shorter exposure.			
Calculations:				
<u>10-min AEGL-1</u> <u>30-min AEGL-1</u> <u>1-hour AEGL-1</u> <u>4-hour AEGL-1</u> <u>8-hour AEGL-1</u>	$\begin{array}{lll} C^3 * 10 \mbox{ min} = 208308334.3 \mbox{ ppm}^3 * \mbox{ min}; & C = 275.2 \mbox{ ppm} \\ C^3 * 30 \mbox{ min} = 208308334.3 \mbox{ ppm}^3 * \mbox{ min}; & C = 190.8 \mbox{ ppm} \\ C^3 * 60 \mbox{ min} = 208308334.3 \mbox{ ppm}^3 * \mbox{ min}; & C = 151.4 \mbox{ ppm} \\ C^3 * 240 \mbox{ min} = 208308334.3 \mbox{ ppm}^3 * \mbox{ min}; & C = 95.4 \mbox{ ppm} \\ C^1 * 480 \mbox{ min} = 29998.8 \mbox{ ppm} * \mbox{ min}; & C = 62.5 \mbox{ ppm} \end{array}$			

### Table 7 - Extant Standards and Guidelines for Dimethylamine

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Exposure	Exposure Duration				
Duration	10-minute	10-minute	10-minute	10-minute	10-minute
	3.3 ppm	3.3 ppm	3.3 ppm	3.3 ppm	3.3 ppm
AEGL-1	$6.1 (mg/m^3)$	$6.1 (mg/m^3)$	$6.1 (mg/m^3)$	$6.1 (mg/m^3)$	$6.1 (mg/m^3)$
	19.3 ppm	13.4 ppm	10.6 ppm	6.7 ppm	4.4 ppm
AEGL-2	$35.6(mg/m^3)$	$24.7(mg/m^3)$	$19.5(mg/m^3)$	$12.4(mg/m^3)$	$8.1 (mg/m^3)$
	275.2 ppm	190.8 ppm	151.4 ppm	95.4 ppm	62.5 ppm
AEGL-3	$507.4(mg/m^3)$	$351.8(mg/m^3)$	$279.1(mg/m^3)$	$175.9(mg/m^3)$	$115.2(mg/m^3)$
ERPG-1			0.6 ppm		
(AIHA)					
ERPG-2					
(AIHA)			100 ppm		
ERPG-3					
(AIHA)			350 ppm		
EEGL				Ň	
(NRC)					
PEL-TWA					10 ppm
(OSHA)					$(18 \text{ mg/m}^3)$
PEL-STEL					
(OSHA)					10 ppm
IDLH					
(NIOSH)		2,000 ppm			
REL-TWA					10ppm
(NIOSH)		· • •	<u> </u>		$(18 \text{ mg/m}^3)$
REL-STEL					
(NIOSH)					
TLV-TWA					5 ppm
(ACGIH)					$(9.2 \text{ mg/m}^3)$
TLV-STEL	15 ppm				
(ACGIH)	$(27.6 \text{mg/m}^3)$				
MAK Peak	2 ppm				
Limit	$(3.7) \text{ mg/m}^3$				
(Germany)					
MAC					1 ppm
(Nederland's)					$(1.8) \text{ mg/m}^3$ .

### ACUTE EXPOSURE GUIDELINE LEVELS (AEGLs) FOR ETHYL MERCAPTAN

NAC/AEGL-35 December 13-15, 2004 Washington, D. C.

**ORNL Staff Scientist: Cheryl Bast** 

Chemical Manager: Iris Camacho

**Chemical Reviewers: Steve Barbee and George Rusch** 

**Mechanism of Toxicity** 

Acts similarly to hydrogen sulfide, methyl mercaptan and cyanide

Interrupts electron transport through inhibition of cytochrome oxidase

**Relative Toxicity (Rat Lethality Data)** 

Acute toxicity of ethyl mercaptan is much less than that of

Methyl mercaptan

Hydrogen sulfide

Acute toxicity of methyl mercaptan is similar to or slightly less than that of hydrogen sulfide

4-Hour Rat LC <sub>50</sub> Values (Tansy et al., 1981)			
Ethyl Mercaptan 4740 ppm			
Methyl Mercaptan 675 ppm			
Hydrogen Sulfide 444 ppm			

AEGL-1 VALUES: ETHYL MERCAPTAN					
10 minute 30 minute 1 hour 4 hour 8 hour					
1 ppm 1 ppm 1 ppm 1 ppm 1 ppm					

Species:RabbitConcentration:10 ppmTime:20 minutesEndpoint:NOEL for IrritationReference:Shibata, 1966b

Time Scaling: Values held constant across time because minor irritation does not vary greatly over time.

**Uncertainty Factors:** 

Interspecies = 3

Intraspecies = 3

Use of the full factor of 10 for either interspecies or intraspecies variability would yield AEGL-1 values  $\leq 0.3$  ppm which is inconsistent with the available human data. No mucosal irritation was noted in humans exposed to 0.4 ppm ethyl mercaptan 3 hours/day for 5 or 10 days (Blinova,1965).

<b>AEGL-2 VALUES: ETHYL MERCAPTAN</b>						
10 minute	30 minute	1 hour	4 hour	8 hour		
150 ppm	150 ppm	120 ppm	77 ppm	37 ppm		

**Endpoint:** 

Three-fold reduction of AEGL-3 values. Estimated threshold for the inability to escape.

**Reference:** 

Time Scaling: See AEGL-3 derivation.

**Uncertainty Factors:** 

Interspecies = 3 See AEGL-3 justification.

Intraspecies = 3 See AEGL-3 justification.

	AEGL-3 VAL	LUES: ETHYL N	IERCAPTAN	
10 minute	30 minute	1 hour	4 hour	8 hour
450 ppm	450 ppm	360 ppm	230 ppm	110 ppm

Species:	Mouse
<b>Concentration:</b>	2250 ppm
Time:	4 hours
Endpoint:	LC <sub>01</sub> (Estimated threshold for death. Used instead of BMCL <sub>05</sub>
	for consistency with methyl mercaptan)
Reference:	Fairchild and Stokinger, 1958
Time Scaling:	$c^n x t = k$ , where the exponent, n, is the conservative default of 1 (8-hr) or 3 (30-min, 1-hr, 4-hr). The 30-min value is adopted as the 10-min AEGL-3 value.

**Uncertainty Factors:** 

Interspecies = 3 The mouse is the most sensitive species.

Intraspecies = 3 Considered sufficient due to the steepness of the lethal response curve which implies limited individual variability.

<u>Mouse (4-hr):</u> 2600 ppm: 40% lethality 2770 ppm: LC<sub>50</sub> 3573 ppm: 100% lethality <u>Rat (4-hr):</u> 3808 ppm: LC<sub>01</sub> 4740 ppm: LC<sub>50</sub>

**Support for Total UF of 10:** 

A total UF of 30 would yield AEGL-3 values inconsistent with the total data set.

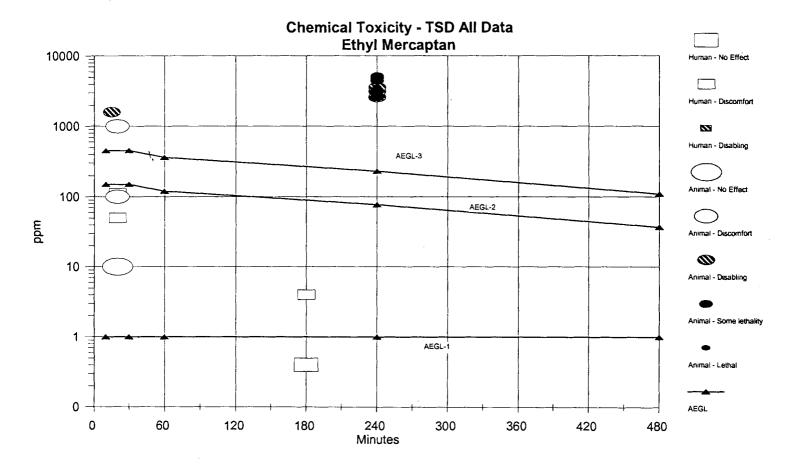
Values would approach AEGL-3 values derived for hydrogen sulfide 8-hr H<sub>2</sub>S AEGL-3 = 31 ppm 8-hr ethyl mercaptan AEGL-3 would be 37 ppm

Ethyl mercaptan is less toxic than hydrogen sulfide, [the 4-hour rat  $LC_{50}$  value for ethyl mercaptan was 4740 ppm. The 4-hour  $LC_{50}$  value for hydrogen sulfide was 444 ppm]

A 30-minute AEGL-3 of 150 ppm would be derived. Humans exposed to 112 ppm ethyl mercaptan for 20 minutes exhibited only a slightly irregular, and decreased breathing rate.

Extant Standards and Guidelines for Ethyl Mercaptan							
	Exposure Duration						
Guideline	10 minute	30 minute	1 hour	4 hour	8 hour		
AEGL-1	1 ppm	1 ppm	1 ppm	1 ppm	1 ppm		
AEGL-2	150 ppm	150 ppm	120 ppm	77 ppm	37 ppm		
AEGL-3	450 ppm	450 ppm	360 ppm	230 ppm	110 ppm		
NIOSH IDLH		500 ppm					
NIOSH REL	0.5 ppm						
OSHA PEL					10 ppm (ceiling)		
ACGIH-TLV TWA					0.5 ppm		
MAK (German)					0.5 ppm		
MAC (Dutch)					0.5 ppm		

.



