

# Using SOAS & related field study data for scientific and regulatory modeling

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## Contributors:

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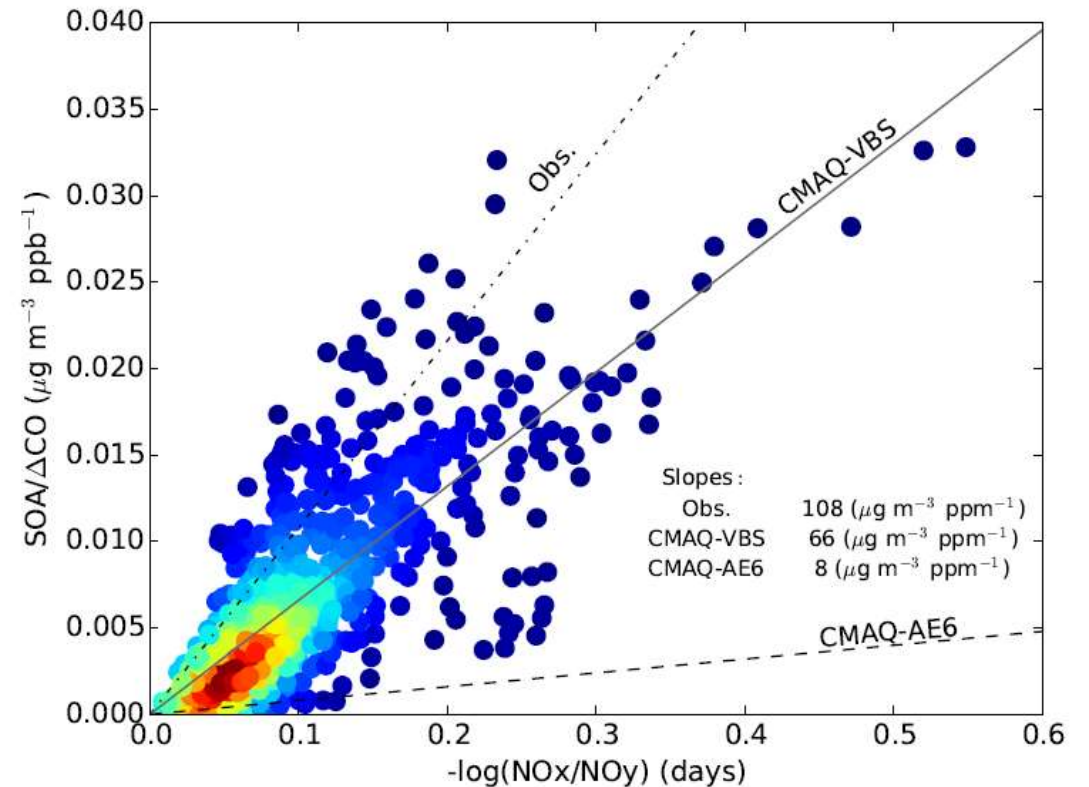
# Using SOAS & related field study data for scientific and regulatory modeling:

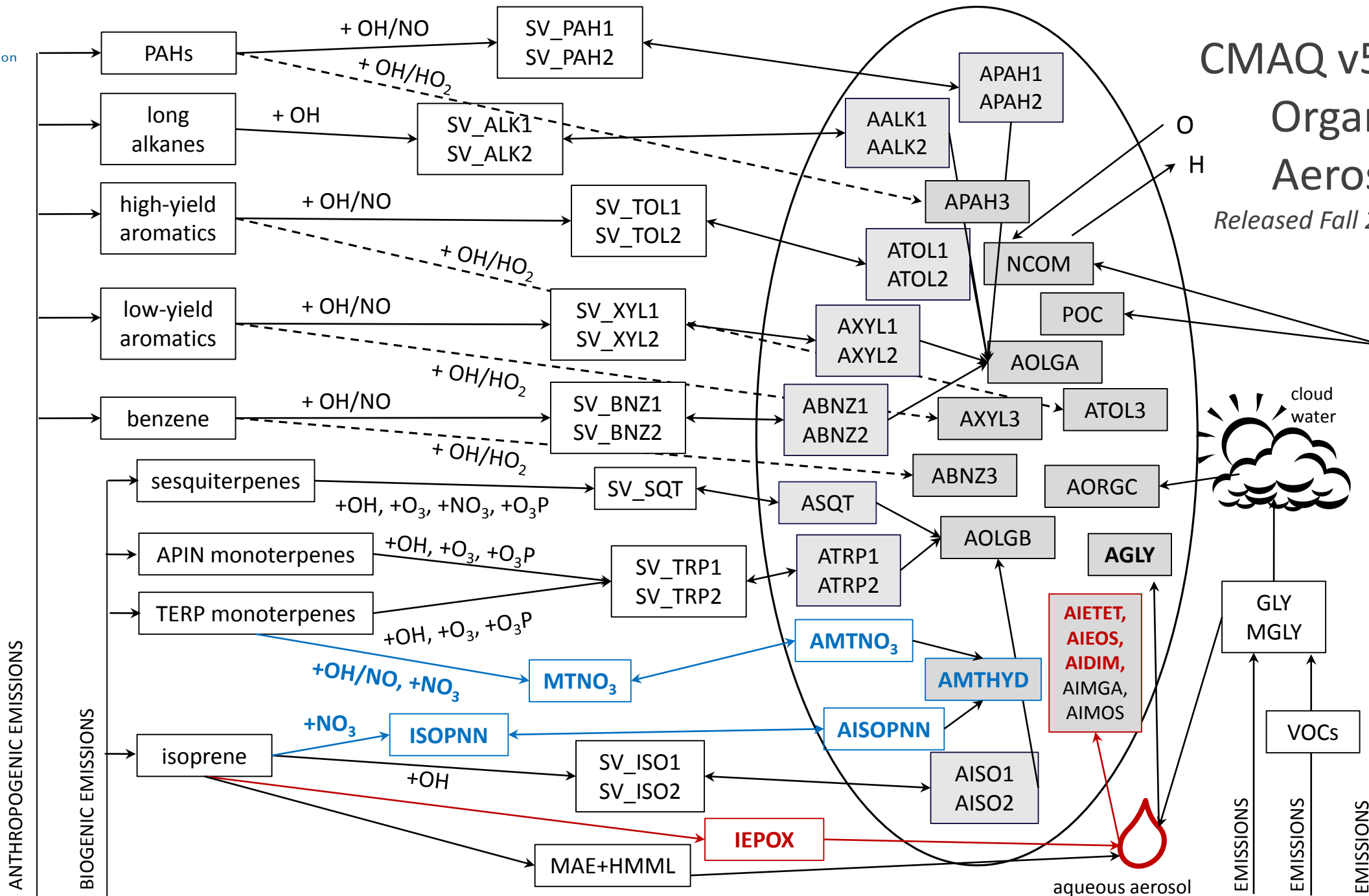
*How field and laboratory data has moved us from surrogates to explicit SOA mechanisms*

# Development of State of the Science SOA Models

- CMAQ model is often used for regulatory decision making in a relative sense (for ozone,  $\text{SO}_2$ , etc)
- Current regulatory uses do not specifically focus on SOA
  - Large uncertainties in anthropogenically dominated locations
  - More mechanistic information known in BVOC dominated locations (such as the southeast United States)
- Future lower  $\text{PM}_{2.5}$  NAAQS, component specific  $\text{PM}_{2.5}$  standards, or climate relevant work could require mechanistic SOA models

