APPENDIX A

REFERENCES FOR CONSEQUENCE ANALYSIS METHODS
APPENDIX A REFERENCES FOR CONSEQUENCE ANALYSIS METHODS

Exhibit A-1 lists references that may provide useful information for modeling or calculation methods that could be used in the offsite consequence analyses. This exhibit is not intended to be a complete listing of references that may be used in the consequence analysis; any appropriate model or method may be used.
Appendix A
References for Consequence Analysis Methods

Exhibit A-1
Selected References for Information on Consequence Analysis Methods


TNO Bureau for Industrial Safety, Netherlands Organization for Applied Scientific Research. *Methods for the Determination of Possible Damage to People and Objects Resulting from Releases of*


APPENDIX B

TOXIC SUBSTANCES
APPENDIX B      TOXIC SUBSTANCES

B.1      Data for Toxic Substances

The exhibits in this section of Appendix B provide the data needed to carry out the calculations for
regulated toxic substances using the methods presented in the text of this guidance. Exhibit B-1 presents
data for toxic gases, Exhibit B-2 presents data for toxic liquids, and Exhibit B-3 presents data for several
toxic substances commonly found in water solution and for oleum. Exhibit B-4 provides temperature
correction factors that can be used to correct the release rates estimated for pool evaporation of toxic liquids
that are released at temperatures between 25 °C to 50 °C.

The derivation of the factors presented in Exhibits B-1 - B-4 is discussed in Appendix D. The data
used to develop the factors in Exhibits B-1 and B-2 are primarily from Design Institute for Physical Property
Data (DIPPR), American Institute of Chemical Engineers, Physical and Thermodynamic Properties of Pure
Chemicals, Data Compilation. Other sources, including the National Library of Medicine’s Hazardous
Substances Databank (HSDB) and the Kirk-Othmer Encyclopedia of Chemical Technology, were used for
Exhibits B-1 and B-2 if data were not available from the DIPPR compilation. The factors in Exhibit B-3
were developed using data primarily from Perry's Chemical Engineers' Handbook and the Kirk-Othmer
Encyclopedia of Chemical Technology. The temperature correction factors in Exhibit B-4 were developed
using vapor pressure data derived from the vapor pressure coefficients in the DIPPR compilation.
## Exhibit B-1
### Data for Toxic Gases

<table>
<thead>
<tr>
<th>CAS Number</th>
<th>Chemical Name</th>
<th>Molecular Weight</th>
<th>Ratio of Specific Heats</th>
<th>Toxic Endpoint&lt;sup&gt;a&lt;/sup&gt;</th>
<th>Liquid Factor Boiling (LFB)</th>
<th>Density Factor (DF) (Boiling)</th>
<th>Gas Factor (GF)&lt;sup&gt;b&lt;/sup&gt;</th>
<th>Vapor Pressure @ 25 °C (psia)</th>
<th>Reference Table&lt;sup&gt;b&lt;/sup&gt;</th>
</tr>
</thead>
<tbody>
<tr>
<td>7664-41-7</td>
<td>Ammonia (anhydrous)&lt;sup&gt;c&lt;/sup&gt;</td>
<td>17.03</td>
<td>1.31</td>
<td>0.14</td>
<td>200 ERPG-2</td>
<td>0.073</td>
<td>0.71</td>
<td>14</td>
<td>145</td>
</tr>
<tr>
<td>7784-42-1</td>
<td>Arsine</td>
<td>77.95</td>
<td>1.28</td>
<td>0.0019</td>
<td>0.6 EHS-LOC (IDLH)</td>
<td>0.23</td>
<td>0.30</td>
<td>30</td>
<td>239</td>
</tr>
<tr>
<td>10294-34-5</td>
<td>Boron trichloride</td>
<td>117.17</td>
<td>1.15</td>
<td>0.010</td>
<td>2 EHS-LOC (Tox)&lt;sup&gt;c&lt;/sup&gt;</td>
<td>0.22</td>
<td>0.36</td>
<td>36</td>
<td>22.7</td>
</tr>
<tr>
<td>7637-07-2</td>
<td>Boron trifluoride</td>
<td>67.81</td>
<td>1.20</td>
<td>0.028</td>
<td>10 EHS-LOC (IDLH)</td>
<td>0.25</td>
<td>0.31</td>
<td>28</td>
<td>113</td>
</tr>
<tr>
<td>7782-50-5</td>
<td>Chlorine</td>
<td>70.91</td>
<td>1.32</td>
<td>0.0087</td>
<td>3 ERPG-2</td>
<td>0.19</td>
<td>0.31</td>
<td>29</td>
<td>113</td>
</tr>
<tr>
<td>10049-04-4</td>
<td>Chlorine dioxide</td>
<td>67.45</td>
<td>1.25</td>
<td>0.0028</td>
<td>1 EHS-LOC equivalent (IDLH)&lt;sup&gt;d&lt;/sup&gt;</td>
<td>0.15</td>
<td>0.30</td>
<td>28</td>
<td>24.3</td>
</tr>
<tr>
<td>506-77-4</td>
<td>Cyanogen chloride</td>
<td>61.47</td>
<td>1.22</td>
<td>0.030</td>
<td>12 EHS-LOC equivalent (Tox)&lt;sup&gt;e&lt;/sup&gt;</td>
<td>0.14</td>
<td>0.41</td>
<td>26</td>
<td>23.7</td>
</tr>
<tr>
<td>19287-45-7</td>
<td>Diborane</td>
<td>27.67</td>
<td>1.17</td>
<td>0.0011</td>
<td>1 ERPG-2</td>
<td>0.13</td>
<td>1.13</td>
<td>17</td>
<td>145</td>
</tr>
<tr>
<td>75-21-8</td>
<td>Ethylene oxide</td>
<td>44.05</td>
<td>1.21</td>
<td>0.090</td>
<td>50 ERPG-2</td>
<td>0.12</td>
<td>0.55</td>
<td>22</td>
<td>25.4</td>
</tr>
<tr>
<td>7782-41-4</td>
<td>Fluorine</td>
<td>38.00</td>
<td>1.36</td>
<td>0.0039</td>
<td>2.5 EHS-LOC (IDLH)</td>
<td>0.35</td>
<td>0.32</td>
<td>22</td>
<td>14.8</td>
</tr>
<tr>
<td>50-00-0</td>
<td>Formaldehyde (anhydrous)&lt;sup&gt;f&lt;/sup&gt;</td>
<td>30.03</td>
<td>1.31</td>
<td>0.012</td>
<td>10 ERPG-2</td>
<td>0.10</td>
<td>0.59</td>
<td>19</td>
<td>75.2</td>
</tr>
<tr>
<td>74-90-8</td>
<td>Hydrocyanic acid</td>
<td>27.03</td>
<td>1.30</td>
<td>0.011</td>
<td>10 ERPG-2</td>
<td>0.079</td>
<td>0.72</td>
<td>18</td>
<td>14.8</td>
</tr>
<tr>
<td>7647-01-0</td>
<td>Hydrogen chloride (anhydrous)&lt;sup&gt;c&lt;/sup&gt;</td>
<td>36.46</td>
<td>1.40</td>
<td>0.030</td>
<td>20 ERPG-2</td>
<td>0.15</td>
<td>0.41</td>
<td>21</td>
<td>684</td>
</tr>
<tr>
<td>7664-39-3</td>
<td>Hydrogen fluoride (anhydrous)&lt;sup&gt;c&lt;/sup&gt;</td>
<td>20.01</td>
<td>1.40</td>
<td>0.016</td>
<td>20 ERPG-2</td>
<td>0.066</td>
<td>0.51</td>
<td>16</td>
<td>17.7</td>
</tr>
<tr>
<td>7783-07-5</td>
<td>Hydrogen selenide</td>
<td>80.98</td>
<td>1.32</td>
<td>0.00066</td>
<td>0.2 EHS-LOC (IDLH)</td>
<td>0.21</td>
<td>0.25</td>
<td>31</td>
<td>151</td>
</tr>
<tr>
<td>7783-06-4</td>
<td>Hydrogen sulfide</td>
<td>34.08</td>
<td>1.32</td>
<td>0.042</td>
<td>30 ERPG-2</td>
<td>0.13</td>
<td>0.51</td>
<td>20</td>
<td>302</td>
</tr>
<tr>
<td>74-87-3</td>
<td>Methyl chloride</td>
<td>50.49</td>
<td>1.26</td>
<td>0.82</td>
<td>400 ERPG-2</td>
<td>0.14</td>
<td>0.48</td>
<td>24</td>
<td>83.2</td>
</tr>
<tr>
<td>74-93-1</td>
<td>Methyl mercaptan</td>
<td>48.11</td>
<td>1.20</td>
<td>0.049</td>
<td>25 ERPG-2</td>
<td>0.12</td>
<td>0.55</td>
<td>23</td>
<td>29.2</td>
</tr>
<tr>
<td>10102-43-9</td>
<td>Nitric oxide</td>
<td>30.01</td>
<td>1.38</td>
<td>0.031</td>
<td>25 EHS-LOC (TLV)&lt;sup&gt;f&lt;/sup&gt;</td>
<td>0.21</td>
<td>0.38</td>
<td>19</td>
<td>145</td>
</tr>
<tr>
<td>75-44-5</td>
<td>Phosgene</td>
<td>98.92</td>
<td>1.17</td>
<td>0.00081</td>
<td>0.2 ERPG-2</td>
<td>0.20</td>
<td>0.35</td>
<td>33</td>
<td>27.4</td>
</tr>
</tbody>
</table>

<sup>a</sup> Toxic endpoint selected by HHETC, with the default basis of mg/L. Vapor pressures are based on the assumption of 1 atmosphere pressure. Values in parentheses are default values for the basis of ppm. Liquid and gas factors are based on the assumption of 1 atmosphere pressure. Values in parentheses are default values for the basis of ppm. Values in parentheses are default values for the basis of mg/L.


<sup>c</sup> Data from National Institute of Environmental Health Sciences, 1990.

<sup>d</sup> Buoyant: Substance floats on water.

<sup>e</sup> EHS-LOC (Tox): EHS-LOC (Threshold Limit Value) for toxic effects.

<sup>f</sup> TLV: Threshold Limit Value.
Exhibit B-1 (continued)

<table>
<thead>
<tr>
<th>CAS Number</th>
<th>Chemical Name</th>
<th>Molecular Weight</th>
<th>Ratio of Specific Heats</th>
<th>Toxic Endpoint*</th>
<th>Liquid Factor Boiling (LFB)</th>
<th>Density Factor (DF) (Boiling)</th>
<th>Gas Factor (GF)</th>
<th>Vapor Pressure @ 25 °C (psia)</th>
<th>Reference Table</th>
</tr>
</thead>
<tbody>
<tr>
<td>7803-51-2</td>
<td>Phosphine</td>
<td>34.00</td>
<td>1.29</td>
<td>0.0035</td>
<td>2.5 ERPG-2</td>
<td>0.15</td>
<td>0.66</td>
<td>20</td>
<td>567</td>
</tr>
<tr>
<td>7446-09-5</td>
<td>Sulfur dioxide (anhydrous)</td>
<td>64.07</td>
<td>1.26</td>
<td>0.0078</td>
<td>3 ERPG-2</td>
<td>0.16</td>
<td>0.33</td>
<td>27</td>
<td>58.0</td>
</tr>
<tr>
<td>7783-60-0</td>
<td>Sulfur tetrafluoride</td>
<td>108.06</td>
<td>1.30</td>
<td>0.0092</td>
<td>2 EHS-LOC (Tox⁵)</td>
<td>0.25 (at -73 °C)</td>
<td>36</td>
<td>293</td>
<td>Dense</td>
</tr>
</tbody>
</table>

Notes:

* Toxic endpoints are specified in Appendix A to 40 CFR part 68 in units of mg/L. To convert from units of mg/L to mg/m³, multiply by 1,000. To convert mg/L to ppm, use the following equation:

\[
\text{Endpoint}_{\text{ppm}} = \frac{\text{Endpoint}_{\text{mg/L}} \times 1,000 \times 24.5}{\text{Molecular Weight}}
\]

b "Buoyant" in the Reference Table column refers to the tables for neutrally buoyant gases and vapors; "Dense" refers to the tables for dense gases and vapors. See Appendix D, Section D.4.4, for more information on the choice of reference tables.

c See Exhibit B-3 of this appendix for data on water solutions.

d Gases that are lighter than air may behave as dense gases upon release if liquefied under pressure or cold; consider the conditions of release when choosing the appropriate table.

e LOC is based on the IDLH-equivalent level estimated from toxicity data.

f Cannot be liquefied at 25 °C.

g Not an EHS; LOC-equivalent value was estimated from one-tenth of the IDLH.

h Not an EHS; LOC-equivalent value was estimated from one-tenth of the IDLH-equivalent level estimated from toxicity data.

i Hydrogen fluoride is lighter than air, but may behave as a dense gas upon release under some circumstances (e.g., release under pressure, high concentration in the released cloud) because of hydrogen bonding; consider the conditions of release when choosing the appropriate table.

j LOC based on Threshold Limit Value (TLV) - Time-weighted average (TWA) developed by the American Conference of Governmental Industrial Hygienists (ACGIH).

k Use GF for gas leaks under choked (maximum) flow conditions.
### Exhibit B-2

**Data for Toxic Liquids**

<table>
<thead>
<tr>
<th>CAS</th>
<th>Chemical Name</th>
<th>Molecular Weight</th>
<th>Vapor Pressure at 25 °C (mm Hg)</th>
<th>Toxic Endpoint*</th>
<th>Liquid Factors</th>
<th>Density Leak Factor (LLF)</th>
<th>Reference Table</th>
<th>Worst Case</th>
<th>Alternative Case</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td></td>
<td>mg/L</td>
<td>ppm</td>
<td>Basis</td>
<td>Ambient (LFA)</td>
<td>Boiling (LFB)</td>
<td></td>
<td></td>
</tr>
<tr>
<td>107-02-8</td>
<td>Acrolein</td>
<td>56.06</td>
<td>274</td>
<td>0.0011</td>
<td>0.5 ERPG-2</td>
<td>0.047</td>
<td>0.12</td>
<td>0.58</td>
<td>40 Dense</td>
</tr>
<tr>
<td>107-13-1</td>
<td>Acrylonitrile</td>
<td>53.06</td>
<td>108</td>
<td>0.0076</td>
<td>35 ERPG-2</td>
<td>0.018</td>
<td>0.11</td>
<td>0.61</td>
<td>39 Dense</td>
</tr>
<tr>
<td>814-68-6</td>
<td>Acryl chloride</td>
<td>90.51</td>
<td>110</td>
<td>0.00090</td>
<td>0.2 EHS-LOC (Tox^3)</td>
<td>0.026</td>
<td>0.15</td>
<td>0.44</td>
<td>54 Dense</td>
</tr>
<tr>
<td>107-18-6</td>
<td>Allyl alcohol</td>
<td>58.08</td>
<td>26.1</td>
<td>0.036</td>
<td>15 EHS-LOC (IDLH)</td>
<td>0.0046</td>
<td>0.11</td>
<td>0.58</td>
<td>41 Dense Buoyant^d</td>
</tr>
<tr>
<td>107-11-9</td>
<td>Allylamine</td>
<td>57.10</td>
<td>242</td>
<td>0.0032</td>
<td>1 EHS-LOC (Tox^3)</td>
<td>0.042</td>
<td>0.12</td>
<td>0.64</td>
<td>36 Dense</td>
</tr>
<tr>
<td>7784-34-1</td>
<td>Arsenous trichloride</td>
<td>181.28</td>
<td>10</td>
<td>0.01</td>
<td>1 EHS-LOC (Tox^3)</td>
<td>0.0037</td>
<td>0.21</td>
<td>0.23</td>
<td>100 Dense</td>
</tr>
<tr>
<td>353-42-4</td>
<td>Boron trifluoride</td>
<td>113.89</td>
<td>11</td>
<td>0.023</td>
<td>5 EHS-LOC (Tox^3)</td>
<td>0.0030</td>
<td>0.16</td>
<td>0.49</td>
<td>48 Dense Buoyant^d</td>
</tr>
<tr>
<td>7726-95-6</td>
<td>Bromine</td>
<td>159.81</td>
<td>212</td>
<td>0.0065</td>
<td>1 ERPG-2</td>
<td>0.073</td>
<td>0.23</td>
<td>0.16</td>
<td>150 Dense</td>
</tr>
<tr>
<td>75-15-0</td>
<td>Carbon disulfide</td>
<td>76.14</td>
<td>359</td>
<td>0.16</td>
<td>50 ERPG-2</td>
<td>0.075</td>
<td>0.15</td>
<td>0.39</td>
<td>60 Dense</td>
</tr>
<tr>
<td>67-66-3</td>
<td>Chloroform</td>
<td>119.38</td>
<td>196</td>
<td>0.49</td>
<td>100 EHS-LOC (IDLH)</td>
<td>0.055</td>
<td>0.19</td>
<td>0.33</td>
<td>71 Dense</td>
</tr>
<tr>
<td>542-88-1</td>
<td>Chloromethyl ether</td>
<td>114.96</td>
<td>29.4</td>
<td>0.00025</td>
<td>0.05 EHS-LOC (Tox^3)</td>
<td>0.0080</td>
<td>0.17</td>
<td>0.37</td>
<td>63 Dense</td>
</tr>
<tr>
<td>107-30-2</td>
<td>Chloromethyl methyl ether</td>
<td>80.51</td>
<td>199</td>
<td>0.0018</td>
<td>0.6 EHS-LOC (Tox^3)</td>
<td>0.043</td>
<td>0.15</td>
<td>0.46</td>
<td>51 Dense</td>
</tr>
<tr>
<td>4170-30-3</td>
<td>Crotonaldehyde</td>
<td>70.09</td>
<td>33.1</td>
<td>0.029</td>
<td>10 ERPG-2</td>
<td>0.0066</td>
<td>0.12</td>
<td>0.58</td>
<td>41 Dense Buoyant^d</td>
</tr>
<tr>
<td>123-73-9</td>
<td>Crotonaldehyde, (E)-</td>
<td>70.09</td>
<td>33.1</td>
<td>0.029</td>
<td>10 ERPG-2</td>
<td>0.0066</td>
<td>0.12</td>
<td>0.58</td>
<td>41 Dense Buoyant^d</td>
</tr>
<tr>
<td>108-91-8</td>
<td>Cyclohexylamine</td>
<td>99.18</td>
<td>10.1</td>
<td>0.16</td>
<td>39 EHS-LOC (Tox^3)</td>
<td>0.0025</td>
<td>0.14</td>
<td>0.56</td>
<td>41 Dense Buoyant^d</td>
</tr>
<tr>
<td>75-78-5</td>
<td>Dimethylchlorosilane</td>
<td>129.06</td>
<td>141</td>
<td>0.026</td>
<td>5 ERPG-2</td>
<td>0.042</td>
<td>0.20</td>
<td>0.46</td>
<td>51 Dense</td>
</tr>
<tr>
<td>57-14-7</td>
<td>1,1-Dimethylhydrazine</td>
<td>60.10</td>
<td>157</td>
<td>0.012</td>
<td>5 EHS-LOC (IDLH)</td>
<td>0.028</td>
<td>0.12</td>
<td>0.62</td>
<td>38 Dense</td>
</tr>
<tr>
<td>106-89-8</td>
<td>Epichlorohydrin</td>
<td>92.53</td>
<td>17.0</td>
<td>0.076</td>
<td>20 ERPG-2</td>
<td>0.0040</td>
<td>0.14</td>
<td>0.42</td>
<td>57 Dense Buoyant^d</td>
</tr>
<tr>
<td>107-15-3</td>
<td>Ethylenediamine</td>
<td>60.10</td>
<td>12.2</td>
<td>0.49</td>
<td>200 EHS-LOC (IDLH)</td>
<td>0.0022</td>
<td>0.13</td>
<td>0.54</td>
<td>43 Dense Buoyant^d</td>
</tr>
<tr>
<td>151-56-4</td>
<td>Ethyleneimine</td>
<td>43.07</td>
<td>211</td>
<td>0.018</td>
<td>10 EHS-LOC (IDLH)</td>
<td>0.030</td>
<td>0.10</td>
<td>0.58</td>
<td>40 Dense</td>
</tr>
<tr>
<td>110-00-9</td>
<td>Furan</td>
<td>68.08</td>
<td>600</td>
<td>0.0012</td>
<td>0.4 EHS-LOC (Tox^3)</td>
<td>0.12</td>
<td>0.14</td>
<td>0.52</td>
<td>45 Dense</td>
</tr>
<tr>
<td>302-01-2</td>
<td>Hydrazine</td>
<td>32.05</td>
<td>14.4</td>
<td>0.011</td>
<td>8 EHS-LOC (IDLH)</td>
<td>0.0017</td>
<td>0.069</td>
<td>0.48</td>
<td>48 Buoyant^d</td>
</tr>
</tbody>
</table>

*Exhibit B-2, Data for Toxic Liquids*
## Exhibit B-2 (continued)