The odor detection threshold  $(OT_{50})$ : 0.00076 ppm (Amoore and Hautala (1983).

The concentration (C) leading to an odor intensity (I) of distinct odor detection (I=3) is derived using the Fechner function:

 $I = k_w x \log (C / OT_{50}) + 0.5$ 

For the Fechner coefficient, the default of  $k_w = 2.33$  will be used due to the lack of chemical-specific data:

 $3 = 2.33 \text{ x} \log (C / 0.00076) + 0.5$  which can be rearranged to  $\log (C / 0.00076) = (3 - 0.5) / 2.33 = 1.07$  and results in  $C = (10^{1.07}) \text{ x} 0.00076 = 0.0089 \text{ ppm}$ 

LOA = C x 1.33 = 0.0089 ppm x 1.33 = 0.011837 ppm

The LOA for ethyl mercaptan is 0.012 ppm.

## ACUTE EXPOSURE GUIDELINE LEVELS (AEGLs)

### FOR

### NITROGEN MUSTARDS (HN1 CAS Reg. No. 538-07-8) (HN2 CAS Reg. No. 51-75-2) (HN3 CAS Reg. No. 555-77-1)

December 13-16, 2004

Nitrogen Mustards - Nonlethal Toxicity in Humans

- eyes are sensitive target [similar to agent HD (sulfur mustard)]
- respiratory tract effects possible but not reported at exposures inducing ocular effects

Es	timated effect threshol	ds in humans expose	d to nitrogen mustard vapors.
HN1	HN2	HN3	Effect
-	0.012 mg-min/m <sup>3</sup>	-	No observable effect level during therapeutic use of HN2 (Van Vloten et al., 1993)
90 mg-min/m <sup>3</sup>	70 mg-min/m <sup>3</sup>	42 mg-min/m <sup>3</sup>	Moderate but reversible ocular effects (Porton report, 1942a, 1943d; U.S. Army Med. Div., 1945c,d; NDRC, 1946)
>21,170 mg- min/m <sup>3</sup>	5800 mg-min/m <sup>3</sup>	1800 mg-min/m <sup>3</sup> 1300 mg-min/m <sup>3</sup>	Median blistering Ct (10-min or 20-min exposure) for normal skin Median blistering Ct (20-min exposure) for sweating skin (NDRC, 1944)

# Nitrogen Mustards - Lethal Toxicity in Humans

No quantitative data regarding lethal toxicity of HN

# Nitrogen Mustards - Nonlethal Toxicity in Animals

Studies in animals focused on lethality; no information available regarding nonlethal effects

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### Nitrogen Mustards - Lethal Toxicity in Animals

- LCt<sub>50</sub> values for multiple species; various concentrations and durations
  - **HN1**

Monkey:	1500 mg-min/m <sup>3</sup>
Dog:	800 mg-min/m <sup>3</sup>
Rat:	750-1200 mg-min/m <sup>3</sup>
Mouse:	900-1300 mg-min/m <sup>3</sup>
Rabbit:	900->4000 mg-min/m <sup>3</sup>
Cat:	400 mg-min/m <sup>3</sup>
Guinea pig:	1500-3000 mg-min/m <sup>3</sup>

• **HN2** 

Dog:	2000 mg-min/m <sup>3</sup>
Rat:	600-4000 mg-min/m <sup>3</sup>
Mouse:	1500-7000 mg-min/m <sup>3</sup>
Rabbit:	1000-8000 mg-min/m <sup>3</sup>
Guinea pig:	>1200-8000 mg-min/m <sup>3</sup>

### Nitrogen Mustards - Lethal Toxicity in Animals

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• **HN3** 

Dog:	400-1500 mg-min/m <sup>3</sup>
Rat:	670-1700 mg-min/m <sup>3</sup>
Mouse:	165-600 mg-min/m <sup>3</sup>
Rabbit:	500-3000 mg-min/m <sup>3</sup>
Cat:	400 mg-min/m <sup>3</sup>
Guinea pig:	>1000->23000 mg-min/m <sup>3</sup>

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# Nitrogen Mustards - Data Evaluation/Study Selection Criteria

- analytical vs nominal exposure concentrations
- exposure duration data
- number of animals
- post-exposure observation period
- environmental conditions (temp., humidity)
- species sensitivity

### **Nitrogen Mustards - Special Considerations**

- Metabolism/Disposition
  - dermal penetration of HN vapor
    - linear with time
    - enhanced with increasing temperature & humidity

- Mechanism of action
  - formation of immonium ion which is reactive with nucelophiles

- all HN alkylators
- precise mechanism unclear

	Summary of	AEGL Values for	r Nitrogen Mus	tards (mg/m <sup>3</sup> )	
Classification	10-minute	30-minute	1-hour	4-hour	8-hour
AEGL-1					······································
Nondisabling)					
HN1	NR <sup>a</sup>	NR <sup>a</sup>	NR <sup>a</sup>	NR <sup>a</sup>	NRª
HN2	NR <sup>a</sup>	NR <sup>a</sup>	NR <sup>a</sup>	NR <sup>a</sup>	NR <sup>a</sup>
HN3	NR <sup>a</sup>	NR <sup>a</sup>	NR <sup>a</sup>	NR <sup>a</sup>	NR <sup>a</sup>
AEGL-2	· · · · · · · · · · · · · · · · · · ·		· · · · ·		
(Disabling)					
HN1	0.90	0.30	0.15	0.038	0.019
HN2	0.55	0.18	0.092	0.023	0.011
HN3	0.42	0.14	0.070	0.018	0.0088
AEGL-3	· · · · · · · · · · · · · · · · · · ·				
(Lethality)			,		
HN1	1.8	0.96	0.48	0.12	0.060
HN2	1.3	0.88	0.70	0.28	0.14
HN3	2.2	0.74	0.37	0.093	0.047

	AE	<b>GL-2 VALUES FOR</b>	HN1	· ····
10 minutes	30 minutes	1 hour	4 hours	8 hours
0.90 mg/m <sup>3</sup>	0.30 mg/m <sup>3</sup>	0.15 mg/m <sup>3</sup>	0.038 mg/m <sup>3</sup>	0.019 mg/m <sup>3</sup>
Reference: Porton Ro 1946.	eport. 1943d. The effects of HN1 v	apour on human and rab	bit eyes. No. 2563. November	• 18, 1943. Cited in NDRC,
Test Species/Strain/S	ex/Number: Human volunteers/m	ales/21		
Exposure Route/Con minutes.	centrations/Durations: ocular exp	osure to vapors; CT deter	mined based upon exposure	durations of 5 to 67
Effects: Ocular irrita conjunctival injection	tion in human volunteer subjects; 1.	lacrimation, feeling of gr	ittiness in eyes, belpharospas	sm, photophobia,
Endpoint/Concentrat	tion/Rationale: 90 mg-min/m <sup>3</sup> base	ed upon exposure duration	1s of 5-67 minutes.	
Uncertainty Factors/ Total uncertainty fa Interspecies: Intraspecies:	ctor: 3 none; human subjects			d the result of direct-contact
	some of the tests were apparently ratory tract effects. Therefore, a r			
Animal to Human Do	osimetric Adjustment: Not applica	ble		
valu AEC not l min	the 10-min., 30-min, and 1-hr AE e of 90 mg-min/m <sup>3</sup> . The exposure GL time points) is uncertain and a be developed. Consistent with AE ute experimental exposure of 1.5 m osures of 0.38 mg/m <sup>3</sup> and 1.88 mg/	concentration-time relati n empirically-derived valu GL methodologies (NRC, ng/m <sup>3</sup> period to the 4-hour	onship for longer durations ( ie for the exponent, <i>n</i> , in the 2001), an <i>n</i> of 1 was used in	e.g., the 4-hr and 8-hr equation $C^n x t = k$ could extrapolating from the 60-

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Data Adequacy: The available data provide exposure-response data characterizing a sensitive critical effect in human volunteer subjects. The effect is consistent with the continuum of effects observed for this class of compounds. The data are considered appropriate for setting AEGL-2 values for HN1.

