	AsH ₃	MW: 77.95	CAS: 7784-42-1	RTECS: CG6475000
METHOD: 6001, Issue 2		EVALUATION: FULL	Issue 1: 15 May 1985 Issue 2: 15 August 1994	
NIOSH:	0.05 ppm C 0.002 mg/m³/1 0.05 ppm; carcino (1 ppm = 3.19 mg		PROPERTIES:	gas; d 3.484 g/L @ 20 °C; BP -62.5 °C; MP -116.3 °C

SYNONYMS: hydrogen arsenide; arsenic trihydride.

	SAMPLING		MEASUREMENT
SAMPLER:	SOLID SORBENT TUBE (coconut shell charcoal, 100 mg/50 mg)	TECHNIQUE:	ATOMIC ABSORPTION, GRAPHITE FURNACE
FLOW RATE:	0.01 to 0.2 L/min	ANALYTE:	arsenic
VOL-MIN: -MAX:	0.1 L @ 0.05 ppm 10 L	DESORPTION:	1 mL 0.01 <u>M</u> HNO ₃ ; 30 min in ultrasonic bath
SHIPMENT:	routine	MATRIX MODIFIER:	Ni ²⁺ , 1000 µg/mL
SAMPLE STABILITY:	at least 6 days @ 25 °C [1]	WAVELENGTH:	193.7 nm; D ₂ or H ₂ correction
BLANKS:	2 to 10 field blanks per set	GRAPHITE FURNACE:	DRY: 40 sec @ 110 °C; CHAR: 15 sec @ 1200 °C; ATOMIZE 7 sec @ 2540 °C
	ACCURACY	INJECTION:	50 µL
RANGE STUDIE	(10-L samples);	CALIBRATION:	As(III) in 0.01 <u>M</u> HNO ₃ with 100 mg charcoal present
	0.001 to 0.01 mg/m ³ [2]	RANGE:	0.01 to 0.3 µg per sample [2]
BIAS:	-6.13% at 0.01 to 0.2 L/min rates [1]; -11% @ 0.876 L/min [2]	ESTIMATED LOD	: 0.004 µg per sample
OVERALL PREC	CISION (Ŝ _{rt}): 0.087 [2]	PRECISION (Ŝ _r):	0.060 @ 0.012 to 0.11 µg
ACCURACY:	± 23.2%		per sample [2]

APPLICABILITY: The working range is 0.0003 to 0.06 ppm (0.001 to 0.2 mg/m³) for a 10-L air sample. This is an elemental analysis and is not compound-specific.

INTERFERENCES: Use background correction to control molecular absorption. Other arsenic compounds (gases or aerosols) may be collected on the sampler and would be erroneously reported as arsine. A cellulose ester filter in front of the ch arcoal tube may be used to remove aerosols [3,4]. The effect of relative humidity on the capacity of charcoal for arsine has not bee n studied.

OTHER METHODS: This method combines and replaces P&CAM 265 [5] and S229 [6] for arsine.

REAGENTS:

- 1. Water, distilled or deionized.
- 2. Nitric acid, conc.*
- 3. Nitric acid, 0.01 <u>M</u>. Dilute 0.4 mL conc. HNO ₃ to 1 L with water.
- 4. Nitric acid, 0.1 <u>M</u>. Dilute 4 mL conc. HNO ₃ to 1 L with water.
- Arsenic stock solution, 1000 μg As(III)/mL.* Commercial standard or dissolve 1.322 g dried, certified reagent As ₂O₃ in 100 mL 0.1 <u>M</u> HNO₃; dilute to 1 L with 0.1 <u>M</u> HNO₃.
- Calibration stock solution, 1.0 μg As(III)/mL.* Dilute 0.1 mL arsenic stock solution (1000 μg As/mL) to 100 mL with 0.01 <u>M</u> HNO₃. Prepare fresh daily.
- Nickel nitrate solution, 1000 μg Ni/mL. Commercial nickel atomic absorption standard or dissolve 3.112 g dried reagent Ni(NO ₃)₂ in 100 mL 0.1 <u>M</u> HNO₃; dilute to 1 L with water.
- 8. Argon, compressed.
- 9. Arsine, *99%, or certified mixture in nitrogen.
 - * See Special Precautions

EQUIPMENT:

- Sampler: glass tube, 7 cm long, 6-mm OD, 4-mm ID, flame-sealed ends, containing two sections of activated (600 °C) coconut shell charcoal (front = 100 mg; back = 50 mg) separated by a 2-mm urethane foam plug. A silylated glass wool plug precedes the front section and a 3-mm urethane foam plug follows the back section. Pressure drop across the tube at 1 L/min airflow must be less than 3.4 kPa. Tubes are commercially available.
 - NOTE: Use a cellulose ester membrane filter in front of the sampler if particulate arsenic is present [3,4].
- 2. Personal sampling pump, 0.01 to 0.2 L/min, with flexible connecting tubing.
- 3. Atomic absorption spectrophotometer with graphite furnace, non-pyrolytic tubes, background correction and electrodeless discharge (and power supply) or hollow cathode lamp for arsenic.
- 4. Volumetric flasks, 1-L and 100-mL.*
- 5. Micropipets, 5- to 500-µL.*
- 6. Centrifuge tubes, 10- or 15-mL.*
- 7. Ultrasonic bath.
- 8. Centrifuge.
- 9. Syringe, gas, 0.1-mL, readable to 1 μ L.