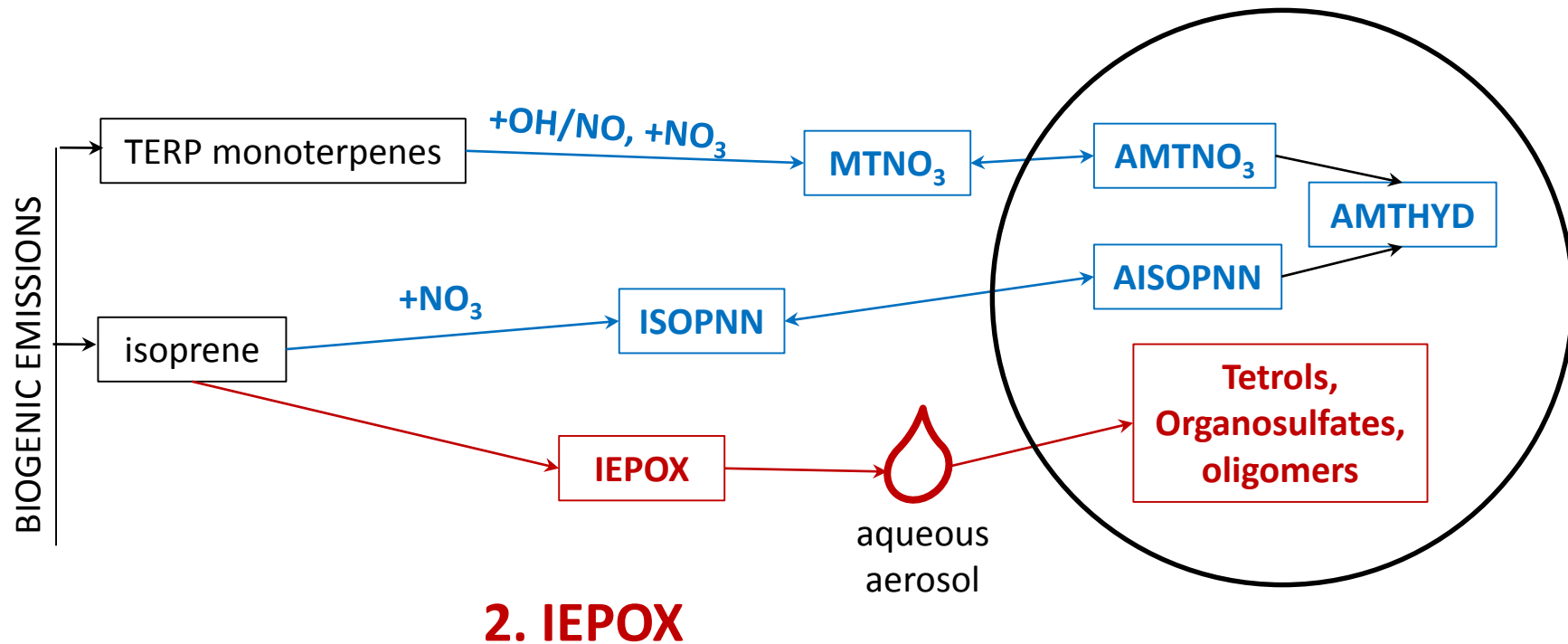
## SOA Predictions During CalNex 2010 (Woody et al. 2015 ACPD)