10 minutes	30 minutes	1 hour	4 hours	8 hours
<u> </u>	0.96 mg/m <sup>3</sup>	0.48 mg/m <sup>3</sup>	0.12 mg/m <sup>3</sup>	0.060 mg/m <sup>3</sup>
Reference: U.S. Army	Medical Division. 1945a. Med	ical Division monthly progr	ess report. September, 1945	. Cited in NRDC, 1946.
Test Species/Strain/Se	x/Number: 84 male rats			
	entrations/Durations: inhalatio chamber temp., 10-15 day obse		arations of 20-100 minutes/ a	nalytically determined
Effects: Lethality resp	onse data only			
	onse data only on/Rationale: Lethality thresho	ld of 287 mg-min/m <sup>3</sup> in rat	s estimated by 3-fold reducti	on of inhalation LCt <sub>50</sub> of 8
Endpoint/Concentrati mg-min/m <sup>3</sup> Uncertainty Factors/F	on/Rationale: Lethality thresho Rationale:	ld of 287 mg-min/m <sup>3</sup> in rat	s estimated by 3-fold reducti	on of inhalation LCt <sub>50</sub> of 8
Endpoint/Concentrati mg-min/m <sup>3</sup> Uncertainty Factors/F Total uncertainty fac	on/Rationale: Lethality thresho Rationale: etor: 10			
Endpoint/Concentrati mg-min/m <sup>3</sup> Uncertainty Factors/F	on/Rationale: Lethality thresho Rationale:	alues among seven species (	including nonhuman primat	
Endpoint/Concentrati mg-min/m <sup>3</sup> Uncertainty Factors/F Total uncertainty fac	on/Rationale: Lethality thresho Rationale: etor: 10 Limited to 3 because LCt <sub>50</sub> v	alues among seven species ( rat being somewhat more s direct action of nitrogen mu AEGL-3 values inconsistent	including nonhuman primat sensitive. Istards on tissue and because t with AEGL-2 values and av	es) did not appear to vary e additional downward vailable human data (ocula

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	AE	GL-2 VALUES FOR	HN2	
10 minutes	30 minutes	1 hour	4 hours	8 hours
0.55 mg/m <sup>3</sup>	0.18 mg/m <sup>3</sup>	0.092 mg/m <sup>3</sup>	0.023 mg/m <sup>3</sup>	0.011 mg/m <sup>3</sup>
Reference: Porton Report 1942. Cited in NDRC, 194	. 1942a. On the action of S or 6	n the eye; its comparison w	ith allied compounds and w	ith H. No. 2402. August 7,
Test Species/Strain/Sex/N	umber: Human male volunte	ers/number not specified		
Exposure Route/Concentr subjects wore oronasal ma	ations/Durations: 10-55 mg/n nsks	m <sup>3</sup> ; exposure durations of	0.5 min to 10 min.; Ct value	s of 40-55 mgpmin/m <sup>3</sup> ;
Effects: ocular irritation f	ollowing exposure (grittiness	in eyes; photophobia, belp	oharospasm; ocular pain).	
Endpoint/Concentration/H	Rationale: 55 mg-min/m <sup>3</sup> cons	sidered threshold for indu	cing military fine-skill opera	ational ineffectiveness
Intraspecies: 3			-	d the result of direct-couta
	e of the tests were apparently y tract effects. Therefore, a r			
Animal to Human Dosime	tric Adjustment: Not applica	able		
mg-min/r uncertain	0-min. AEGL-2, concentrations n <sup>3</sup> . The exposure concentrations and an empirically-derived at with AEGL methodologies	ion-time relationship for r value for the exponent, <i>n</i> ,	emaining AEGL-specific tin in the equation C <sup>n</sup> x t = k con	ne points durations is uld not be developed.
	ilable data provide exposure sistent with the continuum o EGL-2 values for HN2.			

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Time Scaling:	$C^n x t = k$ ; data were unavailable for empirical derivation of a scaling factor. The exposure concentration-time relationship for many irritant and systemically acting vapors and gases may described by $C^n x t = k$ , where the exponent <i>n</i> ranges from 0.8 to 3.5. In the absence of chemical-specific data, temporal scaling was performed using $n = 3$ when extrapolating to shorter time points and $n = 1$ when extrapolating to longer time points using the $C^n x t = k$ equation (NRC, 2001).
	For 10-min. AEGL-3: point-of-departure based upon estimated lethality threshold of 287 mg-min/m <sup>3</sup> resulting from 20-
	minute exposure (14.4 mg/m <sup>3</sup> )
	$(14.4 \text{ mg/m}^3)^3 \ge 20 \text{ min.} = 59,719 \text{ mg-min/m}^3$

Data Adequacy: The AEGL-3 values were based upon lethality assessment (analytically determined concentrations) using the most sensitive species exposed to high temperature conditions optimal for enhancing HN1 activity (i.e., worst-case scenario). A 10 to15-day post exposure observation period accounted for kown latency in toxic responses to HN1

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10 minutes	30 minutes	1 hour	4 hours	8 hours
1.3 mg/m <sup>3</sup>	0.88 mg/m <sup>3</sup>	0.70 mg/m <sup>3</sup>	0.28 mg/m <sup>3</sup>	0.14 mg/m <sup>3</sup>
Reference: Porton Rep February 9, 1943. Cited	ort. 1943b. Toxicity of S vapour I in NDRC, 1946.	•. Further experiments on	the exposure of animals to S	5 vapour. No. 2464.
Test Species/Strain/Sex	/Number: rat/gender not specif	ied/56		
Exposure Route/Conce exposures of 2000 mg-n	ntrations/Durations: inhalation nin/m <sup>3</sup>	/experimental exposure du	irations of 120-360 minutes	resulting in cumulative
Effects: Lethality only				
Endpoint/Concentratio mg-min/m <sup>3</sup> .	n/Rationale: Lethality threshol	d of 667 mg-min/m³ in rats	s estimated by 3-fold reduct	ion of $LCt_{50}$ of 2000
Uncertainty Factors/Ra				
Total uncertainty fact Interspecies:		luce among cours energies ()		
interspecies:	Limited to 3 because LCt <sub>50</sub> va by more than three-fold; the			es) and not appear to vary
Intraspecies:	Limited to 3 because of the d			e additional downward
	adjustment would result in A		t with AEGL-2 values and a for therapeutic use of nitrog	
	(ocular and dermai response	data and monitoring data		

Time Scaling:	$C^n x t = k$ ; data were unavailable for empirical derivation of a scaling factor. The concentration-time relationship for many irritant and systemically acting vapors and gases may described by $C^n x t = k$ , where the exponent n ranges from 0.8 to 3.5. In the absence of chemical-specific data, temporal scaling was performed using $n = 3$ when extrapolating to shorter time points and $n = 1$ when extrapolating to longer time points using the $Cn x t = k$ equation (NRC, 2001).
	For 10-min., 30-min, and 1-hr AEGL-3: point-of-departure based upon estimated lethality threshold of 667 mg-min/m <sup>3</sup> resulting from 120-minute exposure (5.56 mg/m <sup>3</sup> ) (5.56 mg/m <sup>3</sup> ) <sup>3</sup> x 120 min. = 20,625.6 mg-min/m <sup>3</sup>

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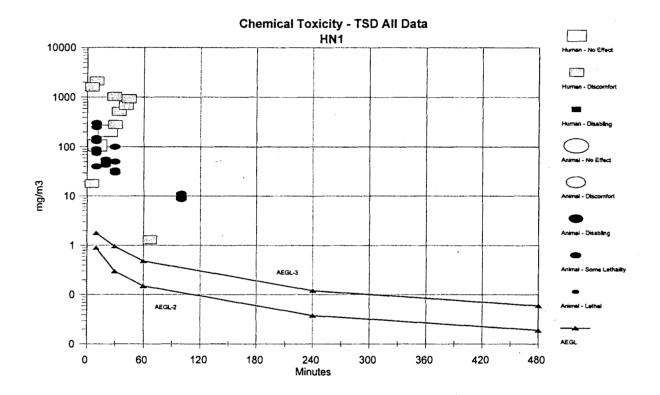
sensitive species. A 14-day post exposure observation period accounted for known latency in toxic responses to HN2.

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	AE	<b>GL-2 VALUES FOR</b>	HN3	
10 minutes	30 minutes	1 hour	4 hours	8 hours
0.42 mg/m <sup>3</sup>	0.14 mg/m <sup>3</sup>	0.070 mg/m <sup>3</sup>	0.018 mg/m <sup>3</sup>	0.0088 mg/m <sup>3</sup>
	y Medical Division. 1945c. Ń y Medical Division. 1945d. N			
Test Species/Strain/Sex/N	umber: Human volunteer sul	ojects/male/7		
Exposure Route/Concentr	ations/Durations: inhalation	/20-40 mg-min/m <sup>3</sup> ; 7 min.	· ·	
being reported by exposure to 40-m Endpoint/Concentration/J Uncertainty Factors/Ratio Total uncertainty factors Interspecies: hum Intraspecies: adju	ng-min/m <sup>3</sup> produced lacrimati Rationale:40-mg-min/m <sup>3</sup> cons onale:	ion, feeling of grittiness, p idered threshold for comp use the ocular response is	hotophobia, marked conjun promised task efficiency.	ctival injection
	e of the tests may have been ry tract effects. Therefore, a r			
Animal to Human Dosime	etric Adjustment: Not applica	ible		
empirically-derived value	re-time response relationship for the exponent, <i>n</i> , in the eq <u>is used in extrapolating from</u>	uation $C^n x t = k$ could no	t be developed. Consistent	with AEGL methodologies

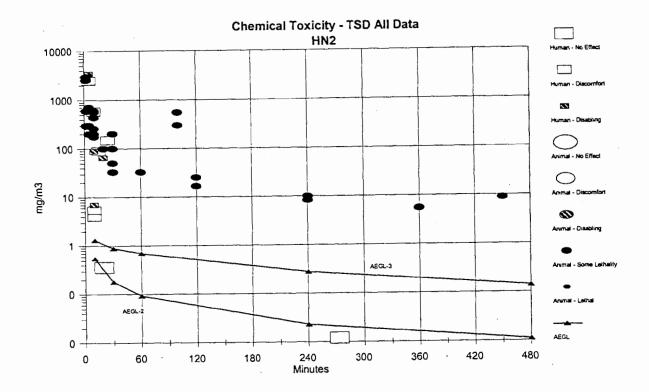
Furthermore, the critical effect is a conservative point-of-departure for AEGL-2 severity effects. The data are considered appropriate subjects. The effect is consistent with the continuum of effects observed for this class of compounds. Although the short exposure duration results in extensive extrapolation, an n of 1 was applied to provide more conservative exposure concentration estimates. Data Adequacy: The available data provide exposure-response data characterizing a sensitive critical effect in human volunteer for setting AEGL-2 values for HN3.