<table>
<thead>
<tr>
<th>CAS Number</th>
<th>Chemical Name</th>
<th>Molecular Weight</th>
<th>Vapor Pressure at 25°C (mm Hg)</th>
<th>Toxic Endpoint&lt;sup&gt;a&lt;/sup&gt;</th>
<th>Liquid Factors</th>
<th>Density Leak Factor (LLF)&lt;sup&gt;d&lt;/sup&gt;</th>
<th>Reference Table&lt;sup&gt;b&lt;/sup&gt;</th>
<th>Worst Case</th>
<th>Alternative Case</th>
</tr>
</thead>
<tbody>
<tr>
<td>13463-40-6</td>
<td>Iron, pentacarbonyl-</td>
<td>195.90</td>
<td>40</td>
<td>0.00044 0.05 EHS-LOC (Tox&lt;sup&gt;c&lt;/sup&gt;)</td>
<td>0.016 0.24 0.33</td>
<td>70 Dense</td>
<td>Dense</td>
<td></td>
<td></td>
</tr>
<tr>
<td>78-82-0</td>
<td>Isobutynitrile</td>
<td>69.11</td>
<td>32.7</td>
<td>0.14 50 ERPG-2</td>
<td>0.0064 0.12 0.63</td>
<td>37 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>108-23-6</td>
<td>Isopropyl chloroformate</td>
<td>122.55</td>
<td>28</td>
<td>0.10 20 EHS-LOC (Tox&lt;sup&gt;c&lt;/sup&gt;)</td>
<td>0.0080 0.17 0.45</td>
<td>52 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>126-98-7</td>
<td>Methacrylonitrile</td>
<td>67.09</td>
<td>71.2</td>
<td>0.0027 1 EHS-LOC (TLV&lt;sup&gt;b&lt;/sup&gt;)</td>
<td>0.014 0.12 0.61</td>
<td>38 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>79-22-1</td>
<td>Methyl chloroformate</td>
<td>94.50</td>
<td>108</td>
<td>0.0019 0.5 EHS-LOC (Tox&lt;sup&gt;c&lt;/sup&gt;)</td>
<td>0.026 0.16 0.40</td>
<td>58 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>60-34-4</td>
<td>Methyl hydrazine</td>
<td>46.07</td>
<td>49.6</td>
<td>0.0094 5 EHS-LOC (IDLH)</td>
<td>0.0074 0.094 0.56</td>
<td>42 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>624-83-9</td>
<td>Methyl isocyanate</td>
<td>57.05</td>
<td>457</td>
<td>0.0012 0.5 ERPG-2</td>
<td>0.079 0.13 0.52</td>
<td>45 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>556-64-9</td>
<td>Methyl thiocyanate</td>
<td>73.12</td>
<td>10</td>
<td>0.085 29 EHS-LOC (Tox&lt;sup&gt;c&lt;/sup&gt;)</td>
<td>0.0020 0.11 0.45</td>
<td>51 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>75-79-6</td>
<td>Methyltrichlorosilane</td>
<td>149.48</td>
<td>173</td>
<td>0.018 3 ERPG-2</td>
<td>0.057 0.22 0.38</td>
<td>61 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>13463-39-3</td>
<td>Nickel carbonyl</td>
<td>170.73</td>
<td>400</td>
<td>0.00067 0.1 EHS-LOC (Tox&lt;sup&gt;c&lt;/sup&gt;)</td>
<td>0.14 0.26 0.37</td>
<td>63 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>7697-37-2</td>
<td>Nitric acid (100%)&lt;sup&gt;3&lt;/sup&gt;</td>
<td>63.01</td>
<td>63.0</td>
<td>0.026 10 EHS-LOC (Tox&lt;sup&gt;c&lt;/sup&gt;)</td>
<td>0.012 0.12 0.32</td>
<td>73 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>79-21-0</td>
<td>Peracetic acid</td>
<td>76.05</td>
<td>13.9</td>
<td>0.0045 1.5 EHS-LOC (Tox&lt;sup&gt;c&lt;/sup&gt;)</td>
<td>0.0029 0.12 0.40</td>
<td>58 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>594-42-3</td>
<td>Perchloromethymercaptan</td>
<td>185.87</td>
<td>6</td>
<td>0.0076 1 EHS-LOC (IDLH)</td>
<td>0.0023 0.20 0.29</td>
<td>81 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>10025-87-3</td>
<td>Phosphorus oxychloride</td>
<td>153.33</td>
<td>35.8</td>
<td>0.0030 0.5 EHS-LOC (Tox&lt;sup&gt;c&lt;/sup&gt;)</td>
<td>0.012 0.20 0.29</td>
<td>80 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
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</tr>
<tr>
<td>7719-12-2</td>
<td>Phosphorus trichloride</td>
<td>137.33</td>
<td>120</td>
<td>0.028 5 EHS-LOC (IDLH)</td>
<td>0.037 0.20 0.31</td>
<td>75 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
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<td></td>
</tr>
<tr>
<td>110-89-4</td>
<td>Piperidine</td>
<td>85.15</td>
<td>32.1</td>
<td>0.022 6 EHS-LOC (Tox&lt;sup&gt;c&lt;/sup&gt;)</td>
<td>0.0072 0.13 0.57</td>
<td>41 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>107-12-0</td>
<td>Propionitrile</td>
<td>55.08</td>
<td>47.3</td>
<td>0.0037 1.6 EHS-LOC (Tox&lt;sup&gt;c&lt;/sup&gt;)</td>
<td>0.0080 0.10 0.63</td>
<td>37 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>109-61-5</td>
<td>Propyl chloroformate</td>
<td>122.56</td>
<td>20.0</td>
<td>0.010 2 EHS-LOC (Tox&lt;sup&gt;c&lt;/sup&gt;)</td>
<td>0.0058 0.17 0.45</td>
<td>52 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>75-55-8</td>
<td>Propyleneimine</td>
<td>57.10</td>
<td>187</td>
<td>0.12 50 EHS-LOC (IDLH)</td>
<td>0.032 0.12 0.61</td>
<td>39 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>75-56-9</td>
<td>Propylene oxide</td>
<td>58.08</td>
<td>533</td>
<td>0.59 250 ERPG-2</td>
<td>0.093 0.13 0.59</td>
<td>40 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>7446-11-9</td>
<td>Sulfur trioxide</td>
<td>80.06</td>
<td>263</td>
<td>0.010 3 ERPG-2</td>
<td>0.057 0.15 0.26</td>
<td>91 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>75-74-1</td>
<td>Tetramethyllead</td>
<td>267.33</td>
<td>22.5</td>
<td>0.0040 0.4 EHS-LOC (IDLH)</td>
<td>0.011 0.29 0.24</td>
<td>96 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
<tr>
<td>509-14-8</td>
<td>Tetranitromethane</td>
<td>196.04</td>
<td>11.4</td>
<td>0.0040 0.5 EHS-LOC (IDLH)</td>
<td>0.0045 0.22 0.30</td>
<td>78 Dense</td>
<td>Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

April 15, 1999

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### Exhibit B-2 (continued)

<table>
<thead>
<tr>
<th>CAS Number</th>
<th>Chemical Name</th>
<th>Molecular Weight</th>
<th>Vapor Pressure at 25 °C (mm Hg)</th>
<th>Toxic Endpoint*</th>
<th>Liquid Factors</th>
<th>Density Factor (DF)</th>
<th>Liquid Leak Factor (LLF)</th>
<th>Reference Table</th>
</tr>
</thead>
<tbody>
<tr>
<td>7550-45-0</td>
<td>Titanium tetrachloride</td>
<td>189.69</td>
<td>12.4</td>
<td>0.020</td>
<td>2.6</td>
<td>ERPG-2</td>
<td>0.0048</td>
<td>82</td>
</tr>
<tr>
<td>584-84-9</td>
<td>Toluene 2,4-diisocyanate</td>
<td>174.16</td>
<td>0.017</td>
<td>0.0070</td>
<td>1</td>
<td>EHS-LOC (IDLH)</td>
<td>0.000006</td>
<td>59</td>
</tr>
<tr>
<td>91-08-7</td>
<td>Toluene 2,6-diisocyanate</td>
<td>174.16</td>
<td>0.05</td>
<td>0.0070</td>
<td>1</td>
<td>EHS-LOC (IDLH)</td>
<td>0.000018</td>
<td>59</td>
</tr>
<tr>
<td>26471-62-5</td>
<td>Toluene diisocyanate (unspecified isomer)</td>
<td>174.16</td>
<td>0.017</td>
<td>0.0070</td>
<td>1</td>
<td>EHS-LOC equivalent (IDLH)</td>
<td>0.000006</td>
<td>59</td>
</tr>
<tr>
<td>75-77-4</td>
<td>Trimethylchlorosilane</td>
<td>108.64</td>
<td>231</td>
<td>0.050</td>
<td>11</td>
<td>EHS-LOC (Tox²)</td>
<td>0.061</td>
<td>41</td>
</tr>
<tr>
<td>108-05-4</td>
<td>Vinyl acetate monomer</td>
<td>86.09</td>
<td>113</td>
<td>0.26</td>
<td>75</td>
<td>ERPG-2</td>
<td>0.026</td>
<td>45</td>
</tr>
</tbody>
</table>

Notes:

* Toxic endpoints are specified in the Appendix A to 40 CFR part 68 in units of mg/L. To convert from units of mg/L to mg/m³, multiply by 1,000. To convert mg/L to ppm, use the following equation:

\[
\text{Endpoint}_\text{ppm} = \frac{\text{Endpoint}_\text{mg/L} \times 1,000 \times 24.5}{\text{Molecular Weight}}
\]

* "Buoyant" in the Reference Table column refers to the tables for neutrally buoyant gases and vapors; "Dense" refers to the tables for dense gases and vapors. See Appendix D, Section D.4.4, for more information on the choice of reference tables.
* LOC is based on IDLH-equivalent level estimated from toxicity data.
* Use dense gas table if substance is at an elevated temperature.
* LOC based on Threshold Limit Value (TLV) - Time-weighted average (TWA) developed by the American Conference of Governmental Industrial Hygienists (ACGIH).
* See Exhibit B-3 of this appendix for data on water solutions.
* LOC for this isomer is based on IDLH for toluene 2,4-diisocyanate.
* Not an EHS; LOC-equivalent value is based on IDLH for toluene 2,4-diisocyanate.
* Use the LLF only for leaks from tanks at atmospheric pressure.
## Exhibit B-3

Data for Water Solutions of Toxic Substances and for Oleum
For Wind Speeds of 1.5 and 3.0 Meters per Second (m/s)

<table>
<thead>
<tr>
<th>CAS Number</th>
<th>Regulated Substance in Solution</th>
<th>Molecular Weight</th>
<th>Toxic Endpoint&lt;sup&gt;a&lt;/sup&gt;</th>
<th>Initial Concentration (Wt %)</th>
<th>10-min. Average Vapor Pressure (mm Hg)</th>
<th>Liquid Factor at 25°C (LFA)</th>
<th>Density Factor (DF)</th>
<th>Liquid Leak Factor (LLF)</th>
<th>Reference Table&lt;sup&gt;b&lt;/sup&gt;</th>
<th>Worst</th>
<th>Alternative</th>
</tr>
</thead>
<tbody>
<tr>
<td>7664-41-7</td>
<td>Ammonia</td>
<td>17.03</td>
<td>ERPG-2</td>
<td>30: 332; 248; 0.026; 0.019; 0.55</td>
<td>43 Buoyant Buoyant</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>24: 241; 184; 0.019; 0.014; 0.54</td>
<td>44 Buoyant Buoyant</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>20: 190; 148; 0.015; 0.011; 0.53</td>
<td>44 Buoyant Buoyant</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>50-00-0</td>
<td>Formaldehyde</td>
<td>30.027</td>
<td>ERPG-2</td>
<td>37: 1.5; 1.4; 0.0002; 0.0002; 0.44</td>
<td>53 Buoyant Buoyant</td>
<td></td>
<td></td>
<td></td>
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</tr>
<tr>
<td>7647-01-0</td>
<td>Hydrochloric acid</td>
<td>36.46</td>
<td>ERPG-2</td>
<td>38: 78; 55; 0.010; 0.0070; 0.41</td>
<td>57 Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>37: 67; 48; 0.0085; 0.0062; 0.42</td>
<td>57 Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>36°: 56; 42; 0.0072; 0.0053; 0.42</td>
<td>57 Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>34°: 38; 29; 0.0048; 0.0037; 0.42</td>
<td>56 Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
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<td></td>
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</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>30°: 13; 12; 0.0016; 0.0015; 0.42</td>
<td>55 Buoyant&lt;sup&gt;d&lt;/sup&gt; Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
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</tr>
<tr>
<td>7664-39-3</td>
<td>Hydrofluoric acid</td>
<td>20.01</td>
<td>ERPG-2</td>
<td>70: 124; 107; 0.011; 0.010; 0.39</td>
<td>61 Buoyant Buoyant</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
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</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>50: 16; 15; 0.0014; 0.0013; 0.41</td>
<td>58 Buoyant Buoyant</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>7697-37-2</td>
<td>Nitric acid</td>
<td>63.01</td>
<td>EHS-LOC (IDLH)</td>
<td>90: 25; 22; 0.0046; 0.0040; 0.33</td>
<td>71 Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
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<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>85: 17; 16; 0.0032; 0.0029; 0.33</td>
<td>70 Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
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</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td>80: 10.2; 10; 0.0019; 0.0018; 0.33</td>
<td>70 Dense Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>8014-95-7</td>
<td>Oleum - based on SO₃</td>
<td>80.06 (SO₃)</td>
<td>ERPG-2</td>
<td>30 (SO₃); 3.5 (SO₃); 3.4 (SO₃); 0.0008; 0.0007; 0.25</td>
<td>93 Buoyant&lt;sup&gt;d&lt;/sup&gt; Buoyant&lt;sup&gt;d&lt;/sup&gt;</td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>

**Notes:**

<sup>a</sup> Toxic endpoints are specified in the Appendix A to 40 CFR part 68 in units of mg/L. See Notes to Exhibit B-1 or B-2 for converting to other units.

<sup>b</sup> "Buoyant" in the Reference Table column refers to the tables for neutrally buoyant gases and vapors; "Dense" refers to the tables for dense gases and vapors. See Appendix D, Section D.4.4, for more information on the choice of reference tables.

<sup>c</sup> Hydrochloric acid in concentrations below 37 percent is not regulated.

<sup>d</sup> Use dense gas table if substance is at an elevated temperature.
### Exhibit B-4
Temperature Correction Factors for Liquids Evaporating from Pools at Temperatures Between 25 °C and 50 °C (77 °F and 122 °F)

<table>
<thead>
<tr>
<th>CAS Number</th>
<th>Chemical Name</th>
<th>Boiling Point (°C)</th>
<th>Temperature Correction Factor (TCF)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td></td>
<td>30 °C (86 °F)</td>
</tr>
<tr>
<td>107-02-8</td>
<td>Acrolein</td>
<td>52.69</td>
<td>1.2</td>
</tr>
<tr>
<td>107-13-1</td>
<td>Acrylonitrile</td>
<td>77.35</td>
<td>1.2</td>
</tr>
<tr>
<td>814-68-6</td>
<td>Acryl chloride</td>
<td>75.00</td>
<td>ND</td>
</tr>
<tr>
<td>107-18-6</td>
<td>Allyl alcohol</td>
<td>97.08</td>
<td>1.3</td>
</tr>
<tr>
<td>107-11-9</td>
<td>Allylamine</td>
<td>53.30</td>
<td>1.2</td>
</tr>
<tr>
<td>7784-34-1</td>
<td>Arsenous trichloride</td>
<td>130.06</td>
<td>ND</td>
</tr>
<tr>
<td>353-42-4</td>
<td>Boron trifluoride compound with methyl ether (1:1)</td>
<td>126.85</td>
<td>ND</td>
</tr>
<tr>
<td>7726-95-6</td>
<td>Bromine</td>
<td>58.75</td>
<td>1.2</td>
</tr>
<tr>
<td>75-15-0</td>
<td>Carbon disulfide</td>
<td>46.22</td>
<td>1.2</td>
</tr>
<tr>
<td>67-66-3</td>
<td>Chloroform</td>
<td>61.18</td>
<td>1.2</td>
</tr>
<tr>
<td>542-88-1</td>
<td>Chloromethyl ether</td>
<td>104.85</td>
<td>1.3</td>
</tr>
<tr>
<td>107-30-2</td>
<td>Chloromethyl methyl ether</td>
<td>59.50</td>
<td>1.2</td>
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<tr>
<td>4170-30-3</td>
<td>Crotonaldehyde</td>
<td>104.10</td>
<td>1.3</td>
</tr>
<tr>
<td>123-73-9</td>
<td>Crotonaldehyde, (E)-</td>
<td>102.22</td>
<td>1.3</td>
</tr>
<tr>
<td>108-91-8</td>
<td>Cyclohexylamine</td>
<td>134.50</td>
<td>1.3</td>
</tr>
<tr>
<td>75-78-5</td>
<td>Dimethyl dichlorosilane</td>
<td>70.20</td>
<td>1.2</td>
</tr>
<tr>
<td>57-14-7</td>
<td>1,1-Dimethylhydrazine</td>
<td>63.90</td>
<td>ND</td>
</tr>
<tr>
<td>106-89-8</td>
<td>Epichlorohydrin</td>
<td>118.50</td>
<td>1.3</td>
</tr>
<tr>
<td>107-15-3</td>
<td>Ethylenediamine</td>
<td>36.26</td>
<td>1.3</td>
</tr>
<tr>
<td>151-56-4</td>
<td>Ethyleneimine</td>
<td>55.85</td>
<td>1.2</td>
</tr>
<tr>
<td>110-00-9</td>
<td>Furan</td>
<td>31.35</td>
<td>1.2</td>
</tr>
<tr>
<td>302-01-2</td>
<td>Hydrazine</td>
<td>113.50</td>
<td>1.3</td>
</tr>
<tr>
<td>13463-40-6</td>
<td>Iron, pentacarbonyl-</td>
<td>102.65</td>
<td>ND</td>
</tr>
<tr>
<td>78-82-0</td>
<td>Isobutynitrile</td>
<td>103.61</td>
<td>1.3</td>
</tr>
<tr>
<td>108-23-6</td>
<td>Isopropyl chloroformate</td>
<td>104.60</td>
<td>ND</td>
</tr>
<tr>
<td>126-98-7</td>
<td>Methacrylonitrile</td>
<td>90.30</td>
<td>1.2</td>
</tr>
<tr>
<td>79-22-1</td>
<td>Methyl chloroformate</td>
<td>70.85</td>
<td>1.3</td>
</tr>
<tr>
<td>60-34-4</td>
<td>Methyl hydrazine</td>
<td>87.50</td>
<td>ND</td>
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<tr>
<td>624-83-9</td>
<td>Methyl isocyanate</td>
<td>38.85</td>
<td>1.2</td>
</tr>
<tr>
<td>556-64-9</td>
<td>Methyl thiocyanate</td>
<td>130.00</td>
<td>ND</td>
</tr>
<tr>
<td>75-79-6</td>
<td>Methyltrichlorosilane</td>
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<td>1.2</td>
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### Exhibit B-4 (continued)

<table>
<thead>
<tr>
<th>CAS Number</th>
<th>Chemical Name</th>
<th>Boiling Point (°C)</th>
<th>Temperature Correction Factor (TCF)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td></td>
<td>30 °C (86 °F)</td>
</tr>
<tr>
<td>13463-39-3</td>
<td>Nickel carbonyl</td>
<td>42.85</td>
<td>ND</td>
</tr>
<tr>
<td>7697-37-2</td>
<td>Nitric acid</td>
<td>83.00</td>
<td>1.3</td>
</tr>
<tr>
<td>79-21-0</td>
<td>Peracetic acid</td>
<td>109.85</td>
<td>1.3</td>
</tr>
<tr>
<td>594-42-3</td>
<td>Perchloromethylmercaptan</td>
<td>147.00</td>
<td>ND</td>
</tr>
<tr>
<td>10025-87-3</td>
<td>Phosphorus oxychloride</td>
<td>105.50</td>
<td>1.3</td>
</tr>
<tr>
<td>7719-12-2</td>
<td>Phosphorus trichloride</td>
<td>76.10</td>
<td>1.2</td>
</tr>
<tr>
<td>110-89-4</td>
<td>Piperidine</td>
<td>106.40</td>
<td>1.3</td>
</tr>
<tr>
<td>107-12-0</td>
<td>Propionitrile</td>
<td>97.35</td>
<td>1.3</td>
</tr>
<tr>
<td>109-61-5</td>
<td>Propyl chloroformate</td>
<td>112.40</td>
<td>ND</td>
</tr>
<tr>
<td>75-55-8</td>
<td>Propyleneimine</td>
<td>60.85</td>
<td>1.2</td>
</tr>
<tr>
<td>75-56-9</td>
<td>Propylene oxide</td>
<td>33.90</td>
<td>1.2</td>
</tr>
<tr>
<td>7446-11-9</td>
<td>Sulfur trioxide</td>
<td>44.75</td>
<td>1.3</td>
</tr>
<tr>
<td>75-74-1</td>
<td>Tetramethylethyl</td>
<td>110.00</td>
<td>ND</td>
</tr>
<tr>
<td>509-14-8</td>
<td>Tetranitromethane</td>
<td>125.70</td>
<td>1.3</td>
</tr>
<tr>
<td>7550-45-0</td>
<td>Titanium tetrachloride</td>
<td>135.85</td>
<td>1.3</td>
</tr>
<tr>
<td>584-84-9</td>
<td>Toluene 2,4-diisocyanate</td>
<td>251.00</td>
<td>1.6</td>
</tr>
<tr>
<td>91-08-7</td>
<td>Toluene 2,6-diisocyanate</td>
<td>244.85</td>
<td>ND</td>
</tr>
<tr>
<td>26471-62-5</td>
<td>Toluene diisocyanate (unspe-</td>
<td>250.00</td>
<td>1.6</td>
</tr>
<tr>
<td></td>
<td>cified isomer)</td>
<td></td>
<td></td>
</tr>
<tr>
<td>75-77-4</td>
<td>Trimethylchlorosilane</td>
<td>57.60</td>
<td>1.2</td>
</tr>
<tr>
<td>108-05-4</td>
<td>Vinyl acetate monomer</td>
<td>72.50</td>
<td>1.2</td>
</tr>
</tbody>
</table>

**Notes:**

ND: No data available.
LFB: Chemical above boiling point at this temperature; use LFB for analysis.
B.2 Mixtures Containing Toxic Liquids

In case of a spill of a liquid mixture containing a regulated toxic substance (with the exception of common water solutions, discussed in Section 3.3 in the text), the area of the pool formed by the entire liquid spill is determined as described in Section 3.2.2 or 3.2.3. For the area determination, if the density of the mixture is unknown, the density of the regulated substance in the mixture may be assumed as the density of the entire mixture.