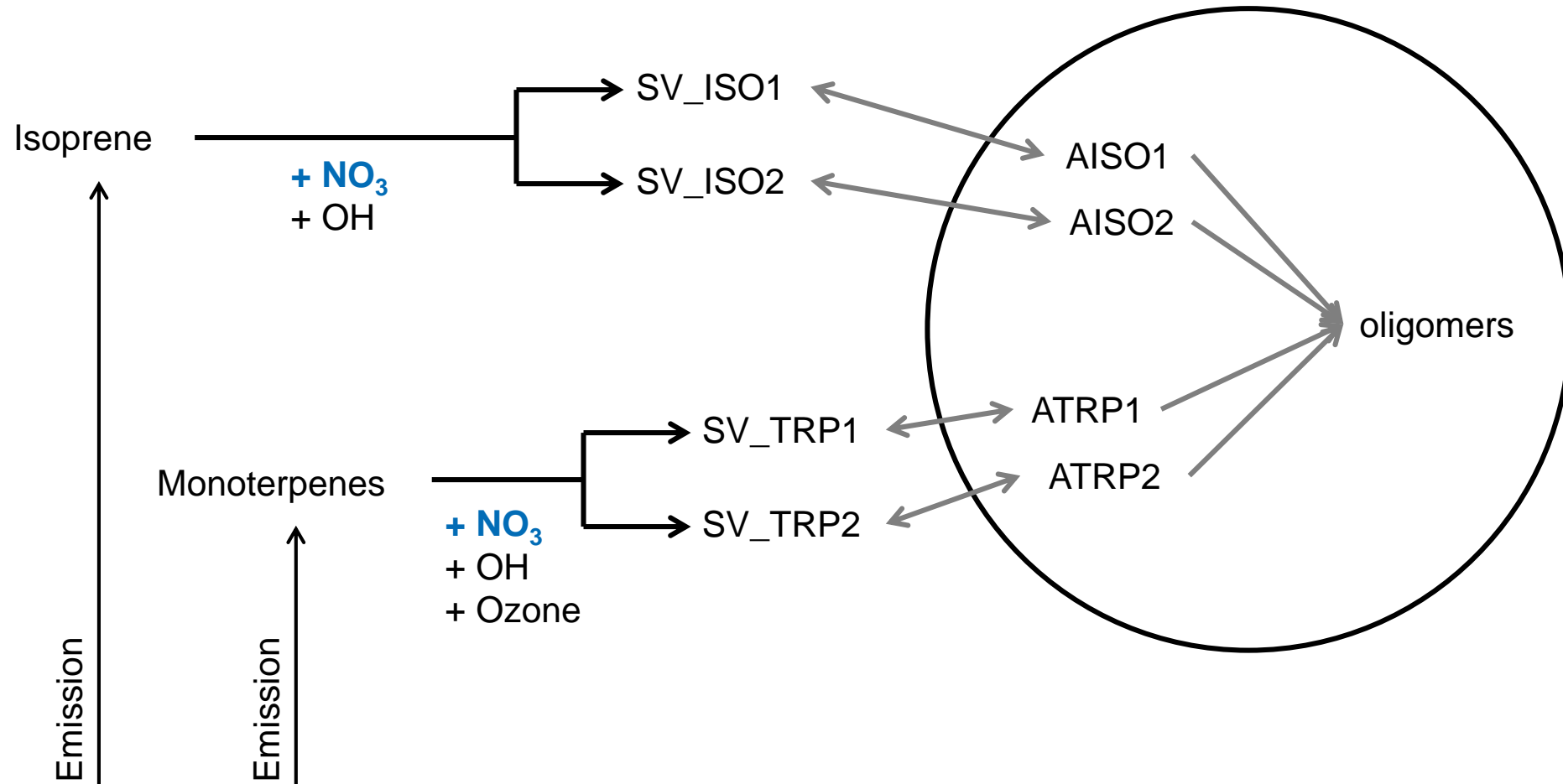


# Pathways to SOA in CMAQ informed by SOAS

## 1. Organic Nitrates

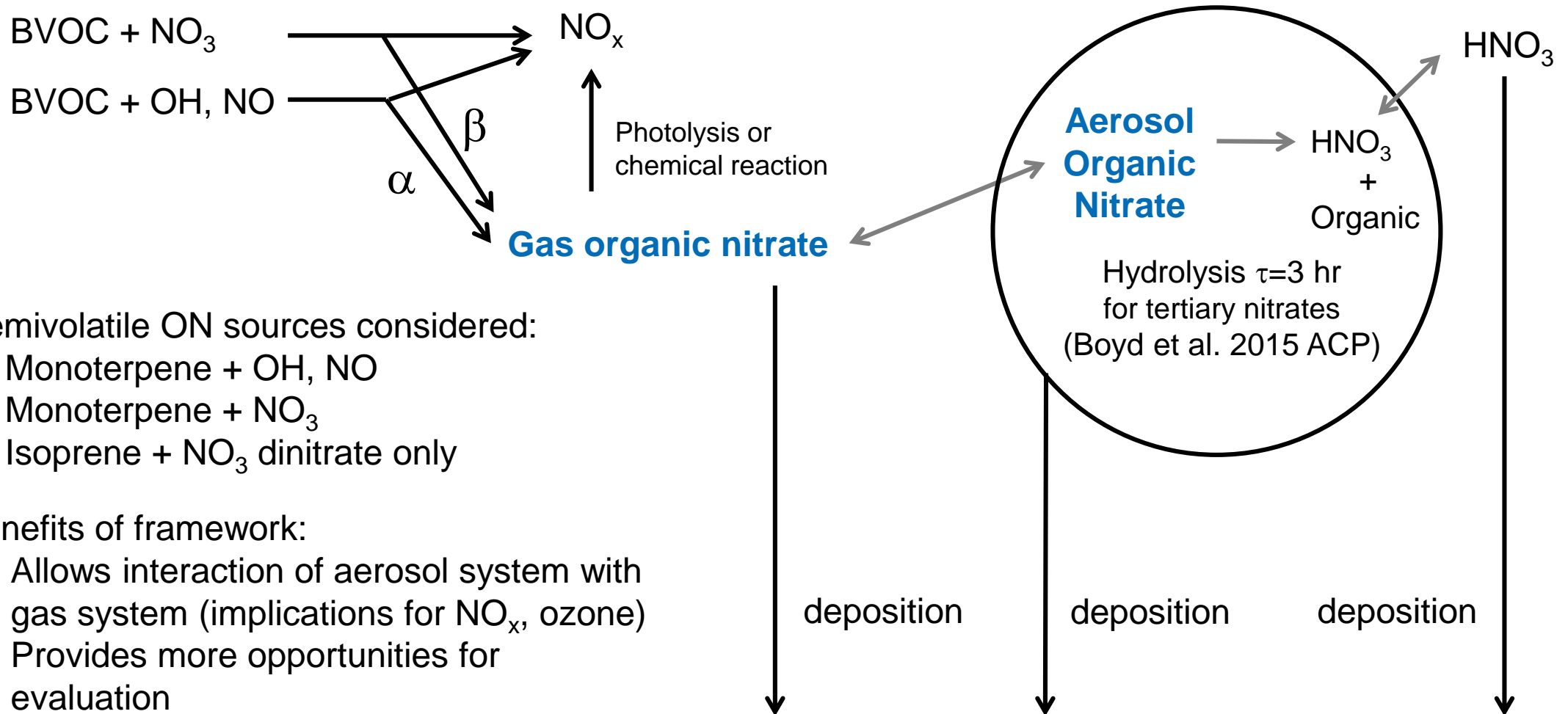


# Traditional BVOC+NO<sub>3</sub> OA model



Monoterpenes: Carlton et al. 2010 ES&T  
Isoprene: v5.1 unpublished

# Revised organic nitrogen SOA model



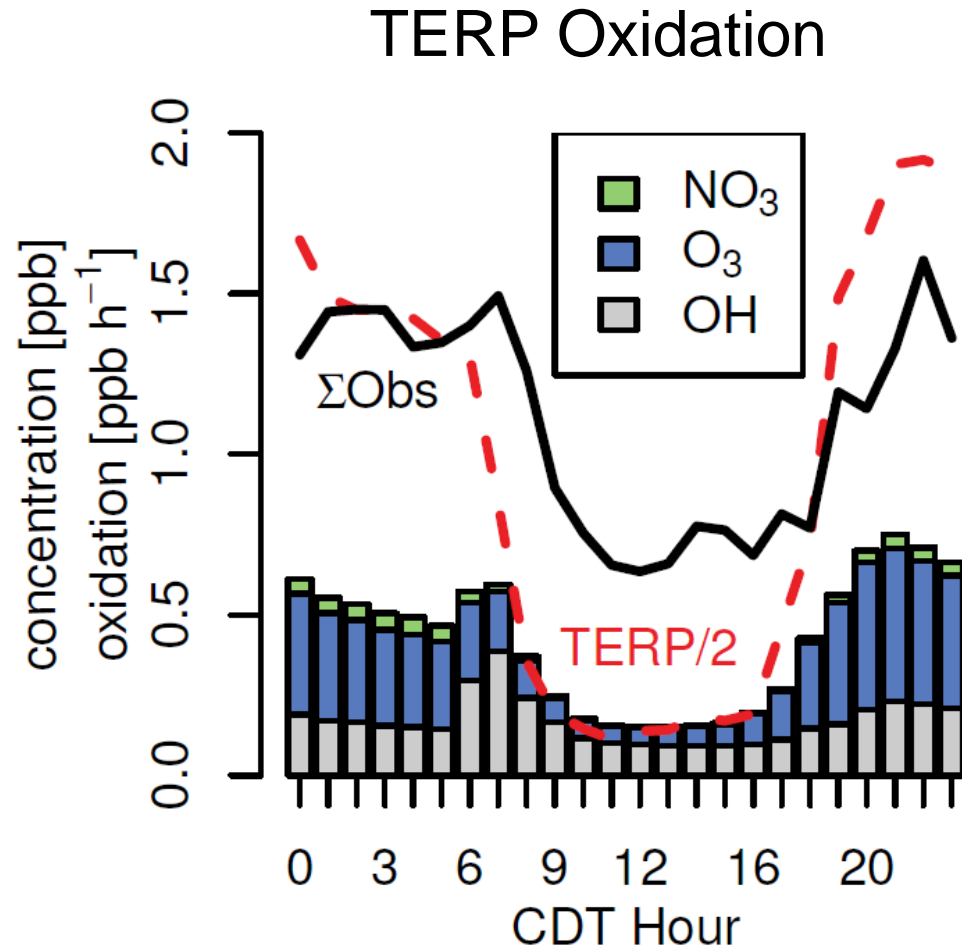
Semivolatile ON sources considered:

- Monoterpene + OH, NO
- Monoterpene + NO<sub>3</sub>
- Isoprene + NO<sub>3</sub> dinitrate only

Benefits of framework:

- Allows interaction of aerosol system with gas system (implications for NO<sub>x</sub>, ozone)
- Provides more opportunities for evaluation

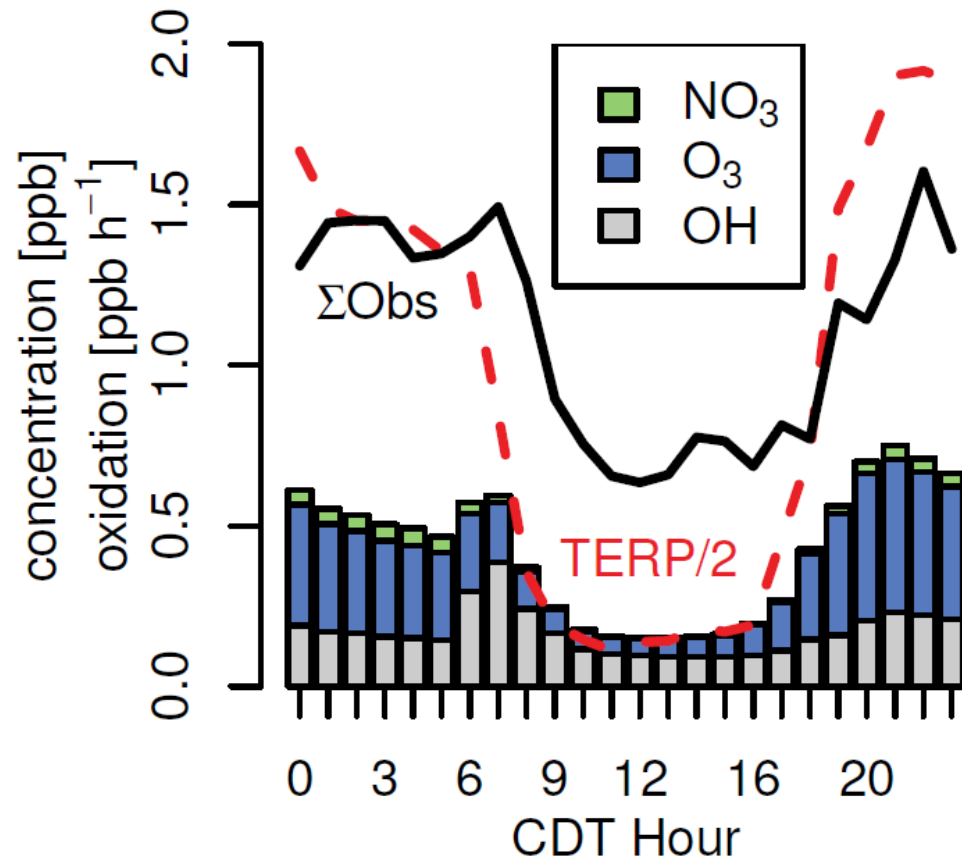
# O<sub>3</sub> is the dominant nocturnal oxidant