	AEC	GL-3 VALUES FOR H	IN3	
10 minutes	30 minutes	1 hour	4 hours	8 hours
2.2 mg/m <sup>3</sup>	0.74 mg/m <sup>3</sup>	0.37 mg/m <sup>3</sup>	0.093 mg/m <sup>3</sup>	0.047 mg/m <sup>3</sup>
Reference: Porton R	Report, 1943c. Toxicity and pa	athology of HN3. No. 2548.	November 18, 1944. Cited	in NDRC, 1946
Test Species/Strain/Sex/N	Number: 69 rats/gender not sp	pecified/exposure group		······································
Exposure Route/Concent	rations/Durations: inhalation	LCt <sub>50</sub> of 670 mg-min/m <sup>3</sup> ; of	exposure durations of 10-10	0 <u>min.</u>
Effects: Lethality respons	se data only			·
	Rationale: Lethality threshold Il exposure durations of 10-10		ats estimated by 3-fold redu	ction of $LCt_{50}$ of 670
Uncertainty Factors/Rati Total uncertainty factor				
Intraspecies:	Limited to 3 because LCt <sub>50</sub> va by more than three-fold; the I Limited to 3 because of the di adjustment would result in A (ocular and dermal response	rat being somewhat more s irect action of nitrogen mu EGL-3 values inconsistent	ensitive. stards on tissue and becaus with AEGL-2 values and a	e additional downward vailable human data
Modifying Factor: Not a	pplicable			
Animal to Human Dosim	etric Adjustment: Not applica	able	······	·····
	-departure concentrations for d value of 223.3 mg-min/m <sup>3</sup> .			n cumulative exposure
	The AEGL-3 values were bas most sensitive species and a c 15-day post exposure observa	hamber temperature (85°)	F) which would represent a	worst-case scenario. A

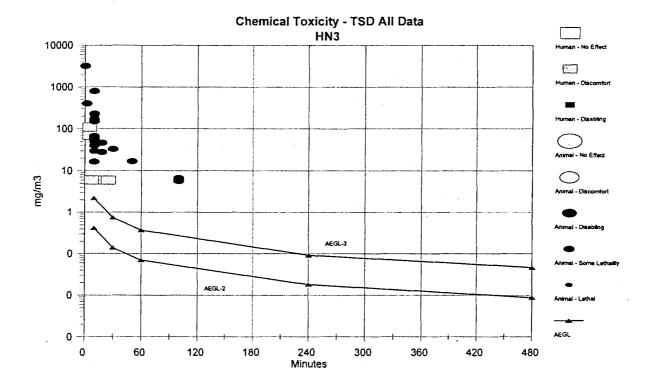


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### **Application of Acute Exposure Guideline Levels**

The Acute Exposure Level Guidelines have been developed primarily to provide guidance in situations were there can be a rare, typically accidental exposure to a particular chemical that can involve the general public. They, therefore, differ from PELs, TLV®s, WEEL®s, RELs or MAK values etc. in that they are based primarily on acute toxicology data and not subchronic or chronic data. The guidance therefore does not reflect the effects that could result from frequent exposure. Also, they are designed to protect the general population including the elderly and children, groups that are generally not considered in the development of workplace exposure levels. Users of the AEGL TSDs should first determine if there are legally enforceable standards that apply to the situation. Other organizations may also have recommended levels of exposure that more appropriately apply to the scenarios under evaluation.

It is however recognized that there may be an occasion where it may seem desirable to use these values for other exposure scenarios. In these cases, one should consult the technical support document. This document contains a comprehensive review of all identified acute toxicology data on the subject chemical and the basis for the development of the AEGL values. From this review one will have the information to determine the applicability of the AEGL to their particular situation.

### Presentation given to NAC/AEGL Committee on September 23, 2004 regarding language on AEGL definition

### Current main AEGL web page http://www.epa.gov/oppt/aegl/

The Development of Acute Exposure Guideline Levels (AEGLs) A collaborative effort of the public and private sectors worldwide

Acute Exposure Guideline Levels, or AEGLs, describe the dangers to humans resulting from short-term exposure to airborne chemicals. The National Advisory Committee for AEGLs is developing these guidelines to help both federal and local authorities, as well as private companies, deal with emergencies involving spills, or other accidental exposures.

### SOP

The AEGL Standard Operating Procedures section "Purpose and Objectives of the AEGL Program and the NAC/AEGL Committee" (page 21) states:

"The primary purpose of the AEGL program and the NAC/AEGL Committee is to develop guideline levels for once-in-a-lifetime, shortterm exposures to airborne concentrations of acutely toxic, high-priority chemicals."

### NEW DEFINITION FOR AEGL WEBSITE

Acute\* Exposure Guideline Levels are intended to describe the risk to humans resulting from once-in-a-lifetime, or rare, exposure to airborne chemicals. The National Advisory Committee for AEGLs is developing these guidelines to help both federal and local authorities, as well as private companies, deal with emergencies involving spills, or other catastrophic exposures.

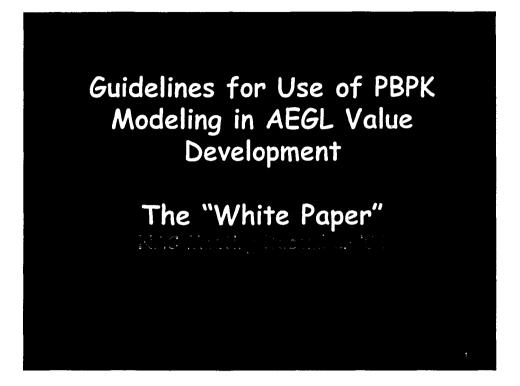
\*Definition = Acute exposures are single, non-repetitive exposures.

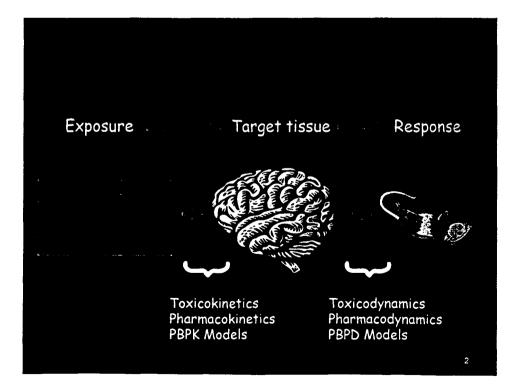
### Changes adopted to AEGL definition after collecting formal vote on September 23, 2004

Acute\* Exposure Guideline Levels, or AEGLs, are intended to describe the risk to humans resulting from once-in-a-lifetime, or rare, exposure to airborne chemicals. The National Advisory Committee for AEGLs is developing these guidelines to help both federal national and local authorities, as well as private companies, deal with emergencies involving spills, or other catastrophic exposures.

\*Definition = Acute exposures are single, non-repetitive exposures for not more than 8 hrs.  $\int_{n_c}^{n_c} \int_{n_c}^{n_c} \int_{n_c$ 

### ATTACHMENT 23



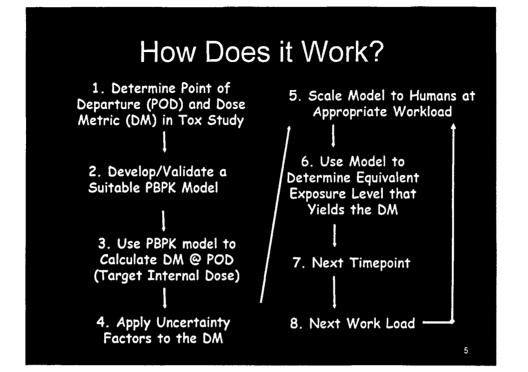


### Overview

- Until we can do all toxicology through computer simulation, we will have to extrapolate from animal studies
- Extrapolating using biological properties of the animals is the logical method
- We use UFs when we don't have an adequate biological description
- For some risk assessments, use of PBPK modeling is the default, use of UF is the backup

### Principal Advantages

- 1. Improve the quality of the risk assessment
- 2. Avoid the difficult issue of setting UF
- 3. For chemicals that PBPK-AEGL is higher than UF-AEGL, avoid issue of "running into OELs"
- 4. For chemicals where PBPK-AEGL is lower than UF-AEGL, people are protected
- 5. Confidence in AEGL is higher
  - Extrapolate from animal study to human
  - Extrapolate from one duration to another
  - Take exercise into account