If the partial vapor pressure of the regulated substance in the mixture is known, that vapor pressure may be used to derive a release rate using the equations in Section 3.2. If the partial vapor pressure of the regulated toxic substance in the mixture is unknown, it may be estimated from the vapor pressure of the pure substance (listed in Exhibit B-2, Appendix B) and the concentration in the mixture, if you assume the mixture is an ideal solution, where an ideal solution is one in which there is complete uniformity of cohesive forces. This method may overestimate or underestimate the partial pressure for a regulated substance that interacts with the other components of a mixture or solution. For example, water solutions are generally not ideal. This method is likely to overestimate the partial pressure of regulated substances in water solution if there is hydrogen bonding in the solution (e.g., solutions of acids or alcohols in water).

To estimate partial pressure for a regulated substance in a mixture or solution, use the following steps, based on Raoults's Law for ideal solutions:

- Determine the mole fraction of the regulated substance in the mixture.
  - The mole fraction of the regulated substance in the mixture is the number of moles of the regulated substance in the mixture divided by the total number of moles of all substances in the mixture.
  - If the molar concentration (moles per liter) of each component of the mixture is known, the mole fraction may be determined as follows:

    \[
    X_r = \frac{M_r \times V_i}{\sum_{i=1}^{n} (M_i \times V_i)} \quad \text{(B-1)}
    \]

    or (canceling out \(V_i\)):

    \[
    X_r = \frac{M_r}{\sum_{i=1}^{n} M_i} \quad \text{(B-2)}
    \]
where: \( X_r \) = Mole fraction of regulated substance in mixture (unitless)  
\( M_r \) = Molar concentration of regulated substance in mixture (moles per liter)  
\( V_t \) = Total volume of mixture (liters)  
n = Number of components of mixture  
\( M_i \) = Molar concentration of each component of mixture (moles per liter)

For a mixture with three components, this would correspond to:

\[
X_r = \frac{M_r}{M_r + M_2 + M_3} \quad (B-3)
\]

where: \( X_r \) = Mole fraction of regulated substance in mixture (unitless)  
\( M_r \) = Molar concentration of regulated substance (first component) in mixture (moles per liter)  
\( M_2 \) = Molar concentration of second component of mixture (moles per liter)  
\( M_3 \) = Molar concentration of any other components of mixture (moles per liter)  

-- If the weight of each of the components of the mixture is known, the mole fraction of the regulated substance in the mixture may be calculated as follows:

\[
X_r = \frac{\left( \frac{W_r}{MW_r} \right)}{\sum_{i=1}^{n} \left( \frac{W_i}{MW_i} \right)} \quad (B-4)
\]

where: \( X_r \) = Mole fraction of the regulated substance  
\( W_r \) = Weight of the regulated substance  
\( MW_r \) = Molecular weight of the regulated substance  
n = Number of components of the mixture  
\( W_i \) = Weight of each component of the mixture  
\( MW_i \) = Molecular weight of each component of the mixture

(Note: Weights can be in any consistent units.)

For a mixture with three components, this corresponds to:

\[
X_r = \frac{\left( \frac{W_r}{MW_r} \right)}{\left( \frac{W_r}{MW_r} \right) + \left( \frac{W_2}{MW_2} \right) + \left( \frac{W_3}{MW_3} \right)} \quad (B-5)
\]
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where:  
\[ X_r = \text{Mole fraction of the regulated substance} \]
\[ W_r = \text{Weight of the regulated substance (first component of the mixture)} \]
\[ MW_r = \text{Molecular weight of the regulated substance} \]
\[ W_2 = \text{Weight of the second component of the mixture} \]
\[ MW_2 = \text{Molecular weight of the second component of the mixture} \]
\[ W_3 = \text{Weight of the third component of the mixture} \]
\[ MW_3 = \text{Molecular weight of the third component of the mixture} \]

(Note: Weights can be in any consistent units.)

- Estimate the partial vapor pressure of the regulated substance in the mixture as follows:

\[ VP_m = X_r \times VP_p \]  \hspace{1cm} \text{(B-6)}

where:  
\[ VP_m = \text{Partial vapor pressure of the regulated substance in the mixture (millimeters of mercury (mm Hg))} \]
\[ X_r = \text{Mole fraction of the regulated substance (unitless)} \]
\[ VP_p = \text{Vapor pressure of the regulated substance in pure form at the same temperature as the mixture (mm Hg) (vapor pressure at 25 °C is given in Exhibit B-1, Appendix B)} \]

The evaporation rate for the regulated substance in the mixture is determined as for pure substances, with \( VP_m \) as the vapor pressure. If the mixture contains more than one regulated toxic substance, carry out the analysis individually for each of the regulated components. The release rate equation is:

\[ QR = \frac{0.0035 \times U^{0.78} \times MW^{2/3} \times A \times VP}{T} \]  \hspace{1cm} \text{(B-7)}

where:  
\[ QR = \text{Evaporation rate (pounds per minute)} \]
\[ U = \text{Wind speed (meters per second)} \]
\[ MW = \text{Molecular weight (given in Exhibit B-2, Appendix B)} \]
\[ A = \text{Surface area of pool formed by the entire quantity of the mixture (square feet) (determined as described in 3.2.2)} \]
\[ VP = \text{Vapor pressure (mm Hg) (VP_m from Equation B-4 above)} \]
\[ T = \text{Temperature (Kelvin (K); temperature in °C plus 273, or 298 for 25 °C)} \]

See Appendix D, Section D.2.1 for more discussion of the evaporation rate equation. Equation B-7 is derived from Equation D-1.

Worst-case consequence distances to the toxic endpoint may be estimated from the release rate using the tables and instructions presented in Chapter 4.
APPENDIX C

FLAMMABLE SUBSTANCES

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APPENDIX C FLAMMABLE SUBSTANCES

C.1 Equation for Estimation of Distance to 1 psi Overpressure for Vapor Cloud Explosions

For a worst-case release of flammable gases and volatile flammable liquids, the release rate is not considered. The total quantity of the flammable substance is assumed to form a vapor cloud. The entire contents of the cloud is assumed to be within the flammability limits, and the cloud is assumed to explode. For the worst-case analysis, 10 percent of the flammable vapor in the cloud is assumed to participate in the explosion (i.e., the yield factor is 0.10). Consequence distances to an overpressure level of 1 pound per square inch (psi) may be determined using the following equation, which is based on the TNT-equivalency method:

\[ D = 17 \times \left( 0.1 \times W_f \times \frac{Hc_f}{HC_{TNT}} \right)^{1/3} \]  
(C-1)

where:
- \( D \) = Distance to overpressure of 1 psi (meters)
- \( W_f \) = Weight of flammable substance (kilograms or pounds/2.2)
- \( Hc_f \) = Heat of combustion of flammable substance (kilojoules per kilogram)
- (listed in Exhibit C-1)
- \( HC_{TNT} \) = Heat of explosion of trinitrotoluene (TNT) (4,680 kilojoules per kilogram)

The factor 17 is a constant for damages associated with 1.0 psi overpressures. The factor 0.1 represents an explosion efficiency of 10 percent. To convert distances from meters to miles, multiply by 0.00062.

Alternatively, use the following equation for quantity in pounds and distance in miles:

\[ D_{mi} = 0.0081 \times \left( 0.1 \times W_{lb} \times \frac{Hc_f}{HC_{TNT}} \right)^{1/3} \]  
(C-2)

where:
- \( D_{mi} \) = Distance to overpressure of 1 psi (miles)
- \( W_{lb} \) = Weight of flammable substance (pounds)

These equations were used to derive Reference Table 13 for worst-case distances to the overpressure endpoint (1 psi) for vapor cloud explosions.

C.2 Mixtures of Flammable Substances

For a mixture of flammable substances, you may estimate the heat of combustion of the mixture from the heats of combustion of the components of the mixture using the equation below and then use the equation given in the previous section of this appendix to determine the vapor cloud explosion distance. The heat of combustion of the mixture may be estimated as follows:

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\[ HC_m = \frac{W_x}{W_m} \times HC_x + \frac{W_y}{W_m} \times HC_y \quad (C-3) \]

where:

- \( HC_m \) = Heat of combustion of mixture (kilojoules per kilogram)
- \( W_x \) = Weight of component "X" in mixture (kilograms or pounds/2.2)
- \( W_m \) = Total weight of mixture (kilograms or pounds/2.2)
- \( HC_x \) = Heat of combustion of component "X" (kilojoules per kilogram)
- \( W_y \) = Weight of component "Y" in mixture (kilograms or pounds/2.2)
- \( HC_y \) = Heat of combustion of component "Y" (kilojoules per kilogram)

Heats of combustion for regulated flammable substances are listed in Exhibit C-1 in the next section (Section C.3) of this appendix.

C.3 Data for Flammable Substances

The exhibits in this section of Appendix C provide the data needed to carry out the calculations for regulated flammable substances using the methods presented in the text of this guidance. Exhibit C-1 presents heat of combustion data for all regulated flammable substances, Exhibit C-2 presents additional data for flammable gases, and Exhibit C-3 presents additional data for flammable liquids. The heats of combustion in Exhibit C-1 and the data used to develop the factors in Exhibits C-2 and C-3 are primarily from Design Institute for Physical Property Data, American Institute of Chemical Engineers, Physical and Thermodynamic Properties of Pure Chemicals, Data Compilation. The derivation of the factors presented in Exhibits C-2 and C-3 is discussed in Appendix D.
### Exhibit C-1
#### Heats of Combustion for Flammable Substances

<table>
<thead>
<tr>
<th>CAS No.</th>
<th>Chemical Name</th>
<th>Physical State at 25° C</th>
<th>Heat of Combustion (k joule/kg)</th>
</tr>
</thead>
<tbody>
<tr>
<td>75-07-0</td>
<td>Acetaldehyde</td>
<td>Gas</td>
<td>25,072</td>
</tr>
<tr>
<td>74-86-2</td>
<td>Acetylene [Ethyne]</td>
<td>Gas</td>
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</tr>
<tr>
<td>598-73-2</td>
<td>Bromotrifluoroethylene [Ethene, bromotrifluoro-]</td>
<td>Gas</td>
<td>1,967</td>
</tr>
<tr>
<td>106-99-0</td>
<td>1,3-Butadiene</td>
<td>Gas</td>
<td>44,548</td>
</tr>
<tr>
<td>106-97-8</td>
<td>Butane</td>
<td>Gas</td>
<td>45,719</td>
</tr>
<tr>
<td>25167-67-3</td>
<td>Butene</td>
<td>Gas</td>
<td>45,200’</td>
</tr>
<tr>
<td>590-18-1</td>
<td>2-Butene-cis</td>
<td>Gas</td>
<td>45,171</td>
</tr>
<tr>
<td>624-64-6</td>
<td>2-Butene-trans [2-Butene, (E)]</td>
<td>Gas</td>
<td>45,069</td>
</tr>
<tr>
<td>106-98-9</td>
<td>1-Butene</td>
<td>Gas</td>
<td>45,292</td>
</tr>
<tr>
<td>107-01-7</td>
<td>2-Butene</td>
<td>Gas</td>
<td>45,100’</td>
</tr>
<tr>
<td>463-58-1</td>
<td>Carbon oxysulfide [Carbon oxide sulfide (COS)]</td>
<td>Gas</td>
<td>9,126</td>
</tr>
<tr>
<td>7791-21-1</td>
<td>Chlorine monoxide [Chlorine oxide]</td>
<td>Gas</td>
<td>1,011’</td>
</tr>
<tr>
<td>590-21-6</td>
<td>1-Chloropropylene [1-Propene, 1-chloro-]</td>
<td>Liquid</td>
<td>23,000’</td>
</tr>
<tr>
<td>557-98-2</td>
<td>2-Chloropropylene [1-Propene, 2-chloro-]</td>
<td>Gas</td>
<td>22,999</td>
</tr>
<tr>
<td>460-19-5</td>
<td>Cyanogen [Ethanedinitrile]</td>
<td>Gas</td>
<td>21,064</td>
</tr>
<tr>
<td>75-19-4</td>
<td>Cyclopropane</td>
<td>Gas</td>
<td>46,560</td>
</tr>
<tr>
<td>4109-96-0</td>
<td>Dichlorosilane [Silane, dichloro-]</td>
<td>Gas</td>
<td>8,225</td>
</tr>
<tr>
<td>75-37-6</td>
<td>Difluoroethane [Ethane, 1,1-difluoro-]</td>
<td>Gas</td>
<td>11,484</td>
</tr>
<tr>
<td>124-40-3</td>
<td>Dimethylamine [Methanamine, N-methyl-]</td>
<td>Gas</td>
<td>35,813</td>
</tr>
<tr>
<td>463-82-1</td>
<td>2,2-Dimethylpropane [Propane, 2,2-dimethyl-]</td>
<td>Gas</td>
<td>45,051</td>
</tr>
<tr>
<td>74-84-0</td>
<td>Ethane</td>
<td>Gas</td>
<td>47,509</td>
</tr>
<tr>
<td>107-00-6</td>
<td>Ethyl acetylene [1-Butyne]</td>
<td>Gas</td>
<td>45,565</td>
</tr>
<tr>
<td>75-04-7</td>
<td>Ethylamine [Ethanamine]</td>
<td>Gas</td>
<td>35,210</td>
</tr>
<tr>
<td>75-00-3</td>
<td>Ethyl chloride [Ethane, chloro-]</td>
<td>Gas</td>
<td>19,917</td>
</tr>
<tr>
<td>74-85-1</td>
<td>Ethylene [Ethene]</td>
<td>Gas</td>
<td>47,145</td>
</tr>
</tbody>
</table>
**Exhibit C-1 (continued)**

<table>
<thead>
<tr>
<th>CAS No.</th>
<th>Chemical Name</th>
<th>Physical State at 25°C</th>
<th>Heat of Combustion (k joule/kg)</th>
</tr>
</thead>
<tbody>
<tr>
<td>60-29-7</td>
<td>Ethyl ether [Ethane, 1,1’-oxybis-]</td>
<td>Liquid</td>
<td>33,775</td>
</tr>
<tr>
<td>75-08-1</td>
<td>Ethyl mercaptan [Ethanethiol]</td>
<td>Liquid</td>
<td>27,948</td>
</tr>
<tr>
<td>109-95-5</td>
<td>Ethyl nitrite [Nitrous acid, ethyl ester]</td>
<td>Gas</td>
<td>18,000</td>
</tr>
<tr>
<td>1333-74-0</td>
<td>Hydrogen</td>
<td>Gas</td>
<td>119,950</td>
</tr>
<tr>
<td>75-28-5</td>
<td>Isobutane [Propane, 2-methyl]</td>
<td>Gas</td>
<td>45,576</td>
</tr>
<tr>
<td>78-78-4</td>
<td>Isopentane [Butane, 2-methyl-]</td>
<td>Liquid</td>
<td>44,911</td>
</tr>
<tr>
<td>78-79-5</td>
<td>Isoprene [1,3-Butadiene, 2-methyl-]</td>
<td>Liquid</td>
<td>43,809</td>
</tr>
<tr>
<td>75-31-0</td>
<td>Isopropylamine [2-Propanamine]</td>
<td>Liquid</td>
<td>36,484</td>
</tr>
<tr>
<td>75-29-6</td>
<td>Isopropyl chloride [Propane, 2-chloro-]</td>
<td>Liquid</td>
<td>23,720</td>
</tr>
<tr>
<td>74-82-8</td>
<td>Methane</td>
<td>Gas</td>
<td>50,029</td>
</tr>
<tr>
<td>74-89-5</td>
<td>Methylamine [Methanamine]</td>
<td>Gas</td>
<td>31,396</td>
</tr>
<tr>
<td>563-45-1</td>
<td>3-Methyl-1-butene</td>
<td>Gas</td>
<td>44,559</td>
</tr>
<tr>
<td>563-46-2</td>
<td>2-Methyl-1-butene</td>
<td>Liquid</td>
<td>44,414</td>
</tr>
<tr>
<td>115-10-6</td>
<td>Methyl ether [Methane, oxybis-]</td>
<td>Gas</td>
<td>28,835</td>
</tr>
<tr>
<td>107-31-3</td>
<td>Methyl formate [Formic acid, methyl ester]</td>
<td>Liquid</td>
<td>15,335</td>
</tr>
<tr>
<td>115-11-7</td>
<td>2-Methylpropene [1-Propene, 2-methyl-]</td>
<td>Gas</td>
<td>44,985</td>
</tr>
<tr>
<td>504-60-9</td>
<td>1,3-Pentadiene</td>
<td>Liquid</td>
<td>43,834</td>
</tr>
<tr>
<td>109-66-0</td>
<td>Pentane</td>
<td>Liquid</td>
<td>44,697</td>
</tr>
<tr>
<td>109-67-1</td>
<td>1-Pentene</td>
<td>Liquid</td>
<td>44,625</td>
</tr>
<tr>
<td>646-04-8</td>
<td>2-Pentene, (E)-</td>
<td>Liquid</td>
<td>44,458</td>
</tr>
<tr>
<td>627-20-3</td>
<td>2-Pentene, (Z)-</td>
<td>Liquid</td>
<td>44,520</td>
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* Estimated heat of combustion
## Exhibit C-2

### Data for Flammable Gases

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<th>Molecular Weight</th>
<th>Ratio of Specific Heats</th>
<th>Flammability Limits (Vol %)</th>
<th>LFL (mg/L)</th>
<th>Gas Factor (GF)</th>
<th>Liquid Factor Boiling (LFB)</th>
<th>Density Factor Boiling (DF)</th>
<th>Reference Table</th>
<th>Pool Fire Factor (PFF)</th>
<th>Flash Fraction Factor (FFF)</th>
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*April 15, 1999*  

*C - 6*
### Exhibit C-2 (continued)

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<th>Gas Factor (GF)</th>
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<th>Density Factor (Boiling) (DF)</th>
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April 15, 1999
### Exhibit C-2 (continued)

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<th>CAS Number</th>
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<th>Molecular Weight</th>
<th>Ratio of Specific Heats</th>
<th>Flammability Limits (Vol %)</th>
<th>LFL (mg/L)</th>
<th>Gas Factor (GF)</th>
<th>Liquid Factor Boiling (LFB)</th>
<th>Density Factor Boiling (DF)</th>
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<th>Pool Fire Factor (PFF)</th>
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<td>25</td>
<td>0.17</td>
<td>0.57</td>
<td>Dense</td>
<td>3.7</td>
<td>0.093</td>
</tr>
</tbody>
</table>

**Notes:**

NA: Data not available

a “Buoyant” in the Reference Table column refers to the tables for neutrally buoyant gases and vapors; “Dense” refers to the tables for dense gases and vapors. See Appendix D, Section D.4.4, for more information on the choice of reference tables.

b Gases that are lighter than air may behave as dense gases upon release if liquefied under pressure or cold; consider the conditions of release when choosing the appropriate table.

c Reported to be spontaneously combustible.

d Much lighter than air; table of distances for neutrally buoyant gases not appropriate.

e Pool formation unlikely.

f Calculated at 298 K (25 °C) with the following exceptions:

   - Ethylene factor calculated at critical temperature, 282 K.
   - Methane factor calculated at critical temperature, 191 K.
   - Silane factor calculated at critical temperature, 270 K.

g Use GF for gas leaks under choked (maximum) flow conditions.
# Exhibit C-3

## Data for Flammable Liquids

<table>
<thead>
<tr>
<th>CAS Number</th>
<th>Chemical Name</th>
<th>Molecular Weight</th>
<th>Flammability Limit (Vol%)</th>
<th>LFL (mg/L)</th>
<th>Liquid Factors</th>
<th>Density Factor</th>
<th>Liquid Leak Factor (LFL) (^a)</th>
<th>Reference Table (^b)</th>
<th>Pool Fire Factor (PFF)</th>
</tr>
</thead>
<tbody>
<tr>
<td>590-21-6</td>
<td>1-Chloropropylene</td>
<td>76.53</td>
<td>4.5</td>
<td>16.0</td>
<td>140</td>
<td>0.11</td>
<td>0.15</td>
<td>0.52</td>
<td>45</td>
</tr>
<tr>
<td>60-29-7</td>
<td>Ethyl ether</td>
<td>74.12</td>
<td>1.9</td>
<td>48.0</td>
<td>57</td>
<td>0.11</td>
<td>0.15</td>
<td>0.69</td>
<td>34</td>
</tr>
<tr>
<td>75-08-1</td>
<td>Ethyl mercaptan</td>
<td>62.14</td>
<td>2.8</td>
<td>18.0</td>
<td>71</td>
<td>0.10</td>
<td>0.13</td>
<td>0.58</td>
<td>40</td>
</tr>
<tr>
<td>78-78-4</td>
<td>Isopentane</td>
<td>72.15</td>
<td>1.4</td>
<td>7.6</td>
<td>41</td>
<td>0.14</td>
<td>0.15</td>
<td>0.79</td>
<td>30</td>
</tr>
<tr>
<td>78-79-5</td>
<td>Isoprene</td>
<td>68.12</td>
<td>2.0</td>
<td>9.0</td>
<td>56</td>
<td>0.11</td>
<td>0.14</td>
<td>0.72</td>
<td>32</td>
</tr>
<tr>
<td>75-31-0</td>
<td>Isopropylamine</td>
<td>59.11</td>
<td>2.0</td>
<td>10.4</td>
<td>48</td>
<td>0.10</td>
<td>0.13</td>
<td>0.71</td>
<td>33</td>
</tr>
<tr>
<td>75-29-6</td>
<td>Isopropyl chloride</td>
<td>78.54</td>
<td>2.8</td>
<td>10.7</td>
<td>90</td>
<td>0.11</td>
<td>0.16</td>
<td>0.57</td>
<td>41</td>
</tr>
<tr>
<td>563-46-2</td>
<td>2-Methyl-1-butene</td>
<td>70.13</td>
<td>1.4</td>
<td>9.6</td>
<td>40</td>
<td>0.12</td>
<td>0.15</td>
<td>0.75</td>
<td>31</td>
</tr>
<tr>
<td>107-31-3</td>
<td>Methyl formate</td>
<td>60.05</td>
<td>5.9</td>
<td>20.0</td>
<td>140</td>
<td>0.10</td>
<td>0.13</td>
<td>0.50</td>
<td>46</td>
</tr>
<tr>
<td>504-60-9</td>
<td>1,3-Pentadiene</td>
<td>68.12</td>
<td>1.6</td>
<td>13.1</td>
<td>44</td>
<td>0.077</td>
<td>0.14</td>
<td>0.72</td>
<td>33</td>
</tr>
<tr>
<td>109-66-0</td>
<td>Pentane</td>
<td>72.15</td>
<td>1.3</td>
<td>8.0</td>
<td>38</td>
<td>0.10</td>
<td>0.15</td>
<td>0.78</td>
<td>30</td>
</tr>
<tr>
<td>109-67-1</td>
<td>1-Pentene</td>
<td>70.13</td>
<td>1.5</td>
<td>8.7</td>
<td>43</td>
<td>0.13</td>
<td>0.15</td>
<td>0.77</td>
<td>31</td>
</tr>
<tr>
<td>646-04-8</td>
<td>2-Pentene, (E)-</td>
<td>70.13</td>
<td>1.4</td>
<td>10.6</td>
<td>40</td>
<td>0.10</td>
<td>0.15</td>
<td>0.76</td>
<td>31</td>
</tr>
<tr>
<td>627-20-3</td>
<td>2-Pentene, (Z)-</td>
<td>70.13</td>
<td>1.4</td>
<td>10.6</td>
<td>40</td>
<td>0.10</td>
<td>0.15</td>
<td>0.75</td>
<td>31</td>
</tr>
<tr>
<td>75-76-3</td>
<td>Tetramethyilsilane</td>
<td>88.23</td>
<td>1.5</td>
<td>NA</td>
<td>54</td>
<td>0.17</td>
<td>0.17</td>
<td>0.59</td>
<td>40</td>
</tr>
<tr>
<td>10025-78-2</td>
<td>Trichlorosilane</td>
<td>135.45</td>
<td>1.2</td>
<td>90.5</td>
<td>66</td>
<td>0.18</td>
<td>0.23</td>
<td>0.37</td>
<td>64</td>
</tr>
<tr>
<td>109-92-2</td>
<td>Vinyl ethyl ether</td>
<td>72.11</td>
<td>1.7</td>
<td>28.0</td>
<td>50</td>
<td>0.10</td>
<td>0.15</td>
<td>0.65</td>
<td>36</td>
</tr>
<tr>
<td>75-35-4</td>
<td>Vinylvindene chloride</td>
<td>96.94</td>
<td>7.3</td>
<td>NA</td>
<td>290</td>
<td>0.15</td>
<td>0.18</td>
<td>0.44</td>
<td>54</td>
</tr>
</tbody>
</table>

**Notes:**

NA: Data not available.