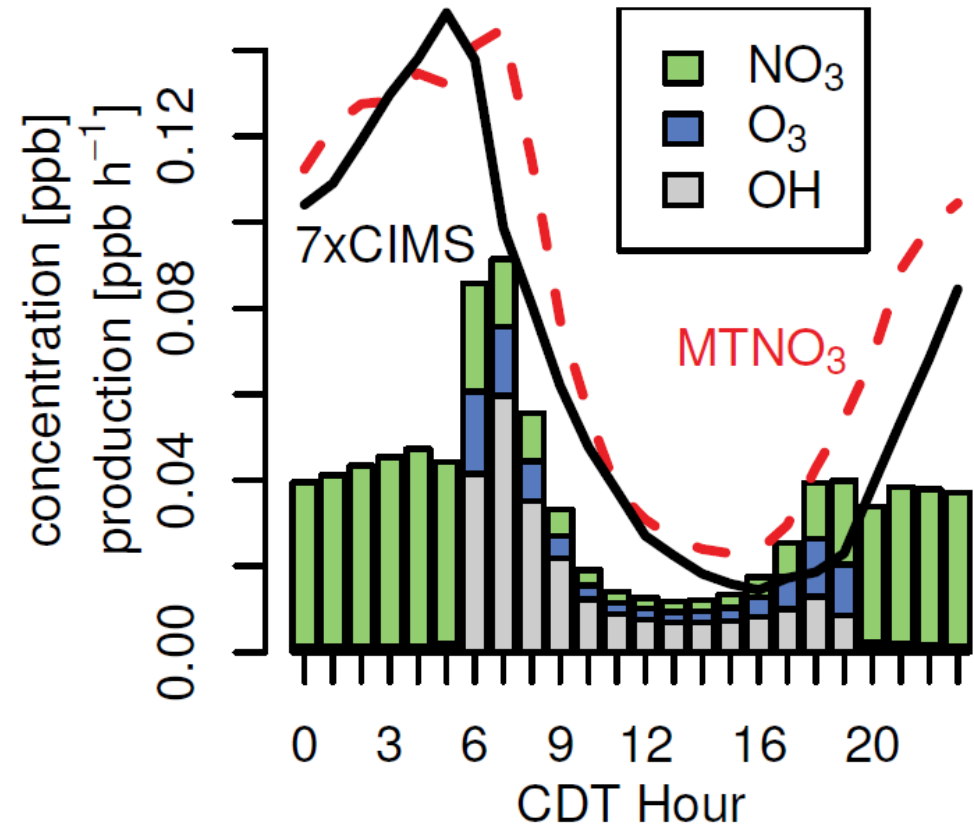


# NO<sub>3</sub> is the dominant ON source

## TERP Oxidation



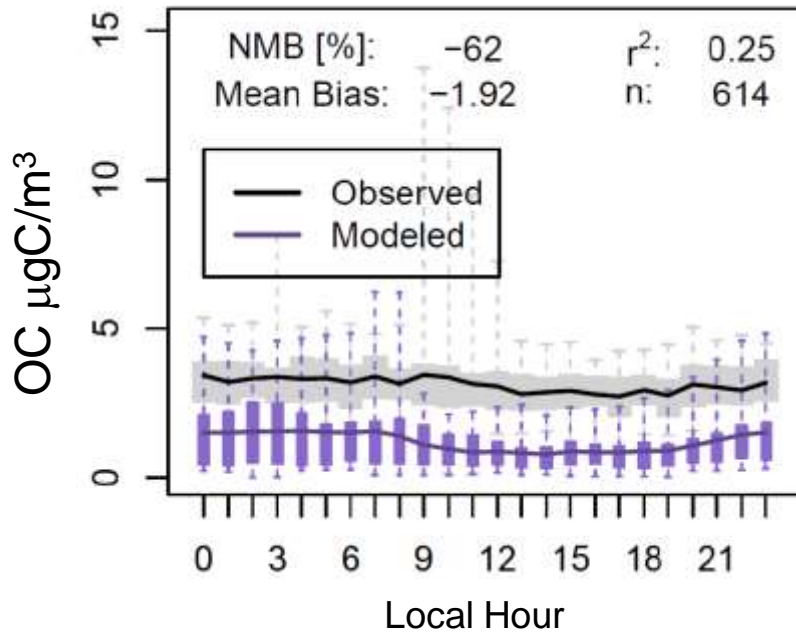
## MTNO<sub>3</sub> Production



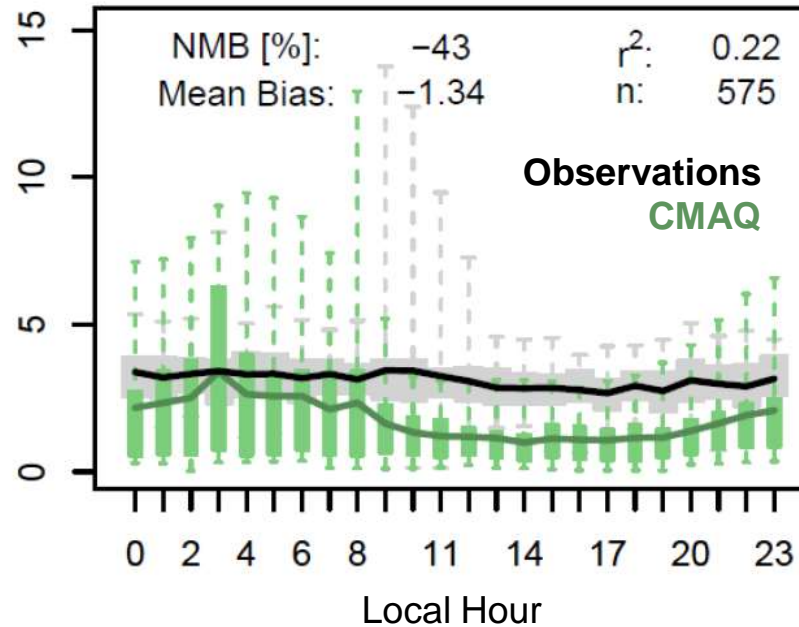
\* CIMS signal is one subset of MTNO<sub>3</sub>

# OC predicted at CTR June 2013

Base v5.1-beta

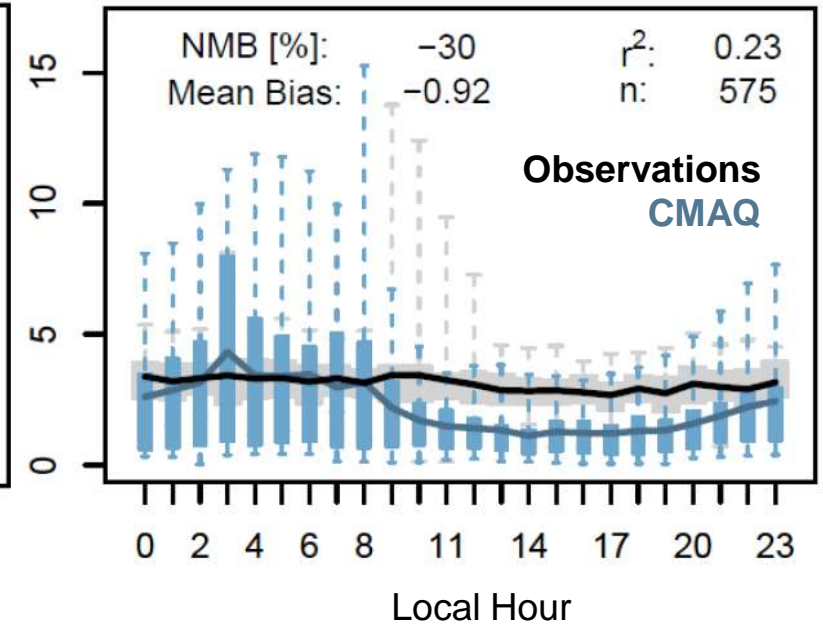


Revised w/  $\tau_{\text{hydrolysis}}=30$  hr



Bias in **OA** vs AMS: -35%  
 (-1.93  $\mu\text{g}/\text{m}^3$ )