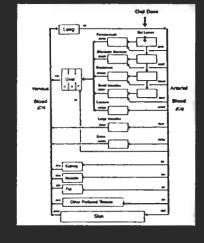


### An Example-Toluene: Step 1

AEGL	Study	Species	NOAEL	Duration
AEGL1	Weight-of-Evidence	Human	200 ppm	8 Hours
AEGL2	Gamberale et al., 1972	Human	700 ppm	20 Minutes
AEGL3	Mullin and Krivanek, 1982	Rat	6250 ppm	2 Hours

### Step 2: Develop PBPK Model

### Model Structure

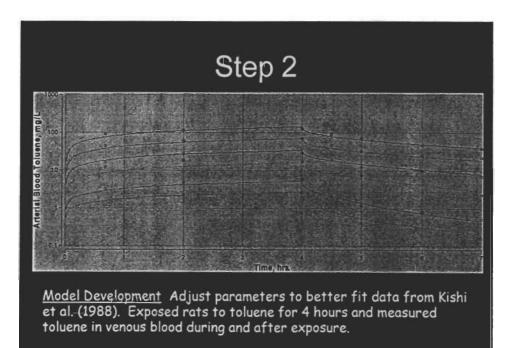


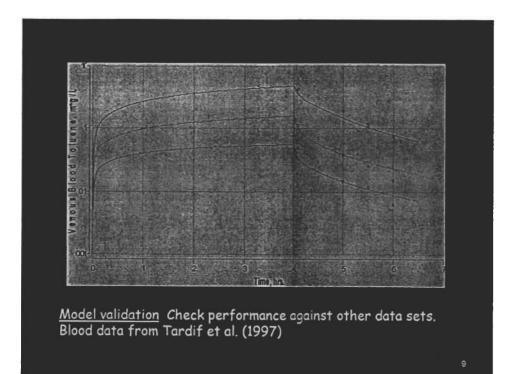
Model Equations, e.g.

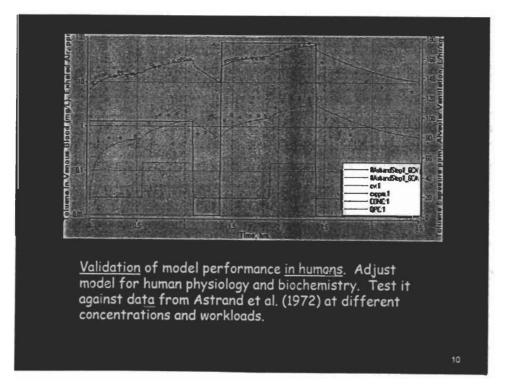
;Chemical in blood AB' = QP\*(CI-CX) + QC\*(CV-CA) INIT AB = 0 CA = AB/VB CV = (QF\*CVF + QR\*CVR + QL\*CVL + QS\*CVS)/QC

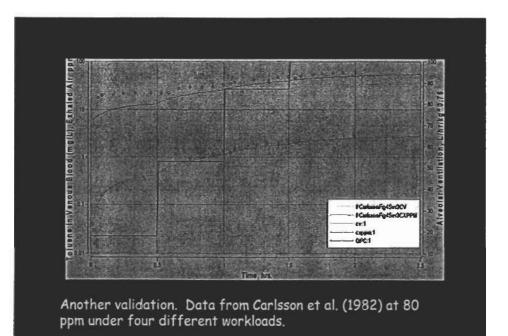
Model Parameter Values, e.g.

VL = .05\*BW BW = 70 kg

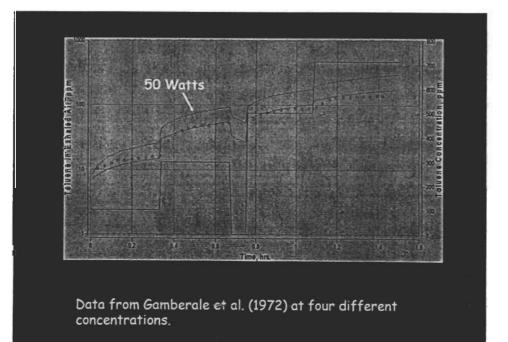












### **AEGL-2** Calculations

Step 1: POD: Gamberale et al. (1972) 700 ppm exposure for 20 minutes in humans

DM: Toluene in venous blood

Step 3: DM @POD = 6.5 mg toluene/L

- Step 4: Apply Uncertainty Factor (1)
- Step 6: Use Human PBPK Model to determine Exposure Concentrations to yield 6.5 mg/L at each timepoint

### How Much Difference Does PBPK Make?

AEGL	-1 Values I	Determined	with PBPF	K and ten B	erge, ppm
PBPK	10 min	30 min	1 hour	4 hour	8 hour
Rest	820	420	330	230	(200)
50W	410	230	160	110	100
75W	360	190	140	100	100
100W	320	170	120	100	90
Ten					
Berge	200	200	200	200	200
			- المر ج:		
					14

					<b>d ten Berge</b> hour 8	, ppm hour
Res 50V	st 1 N	.580 <b>810</b>	780	<b>590</b> 300	<b>410</b> 200	<b>350</b> 190
75\ 100 ten	)W			260 240	190 180	180 170
Bei		990	570	510	510	510
					T T T	
		posteros andress			numerid	15

<b>2440</b> 1430 1370
1270
1370
1350
1,500

### **Guidance Statements**

"... relevant PBPK data can be used to reduce uncertainty in extrapolation and risk assessment" (NRC 1987).

addressing Community Emergency Exposure Levels:

"If PBPK models for calculating delivered dose and cross-species extrapolation have been developed, the pharmacokinetic information should be incorporated into the quantitative risk assessments" (NRC 1993).

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### US EPA Guidance

"The optimal approach for extrapolating from one dose-duration response situation to another is the use of a physiologically based pharmacokinetic model (PBPK) model" (USEPA 2002).

Advocated PBPK modeling for setting AEGLs:

Krewski, et al. (2004) Bruckner et al. (2004) Simmons et al. (submitted)

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### **3 Stages of Consideration**

- Initial determination of PBPK modeling feasibility
- In-depth determination of model adequacyImplementation

### Initial determination of PBPK modeling feasibility

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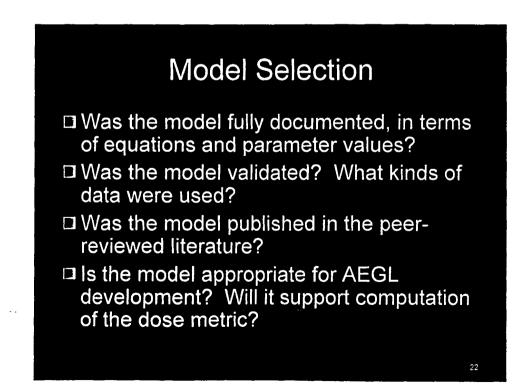
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D We don't want to waste resources

Do we think it will make a difference? Are there existing model(s)? Can the model "inform" the dose metric?