\(^a\) Use the LLF only for leaks from tanks at atmospheric pressure.

\(^b\) "Dense" in the Reference Table column refers to the tables for dense gases and vapors. See Appendix D, Section D.4.4, for more information on the choice of reference tables.
APPENDIX D

TECHNICAL BACKGROUND
APPENDIX D  TECHNICAL BACKGROUND

D.1  Worst-Case Release Rate for Gases

D.1.1  Unmitigated Release

The assumption that the total quantity of toxic gas is released in 10 minutes is the same assumption used in EPA's Technical Guidance for Hazards Analysis (1987).

D.1.2  Gaseous Release Inside Building

The mitigation factor for gaseous release inside a building is based on a document entitled, Risk Mitigation in Land Use Planning: Indoor Releases of Toxic Gases, by S.R. Porter. This paper presented three release scenarios and discussed the mitigating effects that would occur in a building with a volume of 1,000 cubic meters at three different building air exchange rates. There is a concern that a building may not be able to withstand the pressures of a very large release. However, this paper indicated that release rates of at least 2,000 pounds per minute could be withstood by a building.

Analyzing the data in this paper several ways, the value of 55 percent emerged as representing the mitigation that could occur for a release scenario into a building. Data are provided on the maximum release rate in a building and the maximum release rate from a building. Making this direct comparison at the lower maximum release rate (3.36 kg/s) gave a release rate from the building of 55 percent of the release rate into the building. Using information provided on another maximum release rate (10.9 kg/min) and accounting for the time for the release to accumulate in the building, approximately 55 percent emerged again.

The choice of building ventilation rates affects the results. The paper presented mitigation for three different ventilation rates, 0.5, 3, and 10 air changes per hour. A ventilation rate of 0.5 changes per hour is representative of specially designed, “gas-tight” buildings, based on the Porter reference. EPA decided that this ventilation rate was appropriate for this analysis. A mitigation factor of 55 percent may be used in the event of a gaseous release which does not destroy the building into which it is released. This factor may overstate the mitigation provided by a building with a higher ventilation rate.

For releases of ammonia, chlorine, and sulfur dioxide, factors specific to the chemicals, the conditions of the release, and building ventilation rates have been developed to estimate mitigation of releases in buildings. For information on these factors and estimation of mitigated release rates, see Backup Information for the Hazard Assessments in the RMP Offsite Consequence Analysis Guidance, the Guidance for Wastewater Treatment Facilities and the Guidance for Ammonia Refrigeration - Anhydrous Ammonia, Aqueous Ammonia, Chlorine and Sulfur Dioxide. See also the industry-specific guidance documents for ammonia refrigeration and POTWs.

D.2  Worst-Case Release Rate for Liquids

D.2.1  Evaporation Rate Equation

The equation for estimating the evaporation rate of a liquid from a pool is from the Technical Guidance for Hazards Analysis, Appendix G. The same assumptions are made for determination of
maximum pool area (i.e., the pool is assumed to be 1 centimeter (0.033 feet) deep). The evaporation rate equation has been modified to include a different mass transfer coefficient for water, the reference compound. For this document, a value of 0.67 centimeters per second is used as the mass transfer coefficient, instead of the value of 0.24 cited in the Technical Guidance for Hazards Analysis. The value of 0.67 is based on Donald MacKay and Ronald S. Matsugu, "Evaporation Rates of Liquid Hydrocarbon Spills on Land and Water," Canadian Journal of Chemical Engineering, August 1973, p. 434. The evaporation equation becomes:

\[
QR = \frac{0.284 \times U^{0.78} \times MW^{2/3} \times A \times VP}{82.05 \times T}
\]  

(D-1)

where:
- \(QR\) = Evaporation rate (pounds per minute)
- \(U\) = Wind speed (meters per second)
- \(MW\) = Molecular weight (given in Exhibits B-1 and B-2, Appendix B, for toxic substances and Exhibits C-2 and C-3, Appendix C, for flammable substances)
- \(A\) = Surface area of pool formed by the entire quantity of the mixture (square feet) (determined as described in Section 3.2.2 of the text)
- \(VP\) = Vapor pressure (mm Hg)
- \(T\) = Temperature of released substance (Kelvin (K); temperature in °C plus 273, or 298 for 25 °C)

D.2.2 Factors for Evaporation Rate Estimates

Liquid Factors. The liquid factors, Liquid Factor Ambient (LFA) and Liquid Factor Boiling (LFB), used to estimate the evaporation rate from a liquid pool (see Section 3.2 of this guidance document), are derived as described in the Technical Guidance for Hazards Analysis, Appendix G, with the following differences:

- The mass transfer coefficient of water is assumed to be 0.67, as discussed above; the value of the factor that includes conversion factors, the mass transfer coefficient for water, and the molecular weight of water to the one-third power, given as 0.106 in the Technical Guidance is 0.284 in this guidance.

- Density of all substances was assumed to be the density of water in the Technical Guidance; the density was included in the liquid factors. For this guidance document, density is not included in the LFA and LFB values presented in the tables; instead, a separate Density Factor (DF) (discussed below) is provided to be used in the evaporation rate estimation.

With these modifications, the LFA is:

\[
LFA = \frac{0.284 \times MW^{2/3} \times VP}{82.05 \times 298}
\]

(D-2)

where: \(MW\) = Molecular weight
\( VP \) = Vapor pressure at ambient temperature (mm Hg)
298 K (25 °C) = Ambient temperature and temperature of released substance

LFB is:

\[
LFB = \frac{0.284 \times MW^{2/3} \times 760}{82.05 \times BP}
\]  \hspace{1cm} (D-3)

where:

\( MW \) = Molecular weight
760 = Vapor pressure at boiling temperature (mm Hg)
\( BP \) = Boiling point (K)

LFA and LFB values were developed for all toxic and flammable regulated liquids, and LFB values, to be used for analysis of gases liquefied by refrigeration, were developed for toxic and flammable gases.

Density Factor. Because some of the regulated liquids have densities very different from that of water, the density of each substance was used to develop a Density Factor (DF) for the determination of maximum pool area for the evaporation rate estimation. DF values were developed for toxic and flammable liquids at ambient temperature and for toxic and flammable gases at their boiling points. The density factor is:

\[
DF = \frac{1}{d \times 0.033}
\]  \hspace{1cm} (D-4)

where:

\( DF \) = Density factor (1/(lbs/ft\(^3\))
\( d \) = Density of the substance in pounds per cubic foot
0.033 = Depth of pool for maximum area (feet)

Temperature Correction Factors. Temperature correction factors were developed for toxic liquids released at temperatures above 25 °C, the temperature used for development of the LFAs. The temperature correction factors are based on vapor pressures calculated from the coefficients provided in *Physical and Thermodynamic Properties of Pure Chemicals, Data Compilation*, developed by the Design Institute for Physical Property Data (DIPPR), American Institute of Chemical Engineers. The factors are calculated as follows:

\[
TCF_T = \frac{VP_T \times 298}{VP_{298} \times T}
\]  \hspace{1cm} (D-5)

where:

\( TCF_T \) = Temperature Correction Factor at temperature T
\( VP_T \) = Vapor pressure at temperature T
\( VP_{298} \) = Vapor pressure at 298 K
\( T \) = Temperature (K) of released substance

Factors were developed at intervals of 5 °C for temperatures up to 50 °C.
No correction factor was deemed necessary for changes in the density of the regulated toxic liquids with changes in temperature, although the density could affect the pool area and release rate estimates. Analysis of the temperature dependence of the density of these liquids indicated that the changes in density with temperature were very small compared to the changes in vapor pressure with temperature.

### D.2.3 Common Water Solutions and Oleum

Water solutions of regulated toxic substances must be analyzed somewhat differently from pure toxic liquids. Except for solutions of relatively low concentration, the evaporation rate varies with the concentration of the solution. At one specific concentration, the composition of the liquid does not change as evaporation occurs. For concentrated solutions of volatile substances, the evaporation rate from a pool may decrease, very rapidly in some cases, as the toxic substance volatilizes and its concentration in the pool decreases. To analyze these changes, EPA used spreadsheets to estimate the vapor pressure, concentration, and release rate at various time intervals for regulated toxic substances in water solution evaporating from pools. In addition to the spreadsheet analysis, EPA used the ALOHA model with an additional step-function feature (not available in the public version). With this step-function feature, changes in the release rate could be incorporated and the effects of these changes on the consequence distance analyzed. The results of the spreadsheet calculations and the model were found to be in good agreement. The distance results obtained from the spreadsheet analysis and the model for various solutions were compared with the results from various time averages to examine the sensitivity of the results. An averaging time of 10 minutes was found to give reasonable agreement with the step-function model for most substances at various concentrations. The spreadsheet analysis also indicated that the first 10 minutes of evaporation was the most important, and the evaporation rate in the first 10 minutes likely could be used to estimate the distance to the endpoint.

Oleum is a solution of sulfur trioxide in sulfuric acid. Sulfur trioxide evaporating from oleum exhibits release characteristics similar to those of toxic substances evaporating from water solutions. Analysis of oleum releases, therefore, was carried out in the same way as for water solutions.

NOAA developed a computerized calculation method to estimate partial vapor pressures and release rates for regulated toxic substance in solution as a function of concentration, based on vapor pressure data from *Perry's Chemical Engineers' Handbook* and other sources. Using this method and spreadsheet calculations, EPA estimated partial vapor pressures and evaporation rates at one-minute intervals over 10 minutes for solutions of various concentrations. The 10-minute time period was chosen based on the ALOHA results and other calculations. For each one-minute interval, EPA estimated the concentration of the solution based on the quantity evaporated in the previous interval and estimated the partial vapor pressure based on the concentration. These estimated vapor pressures were used to calculate an average vapor pressure over the 10-minute period; this average vapor pressure was used to derive Liquid Factor Ambient (LFA) values, as described above for liquids. Use of these factors is intended to give an evaporation rate that accounts for the decrease in evaporation rate expected to take place as the solution evaporates.

Density Factors (DF) were developed for solutions of various concentrations from data in *Perry's Chemical Engineers' Handbook* and other sources, as discussed above for liquids.

Because solutions do not have defined boiling points, EPA did not develop Liquid Factor Boiling (LFB) values for solutions. As a simple and conservative approach, the quantity of a regulated substance in a solution at an elevated temperatures is treated as a pure substance. The LFB for the pure substance, or the
LFA and a temperature correction factor, is used to estimate the initial evaporation rate of the regulated substance from the solution. Only the first 10 minutes of evaporation are considered, as for solutions at ambient temperatures, because the release rate would decrease rapidly as the substance evaporates and the concentration in the solution decreases. This approach will likely give an overestimate of the release rate and of the consequence distance.

### D.2.4 Releases Inside Buildings

If a liquid is released inside a building, its release to the outside air will be mitigated in two ways. First, the evaporation rate of the liquid may be much lower inside a building than outside. This is due to wind speed, which directly affects the evaporation rate. The second mitigating factor is that the building provides resistance to discharge of contaminated air to the outdoors.

In this method, a conservative wind speed, U, of 0.1 meter per second (m/s) was assumed in the building. (See end of text for a justification of this wind speed.) For a release outdoors in a worst-case scenario, U is set to 1.5 m/s, and for an alternative scenario, U is set to 3 m/s. The evaporation rate equation is:

\[
QR = U^{0.78} \times (LFA, LFB) \times A
\]

(D-6)

where:

- \(QR\) = Release rate (pounds per minute (lbs/min))
- \(U\) = Wind speed (meters per second (m/s))
- \(LFA\) = Liquid Factor Ambient
- \(LFB\) = Liquid Factor Boiling
- \(A\) = Area of pool (square feet (ft\(^2\))

As can be seen, if U inside a building is only 0.1, then the evaporation rate inside a building will be much lower than a corresponding evaporation rate outside (assuming the temperature is the same). The rate will only be \((0.1/1.5)^{0.78}\), about 12 percent of the rate for a worst case, and \((0.1/3)^{0.78}\), about seven percent of the rate for an alternative case.

The evaporated liquid mixes with and contaminates the air in the building. What EPA is ultimately interested in is the rate at which this contaminated air exits the building. In order to calculate the release of contaminated air outside the building, EPA adapted a method from an UK Health and Safety Executive paper entitled, *Risk Mitigation in Land Use Planning: Indoor Releases of Toxic Gases*, by S.R. Porter. EPA assumed that the time for complete evaporation of the liquid pool was one hour. The rate at which contaminated air was released from the building during liquid evaporation (based on the paper) was assumed to be equal to the evaporation rate plus the building ventilation rate (no pressure buildup in building). The building ventilation rate was set equal to 0.5 air changes per hour. This ventilation rate is representative of a specially designed, “gas-tight” building. (The mitigation factor developed based on this type of building would overstate the mitigation provided by a building with higher ventilation rates.) EPA used a building with a volume of 1,000 cubic meters (m\(^3\)) and a floor area of 200 m\(^2\) (2,152 ft\(^2\)) as an example for this analysis. EPA assumed that the liquid pool would cover the entire building floor, representing a conservative scenario.
To provide a conservative estimate, EPA calculated the evaporation rate for a spill of a volatile liquid, carbon disulfide (CS₂), under ambient conditions inside the building:

$$QR = 0.1^{0.78} \times 0.075 \times 2,152 = 26.8 \text{ pounds per minute (lbs/min)}$$

Next, this evaporation rate was converted to cubic meters per minute (m³/min) using the ideal gas law (the molecular weight of CS₂ is 76.1):

$$26.8 \text{ lbs/min} \times 454 \text{ grams per pound (g/lb)} \times 1 \text{ mole CS₂/76.1 g} \times 0.0224 \text{ m³/mole} = 3.58 \text{ m³/min.}$$

The ventilation rate of the building is 0.5 changes per hour, which equals 500 m³ per hour, or 8.33 m³/min. Therefore, during evaporation, contaminated air is leaving the building at a rate of 8.33 + 3.58, or 11.9 m³/min.

EPA used an iterative calculation for carbon disulfide leaving a building using the above calculated parameters. During the first minute of evaporation, 26.8 lbs of pure carbon disulfide evaporates, and EPA assumed this evenly disperses through the building so that the concentration of CS₂ in the building air is 0.0268 lbs/m³ (assuming 1000 m³ volume in the building). Contaminated air is exiting the building at a rate of 11.9 m³/min, so EPA deduced that 11.9 \times 0.0268 = 0.319 lbs of carbon disulfide exit the building in the first minute, leaving 26.5 lbs still evenly dispersed inside. Since this release occurs over one minute, the release rate of the carbon disulfide to the outside is 0.319 lbs/min. During the second minute, another 26.8 lbs of pure carbon disulfide evaporates and disperses, so that the building now contains 26.8 + 26.5 = 53.3 lbs of carbon disulfide, or 0.0533 lbs/m³. Contaminated air is still exiting the building at a rate of 11.9 m³/min, so 11.9 \times 0.05328 = 0.634 lbs of carbon disulfide are released, leaving 52.6 lbs inside. Again, this release occurs over one minute so that the rate of carbon disulfide exiting the building in terms of contaminated air is 0.634 lbs/min. EPA continued to perform this estimation over a period of one hour. The rate of release of carbon disulfide exiting the building in the contaminated air at the sixty minute mark is 13.7 lbs/min. This represents the maximum rate of carbon disulfide leaving the building. After all of the carbon disulfide is evaporated, there is a drop in the concentration of carbon disulfide in the contaminated air leaving the building because the evaporation of carbon disulfide no longer contributes to the overall contamination of the air.

Note that if the same size pool of carbon disulfide formed outside, the release rate for a worst-case scenario would be:

$$QR = 1.5^{0.78} \times 0.075 \times 2,152 = 221 \text{ lbs/min.}$$

and for an alternative case:

$$QR = 3^{0.78} \times 0.075 \times 2,152 = 380 \text{ lbs/min.}$$

The maximum release rate of carbon disulfide in the contaminated building air, assuming a 1,000 m³ building with a building exchange rate of 0.5 air changes per hour, was only about 6 percent (13.7 ÷ 221 lbs/min x 100) of the worst-case scenario rate, and only about 3.6 percent (13.7 ÷ 380 lbs/min x 100) of the alternative scenario rate. EPA set an overall building mitigation factor equal to 10 percent and five percent, respectively, in order to be conservative. Please note that (at a constant ventilation rate of 0.5 changes per
hour) as the size of the building increases, the maximum rate of contaminated air leaving the building will decrease, although only slightly, because of the balancing effect of building volume and ventilation rate. Obviously, a higher ventilation rate will yield a higher maximum release rate of contaminated air from the building.

For a release inside a building, EPA assumed a building air velocity of 0.1 m/s. This conservative value was derived by setting the size of the ventilation fan equal to 1.0 m². This fan is exchanging air from the building with the outside at a rate of 0.5 changes per hour. For a 1,000 m³ building, this value becomes 500 m³/hour, or 0.14 m³/s. Dividing 0.14 m³/s by the area of the fan yields a velocity of 0.14 m/s, which was rounded down to 0.1 m/s.

D.3 Toxic Endpoints

The toxic endpoints for regulated toxic substances, which are specified in the RMP Rule, are presented in Appendix B, Exhibits B-1, B-2, and B-3. The endpoints were chosen as follows, in order of preference:

1. Emergency Response Planning Guideline 2 (ERPG-2), developed by the American Industrial Hygiene Association, if available;

2. Level of Concern (LOC) derived for extremely hazardous substances (EHSs) regulated under section 302 of the Emergency Planning and Community Right-to-Know Act (EPCRA) (see the Technical Guidance for Hazards Analysis for more information on LOCs); the LOC for EHSs is based on:

   -- One-tenth of the Immediately Dangerous to Life and Health (IDLH) level, developed by the National Institute of Occupational Safety and Health (NIOSH), using IDLH values developed before 1994,

   or, if no IDLH value is available,

   -- One-tenth of an estimated IDLH derived from toxicity data; the IDLH is estimated as described in Appendix D of the Technical Guidance for Hazards Analysis.

Note that the LOCs were not updated using IDLHs published in 1994 and later, because NIOSH revised its methodology for the IDLHs. The EHS LOCs based on earlier IDLHs were reviewed by EPA’s Science Advisory Board, and EPA decided to retain the methodology that was reviewed.

ERPG-2 is defined as the maximum airborne concentration below which it is believed nearly all individuals could be exposed for up to one hour without experiencing or developing irreversible or other serious health effects or symptoms that could impair an individual's ability to take protective action.

IDLH (pre-1994) concentrations were defined in the NIOSH Pocket Guide to Chemical Hazards as representing the maximum concentration from which, in the event of respirator failure, one could escape within 30 minutes without a respirator and without experiencing any escape-impairing (e.g., severe eye
irritation) or irreversible health effects. (As noted above, LOCs for EHSs were not updated to reflect 1994 and later IDLHs.)