Revised w/  $\tau_{\text{hydrolysis}}=3$  hr



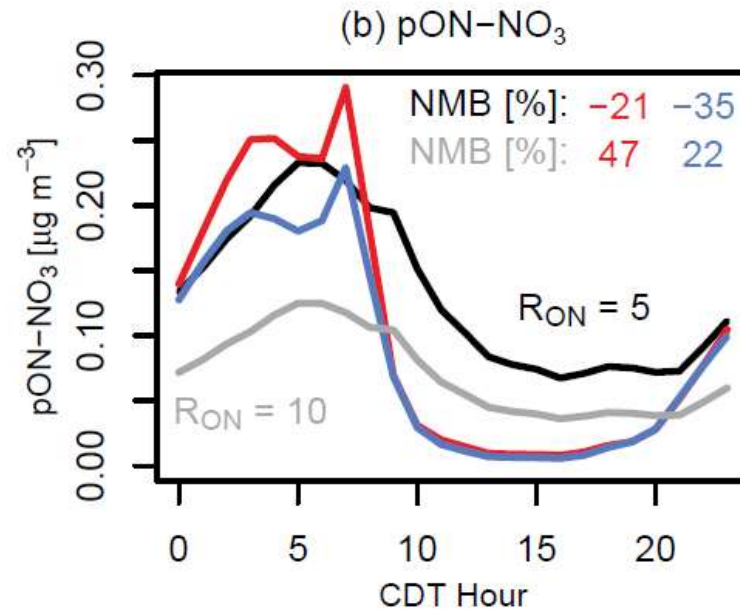
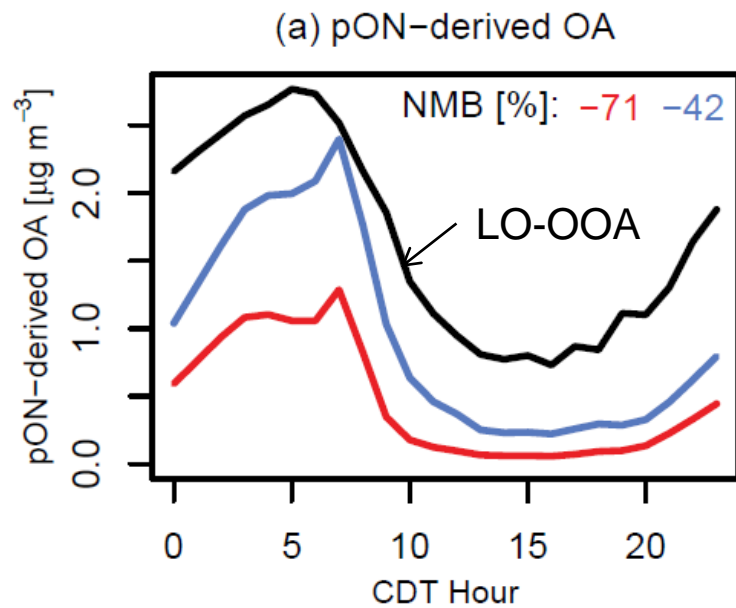
Bias in **OA** vs AMS: -23%  
 (-1.26  $\mu\text{g}/\text{m}^3$ )

# Faster hydrolysis consistent with observations



- Increasing the hydrolysis rate increases the magnitude of modeled LO-OOA

# Faster hydrolysis consistent with observations

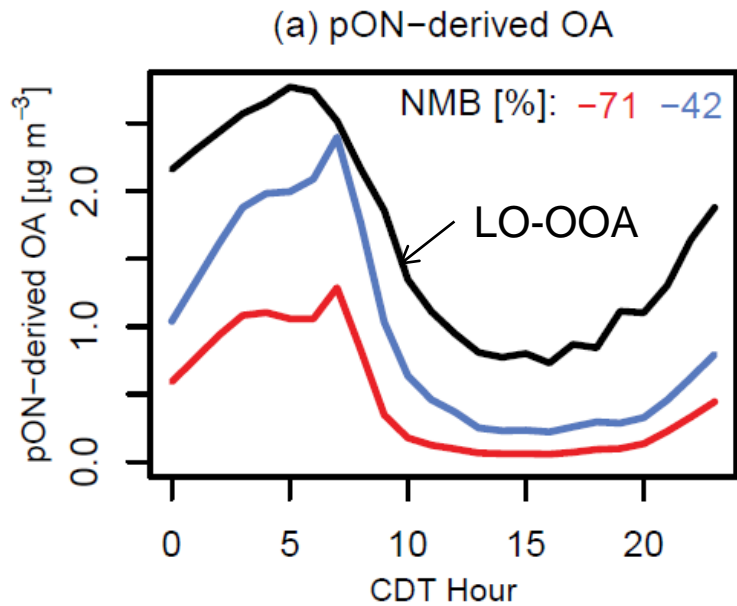


— Observed  
 — 30 h hydrolysis  
 — 3 h hydrolysis

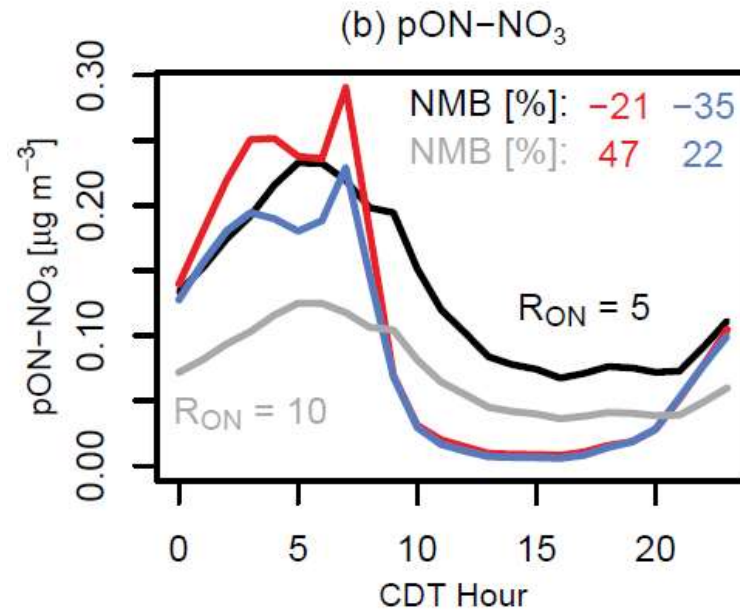
- Increasing the hydrolysis rate increases the magnitude of modeled LO-OOA

- Faster hydrolysis improves the speciation of LO-OOA

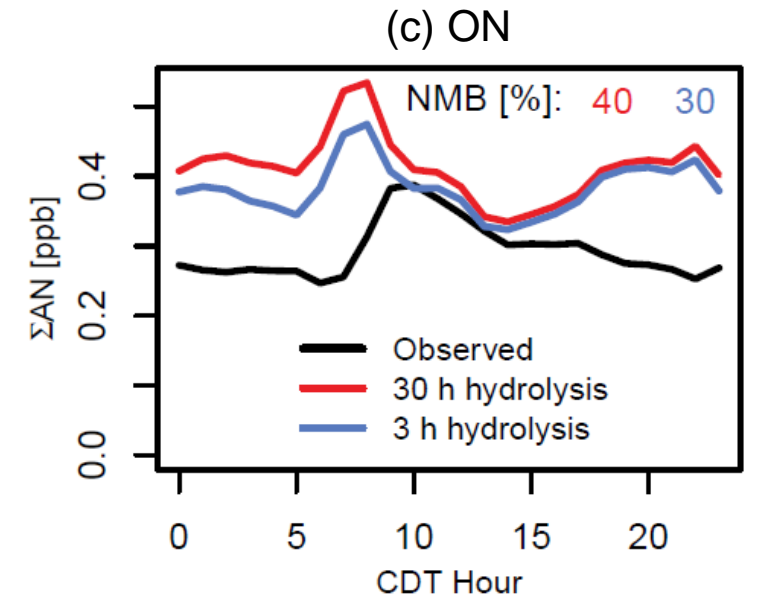
# Faster hydrolysis consistent with observations