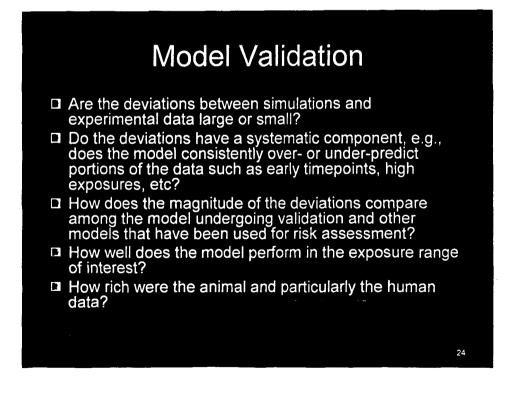
### In-depth determination of adequacy

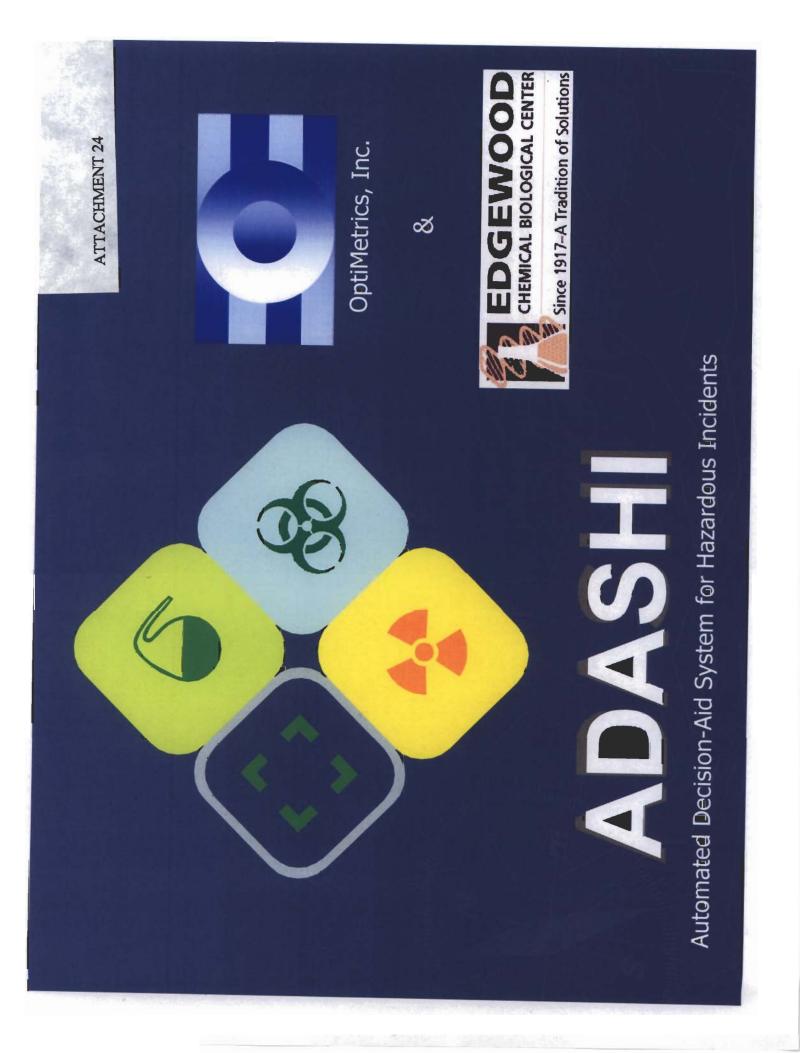
- Can we modify existing models for species of interest?
- Are there sufficient data available to validate the model(s)
- Is the model validated in the exposure range of interest?
- Do we have workload validation?
- The basic difference between this and the initial determination of adequacy is that here, we need to do some modeling to find out



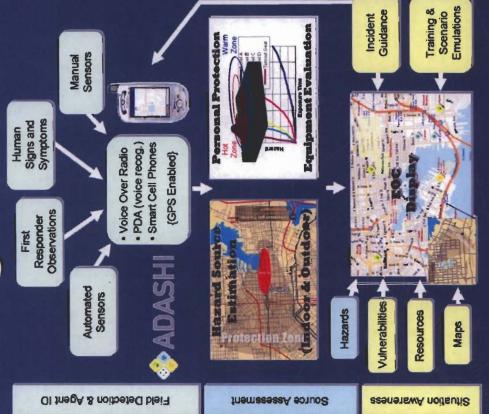
### **Data Selection**

- Do the data involve exposures in the range of interest (i.e., likely range of AEGL values);
- Do the data provide multiple concentrations in one set of studies;
- □ Are data from timecourse studies rather than a single timepoint;
- □ Are there data for more than one tissue;
- □ Are the data collected in the species of interest;
- Are there PBPK model parameters for the experimental species;
- Were body weights reported;
- □ Are exposure conditions clearly defined;
- ⊐ Is the route of exposure appropriate;
- ⊐ Do the data relate to the dose metric;
- Are there data from more than one laboratory;
- Are there data for exercising humans.





### integration with HPAC ITRANS ADASHI is ideally suited for



- ADASHI technology licensed from ECBC
- Commercialization Funded by OptiMetrics
- Designed to handle
- everyday TICs and TIMs
- Designed to handle infrequent WMD events
- Currently uses ALOHA and ERG for Hazard Area Estimation
- Requires no training

ADASHI's data acquisition, assessment,

and mitigation decision aid

# **Evolution of ADASHI Products**

### ADASHI First Response

- Laptop Application for On-Scene Situation Assessment
- Currently Available as Commercial Product for First Responders

### ADASHI Owl

- Situation Assessment, Data Collection and GPS Location Sensing Palmtop or Cell Phone Application for Individual Responder
- Initial Commercial Product Release Planned for Late 04

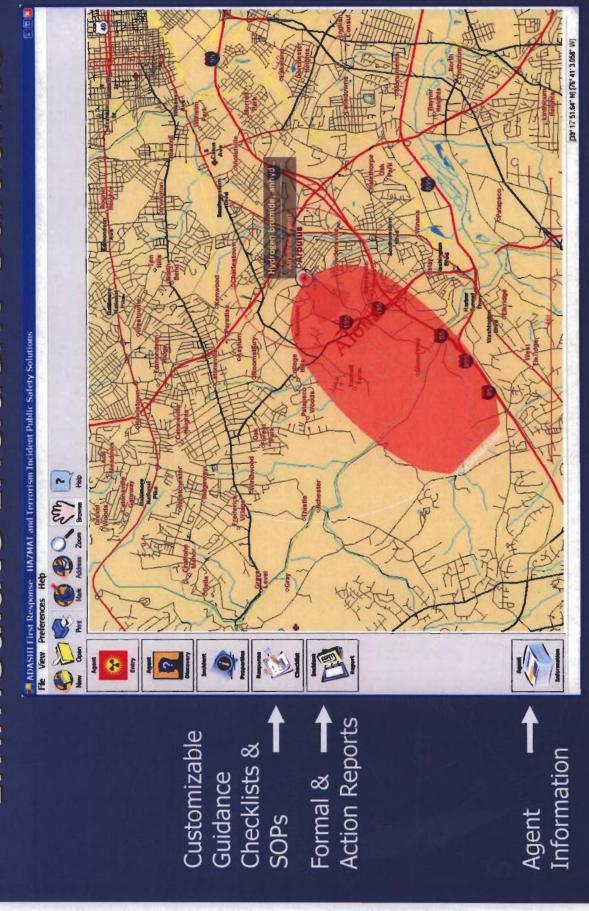
### ADASHI Eagle

- Desktop or Laptop Application for Remote or On-Scene Situation Assessment, Response Guidance and Command, Control and Communications
- Initial Commercial Product Release Planned for Summer 05

## ADASHI Product Line

Feature	First Response	ADASHI Owl	ADASHI Eagle
Source Identification (Chemical, Biological, Radiological, & Nuclear)			
Rail cars, trailers, drums, and tank configurations (HAZMAT Transport - HT)	*	🗸 (НТ)	🗸 (HT)
Human and observed signs and symptoms, manual, and automated detectors		1	1
False alarm and confounding agent identification		1	1
PDA portable data entry within Level A Suit with GPS tracking		1	1
Multiple source identification and interaction analysis		1	1
Automated worst case scenario generation	1	>	1
Automated meteorological data acquisition	>	1	>
Street map and hazard visualization (ESRI and other mapping programs)	>	>	>
ERG 2000 isolation, fire, and spill safe distances	1	1	>
Outdoor atmospheric dispersion modeling	>	1	>
Vulnerability, hazards, and resource tracking with full scheduling capabilities		*	>
Equipment inventories (detector, PPE, decontamination)		>	>
Guidance on detection, public safety, first aid, PPE, decon., and communication		>	1
Microsoft's active directory security integration and responder role assignments			>
Communication (military/civilian, messenger, video, and action tracking/logging)			>





## ALOHA vs. HPAC

### ALOHA

### HPAC

No terrain steering

Static 2-dimentional hazard analysis

Static single sources

Static weather model (wind > 3mph)

Chemical agents only

No fire modeling

No energetic event modeling

Complex terrain inclusion and model effects

Dynamic 3 (4)-dimensional hazard analysis

Static and moving multiple sources

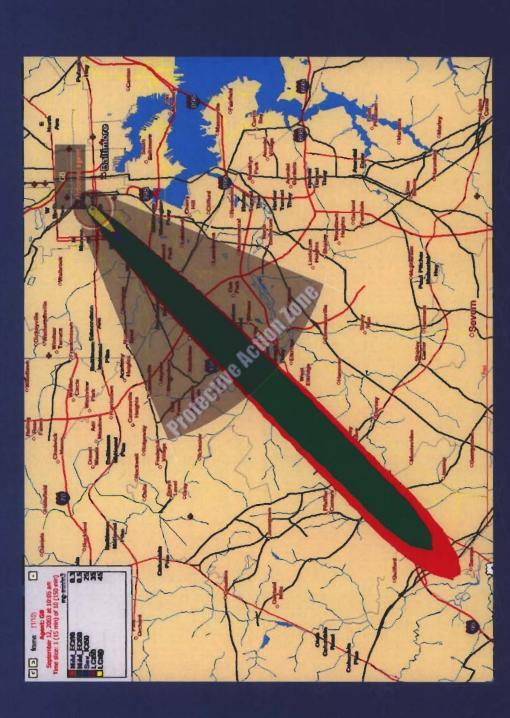
4-dimentional robust dynamic weather inclusion