*Clean all glassware with conc. nitric acid and rinse thoroughly with distilled or deionized water before use.

SPECIAL PRECAUTIONS: Arsenic is a human carcinogen [7]. Perform all concentrated acid handling in a fume hood. Arsine is extremely poisonous by inhalation. Handle in well-ventilated hood.

SAMPLING:

- 1. Calibrate each personal sampling pump with a representative sampler in line.
- 2. Break the ends of the sampler immediately before sampling. Attach sampler to personal sampling pump with flexible tubing.
 - NOTE: Use a cellulose ester membrane prefilter if particulate arsenic compounds may be present [3,4].
- 3. Sample at an accurately known flow rate between 0.01 and 0.2 L/min for a total sample size of 0.1 to 10 L.
- 4. Cap the sampler with plastic (not rubber) caps and pack securely for shipment.

SAMPLE PREPARATION:

- 5. Place the front and back sorbent sections of the sampler tube in separate centrifuge tubes. Discard the glass wool and foam plugs.
- 6. Add 1.0 mL 0.01 \underline{M} HNO₃ to each tube. Cap each tube.

- 7. Agitate for 30 min in an ultrasonic bath.
- 8. Centrifuge each tube.

CALIBRATION AND QUALITY CONTROL:

- 9. Calibrate daily with at least six working standards over the range 0.004 to 0.3 µg arsenic per sample.
 - a. Add known amounts of calibration stock solution and 0.01 \underline{M} HNO₃ for a 1.0-mL final solution volume to centrifuge tubes containing 100 mg activated charcoal from a media blank sampler.
 - b. Analyze standards together with samples and blanks (steps 12 and 13). Analyze a working standard for every five samples to check for instrument drift.
 - c. Prepare a calibration graph (absorbance vs. µg arsenic).
- Determine desorption efficiency (DE) at least once for each batch of charcoal used for sampling in the range 0.004 to 2 µg arsenic per sample. Prepare three tubes at each of five levels plus three media blanks.
 - a. Remove and discard back sorbent section of a media blank sampler.
 - b. Inject a known amount of pure arsine gas (or a certified gas mixture containing arsine) directly onto front sorbent section with a microliter syringe.
 - c. Cap the tube. Allow to stand overnight.
 - d. Desorb (steps 5 through 8) and analyze together with working standards (steps 12 and 13).e. Prepare a graph of DE vs. µg arsenic recovered.
- 11. Analyze three quality control spikes to ensure that the calibration graph is in control.

MEASUREMENT:

- 12. Set the spectrophotometer and furnace to manufacturer's recommendations and to conditions given on page 6001-1.
- 13. Inject a 50-μL aliquot of sample or standard followed by a 50-μL aliquot of nickel nitrate solution prior to initiating the analysis program. Measure peak area.
 - NOTE 1: If sample absorbance is above the linear range of the standards, dilute with 0.01 \underline{M} HNO₃, reanalyze and apply the appropriate dilution factor in calculations.
 - NOTE 2: Monitor the reproducibility of peak area for a working standard throughout the measurements. If erratic results occur, reoptimize instrumental parameters and replace the graphite tube.

CALCULATIONS:

14. Determine the mass, μ g, of arsine found in the sample front (W _f) and back (W _b) sorbent sections, and in the average media blank front (B _f) and back (B _b) sorbent sections by multiplying the mass of arsenic found for each of these sections by 1.040 (M.W. of arsine/M.W. of arsenic).

NOTE: If $W_{h} > W_{f}/10$, report breakthrough and possible sample loss.

15. Calculate concentration, C, of arsine in the air volume sampled, V (L):

$$C = \frac{W_{f} + W_{b} - B_{f} - B_{b}}{V}, mg/m^{3}.$$

EVALUATION OF METHOD:

Method S229 [6] was evaluated over the range 0.094 to 0.404 mg/m 3 using 10-L air samples collected on SKC Lot 105 activated coconut charcoal [1]. Breakthrough (onto the backup section) did not occur after 240 min of sampling at 0.227 L/min from an arsine concentration of 0.405 mg/m 3 (0.022 mg loading). The recovery was found to be 93.7%. Desorption efficiency was 0.90 at 1 µg arsine per sample and 1.00 at 2 and 4 µg arsine per sample.

Method P&CAM 265 [5] was evaluated over the range 0.001 to 0.01 mg/m 3 using 15-L air samples [2]. These samples were collected on SKC Lot 106 activated coconut charcoal at a sampling flow rate of 0.875 L/min for 15 min. At this flow rate, a collection efficiency of 89.1% was found [3]. The effect of high humidity on the sampler capacity was not studied. Desorption efficiency was 0.90 in the range 0.015 to 0.2 µg arsine per sample.

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METHOD REVISED BY:

R. D. Hull, Ph.D., NIOSH/DBBS.