- Increasing the hydrolysis rate increases the magnitude of modeled LO-OOA

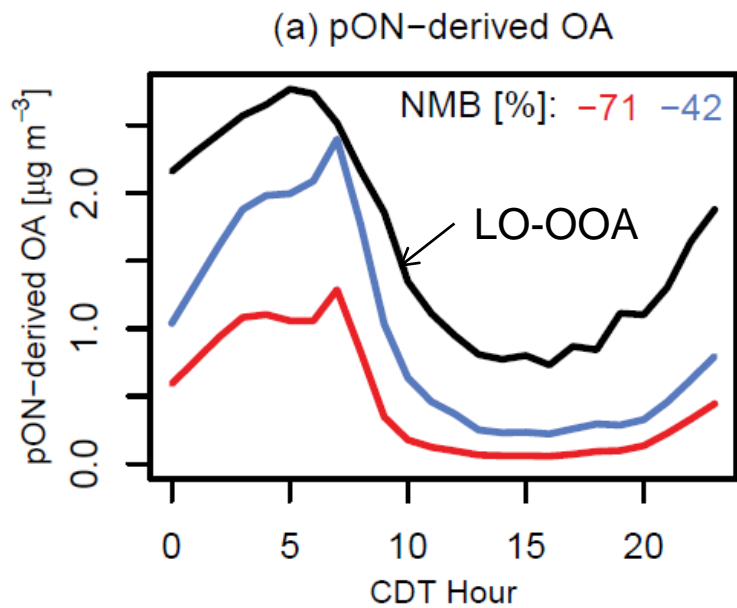


- Faster hydrolysis improves the speciation of LO-OOA

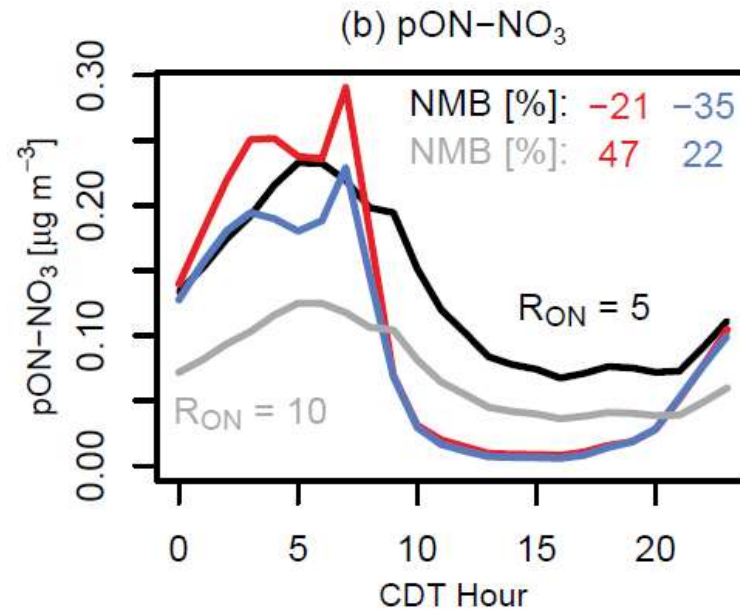


- Faster hydrolysis improves the magnitude of gas-phase organic nitrates

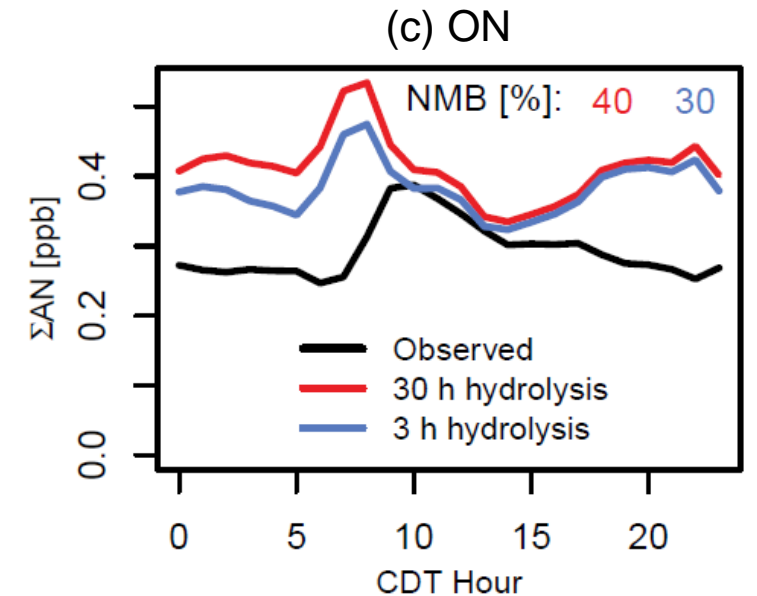
# Faster hydrolysis consistent with observations



- Increasing the hydrolysis rate increases the magnitude of modeled LO-OOA



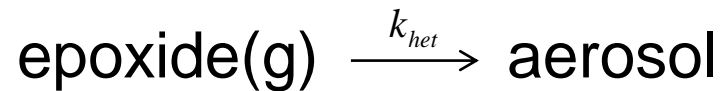
- Faster hydrolysis improves the speciation of LO-OOA



- Faster hydrolysis improves the magnitude of gas-phase organic nitrates

# IEPOX SOA

- Included in all research and regulatory mechanisms of CMAQ as of v5.1
- Modeled as reactive uptake (Pye et al. 2013 ES&T)



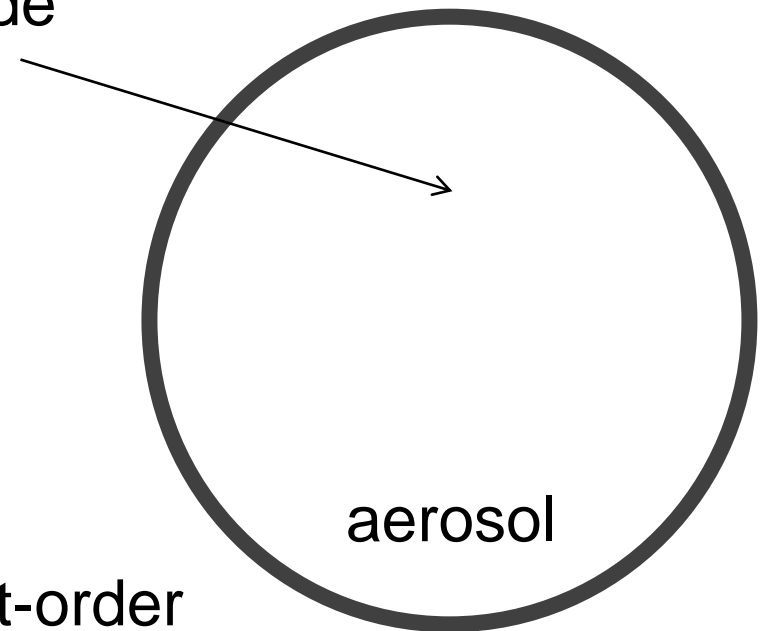
$$k_{het} = \frac{A}{\frac{r_p}{D_g} + \frac{4}{v\gamma_{epoxide}}}$$

$$\gamma = \frac{1}{\frac{1}{\alpha} + \frac{v}{4H_{eff}^*RT\sqrt{D_a k_{particle}}} \cdot \frac{1}{f(q)}}$$