Nuclear, Chemical, Biological, and Radiological

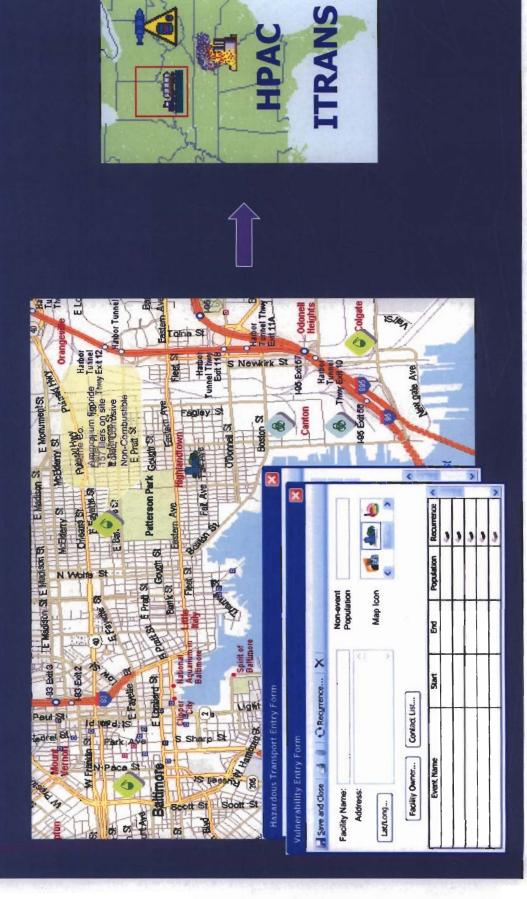
Fireball and jet fire models

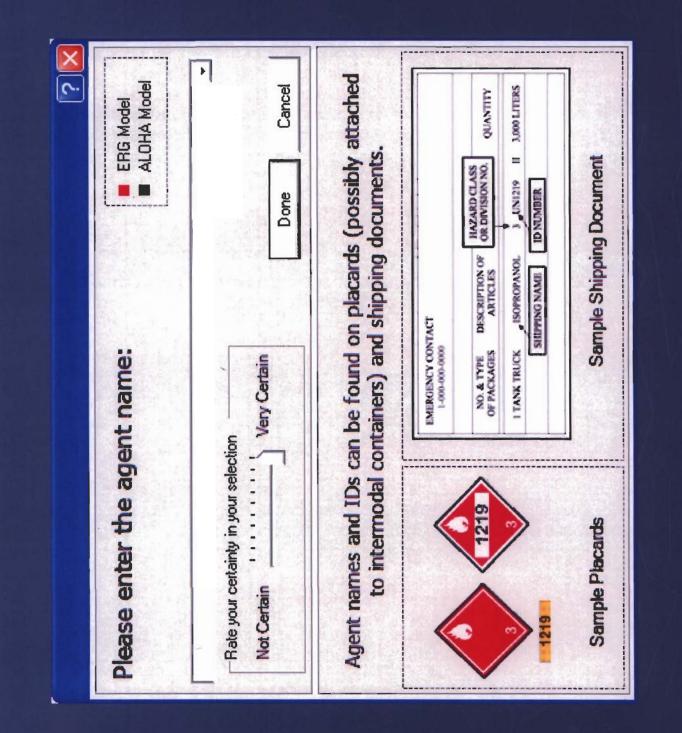
Blast, explosive, and nuclear models

# ADASHI as an Interface to HPAC



### Resources, Vulnerabilities, and Hazards -> ITRANS

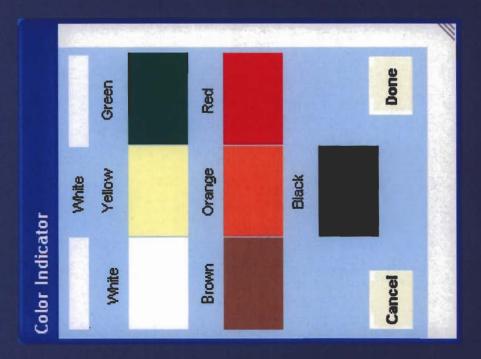


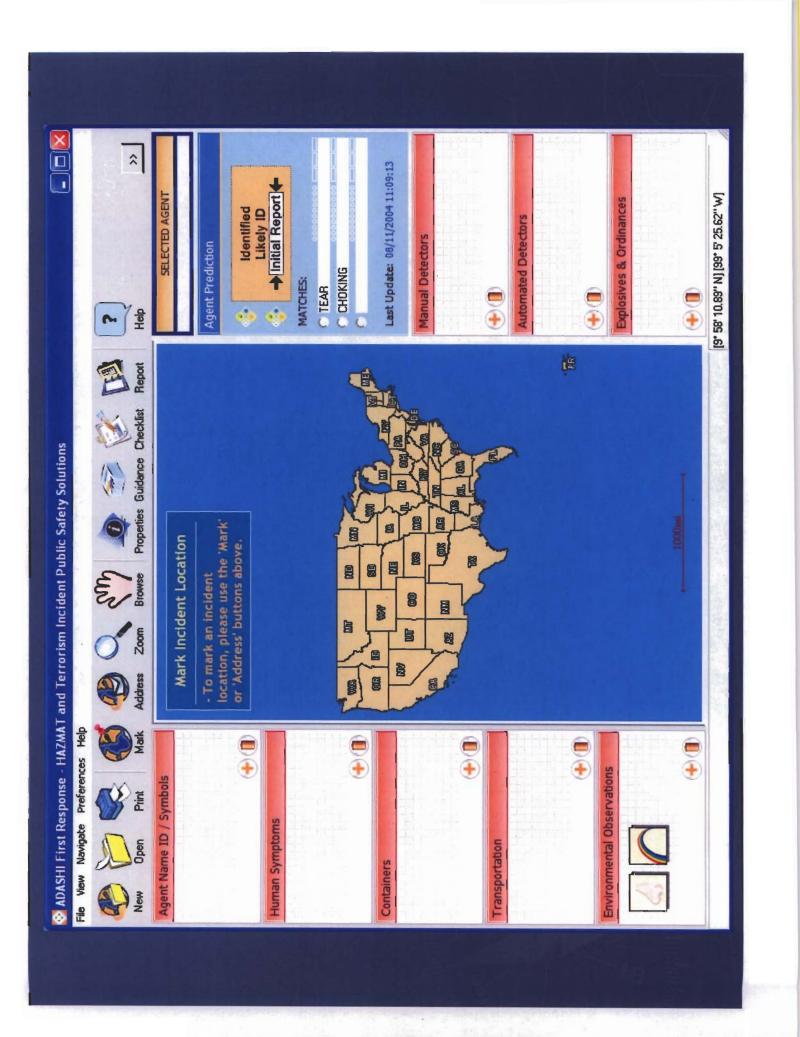


- Sect Intensity **Dnset** Respiratory.Observations Change in Voice Quality Breathing Coughing Normal R : : : Gastrointestinal Cardiovascular Diratory Torso Skin Human Signs and Symptoms 1b12d926-c95f-4407-9ba2-b01b6 Done Number of Patients Select Body Section Patient Name C **Current Symptoms** Cancel