The estimated IDLH is derived from animal toxicity data, in order of preferred data, as follows:

- From median lethal concentration (LC$_{50}$) (inhalation): $0.1 \times \text{LC}_{50}$
- From lowest lethal concentration (LC$_{LO}$) (inhalation): $1 \times \text{LC}_{LO}$
- From median lethal dose (LD$_{50}$) (oral): $0.01 \times \text{LD}_{50}$
- From lowest lethal dose (LD$_{LO}$) (oral): $0.1 \times \text{LD}_{LO}$

The toxic endpoints based on LOCs for EHSs presented in the tables in Appendix B are, in some cases, different from the LOCs listed in the *Technical Guidance for Hazards Analysis*, because some of the LOCs were updated based on IDLHs that were published after the development of the LOCs (and before 1994) or on new or revised toxicity data.

### D.4 Reference Tables for Distances to Toxic and Flammable Endpoints

#### D.4.1 Neutrally Buoyant Gases


Longitudinal dispersion (dispersion in the along-wind direction) is generated mostly by vertical wind shear. Wind shear results from the tendency of the wind speed to assume a wind profile—the speed is lowest next to the ground and increases with height until it reaches an asymptotic value at approximately a few hundred feet above the surface. To account for shear-driven dispersion, any air dispersion model intended for modeling short-duration releases must include either (a) a formulation that accounts, either implicitly or explicitly, for the height-dependence of wind speed or (b) some type of parameterization that converts shear effect into $\sigma_y$, the standard deviation function in the along-wind direction.

Because the standard Gaussian formula does not incorporate $\sigma_y$ (it includes only $\sigma_x$ and $\sigma_z$, the crosswind and horizontal functions), very few alternate ways to formulate $\sigma_y$ have been proposed. The simplest method was proposed by Turner (*Workbook of Atmospheric Dispersion Estimates*, Report PB-191 482, Research Triangle Park, North Carolina: Office of Air Programs, U.S. Environmental Protection Agency, 1970), who suggested simply setting $\sigma_y$ equal to $\sigma_x$. Textbooks such as that by Pasquill and Smith (*Atmospheric Diffusion*, 3rd ed. New York: Halstead Press, 1983) describe a well-known analytic model. However, this model is more complex than a Gaussian model because according to it, dispersion depends on wind shear and the vertical variation of the vertical diffusion coefficient. Wilson (*Along-wind Diffusion of Source Transients, Atmospheric Environment* 15:489-495, 1981) proposed another method in which $\sigma_x$ is
determined as a function of wind shear, but in a form that can then be used in a Gaussian model. However, it
is now believed that Wilson's formulation gives $\sigma_z$ values that are too large.

To avoid the problems of the analytic method and Wilson's formulation, we chose to include a
formulation for $\sigma_z$ derived from work by Beals (1971). We had three reasons for doing so. First, in terms of
magnitude, Beals' $\sigma_z$ fell in the midrange of the alternative formulations that we reviewed. Second, Beals' $\sigma_z$
indirectly accounts for wind shear by using (unpublished) experimental data. Third, both the ALOHA and
DEGADIS models incorporate the Beals methodology.

When a substance is dispersed downwind, the concentration in the air changes over time. To assess
the health effects of potential exposure to the substance, the average concentration of the substance over
some time period is determined. Averaging time is the time interval over which the instantaneous
concentration of the hazardous material in the vapor cloud is averaged. Averaging time should generally be
equal to or shorter than either the release duration or cloud duration and, if possible, should reflect the
exposure time associated with the toxic exposure guideline of interest. The exposure time associated with the
toxic endpoints specified under the RMP Rule include 30 minutes for the Immediately Dangerous to Life and
Health (IDLH) level and 60 minutes for the Emergency Response Planning Guideline (ERPG). For the
neutrally buoyant tables, the 10-minute release scenario was modeled using a 10-minute averaging time. The
60-minute release scenario was modeled using a 30-minute averaging time to be consistent with the 30-
minute exposure time associated with the IDLH. A 60-minute averaging time may have underpredicted
consequence distances and, therefore, was not used for development of the distance tables for this guidance.

Cloud dispersion from a release of finite duration (10 and 60-minute releases) is calculated using an
equation specified in the NOAA publication ALOHA\textsuperscript{TM} 5.0 Theoretical Description, Technical Memorandum
NOS ORCA 65, August 1992.

**Flammable Substances.** The reference tables of distances for vapor cloud fires of neutrally buoyant
flammable substances were derived using the same model as for toxic substances, as described above. The
endpoint for modeling was the lower flammability limit (LFL). For flammable substances, an averaging time
of 0.1 minute (six seconds) was used, because fires are considered to be nearly instantaneous events.

Distances of interest for flammable substances are generally much shorter than for toxic substance,
because the LFL concentrations are much larger than the toxic endpoints. For the short distances found in
modeling the flammable substances, modeling results were found to be the same for 10-minute and longer
releases; therefore, one table of distances for rural conditions and one table for urban conditions, applicable
for both 10-minute and longer releases, were developed for flammable substances.

**D.4.2 Dense Gases**

**Toxic Substances.** The reference tables for dense gases were developed using the widely accepted
SLAB model, developed by Lawrence Livermore National Laboratory. SLAB solves conservation equations
of mass, momentum, energy, and species for continuous, finite duration, and instantaneous releases. The
reference tables were based on the evaporating pool algorithm.

For the reference tables were developed based on modeling releases of hydrogen chloride (HCl). HCl
was chosen based on a SLAB modeling analysis of a range of dispersion behavior for releases of regulated
dense gases or vapors with different molecular weights. This analysis showed that releases of HCl generally provided conservative results under a variety of stability/wind speed combinations, release rates, and toxic endpoints.

Similar to the modeling of neutrally buoyant plumes, the 10-minute release scenario of toxic chemicals was modeled using a 10-minute averaging time. The 60-minute release scenario was modeled using a 30-minute averaging time to be consistent with the 30-minute exposure time associated with the IDLH.

For all dense gas tables, the reference height for the wind speed was 10 meters. Relative humidity was assumed to be 50 percent, and the ambient temperature was 25 °C. The source area was the smallest value that still enabled the model to run for all release rates. The surface roughness factor was one meter for urban scenarios and three centimeters for rural scenarios.

**Flammable Substances.** For the reference tables for dispersion of dense flammable gases and vapors, for analysis of vapor cloud fires, the same model was used as for toxic substances, as described above, and the same assumptions were made. For the dispersion of flammable chemicals, averaging time should be very small (i.e., no more than a few seconds), because flammable vapors need only be exposed to an ignition source for a short period of time to initiate the combustion process. Thus, both the 10-minute and 60-minute reference tables for flammable substances use an averaging time of 10 seconds. The 10-minute and 60-minute tables were combined for flammable substances because the modeling results were found to be the same.

### D.4.3 Chemical-Specific Reference Tables

The chemical-specific reference tables of distances for ammonia, chlorine, and sulfur dioxide were developed for EPA’s risk management program guidance for ammonia refrigeration and for POTWs. For information on the chemical-specific modeling and development of the chemical-specific reference tables, see *Backup Information for the Hazard Assessments in the RMP Offsite Consequence Analysis Guidance, the Guidance for Wastewater Treatment Facilities and the Guidance for Ammonia Refrigeration - Anhydrous Ammonia, Aqueous Ammonia, Chlorine and Sulfur Dioxide*. See also the industry-specific guidance documents for ammonia refrigeration and POTWs.

The modeling carried out for aqueous ammonia also is applied in this guidance to ammonia released as a neutrally buoyant plume in other situations. The tables of distances derived from this modeling would apply to evaporation of ammonia from a water solution, evaporation of ammonia liquefied by refrigeration, or ammonia releases from the vapor space of a vessel, because the ammonia would behave as a neutrally buoyant plume (or possibly buoyant in some cases).

### D.4.4 Choice of Reference Table for Dispersion Distances

**Gases.** Exhibit B-1 of Appendix B indicates whether the reference tables for neutrally buoyant or dense gases should be used for each of the regulated toxic gases. Exhibit C-2, Appendix C, provides this information for flammable gases. The choice of reference table presented in these exhibits is based on the molecular weight of the regulated substance compared to air; however, a number of factors that may cause a substance with a molecular weight similar to or smaller than the molecular weight of air to behave as a dense
gas should be considered in selecting the appropriate table. For example, a cold gas may behave as a dense gas, even if it is lighter than air at ambient temperature. Gases liquefied under pressure may be released as a mixture of vapor and liquid droplets; because of presence of liquid mixed with the vapor, a gas that is lighter than air may behave as a dense gas in such a release. A gas that polymerizes or forms hydrogen bonds (e.g., hydrogen fluoride) also may behave as a dense gas.

**Liquids and Solutions:** Exhibits B-2 and B-3, Appendix B, and Exhibit C-3, Appendix C, indicate the reference table of distances to be used for each regulated liquid. The methodology presented in this guidance for consequence analysis for liquids and solutions assumes evaporation from a pool. All of the liquids regulated under CAA section 112(r) have molecular weights greater than the molecular weight of air; therefore, their vapor would be heavier than air. However, because the vapor from a pool will mix with air as it evaporates, the initial density of the vapor with respect to air may not in all cases indicate whether the vapor released from a pool should be modeled as a dense gas or a neutrally buoyant gas. If the rate of release from the pool is relatively low, the vapor-air mixture that is generated may be neutrally buoyant even if the vapor is denser than air, because the mixture may contain a relatively small fraction of the denser-than-air vapor; i.e., it may be mostly air. This may be the case particularly for some of the regulated toxic liquids with relatively low volatility. All of the regulated flammable substances have relatively high volatility; the reference tables for dense gases are assumed to be appropriate for analyzing dispersion of these flammable liquids.

To identify toxic liquids with molecular weight greater than air that might behave as neutrally buoyant gases when evaporating from a pool, EPA used the ALOHA model for pool evaporation of a number of substances with a range of molecular weights and vapor pressures. Modeling was carried out for F stability and wind speed 1.5 meters per second (worst-case conditions) and for D stability and wind speed 3.0 meters per second (alternative-case conditions). Pool spread to a depth of one centimeter was assumed. Additional modeling was carried out for comparison assuming different pool areas and depths. The molecular weight-vapor pressure combinations at which ALOHA used the neutrally buoyant gas model were used to develop the reference table choices given in Exhibit B-2 (for liquids) and B-3 (for solutions) in Appendix B. The neutrally buoyant tables should generally give reasonable results for pool evaporation under ambient conditions when indicated for liquids. At elevated temperatures, however, evaporation rates will be greater, and the dense gas tables should be used.

The liquids for which the neutrally buoyant table is identified for the worst case probably can be expected to behave as neutrally buoyant vapors when evaporating from pools under ambient conditions in most situations, but there may be cases when they exhibit dense gas behavior. Other liquids, for which the neutrally buoyant tables are not indicated for the worst case, might release neutrally buoyant vapors under some conditions (e.g., relatively small pools, temperature not much above 25 °C). Similarly, the liquids for which the neutrally buoyant tables are indicated as appropriate for alternative scenario analysis probably can be considered to behave as neutrally buoyant vapors under the alternative scenario conditions in most cases; however, there may be cases where they will behave as dense gases, and there may be other liquids that in some cases would exhibit neutrally buoyant behavior when evaporating. The reference table choices shown in Exhibit B-2 are intended to reflect the most likely behavior of the substances; they will not predict behavior of the listed substances evaporating under all conditions.
D.4.5 Additional Modeling for Comparison

Modeling was carried out for two worst-case examples and two alternative-case examples, using two different models, for comparison with the results obtained from the methods and distance tables in this guidance. This modeling is discussed below.

**ALOHA Model.** The Areal Locations of Hazardous Atmospheres (ALOHA) system was developed jointly by NOAA and EPA. ALOHA Version 5.2.1 was used for the comparison modeling. The parameters for ALOHA modeling were the same as specified in this guidance document for worst-case and alternative scenarios. The substances modeled are included in ALOHA’s chemical database, so no chemical data were entered for modeling. For consistency with the methodology used to develop the reference tables of distances, a wind speed height of 10 meters was selected for ALOHA modeling.

For all of the substances modeled, the direct source model was chosen for ALOHA modeling, and the release rate estimated using the guidance methodology was entered as the release rate for ALOHA. ALOHA selected the dense gas model to estimate the distances to the endpoints in all cases.

**WHAZAN Model.** The World Bank Hazard Analysis (WHAZAN) system was developed by Technica International in collaboration with the World Bank. The 1988 version of WHAZAN was used for the comparison modeling. The parameters for atmospheric stability, wind speed, and ambient temperature and humidity were the same as specified in this guidance document. For surface roughness, WHAZAN requires entry of a “roughness parameter,” rather than a height. Based on the discussion of this parameter in the WHAZAN Theory Manual, a roughness parameter of 0.07 (corresponding to flat land, few trees) was chosen as equivalent to the surface roughness of 3 centimeters used to represent rural topography in modeling to develop the distance tables for this guidance. A roughness parameter of 0.17 (for woods or rural area or industrial site) was chosen as equivalent to 1 meter, which was used to develop the urban distance tables. Data were added to the WHAZAN chemical database for acrylonitrile and allyl alcohol; ethylene oxide and chlorine were already included in the database.

For WHAZAN modeling of the gases ethylene oxide and chlorine and the liquid acrylonitrile, the WHAZAN dense cloud dispersion model was used. For the alternative-case release of allyl alcohol, the buoyant plume dispersion model was used for consistency with the guidance methodology. The release rates estimated using the guidance methodology were entered as the release rates for all of the WHAZAN modeling.

The WHAZAN dense cloud dispersion requires a “volume dilution factor” as one of its inputs. This factor was not explained; it was presumed to account for dilution of pressurized gases with air upon release. For the gases modeled, the default dilution factor of 60 was used; for acrylonitrile, a dilution factor of 0 was entered. This factor appears to have little effect on the distance results.

D.5 Worst-Case Consequence Analysis for Flammable Substances

The equation used for the vapor cloud explosion analysis for the worst case involving flammable substances is given in Appendix C. This equation is based on the TNT-equivalency method of the UK Health and Safety Executive, as presented in the publication of the Center for Chemical Process Safety of the American Institute of Chemical Engineers (AIChE), *Guidelines for Evaluating the Characteristics of Vapor*
Cloud Explosions, Flash Fires, and BLEVEs (1994). The assumption was made for the worst case that the total quantity of the released substance is in the flammable part of the cloud. The AIChe document lists this assumption as one of a number that have been used for vapor cloud explosion blast prediction; it was chosen as a conservative assumption for the worst-case analysis. The yield factor of 10 percent was a conservative worst-case assumption, based on information presented in the AIChe document. According to the AIChe document, reported values for TNT equivalency for vapor cloud explosions range from a fraction of one percent to tens of percent; for most major vapor cloud explosions, the range is one to ten percent.

The endpoint for the vapor cloud explosion analysis, 1 psi, is reported to cause damage such as shattering of glass windows and partial demolition of houses. Skin laceration from flying glass also is reported. This endpoint was chosen for the consequence analysis because of the potential for serious injuries to people from the property damage that might result from an explosion.

The TNT equivalent model was chosen as the basis for the consequence analysis because of its simplicity and wide use. This model does not take into account site-specific factors and many chemical-specific factors that may affect the results of a vapor cloud explosion. Other methods are available for vapor cloud explosion modeling; see the list of references in Appendix A for some publications that include information on other vapor cloud explosion modeling methods.

D.6 Alternative Scenario Analysis for Gases

The equation for estimating release rate of a gas from a hole in a tank is based on the equations for gas discharge rate presented in the Handbook of Chemical Hazard Analysis Procedures by the Federal Emergency Management Agency (FEMA), DOT, and EPA, and equations in EPA’s Workbook of Screening Techniques for Assessing Impacts of Toxic Air Pollutants. The equation for an instantaneous discharge under non-choked flow conditions is:

\[ m = C_d A_h \sqrt{2p_0 \rho_0 \left( \frac{\gamma}{\gamma - 1} \right) \left( \frac{p_1}{p_0} \right)^{\gamma - 1} - \left( \frac{p_1}{p_0} \right)^{\gamma} } \]  

(D-7)

where:

- \( m \) = Discharge rate (kg/s)
- \( C_d \) = Discharge coefficient
- \( A_h \) = Opening area (m²)
- \( \gamma \) = Ratio of specific heats
- \( p_0 \) = Tank pressure (Pascals)
- \( p_1 \) = Ambient pressure (Pascals)
- \( \rho_0 \) = Density (kg/m³)

Under choked flow conditions (maximum flow rate), the equation becomes:
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\[ m = C_d \ A_h \ \sqrt{\gamma p_0 \rho_0 \left(\frac{2}{\gamma + 1}\right)^{\frac{\gamma + 1}{\gamma - 1}}} \]  \hspace{1cm} (D-8)

For development of the equation and gas factors presented in this guidance, density (\(\rho\)) was rewritten as a function of pressure and molecular weight, based on the ideal gas law:

\[ \rho = \frac{p_0 \ MW}{R \ T_t} \]  \hspace{1cm} (D-9)

where:  \(MW\) = Molecular weight (kilograms per kilomole)  
\(R\) = Gas constant (8,314 Joules per degree-kilomole)  
\(T_t\) = Tank temperature (K)

The choked flow equation can be rewritten:

\[ m = C_d \ A_h \ p_0 \ \frac{1}{\sqrt{T_t}} \ \sqrt{\gamma \left(\frac{2}{\gamma + 1}\right)^{\frac{\gamma + 1}{\gamma - 1}}} \ \sqrt{\frac{MW}{8314}} \]  \hspace{1cm} (D-10)

To derive the equation presented in the guidance, all the chemical-specific properties, constants, and appropriate conversion factors were combined into the "Gas Factor" (GF). The discharge coefficient was assumed to have a value of 0.8, based on the screening value recommended in EPA's Workbook of Screening Techniques for Assessing Impacts of Toxic Air Pollutants. The GF was derived as follows:

\[ GF = 132.2 \times 6,895 \times 6.4516 \times 10^{-4} \times 0.8 \ \sqrt{\gamma \left(\frac{2}{\gamma + 1}\right)^{\frac{\gamma + 1}{\gamma - 1}}} \ \sqrt{\frac{MW}{8314}} \]  \hspace{1cm} (D-11)

where:  \(132.2\) = Conversion factor for lbs/min to kg/s  
\(6,895\) = Conversion factor for psi to Pascals (\(p_0\))  
\(6.4516 \times 10^{-4}\) = Conversion factor for square inches to square meters (\(A_h\))

GF values were calculated for all gases regulated under CAA section 112(r) and are listed in Appendix B, Exhibit B-1, for toxic gases and Appendix C, Exhibit C-2, for flammable gases.

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From the equation for choked flow above and the equation for the GF above, the initial release rate for a gas from a hole in a tank can be written as:

$$QR = HA \times P_t \times \frac{1}{\sqrt{T_i}} \times GF$$  \hspace{1cm} (D-12)

where:
- \(QR\) = Release rate (pounds per minute)
- \(HA\) = Hole area (square inches)
- \(P_t\) = Tank pressure (psia)
- \(T_i\) = Tank temperature (K)

**D.7 Alternative Scenario Analysis for Liquids**

**D.7.1 Releases from Holes in Tanks**

The equation for estimating release rate of a liquid from a hole in a tank is based on the equations for liquid release rate presented in the *Handbook of Chemical Hazard Analysis Procedures* by FEMA, DOT, and EPA and EPA’s *Workbook of Screening Techniques for Assessing Impacts of Toxic Air Pollutants*. The equation for the instantaneous release rate is:

$$m = A_h C_d \sqrt{\rho_l \left[2g \rho_l (H_L - H_b) + 2(P_0 - P_a)\right]}$$  \hspace{1cm} (D-13)

where:
- \(m\) = Discharge rate (kilograms per second)
- \(A_h\) = Opening area (square meters)
- \(C_d\) = Discharge coefficient (unitless)
- \(g\) = Gravitational constant (9.8 meters per second squared)
- \(\rho_l\) = Liquid density (kilograms per cubic meter)
- \(P_0\) = Storage pressure (Pascals)
- \(P_a\) = Ambient pressure (Pascals)
- \(H_L\) = Liquid height above bottom of container (meters)
- \(H_b\) = Height of opening (meters)

A version of this equation is presented in the guidance for use with data found in Appendix B, for gases liquefied under pressure. The equation in the text was derived using the conversion factors listed below and density factors and equilibrium vapor pressure or tank pressure values listed in Appendix B, Exhibit B-1. Equation D-13 becomes:

$$QR = 132.2 \times 6.4516 \times 10^{-4} \times 0.8 \times HA \sqrt{16.018 \times d \times [2 \times 9.8 \times 16.018 \times d \times L H \times 0.0254 + 2P_g \times 6895]}$$  \hspace{1cm} (D-14)
where:  
- \( QR \) = Release rate (pounds per minute)  
- \( HA \) = Hole area (square inches)  
- 132.2 = Conversion factor for kilograms per second to pounds per minute  
- \( 6.4516 \times 10^{-4} \) = Conversion factor for square inches to square meters (HA)  
- 0.8 = Discharge coefficient (0.8)  
- \( d \) = Liquid density (pounds per cubic foot); can derived by using the density factor: \( 1/(DFx0.033) \)  
- \( 16.018 \) = Conversion factor for pounds per cubic feet to kilograms per cubic meters (D)  
- 9.8 = Gravitational constant (meters per second squared)  
- \( LH \) = Height of liquid above hole (inches)  
- \( 2.54 \times 10^{-2} \) = Conversion factor for inches to meters (LH)  
- \( P_g \) = Gauge pressure in tank (psi)  
- \( 6.895 \) = Conversion factor for psi to Pascals (\( P_g \))

After combining the conversion factors and incorporating the density factor (DF), this equation becomes:

\[
QR = HA \times 6.82 \sqrt{\frac{0.7}{DF^2 \times LH} + \frac{669}{DF} \times P_g}
\]  

(D-15)

For liquids stored at ambient pressure, Equation D-13 becomes:

\[
m = A_h C_d \rho_l \sqrt{\frac{2g}{H_L - H_h}}
\]  

(D-16)

To derive the equation presented in the guidance for liquids under ambient pressure, all the chemical-specific properties, constants, and conversion factors were combined into the ”Liquid Leak Factor” (LLF). The discharge coefficient was assumed to have a value of 0.8, based on the screening value recommended in EPA's *Workbook of Screening Techniques for Assessing Impacts of Toxic Air Pollutants*. The LLF was derived as follows:

\[
LLF = 132.2 \times 6.4516 \times 10^{-4} \times 0.1594 \times 0.8 \times \sqrt{2 \times 9.8 \times \rho_l}
\]  

(D-17)

where:  
- \( LLF \) = Liquid Leak Factor (pounds per minute-inches\(^2\))  
- 132.2 = Conversion factor for kilograms per second to pounds per minute (\( m \))  
- \( 6.4516 \times 10^{-4} \) = Conversion factor for square inches to square meters (\( A_h \))  
- 0.1594 = Conversion factor for square root of inches to square root of meters (\( H_L - H_h \))  
- 0.8 = Discharge coefficient (0.8)  
- 9.8 = Gravitational constant (meters per second squared)  
- \( \rho_l \) = Liquid density (kilograms per cubic meter)
LLF values were calculated for all liquids regulated under CAA section 112(r) and are listed in Appendix B, Exhibit B-2, for toxic liquids and Appendix C, Exhibit C-3, for flammable liquids.