$$f(q) = \coth(q) - \frac{1}{q}$$

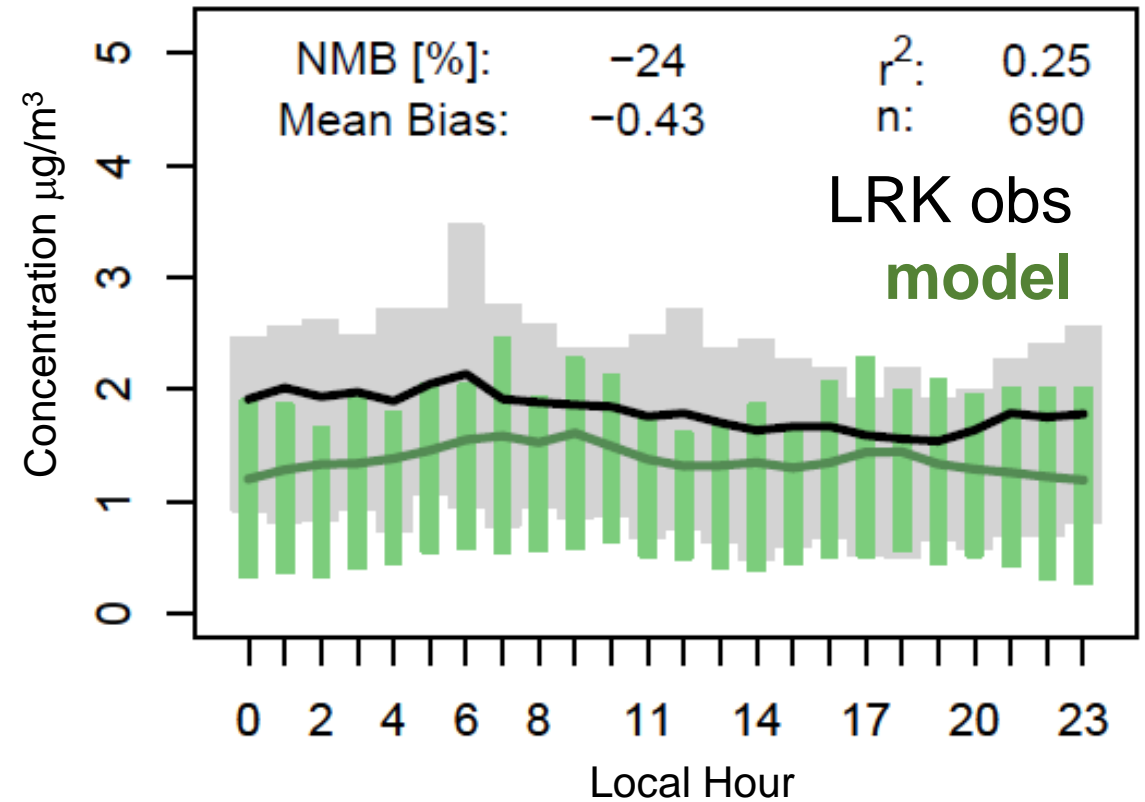
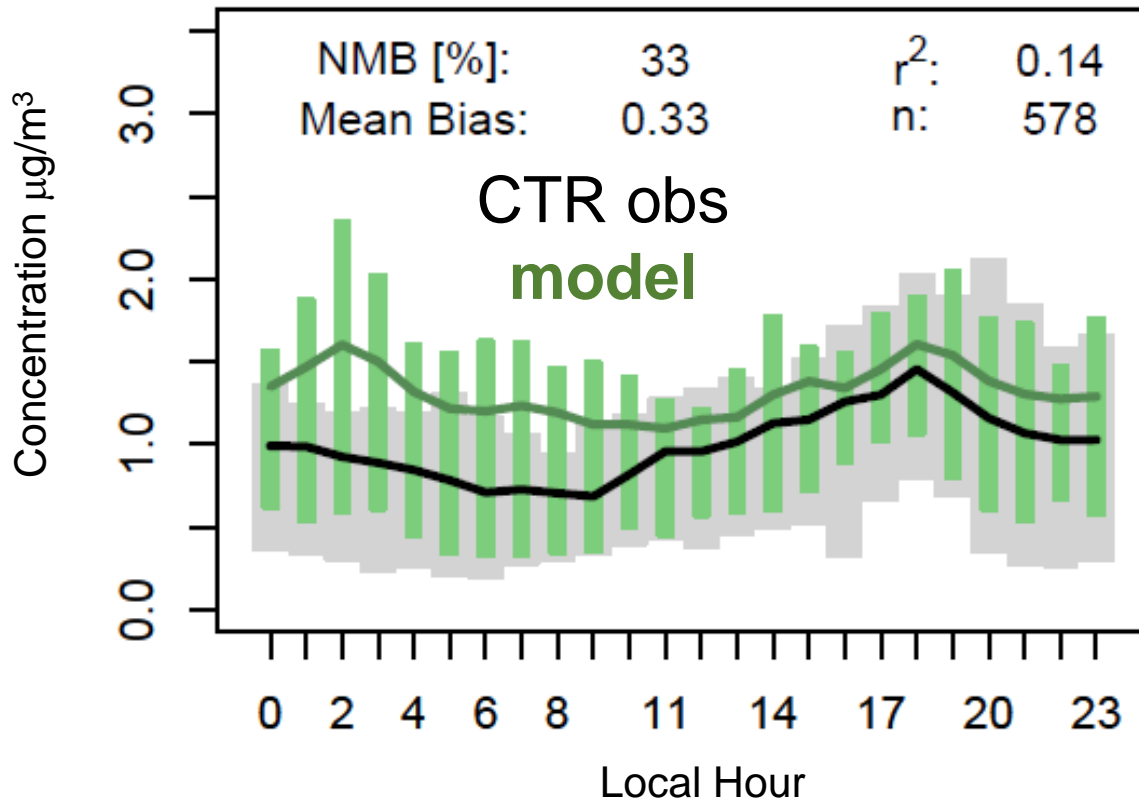
$$q = r_p \sqrt{\frac{k_{particle}}{D_a}}$$