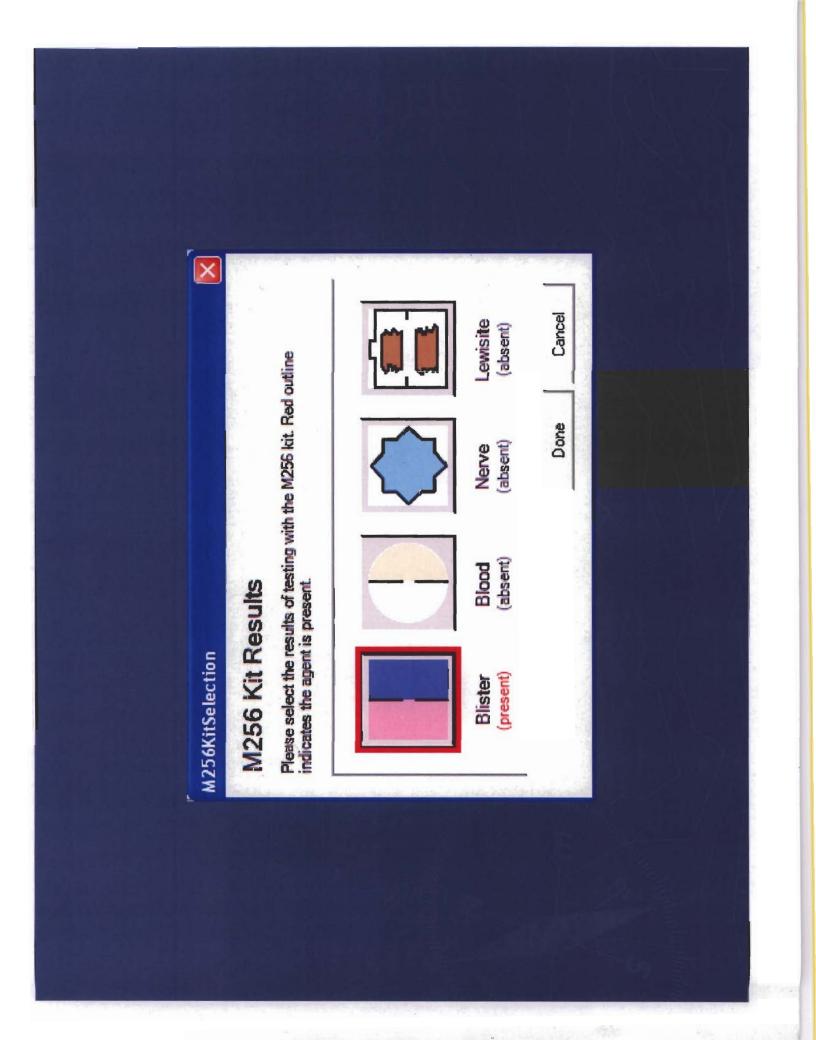




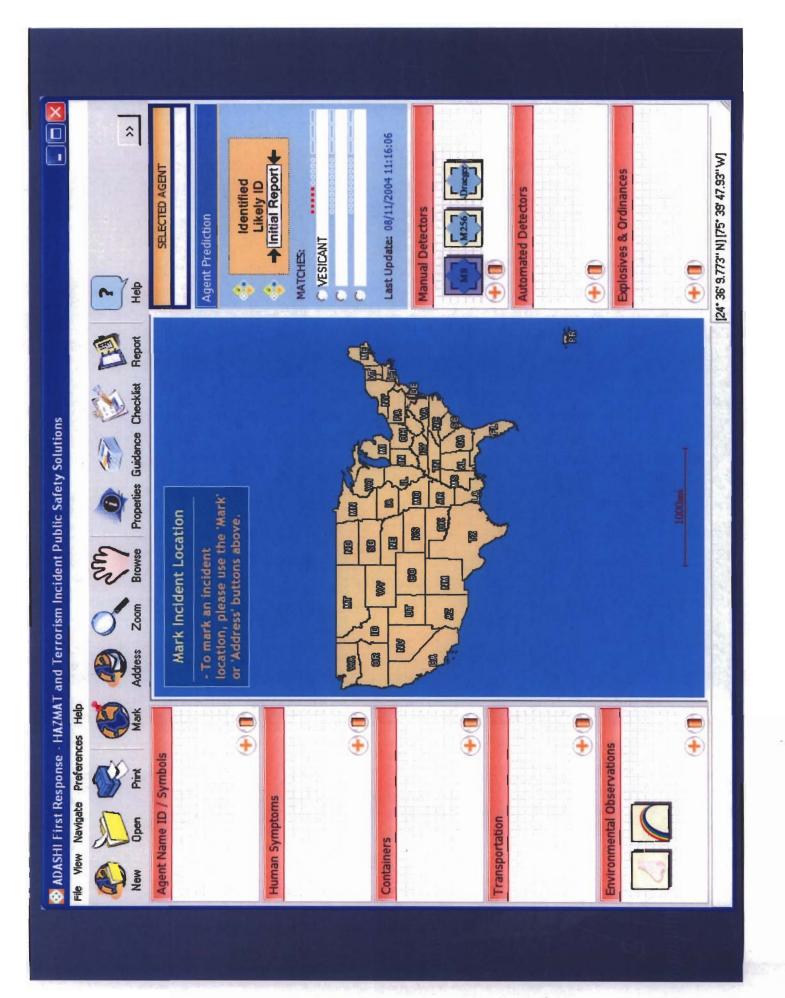


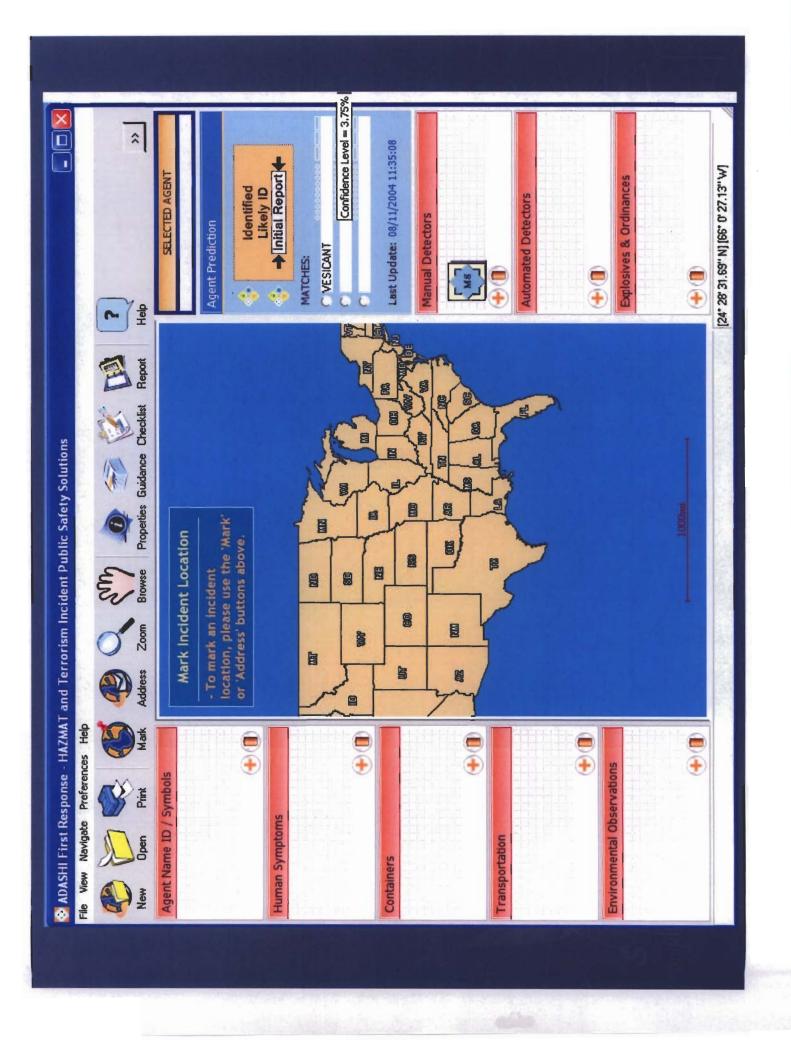


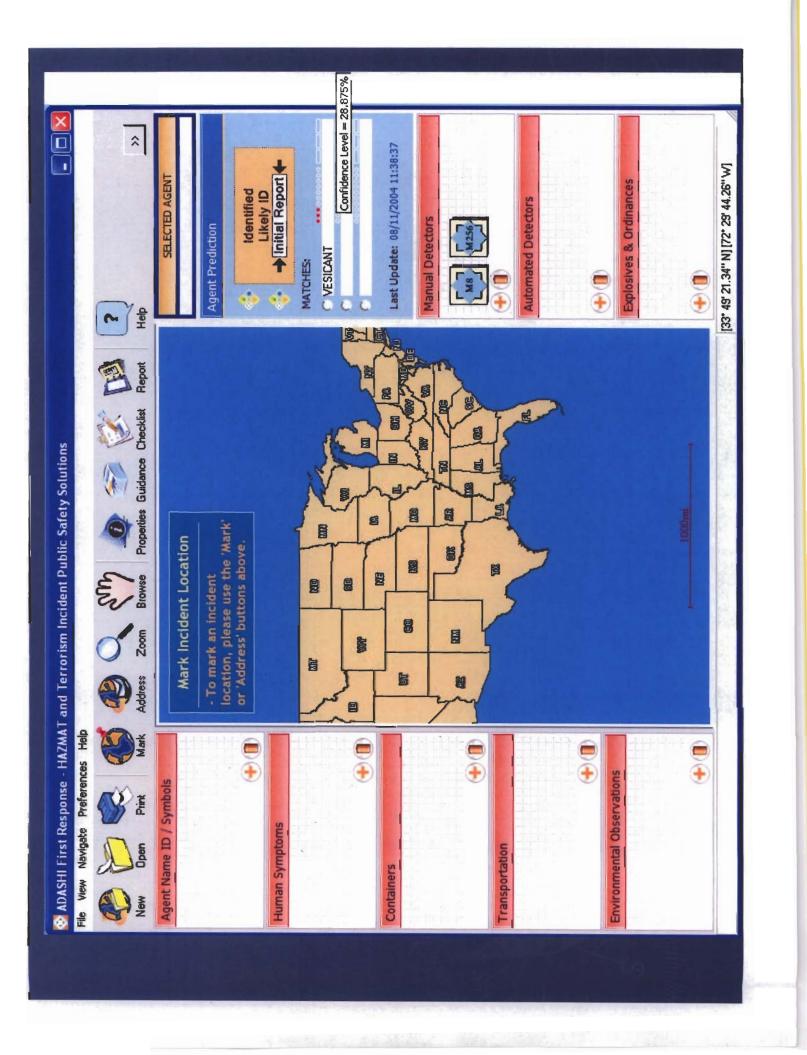




~ Last Update: 08/11/2004 11:13:41 ► Initial Report [22° 34' 0.537" N] [78° 59' 14.77" W] SELECTED AGENT Explosives & Ordinances Identified Automated Detectors **Manual Detectors** Agent Prediction C VESICANT MATCHES: • ø Heb ~ 0 C 1ª Report Properties Guidance Checklist × 😋 ADASHI First Response - HAZMAT and Terrorism Incident Public Safety Solutions 0 C. 8 location, please use the 'Mark' Mark Incident Location or 'Address' buttons above. ä 2 9 H Browse 8 ŝ 8 Zoom 5 Address 8 열 8 -File View Navigate Preferences Help Mark • **e** • • • • Environmental Observations Agent Name ID / Symbols Print Human Symptoms Transportation Containers New







DetectorSelection	
Drager Colorimetric Tubes Results Please input the results obtained from testing with the Drager Colorimetric tubes.	letric tubes.
Thioether (Thioether)	
Thioether	
1 mg/m3	1 mg/m3
Done	Cancel