From the equation for liquid release rate from a hole in a tank at ambient pressure and the equation for the LLF, the initial release rate for a liquid from a tank under atmospheric pressure can be written as:

$$QR_L = HA \times \sqrt{LH} \times LLF$$  \hspace{1cm} (D-18)

where:
- $QR_L$ = Liquid release rate (pounds per minute)
- $HA$ = Hole area (square inches)
- $LH$ = Height of liquid above hole (inches)

### D.7.2 Releases from Pipes

The equation used to estimate releases of liquids from pipes is the Bernoulli equation. It assumes that the density of the liquid is constant and does not account for losses in velocity due to wall friction. The equation follows:

$$\frac{(P_a - P_b)}{D} + g \frac{(Z_a - Z_b)}{g_c} = \frac{(V_b^2 - V_a^2)}{2g_c}$$  \hspace{1cm} (D-19)

where:
- $P_a$ = Pressure at pipe inlet (Pascals)
- $P_b$ = Pressure at pipe outlet (Pascals)
- $Z_a$ = Height above datum plane at pipe inlet (meters)
- $Z_b$ = Height above datum plane at pipe release (meters)
- $g$ = Gravitational acceleration (9.8 meters per second squared)
- $g_c$ = Newton's law proportionality factor (1.0)
- $V_a$ = Operational velocity (meters per second)
- $V_b$ = Release velocity (meters per second)
- $D$ = Density of liquid (kilograms per cubic meter)

Isolating $V_b$ yields:

$$V_b = \sqrt{\frac{2 \ g_c \ (P_a - P_b)}{D} + 2 \ g \ (Z_a - Z_b) + V_a^2}$$  \hspace{1cm} (D-20)

To develop the equation presented in the text, conversion factors for English units and constants were incorporated as follows:
\[ V_b = 197 \sqrt{\frac{2 \times 6895 \times (P_T - 14.7) \times DF \times 0.033}{16.08}} + (2 \times 9.8 \times 0.3048 \times (Z_a - Z_b)) + 0.00508 \times V_a^2 \]  
(D-21)

where:

- \( V_b \) = Release velocity (feet per minute)
- 197 = Conversion factor for meters per second to feet per minute
- 6895 = Conversion factor for psi to Pascals
- \( P_T \) = Total pipe pressure (psi)
- 14.7 = Atmospheric pressure (psi)
- 16.08 = Conversion factor for pounds per cubic foot to kilograms per cubic meter
- \( DF \) = Density factor (1/(0.033 DF) = density in pounds per cubic foot)
- 9.8 = Gravitational acceleration (meters per second\(^2\))
- 0.3048 = Conversion factor for feet to meters
- \( Z_a - Z_b \) = Change in pipe elevation, inlet to outlet (feet)
- 0.00508 = Conversion factor for feet per minute to meters per second
- \( V_a \) = Operational velocity (feet per minute)

### D.8 Vapor Cloud Fires

Factors for leaks from tanks for flammable substances (GF and LLF) were derived as described for toxic substances (see above).

The endpoint for estimating impact distances for vapor cloud fires of flammable substances is the lower flammability limit (LFL). The LFL is one of the endpoints for releases of flammable substances specified in the RMP Rule. It was chosen to provide a reasonable, but not overly conservative, estimation of the possible extent of a vapor cloud fire.

### D.9 Pool Fires

A factor used for estimating the distance to a heat radiation level from a pool fire that could cause second degree burns from a 40-second exposure was developed based on equations presented in the AIChE document, *Guidelines for Evaluating the Characteristics of Vapor Cloud Explosions, Flash Fires, and BLEVEs* and in the Netherlands TNO document, *Methods for the Determination of Possible Damage to People and Objects Resulting from Releases of Hazardous Materials* (1992). The AIChE and TNO documents present a point-source model that assumes that a selected fraction of the heat of combustion is emitted as radiation in all directions. The radiation per unit area received by a target at some distance from the point source is given by:

\[ q = \frac{f m H_c \tau_a}{4 \pi x^2} \]  
(D-22)
where:  
\[ q = \text{Radiation per unit area received by the receptor (Watts per square meter)} \]
\[ m = \text{Rate of combustion (kilograms per second)} \]
\[ \tau_a = \text{Atmospheric transmissivity} \]
\[ H_c = \text{Heat of combustion (Joules per kilogram)} \]
\[ f = \text{Fraction of heat of combustion radiated} \]
\[ x = \text{Distance from point source to receptor (meters)} \]

The fraction of combustion energy dissipated as thermal radiation (\( f \)) in the equation above is reported to range from 0.1 to 0.4. To develop factors for estimating distances for pool fires, this fraction was assumed to be 0.4 for all the regulated flammable substances. The heat radiation level (\( q \)) was assumed to be 5 kilowatts (5,000 Watts) per square meter. This level is reported to cause second degree burns from a 40-second exposure. One of the endpoints for releases of flammable substances specified in the RMP Rule is 5 kilowatts per square meter for 40 seconds. It was assumed that people would be able to escape from the heat in 40 seconds. The atmospheric transmissivity (\( \tau_a \)) was assumed equal to one.

For a pool fire of a flammable substance with a boiling point above the ambient temperature, the combustion rate can be estimated by the following empirical equation:

\[
m = \frac{0.0010 \cdot H_c \cdot A}{H_v + C_p \cdot (T_b - T_a)}
\]

where:
\[ m = \text{Rate of combustion (kilograms per second)} \]
\[ H_c = \text{Heat of combustion (Joules per kilogram)} \]
\[ H_v = \text{Heat of vaporization (Joules per kilogram)} \]
\[ C_p = \text{Liquid heat capacity (Joules per kilogram-degree K)} \]
\[ A = \text{Pool area (square meters)} \]
\[ T_b = \text{Boiling temperature (K)} \]
\[ T_a = \text{Ambient temperature (K)} \]
\[ 0.0010 = \text{Constant} \]

Combining Equations D-22 and D-23 (above), and assuming a heat radiation level of 5,000 Watts per square meter, gives the following equation for liquid pools of substances with boiling points above ambient temperature:

\[
x = H_c \sqrt{\frac{0.0010 \cdot A}{4\pi q}} \left( \frac{H_v + C_p(T_b - T_a)}{H_v + C_p(T_b - T_a)} \right)
\]

or
\[ x = H_c \sqrt{\frac{0.0001 \, A}{5,000\pi \left( H_v + C_p(T_b - T_a) \right)}} \]  \hspace{1cm} (D-25)

where:
- \( x \) = Distance from point source to receptor (meters)
- \( q \) = Radiation per unit area received by the receptor = 5,000 Watts per square meter
- \( H_c \) = Heat of combustion (Joules per kilogram)
- \( f \) = Fraction of heat of combustion radiated = 0.4
- \( H_v \) = Heat of vaporization (Joules per kilogram)
- \( C_p \) = Liquid heat capacity (Joules per kilogram-degree Kelvin)
- \( A \) = Pool area (square meters)
- \( T_b \) = Boiling temperature (K)
- \( T_a \) = Ambient temperature (K)
- 0.0010 = Constant

For a pool fire of a flammable substance with a boiling point below the ambient temperature (i.e., liquefied gases) the combustion rate can be estimated by the following equation, based on the TNO document:

\[ m = \frac{0.0010 \, H_c \, A}{H_v} \]  \hspace{1cm} (D-26)

where:
- \( m \) = Rate of combustion (kilograms per second)
- \( H_v \) = Heat of vaporization (Joules per kilogram)
- \( H_c \) = Heat of combustion (Joules per kilogram)
- \( A \) = Pool area (square meters)
- 0.0010 = Constant

Then the equation for distance at which the radiation received equals 5,000 Watts per square meter becomes:

\[ x = H_c \sqrt{\frac{0.0001 \, A}{5,000\pi \, H_v}} \]  \hspace{1cm} (D-27)

where:
- \( x \) = Distance from point source to receptor (meters)
- 5,000 = Radiation per unit area received by the receptor (Watts per square meter)
- \( H_c \) = Heat of combustion (Joules per kilogram)
- \( H_v \) = Heat of vaporization (Joules per kilogram)
- \( A \) = Pool area (square meters)
- 0.0001 = Derived constant (see equations D-20 and D-21)
A "Pool Fire Factor" (PFF) was calculated for each regulated flammable liquid and gas (to be applied to gases liquefied by refrigeration) to allow estimation of the distance to the heat radiation level that would lead to second degree burns. For the derivation of this factor, ambient temperature was assumed to be 298 K (25 °C). Other factors are discussed above. The PFF for liquids with boiling points above ambient temperature was derived as follows:

\[ PFF = H_c \sqrt{\frac{0.0001}{5000\pi [H_v + C_p(T_b - 298)]}} \] (D-28)

where:
- 5,000 = Radiation per unit area received by the receptor (Watts per square meter)
- \( H_c \) = Heat of combustion (Joules per kilogram)
- \( H_v \) = Heat of vaporization (Joules per kilogram)
- \( C_p \) = Liquid heat capacity (Joules per kilogram-degree K)
- \( T_b \) = Boiling temperature (K)
- 298 = Assumed ambient temperature (K)
- 0.0001 = Derived constant (see above)

For liquids with boiling points below ambient temperature, the PFF is derived as follows:

\[ PFF = H_c \sqrt{\frac{0.0001}{5000\pi H_v}} \] (D-29)

where:
- 5,000 = Radiation per unit area received by the receptor (Watts per square meter)
- \( H_c \) = Heat of combustion (Joules per kilogram)
- \( H_v \) = Heat of vaporization (Joules per kilogram)
- 0.0001 = Derived constant (see above)

Distances where exposed people could potentially suffer second degree burns can be estimated as the PFF multiplied by the square root of the pool area (in square feet), as discussed in the text.

D.10 BLEVEs

Reference Table 30, the table of distances for BLEVEs, was developed based on equations presented in the AIChE document, *Guidelines for Evaluating the Characteristics of Vapor Cloud Explosions, Flash Fires, and BLEVEs*. The Hymes point-source model for a fireball, as cited in the AIChE document, uses the following equation for the radiation received by a receptor:

\[ q = \frac{2.2 \, \tau \, R \, H_c \, m_f^{0.67}}{4\pi L^2} \] (D-30)

where:
- \( q \) = Radiation received by the receptor (Watts per square meter)
- \( m_f \) = Mass of fuel in the fireball (kg)
Appendix D
Technical Background

\[ \tau_s = \text{Atmospheric transmissivity} \]
\[ H_c = \text{Heat of combustion (Joules per kilogram)} \]
\[ R = \text{Radiative fraction of heat of combustion} \]
\[ L = \text{Distance from fireball center to receptor (meters)} \]
\[ \pi = 3.14 \]

Hymes (as cited by AIChE) suggests the following values for R:

\[ R = 0.3 \text{ for vessels bursting below relief valve pressure} \]
\[ R = 0.4 \text{ for vessels bursting at or above relief valve pressure} \]

For development of Reference Table 30, the following conservative assumptions were made:

\[ R = 0.4 \]
\[ \tau_s = 1 \]

The effects of radiant heat on an exposed person depend on both the intensity of the radiation and the duration of the exposure. For development of the table of distances for BLEVEs, it was assumed that the time of exposure would equal the duration of the fireball. The AIChE document gives the following equations for duration of a fireball:

\[ t_c = 0.45 \, m_f^{1/3} \text{ for } m_f < 30,000 \, \text{kg} \]  
(D-31)

and

\[ t_c = 2.6 \, m_f^{1/6} \text{ for } m_f > 30,000 \, \text{kg} \]  
(D-32)

where:
\[ m_f = \text{Mass of fuel (kg)} \]
\[ t_c = \text{Combustion duration (seconds)} \]

According to several sources (e.g., Eisenberg, et al., Vulnerability Model, A Simulation System for Assessing Damage Resulting from Marine Spills; Mudan, Thermal Radiation Hazards from Hydrocarbon Pool Fires (citing K. Buettner)), the effects of thermal radiation are generally proportional to radiation intensity to the four-thirds power times time of exposure. Thus, a thermal "dose" can be estimated using the following equation:

\[ \text{Dose} = t \, q^{4/3} \]  
(D-33)

where:
\[ t = \text{Duration of exposure (seconds)} \]
\[ q = \text{Radiation intensity (Watts/m}^2\text{)} \]

The thermal "dose" that could cause second-degree burns was estimated assuming 40 seconds as the duration of exposure and 5,000 Watts/m² as the radiation intensity. The corresponding dose is 3,420,000 (Watts/m²)⁴/³-second.
For estimating the distance from a fireball at which a receptor might receive enough thermal radiation to cause second degree burns, the dose estimated above was substituted into the equation for radiation received from a fireball:

\[ q = \left[ \frac{3,420,000}{t} \right]^{\frac{3}{4}} \]  \hspace{1cm} (D-34)

\[ \left[ \frac{3,420,000}{t} \right]^{\frac{3}{4}} = \frac{2.2 \; \tau_a \; R \; H_c \; m_f^{0.67}}{4\pi L^2} \]  \hspace{1cm} (D-35)

\[ L = \sqrt{\frac{2.2 \; \tau_a \; R \; H_c \; m_f^{0.67}}{4\pi \left[ \frac{3,420,000}{t} \right]^{\frac{3}{4}}}} \]  \hspace{1cm} (D-36)

where:
- \(L\) = Distance from fireball center to receptor (meters)
- \(q\) = Radiation received by the receptor (Watts per square meter)
- \(m_f\) = Mass of fuel in the fireball (kg)
- \(\tau_a\) = Atmospheric transmissivity (assumed to be 1)
- \(H_c\) = Heat of combustion (Joules per kilogram)
- \(R\) = Radiative fraction of heat of combustion (assumed to be 0.4)
- \(t\) = Duration of the fireball (seconds) (estimated from the equations above); assumed to be duration of exposure

Equation D-36 was used to develop the reference table for BLEVEs presented in the text (Reference Table 30).

**D.11 Alternative Scenario Analysis for Vapor Cloud Explosions**

According to T.A. Kletz, in "Unconfined Vapor Cloud Explosions" (Eleventh Loss Prevention Symposium, sponsored by AIChE, 1977), unconfined vapor cloud explosions almost always result from the release of flashing liquids. For this reason, the quantity in the cloud for the alternative scenario vapor cloud explosion in this guidance is based on the fraction flashed from the release of a flammable gas liquefied under pressure. The guidance provides a method to estimate the quantity in the cloud from the fraction flashed into vapor plus the quantity that might be carried along as aerosol. The recommendation to use twice the quantity flashed as the mass in the cloud (so long as it does not exceed the total amount of flammable substance available) is based on the method recommended by the UK Health and Safety Executive (HSE), as cited in the AIChE document, *Guidelines for Evaluating the Characteristics of Vapor Cloud Explosions, Flash Fires, and BLEVEs*. The factor of two is intended to allow for spray and aerosol formation.
The equation for the flash fraction, for possible use in for the alternative scenario analysis, is based on the Netherlands TNO document, *Methods for the Calculation of the Physical Effects of the Escape of Dangerous Material* (1980), Chapter 4, "Spray Release." The following equation is provided:

\[
X_{\text{vap,a}} = \left( X_{\text{vap,b}} \frac{T_b}{T_l} \right) + \left( \frac{T_b C_i}{h_v} \ln \frac{T_l}{T_b} \right) \tag{D-37}
\]

where:

- \( X_{\text{vap,a}} \) = Weight fraction of vapor after expansion
- \( X_{\text{vap,b}} \) = Weight fraction of vapor before expansion (assumed to be 0 for calculation of the flash fraction)
- \( T_b \) = Boiling temperature of gas compressed to liquid (K)
- \( T_l \) = Temperature of stored gas compressed to liquid (K)
- \( C_i \) = Specific heat of gas compressed to liquid (Joules/kilogram-K)
- \( h_v \) = Heat of evaporation of gas compressed to liquid (Joules/kilogram)

To develop a Flash Fraction Factor (FFF) for use in consequence analysis, compressed gases were assumed to be stored at 25 °C (298 K) (except in cases where the gas could not be liquefied at that temperature). The equation for FFF is:

\[
FFF = \left( \frac{T_b C_i}{h_v} \ln \frac{298}{T_b} \right) \tag{D-38}
\]

where:

- \( T_b \) = Boiling temperature of gas compressed to liquid (K)
- \( C_i \) = Specific heat of gas compressed to liquid (Joules/kilogram-K)
- \( h_v \) = Heat of evaporation of gas compressed to liquid (Joules/kilogram)
- 298 = Temperature of stored gas compressed to liquid (K)

The recommendation to use a yield factor of 0.03 for the alternative scenario analysis for vapor cloud explosions also is based on the UK HSE method cited by AIChE. According to the AIChE document, this recommendation is based on surveys showing that most major vapor cloud explosions have developed between 1 percent and 3 percent of available energy.
APPENDIX E

WORKSHEETS FOR OFFSITE CONSEQUENCE ANALYSIS

Using the Methods in this Guidance

April 15, 1999
# WORKSHEET 1
WORST-CASE ANALYSIS FOR TOXIC GAS

## 1. Select Scenario
(defined by rule for worst case as release of largest quantity over 10 minutes)

<table>
<thead>
<tr>
<th>Identification</th>
<th>Guidance Reference</th>
</tr>
</thead>
</table>
| Identify toxic gas | Name: ____________________________  
CAS number: _______ - ____ - ____  Chapter 2 Section 3.1 |
| Identify largest quantity in largest vessel or pipeline | Quantity (pounds): _______ |
| Identify worst-case meteorological conditions | Atmospheric stability class: F  
Wind speed: 1.5 m/s  
Ambient temperature: 25 °C  
Relative humidity: 50% |

## 2. Determine Release Rate

<table>
<thead>
<tr>
<th>Action</th>
<th>Guidance Reference</th>
</tr>
</thead>
</table>
| Estimate release rate  
Quantity/10 min, except gases liquefied by refrigeration in some cases | Release rate (lbs/min): ___________  
Will release always take place in enclosure? ____  
(If yes, go to next step)  
Can release cause failure of enclosure? ____  
(If yes, use unmitigated release rate)  
Factor to account for enclosure: 0.55  
Mitigated release rate (lbs/min): _______  Section 3.1.1 |
| Revise release rate to account for passive mitigation (enclosure) |

## 3. Determine Distance to the Endpoint Specified by Rule

<table>
<thead>
<tr>
<th>Identification</th>
<th>Guidance Reference</th>
</tr>
</thead>
<tbody>
<tr>
<td>Identify endpoint</td>
<td>Endpoint (mg/L): _______  Exhibit B-1</td>
</tr>
</tbody>
</table>
| Determine gas density  
Consider conditions (e.g., liquefied under pressure) | Dense: ____  
Neutrally buoyant: ____  Exhibit B-1 |
| Determine site topography  
Rural and urban defined by rule | Rural: ____  
Urban: ____  Section 2.1 |
| Determine appropriate reference table of distances  
Use 10-minute tables | Reference table used (number): ____  Chapter 4 Reference Tables 1-12 |
| Find distance on reference table | Release rate/endpoint (neutrally buoyant): ____  
Distance to endpoint (mi): _______  Chapter 4 Reference Tables 1-12 |
### WORKSHEET 2
WORST-CASE ANALYSIS FOR TOXIC LIQUID

<table>
<thead>
<tr>
<th>1. Select Scenario (defined by rule for worst case as release of largest quantity to form an evaporating pool)</th>
<th>Guidance Reference</th>
</tr>
</thead>
<tbody>
<tr>
<td>• Identify toxic liquid</td>
<td>Name: __________________________</td>
</tr>
<tr>
<td>• Identify concentration for solutions or mixtures</td>
<td>CAS number: _____ - <em><strong><strong>-</strong></strong></em></td>
</tr>
<tr>
<td></td>
<td>Concentration in solution or mixture (wt %): ____</td>
</tr>
<tr>
<td>• Identify largest quantity in largest vessel or pipeline</td>
<td>Quantity (pounds): ______</td>
</tr>
<tr>
<td></td>
<td>Quantity of regulated substance in mixture: ____</td>
</tr>
<tr>
<td>• Identify worst-case meteorological conditions</td>
<td>Atmospheric stability class: F</td>
</tr>
<tr>
<td></td>
<td>Wind speed: 1.5 m/s</td>
</tr>
<tr>
<td></td>
<td>Ambient temperature: 25 °C</td>
</tr>
<tr>
<td></td>
<td>Relative humidity: 50%</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>2. Determine Release Rate</th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>• Determine temperature of spilled liquid</td>
<td>Temperature of liquid (°C): ____</td>
</tr>
<tr>
<td>Must be highest maximum daily temperature or process temperature, or boiling point for gases liquefied by refrigeration</td>
<td>Section 3.2</td>
</tr>
<tr>
<td></td>
<td>Temperature of liquid (°C): ____</td>
</tr>
<tr>
<td>• Determine appropriate liquid factors for release rate estimation</td>
<td>LFA: ______</td>
</tr>
<tr>
<td></td>
<td>LFB: ______</td>
</tr>
<tr>
<td></td>
<td>DF: ______</td>
</tr>
<tr>
<td></td>
<td>TCF: ______</td>
</tr>
<tr>
<td></td>
<td>Section 3.2, Exhibits B-2, B-4</td>
</tr>
<tr>
<td></td>
<td>Section 3.3, Exhibit B-3 for water solutions</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Estimate Maximum Pool Area</th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>• Estimate maximum pool area</td>
<td>Maximum pool area (ft²): ____</td>
</tr>
<tr>
<td>Spilled liquid forms pool 1 cm deep</td>
<td>Section 3.2.3</td>
</tr>
<tr>
<td></td>
<td>Equation 3-6</td>
</tr>
</tbody>
</table>
## Estimate Pool Area for Spill into Diked Area