epoxide



Pseudo first-order  
particle-phase  
reaction rate constant

# CMAQ Predictions of IEPOX-OA

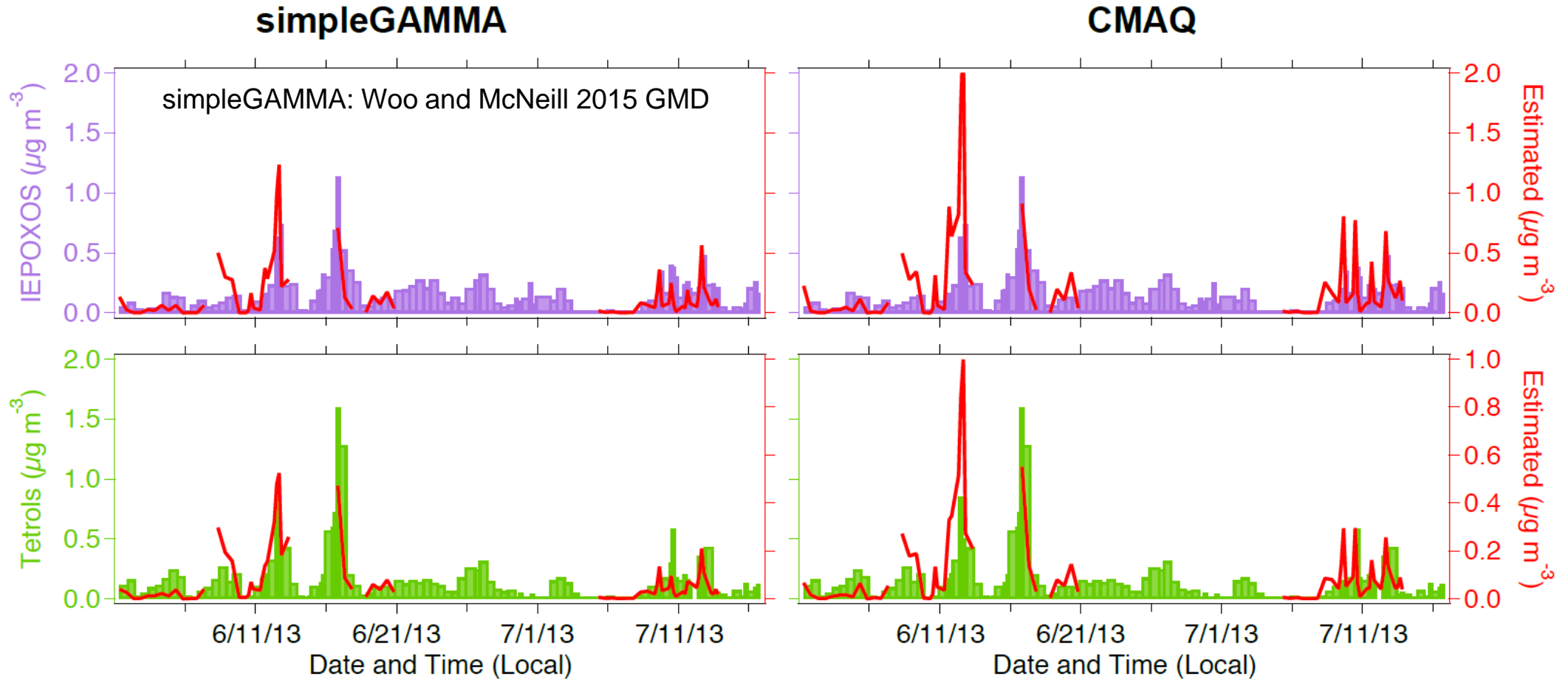


- CMAQ predictions reproduce the observed correlation with sulfate



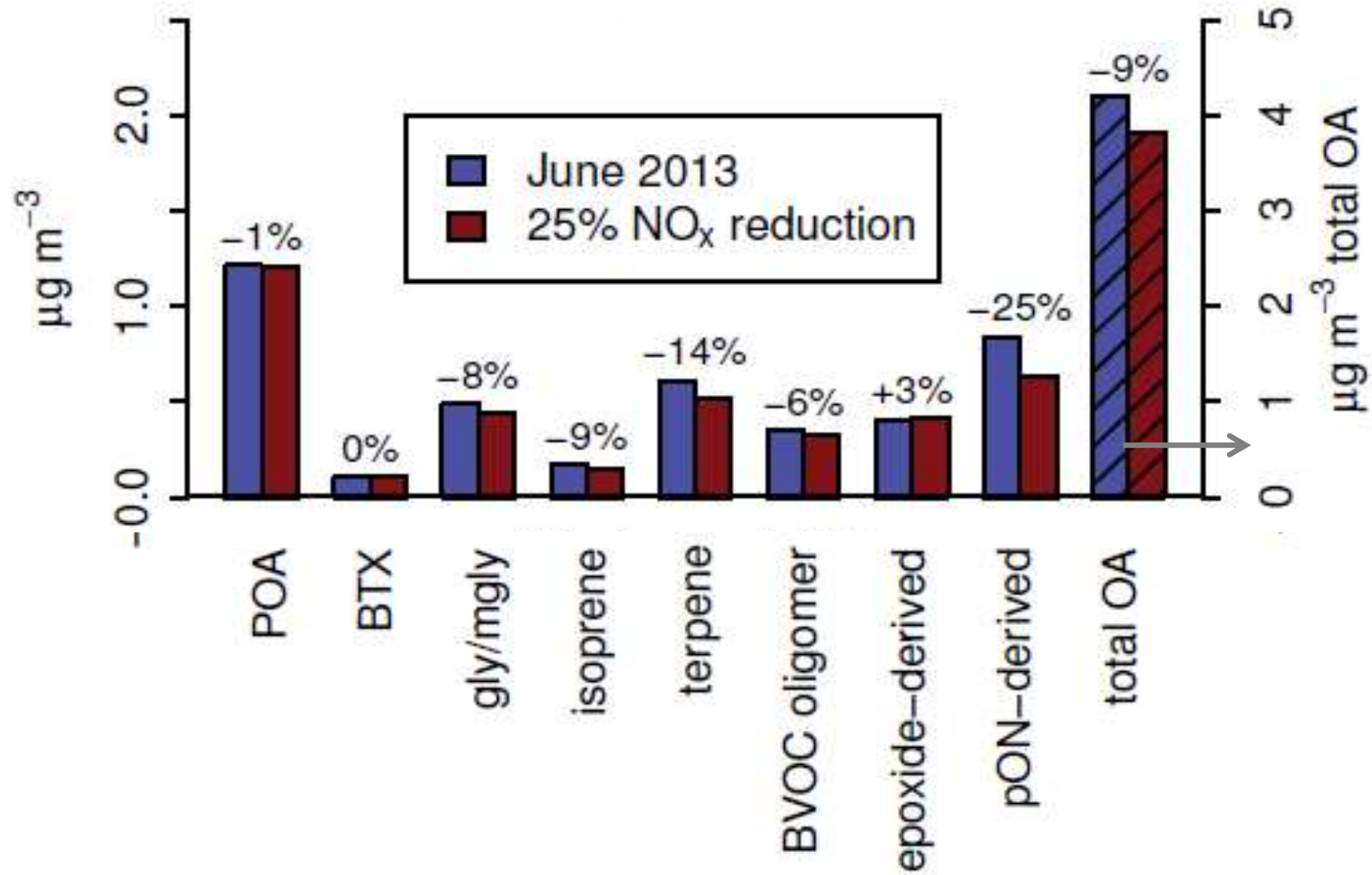
# Modeling of the SOAS-LRK Site

Budisulistiorini et al. in prep



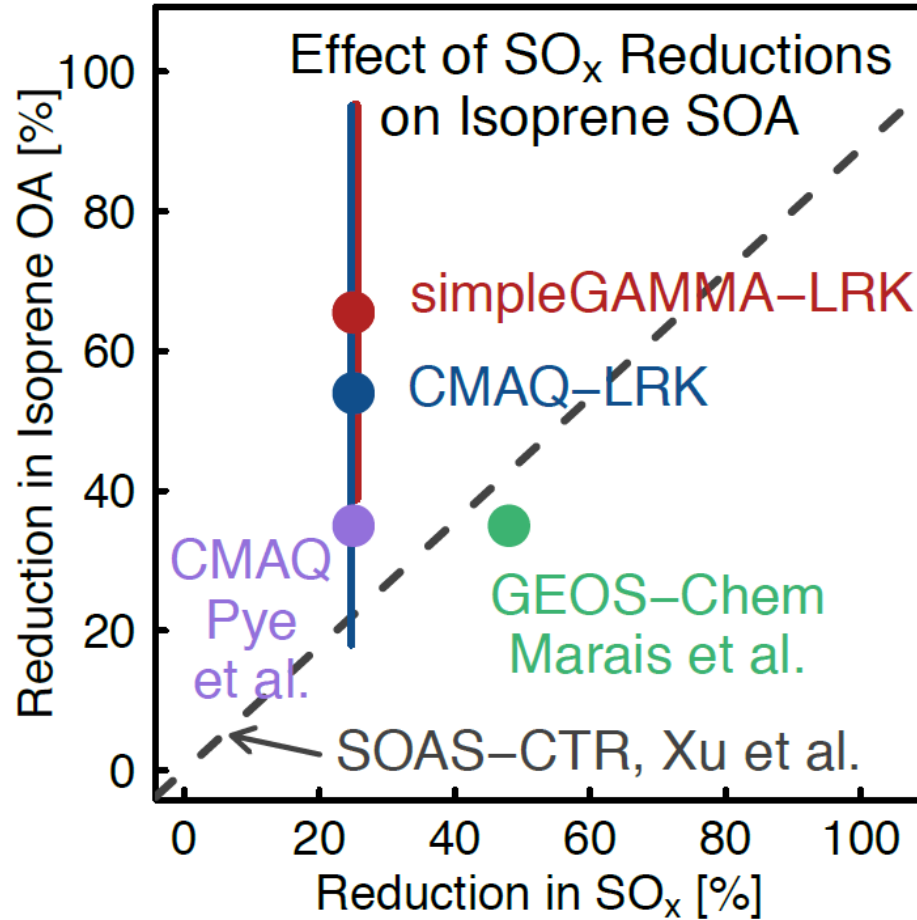
- Models reproduce observed correlation of IEPOX-derived species (tetrols+organosulfates) with aerosol volume and surface area and lack of correlation  $k_{\text{particle}}$

# NO<sub>x</sub> emission reduction leads to OA reduction



# SO<sub>x</sub> Emission Reduction Leads to Isoprene SOA Reduction

Budisulistiorini et al. *in prep.*



Pye et al. 2013 isoprene-OA includes semivolatile and aqueous IEPOX SOA

Marais et al. 2015 isoprene-OA  
58% from IEPOX  
28% from glyoxal

# Conclusions

- Mechanistic SOA parametrizations give us confidence in the predictive capability of models
- For organic nitrate-derived SOA, gas-phase mechanisms should couple with aerosol-phase mechanisms
- CMAQ model predictions are consistent with SOAS observations of  $\text{NO}_y$  components when particle phase organic nitrates undergo fast reaction (3 h)
- Significant progress has been made in modeling isoprene-OA, but uncertainties exist (e.g. the composition of the isoprene-OA factor)
- $\text{NO}_x$  and  $\text{SO}_x$  emission reductions in the Southeast are expected to reduce SOA



# ISSUES IN RESEARCH