- Estimate diked area
  - Consider failure of dikes or overflow of diked area
  - **Diked area (ft²): _______**
  - **Is diked area smaller than maximum area? ___**
    - (If no, use maximum area to estimate release rate)
  - **Diked volume (ft³): _______**
  - **Spilled volume (ft³): _______**
  - **Is spilled volume smaller than diked volume? ___**
    - (If no, estimate overflow)
  - **Overflow volume (ft³): _______**
  - **Overflow area (ft²): _______**
  - Section 3.2.3

- Choose pool area for release rate estimation
  - Maximum area, diked area, or sum of diked area and overflow area
  - **Pool area (ft²): _______**
  - Section 3.2.3

## Estimate Release Rate from Pool

- Estimate release rate for undiked pool (maximum pool area)
  - Based on quantity spilled, LFA or LFB, and DF
  - **Release rate (lbs/min): _______**
  - Section 3.2.2
  - Section 3.2.4 (mixtures)
  - Equation 3-3 or 3-4

- Estimate release rate for diked pool (use pool area from previous section)
  - Based on pool area and LFA or LFB
  - **Release rate (lbs/min): _______**
  - Section 3.2.2
  - Section 3.2.4 (mixtures)
  - Equation 3-7 or 3-8

- Revise release rate for release in building
  - Apply factor to release rate
  - **Release rate if outside (lbs/min) ____**
    - (Use release rate for undiked or diked pool)
  - **Factor to account for enclosure: 0.1**
  - **Revised release rate (lbs/min): _______**
  - Section 3.2.3
  - Equations 3-9, 3-10

- Revise release rate for temperature
  - Apply appropriate TCF to release rate
  - **Revised release rate (lbs/min): _______**
  - Section 3.2.5
  - Equation 3-11

- Estimate duration of release
  - **Release duration (min): _______**
  - Section 3.2.2
  - Equation 3-5
### WORKSHEET 2 (continued)

#### 3. Determine Distance to the Endpoint

<table>
<thead>
<tr>
<th>Step</th>
<th>Description</th>
<th>Reference</th>
</tr>
</thead>
<tbody>
<tr>
<td>- Identify endpoint</td>
<td>Endpoint (mg/L): ___________</td>
<td>Exhibit B-2</td>
</tr>
<tr>
<td><strong>Specified by rule</strong></td>
<td></td>
<td></td>
</tr>
<tr>
<td>- Determine vapor density</td>
<td>Dense: ___________</td>
<td>Exhibit B-2</td>
</tr>
<tr>
<td></td>
<td>Neutrally buoyant: ___________</td>
<td></td>
</tr>
<tr>
<td>- Determine site topography</td>
<td>Rural: ___________</td>
<td>Section 2.1</td>
</tr>
<tr>
<td><strong>Rural and urban defined by rule</strong></td>
<td>Urban: ___________</td>
<td></td>
</tr>
<tr>
<td>- Determine appropriate reference table of distances</td>
<td>Reference table used (number): ___________</td>
<td>Chapter 4 Reference Tables 1-12</td>
</tr>
<tr>
<td><strong>Based on release duration, vapor density, topography</strong></td>
<td></td>
<td></td>
</tr>
<tr>
<td>- Find distance on reference table</td>
<td>Release rate/endpoint (neutrally buoyant): ___________</td>
<td>Chapter 4 Reference Tables 1-12</td>
</tr>
<tr>
<td></td>
<td>Distance to endpoint (mi): ___________</td>
<td></td>
</tr>
</tbody>
</table>
## WORKSHEET 3
WORST-CASE ANALYSIS FOR FLAMMABLE SUBSTANCE

<table>
<thead>
<tr>
<th>1. Select Scenario (defined by rule for worst case as vapor cloud explosion of largest quantity)</th>
<th>Guidance Reference</th>
</tr>
</thead>
<tbody>
<tr>
<td>• Identify flammable substance</td>
<td>Name: ____________________________</td>
</tr>
<tr>
<td></td>
<td>CAS number: _______ - _______ - ____</td>
</tr>
<tr>
<td>• Identify largest quantity in largest vessel or pipeline</td>
<td>Quantity (pounds): ______</td>
</tr>
<tr>
<td>Consider total quantity of flammable substance, including non-regulated substances in flammable mixtures</td>
<td></td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>2. Determine Distance to the Endpoint (endpoint specified by the rule as 1 psi overpressure; yield factor assumed to be 10% for TNT-equivalent model)</th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>• Estimate distance to 1 psi using Reference Table</td>
<td>Distance to 1 psi (mi): ______</td>
</tr>
<tr>
<td>Find quantity, read distance from table</td>
<td></td>
</tr>
<tr>
<td>• Alternatively, estimate distance to 1 psi using equation</td>
<td></td>
</tr>
<tr>
<td>For pure substance:</td>
<td></td>
</tr>
<tr>
<td>Heat of combustion (kJ/kg): ______</td>
<td></td>
</tr>
<tr>
<td>For mixture:</td>
<td></td>
</tr>
<tr>
<td>Heat of combustion of major component (kJ/kg): ______</td>
<td></td>
</tr>
<tr>
<td>Heats of combustion of other components (kJ/kg): ______, ______, ______</td>
<td></td>
</tr>
<tr>
<td>Distance to 1 psi (mi). ______</td>
<td></td>
</tr>
</tbody>
</table>

<p>| | |</p>
<table>
<thead>
<tr>
<th></th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
</tr>
</tbody>
</table>

April 15, 1999
# WORKSHEET 4
ALTERNATIVE SCENARIO ANALYSIS FOR TOXIC GAS

## 1. Select Scenario

<table>
<thead>
<tr>
<th>Guidance Reference</th>
<th>Name: ____________________________</th>
<th>CAS number: <strong><strong><strong>-</strong>__-</strong></strong></th>
</tr>
</thead>
</table>

- Identify toxic gas
  - Identify conditions of storage or processing of toxic gas: Treat gases liquefied by refrigeration as liquids
  - Non-liquefied pressurized gas: _____
  - Gas liquefied under pressure: _____
  - In tank: _____
  - In pipeline: _____
  - Other (describe): __________________________

<table>
<thead>
<tr>
<th>Identify average meteorological conditions</th>
<th>Describe scenario: ____________________________</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>------------------------------------------------</td>
</tr>
</tbody>
</table>

- Develop alternative scenario:
  - More likely than worst case
  - Should reach endpoint off site
  - Atmospheric stability class: D
  - Wind speed: 3.0 m/s
  - Ambient temperature: 25 °C
  - Relative humidity: 50%

## 2. Determine Release Rate

<table>
<thead>
<tr>
<th>Hole area (in²): ______</th>
<th>Tank pressure (psia): ______</th>
<th>Tank temperature (K): ______</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>GF: ______</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Release rate (lbs/min): _____</td>
</tr>
</tbody>
</table>

- Estimate gas release rate from hole in tank (choked/maximum flow) for
  - Pressurized gas
  - Gas liquefied under pressure released from vapor space

<table>
<thead>
<tr>
<th>Hole area (in²): ______</th>
<th>Tank pressure (psig): ______</th>
<th>DF: ______</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>Liquid height above hole (in): ______</td>
</tr>
<tr>
<td></td>
<td></td>
<td>Release rate (lbs/min): _____</td>
</tr>
</tbody>
</table>

- Estimate flashing liquid release rate from hole in tank
  - Gas liquefied under pressure released from liquid space

---

*April 15, 1999*
### WORKSHEET 4 (continued)

<table>
<thead>
<tr>
<th>Task</th>
<th>Information</th>
<th>Source(s)</th>
</tr>
</thead>
</table>
| Estimate flashing liquid release rate from break in long pipeline   | Initial flow rate (lbs/min): _______  
|                                                                      | DF: _______  
|                                                                      | Initial flow velocity (ft/min): _______  
|                                                                      | Pipe pressure (psi): _______  
|                                                                      | Change in pipe elevation (ft): _______  
|                                                                      | Cross-sectional pipe area (ft²): _______  
|                                                                      | Release rate (lbs/min): _______  | Sections 7.1.1 and 7.2.1  
|                                                                      |                                                                        | Exhibit B-1 |
| Estimate release duration                                           | Time to stop release (min): _______  
|                                                                      | Time to empty tank or pipe (min): _______  | Section 7.1.1 |
|                                                                      | Default release duration: 60 min  |
| Revise release rate for passive mitigation (enclosure)              | Release rate if outside (lbs/min): _______  
|                                                                      | Factor to account for enclosure: 0.55  
|                                                                      | Revised release rate (lbs/min): _______  | Sections 7.1.2  
|                                                                      |                                                                        | Section 3.1.2 |
| Revise release rate for active mitigation                            | Active mitigation technique used: _______  
|                                                                      |                                                                        |             |
|                                                                      | Time to stop release using active technique (min): _______  
|                                                                      | Fractional release rate reduction by active technique: _______  
|                                                                      | Revised release rate (lb/min): _______  |
| Estimate release duration (mitigated release)                       | Release duration (min): _______  | Section 7.1.2 |
| Other release rate estimation                                       | Release rate (lb/min): _______  
|                                                                      | Method of release rate estimation (describe): _______  
|                                                                      | Release duration (min): _______  |

### 3. Determine Distance to the Endpoint

<table>
<thead>
<tr>
<th>Task</th>
<th>Information</th>
<th>Source(s)</th>
</tr>
</thead>
</table>
| Identify endpoint  
*Specified by rule*                                             | Endpoint (mg/L): _______  | Exhibit B-1 |
| Determine gas density  
*Consider conditions (e.g., liquefied under pressure, refrigeration)* | Dense: _______  
|                                                                      | Neutrally buoyant: _______  | Exhibit B-1 |
| Determine site topography  
*Rural and urban defined by rule*                               | Rural: _______  
|                                                                      | Urban: _______  | Section 2.1 |
WORKSHEET 4 (continued)

- Determine appropriate reference table of distances
  Based on release duration, vapor density, and topography

| **Reference table used (number): ______** | **Chapter 8 Reference Tables 14-25** |

- Find distance on reference table

| **Release rate/endpoint (neutrally buoyant): _____**  
| **Distance to endpoint (mi): ______** | **Chapter 8 Reference Tables 14-25** |
# WORKSHEET 5
ALTERNATIVE SCENARIO ANALYSIS FOR TOXIC LIQUID

## 1. Select Scenario

| Guidance Reference |
|---------------------|---|
| Chapter 6           | Chapter 7 |
| Section 7.2         | |

- **Identify toxic liquid Include gases liquefied by refrigeration**
- **Identify concentration for solutions or mixtures**

<table>
<thead>
<tr>
<th>Name: ____________________________</th>
</tr>
</thead>
<tbody>
<tr>
<td>CAS number: <em><strong><em><strong><strong>-</strong></strong></em>-</strong></em>_</td>
</tr>
</tbody>
</table>
| Concentration in solution or mixture (wt %): ______

- **Identify conditions of storage or processing of toxic liquid**

<table>
<thead>
<tr>
<th>Atmospheric tank: _____</th>
</tr>
</thead>
<tbody>
<tr>
<td>Pressurized tank: _____</td>
</tr>
<tr>
<td>Pipeline: _____</td>
</tr>
<tr>
<td>Other (describe): ____________________</td>
</tr>
</tbody>
</table>

- **Develop alternative scenario**
  - More likely than worst case
  - Should reach endpoint off site

<table>
<thead>
<tr>
<th>Describe scenario: ____________________</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
</tr>
</tbody>
</table>

- **Identify meteorological conditions**

<table>
<thead>
<tr>
<th>Atmospheric stability class: F</th>
</tr>
</thead>
<tbody>
<tr>
<td>Wind speed: 3.0 m/s</td>
</tr>
<tr>
<td>Ambient temperature: 25 °C</td>
</tr>
<tr>
<td>Relative humidity: 50%</td>
</tr>
</tbody>
</table>

## 2. Determine Release Rate

### Determine Liquid Release Rate and Quantity Released into Pool

- **Estimate liquid release rate from hole in atmospheric tank**

<table>
<thead>
<tr>
<th>Hole area (in²): _____</th>
</tr>
</thead>
<tbody>
<tr>
<td>LLF: _____</td>
</tr>
<tr>
<td>Liquid height above hole (in): _____</td>
</tr>
<tr>
<td>Liquid release rate (lbs/min): _____</td>
</tr>
</tbody>
</table>

- **Estimate liquid release rate from break in long pipeline**

<table>
<thead>
<tr>
<th>Initial flow rate (lbs/min): _____</th>
</tr>
</thead>
<tbody>
<tr>
<td>DF: _____</td>
</tr>
<tr>
<td>Initial flow velocity (ft/min): _____</td>
</tr>
<tr>
<td>Pipe pressure (psi): _____</td>
</tr>
<tr>
<td>Change in pipe elevation (ft): _____</td>
</tr>
<tr>
<td>Cross-sectional pipe area (ft²): _____</td>
</tr>
<tr>
<td>Liquid release rate (lbs/min): _____</td>
</tr>
</tbody>
</table>

- **Estimate liquid release duration**

| Time to stop release (min): _____ |
| Time to empty tank to level of hole (min): _____ |

<table>
<thead>
<tr>
<th>Section 7.2.1</th>
</tr>
</thead>
<tbody>
<tr>
<td>Equations 7-4</td>
</tr>
<tr>
<td>Exhibit B-2</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Section 7.2.1</th>
</tr>
</thead>
<tbody>
<tr>
<td>Equations 7-5</td>
</tr>
<tr>
<td>- 7-7</td>
</tr>
<tr>
<td>Exhibit B-2</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Section 7.2.1</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
</tr>
</tbody>
</table>
**Active mitigation technique** *(describe): ________  Time to stop release (min): ________

**Quantity of liquid released** *(lbs): ________

**Temperature of liquid** *(°C): ________

**LFA: ________**  
**LFB: ________**  
**DF: ________**  
**TCF: ________**

**Maximum pool area** *(ft²): ________

**Diked area** *(ft²): ________**  
**Is diked area smaller than maximum area?____**  
*(If no, use maximum area to estimate release rate)*  
**Diked volume** *(ft³): ________**  
**Spilled volume** *(ft³): ________**  
**Is spilled volume smaller than diked volume?____**  
*(If no, estimate overflow)*  
**Overflow volume** *(ft³): ________**  
**Overflow area** *(ft²): ________**

**Pool area** *(ft²): ________**
### Estimate Release Rate from Pool

- **Estimate release rate for undiked pool**  
  *Based on quantity spilled, LFA or LFB, and DF*
  
  **Release rate (lbs/min): ______**  
  
  [References: Section 7.2.3, Section 3.2.4 (mixtures) Equation 7-8 or 7-9]

- **Estimate release rate for diked pool (use pool area from previous section)**  
  *Based on pool area and LFA or LFB*
  
  **Release rate (lbs/min): ______**  
  
  [References: Sections 7.2.3, 3.2.2 Section 3.2.4 (mixtures) Equation 7-10 or 7-11]

- **Revise release rate for temperature**  
  *Apply appropriate TCF to release rate*
  
  **Revised release rate (lbs/min): ______**  
  
  [References: Sections 7.2.3, 3.2.5 Equation 3-11]

- **Revise release rate for release in building**  
  *Apply factor to release rate*
  
  **Release rate if outside (lbs/min): ______**  
  **Factor to account for enclosure: 0.05**  
  **Revised release rate (lbs/min): ______**  
  
  [References: Sections 7.2.3, 3.2.3]

- **Revise release rate for active mitigation technique**
  
  **Active mitigation technique used: ______**
  
  **Fractional release rate reduction by active technique: ______**  
  **Revised release rate (lb/min): ______**  
  
  [Reference: Section 7.2.3]

- **Compare liquid release rate and pool evaporation rate**  
  **Choose smaller release rate as release rate for analysis**
  
  **Release rate (lb/min): ______**  
  
  [Reference: Section 7.2.3]

### 3. Determine Distance to the Endpoint

- **Identify endpoint**  
  *Specified by rule*
  
  **Endpoint (mg/L): ______**  
  
  [Reference: Exhibit B-2]

- **Determine vapor density**
  
  **Dense: _____**
  **Neutrally buoyant: _____**  
  
  [Reference: Exhibit B-2]

- **Determine site topography**  
  *Rural and urban defined by rule*
  
  **Rural: _____**
  **Urban: _____**  
  
  [Reference: Section 2.1]
**WORKSHEET 5 (continued)**

<table>
<thead>
<tr>
<th>• Determine appropriate reference table of distances Based on release duration, vapor density, and topography</th>
<th><strong>Reference table used</strong> <em>(number): _____</em></th>
<th>Chapter 8 Reference Tables 14-25</th>
</tr>
</thead>
</table>
| • Find distance on reference table | **Release rate/endpoint** *(neutrally buoyant): _____*  
**Distance to endpoint** *(mi): _____* | Chapter 8 Reference Tables 14-25 |
### WORKSHEET 6
**ALTERNATIVE SCENARIO ANALYSIS FOR FLAMMABLE SUBSTANCE**

#### 1. Select Scenario

| Guidance Reference | Name: ____________________________ | CAS number: ______-_____-
|---------------------|--------------------------------------|-----------------------------|

- **Identify flammable substance**
- **Identify conditions of storage or processing of flammable substance**
  - Treat gases liquefied by refrigeration as liquids
  - Non-liquefied pressurized gas: ______
  - Gas liquefied under pressure: ______
  - Gas liquefied by refrigeration: ______
  - Liquid under atmospheric pressure: ______
  - Liquid under pressure greater than atmospheric: ______
  - Other (describe): ____________________________

- **Identify alternative scenario/**
  - Vapor cloud fire
  - Pool fire
  - BLEVEL/fireball
  - Vapor cloud explosion
  - Other (not covered by OCA Guidance)

#### 2. Determine Release Rate

**Determine Release Rate for Vapor Cloud Fire**

- **For gas releases and flashing liquid releases, see Worksheet 4**
  - Release rate (lbs/min): ______

  - Liquid release rate (lbs/min): ______
  - Liquid release duration (min): ______
  - Quantity in pool (lbs): ______
  - Release rate to air (lbs/min): ______

**Determine Pool Area for Pool Fire**

- **Estimate pool area: See Worksheet 5**
  - Quantity in pool (lbs): ______
  - Pool area (ft²): ______

---

**April 15, 1999**

E - 13
### Determine Quantity for BLEVE

Determine quantity in tank  
**Quantity (lbs): ____**

Section 10.3

### Determine Quantity for Vapor Cloud Explosion

Determine quantity in tank  
**Quantity (lbs): ____**

Section 10.4

### 3. Determine Distance to the Endpoint

- Identify endpoint suitable for scenario  
  - LFL  
  - 5 kW/m² for 40 seconds  
  - 1 psi overpressure

**Endpoint: ____**  
Chapter 6  
Exhibits C-2, C-3

### Determine Distance to LFL for Vapor Cloud Fire

- Determine vapor density  
  **Dense: ____**  
  **Neutrally buoyant: ____**

Exhibit B-2

- Determine site topography  
  **Rural: ____**  
  **Urban: ____**

Section 2.1

- Determine appropriate reference table of distances  
  *Based on vapor density and topography*

**Reference table used (number): ____**

Section 10.1  
Reference Tables 26-29

- Find distance on reference table  
  **Release rate/endpoint (neutrally buoyant): ____**  
  **Distance to LFL (mi): ____**

Section 10.1  
Reference Tables 26-29

### Determine Distance to Heat Radiation Endpoint for Pool Fire

- Calculate distance to 5 kW/m²  
  **PFF: ____**  
  **Pool area (ft²): ____**  
  **Distance (ft): ____**

Section 10.2  
Equation 10-1
WORKSHEET 6 (continued)

<table>
<thead>
<tr>
<th>Determine Distance to Heat Radiation Endpoint for BLEVE</th>
</tr>
</thead>
<tbody>
<tr>
<td>Determine distance for radiation from fireball equivalent to 5 kW/m² for 40 seconds</td>
</tr>
<tr>
<td>Quantity (lbs): _______</td>
</tr>
<tr>
<td>Distance (mi): _______</td>
</tr>
<tr>
<td>Section 10.3 Reference Table 30</td>
</tr>
</tbody>
</table>

<table>
<thead>
<tr>
<th>Determine Distance to Overpressure Endpoint For Vapor Cloud Explosion</th>
</tr>
</thead>
<tbody>
<tr>
<td>Determine distance to 1 psi</td>
</tr>
<tr>
<td>Quantity in cloud can be less than total quantity</td>
</tr>
<tr>
<td>Yield factor can be less than 10%</td>
</tr>
<tr>
<td>FFF: _______</td>
</tr>
<tr>
<td>Quantity flashed: _______</td>
</tr>
<tr>
<td>Yield factor: _______</td>
</tr>
<tr>
<td>Distance to 1 psi (mi): _______</td>
</tr>
<tr>
<td>Section 10.4 Exhibit C-2 Reference Table 13</td>
</tr>
</tbody>
</table>
APPENDIX F

CHEMICAL ACCIDENT PREVENTION PROVISIONS

As codified at 40 CFR part 68 as of July 1, 1998
PART 67—TECHNICAL SUPPORT DOCUMENT

APPENDIX A TO PART 67—TECHNICAL SUPPORT DOCUMENT

NOTE: EPA will make copies of appendix A available from: Director, Stationary Source Compliance Division, EN–341, 401 M Street, SW., Washington, DC 20460.

[54 FR 25259, June 20, 1989]

APPENDIX B TO PART 67—INSTRUCTION MANUAL

NOTE: EPA will make copies of appendix B available from: Director, Stationary Source Compliance Division, EN–341, 401 M Street, SW., Washington, DC 20460.

[54 FR 25259, June 20, 1989]

APPENDIX C TO PART 67—COMPUTER PROGRAM

NOTE: EPA will make copies of appendix C available from: Director, Stationary Source Compliance Division, EN–341, 401 M Street, SW., Washington, DC 20460.

[54 FR 25259, June 20, 1989]

PART 68—CHEMICAL ACCIDENT PREVENTION PROVISIONS

Subpart A—General

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68.1 Scope.
68.2 Stayed provisions.
68.3 Definitions.
68.10 Applicability.
68.12 General requirements.
68.15 Management.

Subpart B—Hazard Assessment

68.20 Applicability.
68.22 Offsite consequence analysis parameters.
68.25 Worst-case release scenario analysis.
68.30 Alternative release scenario analysis.
68.33 Defining offsite impacts—population.
68.36 Review and update.
68.39 Documentation.
68.42 Five-year accident history.

Subpart C—Program 2 Prevention Program

68.48 Safety information.
68.50 Hazard review.
68.52 Operating procedures.
68.54 Training.
68.56 Maintenance.
68.58 Compliance audits.
68.60 Incident investigation.

Subpart D—Program 3 Prevention Program

68.65 Process safety information.
68.67 Process hazard analysis.
68.69 Operating procedures.
68.71 Training.
68.73 Mechanical integrity.
68.75 Management of change.
68.77 Pre-startup review.
68.79 Compliance audits.
68.81 Incident investigation.
68.83 Employee participation.
68.85 Hot work permit.
68.87 Contractors.

Subpart E—Emergency Response

68.90 Applicability.
68.95 Emergency response program.

Subpart F—Regulated Substances for Accidental Release Prevention

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68.115 Threshold determination.
68.120 Petition process.
68.125 Exemptions.
68.130 List of substances.

Subpart G—Risk Management Plan

68.150 Submission.
68.155 Executive summary.
68.160 Registration.
68.165 Offsite consequence analysis.
68.168 Five-year accident history.
68.170 Prevention program/Program 2.
68.175 Prevention program/Program 3.
68.180 Emergency response program.
68.185 Certification.
68.190 Updates.

Subpart H—Other Requirements

68.200 Recordkeeping.
68.210 Availability of information to the public.
68.215 Permit content and air permitting authority or designated agency require-
m ents.
68.220 Audits.

APPENDIX A TO PART 68—TABLE OF TOXIC ENDPOINTS

AUTHORITY: 42 U.S.C. 7412(r), 7601(a)(1), 7661–7661f.

SOURCE: 59 FR 4493, Jan. 31, 1994, unless otherwise noted.
Subpart A—General
§ 68.1 Scope.
This part sets forth the list of regulated substances and thresholds, the petition process for adding or deleting substances to the list of regulated substances, the requirements for owners or operators of stationary sources concerning the prevention of accidental releases, and the State accidental release prevention programs approved under section 112(r). The list of substances, threshold quantities, and accident prevention regulations promulgated under this part do not limit in any way the general duty provisions promulgated under section 112(r)(1).

§ 68.2 Stayed provisions.
(a) Notwithstanding any other provision of this part, the effectiveness of the following provisions is stayed from March 2, 1994 to December 22, 1997.
(1) In Sec. 68.3, the definition of “stationary source,” to the extent that such definition includes naturally occurring hydrocarbon reservoirs or transportation subject to oversight or regulation under a state natural gas or hazardous liquid program for which the state has in effect a certification to DOT under 49 U.S.C. 60105;
(2) Section 68.115(b)(2) of this part, to the extent that such provision requires an owner or operator to treat as a regulated flammable substance:
(i) Gasoline, when in distribution or related storage for use as fuel for internal combustion engines;
(ii) Naturally occurring hydrocarbon mixtures prior to entry into a petroleum refining process unit or a natural gas processing plant. Naturally occurring hydrocarbon mixtures include any of the following: condensate, crude oil, field gas, and produced water, each as defined in paragraph (b) of this section;
(iii) Other mixtures that contain a regulated flammable substance and that do not have a National Fire Protection Association flammability hazard rating of 4, the definition of which is in the NFPA 704, Standard System for the Identification of the Fire Hazards of Materials, National Fire Protection Association, Quincy, MA, 1990, available from the National Fire Protection Association, 1 Batterymarch Park, Quincy, MA 02269-9101; and
(3) Section 68.130(a).
(b) From March 2, 1994 to December 22, 1997, the following definitions shall apply to the stayed provisions described in paragraph (a) of this section:
Condensate means hydrocarbon liquid separated from natural gas that condenses because of changes in temperature, pressure, or both, and remains liquid at standard conditions.
Crude oil means any naturally occurring, unrefined petroleum liquid.
Field gas means gas extracted from a production well before the gas enters a natural gas processing plant.
Natural gas processing plant means any processing site engaged in the extraction of natural gas liquids from field gas, fractionation of natural gas liquids to natural gas products, or both. A separator, dehydration unit, heater treater, sweetening unit, compressor, or similar equipment shall not be considered a “processing site” unless such equipment is physically located within a natural gas processing plant (gas plant) site.
Petroleum refining process unit means a process unit used in an establishment primarily engaged in petroleum refining as defined in the Standard Industrial Classification code for petroleum refining (2911) and used for the following: Producing transportation fuels (such as gasoline, diesel fuels, and jet fuels), heating fuels (such as kerosene, fuel gas distillate, and fuel oils), or lubricants; separating petroleum; or separating, cracking, reacting, or reforming intermediate petroleum streams. Examples of such units include, but are not limited to, petroleum based solvent units, alkylation units, catalytic hydrotreating, catalytic hydrocracking, catalytic cracking, crude distillation, lube oil processing, hydrogen production, isomerization, polymerization, thermal processes, and blending, sweetening, and treating processes. Petroleum refining process units include sulfur plants.
Produced water means water extracted from the earth from an oil or natural gas production well, or that is
§ 68.3 Definitions.

For the purposes of this part:

Accidental release means an unanticipated emission of a regulated substance or other extremely hazardous substance into the ambient air from a stationary source.

Act means the Clean Air Act as amended (42 U.S.C. 7401 et seq.)

Administrative controls mean written procedural mechanisms used for hazard control.

Administrator means the administrator of the U.S. Environmental Protection Agency.

AIChE/CCPS means the American Institute of Chemical Engineers/Center for Chemical Process Safety.

API means the American Petroleum Institute.

Article means a manufactured item, as defined under 29 CFR 1910.1200(b), that is formed to a specific shape or design during manufacture, that has end use functions dependent in whole or in part upon the shape or design during end use, and that does not release or otherwise result in exposure to a regulated substance under normal conditions of processing and use.

ASME means the American Society of Mechanical Engineers.

CAS means the Chemical Abstracts Service.

Catastrophic release means a major uncontrolled emission, fire, or explosion, involving one or more regulated substances that presents imminent and substantial endangerment to public health and the environment.

Classified information means “classified information” as defined in the Classified Information Procedures Act, 18 U.S.C. App. 3, section 1(a) as “any information or material that has been determined by the United States Government pursuant to an executive order, statute, or regulation, to require protection against unauthorized disclosure for reasons of national security.”

Condensate means hydrocarbon liquid separated from natural gas that condenses due to changes in temperature, pressure, or both, and remains liquid at standard conditions.

Covered process means a process that has a regulated substance present in more than a threshold quantity as determined under §68.115.

Crude oil means any naturally occurring, unrefined petroleum liquid.

Designated agency means the state, local, or Federal agency designated by the state under the provisions of §68.215(d).

DOT means the United States Department of Transportation.

Environmental receptor means natural areas such as national or state parks, forests, or monuments; officially designated wildlife sanctuaries, preserves, refuges, or areas; and Federal wilderness areas, that could be exposed at any time to toxic concentrations, radiant heat, or overpressures greater than or equal to the endpoints provided in §68.22(a), as a result of an accidental release and that can be identified on local U.S. Geological Survey maps.

Field gas means gas extracted from a production well before the gas enters a natural gas processing plant.

Hot work means work involving electric or gas welding, cutting, brazing, or similar flame or spark-producing operations.

Implementing agency means the state or local agency that obtains delegation for an accidental release prevention program under subpart E, 40 CFR part 63. The implementing agency may, but is not required to, be the state or local air permitting agency. If no state or local agency is granted delegation, EPA will be the implementing agency for that state.

Injury means any effect on a human that results either from direct exposure to toxic concentrations; radiant heat; or overpressures from accidental releases or from the direct consequences of a vapor cloud explosion (such as flying glass, debris, and other projectiles) from an accidental release and that requires medical treatment or hospitalization.

Major change means introduction of a new process, process equipment, or regulated substance, an alteration of process chemistry that results in any change to safe operating limits, or
Mechanical integrity means the process of ensuring that process equipment is fabricated from the proper materials of construction and is properly installed, maintained, and replaced to prevent failures and accidental releases.

Medical treatment means treatment, other than first aid, administered by a physician or registered professional personnel under standing orders from a physician.

Mitigation or mitigation system means specific activities, technologies, or equipment designed or deployed to capture or control substances upon loss of containment to minimize exposure of the public or the environment. Passive mitigation means equipment, devices, or technologies that function without human, mechanical, or other energy input. Active mitigation means equipment, devices, or technologies that need human, mechanical, or other energy input to function.

NFPA means the National Fire Protection Association.

Natural gas processing plant (gas plant) means any processing site engaged in the extraction of natural gas liquids from field gas, fractionation of mixed natural gas liquids to natural gas products, or both, classified as North American Industrial Classification System (NAICS) code 211112 (previously Standard Industrial Classification (SIC) code 1321).

Offsite means areas beyond the property boundary of the stationary source, and areas within the property boundary to which the public has routine and unrestricted access during or outside business hours.

OSHA means the U.S. Occupational Safety and Health Administration. Owner or operator means any person who owns, leases, operates, controls, or supervises a stationary source.

Petroleum refining process unit means a process unit used in an establishment primarily engaged in petroleum refining as defined in NAICS code 324112 (formerly SIC code 1321) and used for the following: Producing transportation fuels (such as gasoline, diesel fuels, and jet fuels), heating fuels (such as kerosene, fuel gas distillate, and fuel oils), or lubricants; Separating petroleum; or Separating, cracking, reacting, or reforming intermediate petroleum streams. Examples of such units include, but are not limited to, petroleum based solvent units, alkylation units, catalytic hydrotreating, catalytic hydroreforming, catalytic hydrocracking, catalytic reforming, catalytic cracking, crude distillation, lube oil processing, hydrogen production, isomerization, polymerization, thermal processes, and blending, sweetening, and treating processes. Petroleum refining process units include sulfur plants.

Population means the public.

Process means any activity involving a regulated substance including any use, storage, manufacturing, handling, or on-site movement of such substances, or combination of these activities. For the purposes of this definition, any group of vessels that are interconnected, or separate vessels that are located such that a regulated substance could be involved in a potential release, shall be considered a single process.

Produced water means water extracted from the earth from an oil or natural gas production well, or that is separated from oil or natural gas after extraction.

Public means any person except employees or contractors at the stationary source.

Public receptor means offsite residences, institutions (e.g., schools, hospitals), industrial, commercial, and office buildings, parks, or recreational areas inhabited or occupied by the public at any time without restriction by the stationary source where members of the public could be exposed to toxic concentrations, radiant heat, or overpressure, as a result of an accidental release.

Regulated substance is any substance listed pursuant to section 112(r)(3) of the Clean Air Act as amended, in §68.130.

Replacement in kind means a replacement that satisfies the design specifications.

RMP means the risk management plan required under subpart G of this part.
§ 68.10 Applicability.

(a) An owner or operator of a stationary source that has more than a threshold quantity of a regulated substance in a process, as determined under §68.115, shall comply with the requirements of this part no later than the latest of the following dates:

1. June 21, 1999;
2. Three years after the date on which a regulated substance is first listed under §68.130; or
3. The date on which a regulated substance is first present above a threshold quantity in a process.

(b) Program 1 eligibility requirements. A covered process is eligible for Program 1 requirements as provided in §68.12(b) if it meets all of the following requirements:

1. For the five years prior to the submission of an RMP, the process has not had an accidental release of a regulated substance where exposure to the substance, its reaction products, overpressure generated by an explosion involving the substance, or radiant heat generated by a fire involving the substance led to any of the following off-site:
   (i) Death;
   (ii) Injury; or
   (iii) Response or restoration activities for an exposure of an environmental receptor;

2. The distance to a toxic or flammable endpoint for a worst-case release assessment conducted under Subpart B and §68.25 is less than the distance to any public receptor, as defined in §68.30; and

3. Emergency response procedures have been coordinated between the stationary source and local emergency planning and response organizations.

(c) Program 2 eligibility requirements. A covered process is subject to Program 2 requirements if it does not meet the eligibility requirements of either paragraph (b) or paragraph (d) of this section.

(d) Program 3 eligibility requirements. A covered process is subject to Program 3 if the process does not meet the requirements of paragraph (b) of this section, and if either of the following conditions is met:
The process is in SIC code 2611, 2812, 2819, 2821, 2865, 2869, 2873, 2879, or 2911; or
(2) The process is subject to the OSHA process safety management standard, 29 CFR 1910.119.
(e) If at any time a covered process no longer meets the eligibility criteria of its Program level, the owner or operator shall comply with the requirements of the new Program level that applies to the process and update the RMP as provided in § 68.190.
(f) The provisions of this part shall not apply to an Outer Continental Shelf ("OCS") source, as defined in 40 CFR 55.2.
§ 68.12 General requirements.
(a) General requirements. The owner or operator of a stationary source subject to this part shall submit a single RMP, as provided in §§ 68.150 to 68.185. The RMP shall include a registration that reflects all covered processes.
(b) Program 1 requirements. In addition to meeting the requirements of paragraph (a) of this section, the owner or operator of a stationary source with a process eligible for Program 1, as provided in § 68.10(b), shall:
(1) Analyze the worst-case release scenario for the process(es), as provided in § 68.25; document that the nearest public receptor is beyond the distance to a toxic or flammable endpoint defined in § 68.22(a); and submit in the RMP the worst-case release scenario as provided in § 68.165;
(2) Complete the five-year accident history for the process as provided in § 68.42 of this part and submit it in the RMP as provided in § 68.168;
(3) Ensure that response actions have been coordinated with local emergency planning and response agencies; and
(4) Certify in the RMP the following: "Based on the criteria in 40 CFR 68.10, the distance to the specified endpoint for the worst-case accidental release scenario for the following process(es) is less than the distance to the nearest public receptor: [list process(es)]." Within the past five years, the process(es) has (have) had no accidental release that caused offsite impacts provided in the risk management program rule (40 CFR 68.10(b)(1)). No additional measures are necessary to prevent offsite impacts from accidental releases. In the event of fire, explosion, or a release of a regulated substance from the process(es), entry within the distance to the specified endpoints may pose a danger to public emergency responders. Therefore, public emergency responders should not enter this area except as arranged with the emergency contact indicated in the RMP. The undersigned certifies that, to the best of my knowledge, information, and belief, formed after reasonable inquiry, the information submitted is true, accurate, and complete. [Signature, title, date signed]."
(c) Program 2 requirements. In addition to meeting the requirements of paragraph (a) of this section, the owner or operator of a stationary source with a process subject to Program 2, as provided in § 68.10(c), shall:
(1) Develop and implement a management system as provided in § 68.15;
(2) Conduct a hazard assessment as provided in §§ 68.20 through 68.42;
(3) Implement the Program 2 prevention steps provided in §§ 68.48 through 68.60 or implement the Program 3 prevention steps provided in §§ 68.65 through 68.87;
(4) Develop and implement an emergency response program as provided in §§ 68.90 to 68.95; and
(5) Submit as part of the RMP the data on prevention program elements for Program 2 processes as provided in § 68.170.
(d) Program 3 requirements. In addition to meeting the requirements of paragraph (a) of this section, the owner or operator of a stationary source with a process subject to Program 3, as provided in § 68.10(d) shall:
(1) Develop and implement a management system as provided in § 68.15;
(2) Conduct a hazard assessment as provided in §§ 68.20 through 68.42;
(3) Implement the prevention requirements of §§ 68.65 through 68.87;
(4) Develop and implement an emergency response program as provided in §§ 68.90 to 68.95 of this part; and
(5) Submit as part of the RMP the data on prevention program elements.
§ 68.15 Management.

(a) The owner or operator of a stationary source with processes subject to Program 2 or Program 3 shall develop a management system to oversee the implementation of the risk management program elements.

(b) The owner or operator shall assign a qualified person or position that has the overall responsibility for the development, implementation, and integration of the risk management program elements.

(c) When responsibility for implementing individual requirements of this part is assigned to persons other than the person identified under paragraph (b) of this section, the names or positions of these people shall be documented and the lines of authority defined through an organization chart or similar document.

§ 68.20 Applicability.

The owner or operator of a stationary source subject to this part shall prepare a worst-case release scenario analysis as provided in §68.25 of this part and complete the five-year accident history as provided in §68.42. The owner or operator of a Program 2 and 3 process must comply with all sections in this subpart for these processes.

§ 68.22 Offsite consequence analysis parameters.

(a) Endpoints. For analyses of offsite consequences, the following endpoints shall be used:

1. Toxics. The toxic endpoints provided in appendix A of this part.

2. Flammables. The endpoints for flammables vary according to the scenarios studied:

   (i) Explosion. An overpressure of 1 psi.

   (ii) Radiant heat/exposure time. A radiant heat of 5 kW/m² for 40 seconds.

   (iii) Lower flammability limit. A lower flammability limit as provided in NFPA documents or other generally recognized sources.

   (b) Wind speed/atmospheric stability class. For the worst-case release analysis, the owner or operator shall use a wind speed of 1.5 meters per second and F atmospheric stability class. If the owner or operator can demonstrate that local meteorological data applicable to the stationary source show a higher minimum wind speed or less stable atmosphere at all times during the previous three years, these minimums may be used. For analysis of alternative scenarios, the owner or operator may use the typical meteorological conditions for the stationary source.

   (c) Ambient temperature/humidity. For worst-case release analysis of a regulated toxic substance, the owner or operator shall use the highest daily maximum temperature in the previous three years and average humidity for the site, based on temperature/humidity data gathered at the stationary source or at a local meteorological station; an owner or operator using the RMP Offsite Consequence Analysis Guidance may use 25°C and 50 percent humidity as values for these variables. For analysis of alternative scenarios, the owner or operator may use typical temperature/humidity data gathered at the stationary source or at a local meteorological station.

   (d) Height of release. The worst-case release of a regulated toxic substance shall be analyzed assuming a ground level (0 feet) release. For an alternative scenario analysis of a regulated toxic substance, release height may be determined by the release scenario.

   (e) Surface roughness. The owner or operator shall use either urban or rural topography, as appropriate. Urban means that there are many obstacles in the immediate area; obstacles include buildings or trees. Rural means there are no buildings in the immediate area and the terrain is generally flat and unobstructed.

   (f) Dense or neutrally buoyant gases. The owner or operator shall ensure that tables or models used for dispersion analysis of regulated toxic substances appropriately account for gas density.
§ 68.25 Worst-case release scenario analysis.

(a) The owner or operator shall analyze and report in the RMP:

(1) For Program 1 processes, one worst-case release scenario for each Program 1 process;

(2) For Program 2 and 3 processes:

(i) One worst-case release scenario that is estimated to create the greatest distance in any direction to an endpoint provided in appendix A of this part resulting from an accidental release of regulated toxic substances from covered processes under worst-case conditions defined in § 68.22;

(ii) One worst-case release scenario that is estimated to create the greatest distance in any direction to an endpoint defined in § 68.22(a) resulting from an accidental release of regulated flammable substances from covered processes under worst-case conditions defined in § 68.22; and

(iii) Additional worst-case release scenarios for a hazard class if a worst-case release from another covered process at the stationary source potentially affects public receptors different from those potentially affected by the worst-case release scenario developed under paragraphs (a)(2)(i) or (a)(2)(ii) of this section.

(b) Determination of worst-case release quantity. The worst-case release quantity shall be the greater of the following:

(1) For substances in a vessel, the greatest amount held in a single vessel, taking into account administrative controls that limit the maximum quantity; or

(2) For substances in pipes, the greatest amount in a pipe, taking into account administrative controls that limit the maximum quantity.

(c) Worst-case release scenario—toxic gases. (1) For regulated toxic substances that are normally gases at ambient temperature and handled as a gas or as a liquid under pressure, the owner or operator shall assume that the quantity in the vessel or pipe, as determined under paragraph (b) of this section, is released as a gas over 10 minutes. The release rate shall be assumed to be the total quantity divided by 10 unless passive mitigation systems are in place.

(2) For gases handled as refrigerated liquids at ambient pressure:

(i) If the released substance is not contained by passive mitigation systems or if the contained pool would have a depth of 1 cm or less, the owner or operator shall assume that the substance is released as a gas in 10 minutes;

(ii) If the released substance is contained by passive mitigation systems in a pool with a depth greater than 1 cm, the owner or operator may assume that the quantity in the vessel or pipe, as determined under paragraph (b) of this section, is spilled instantaneously to form a liquid pool. The volatilization rate (release rate) shall be calculated at the boiling point of the substance and at the conditions specified in paragraph (d) of this section.

(d) Worst-case release scenario—toxic liquids. (1) For regulated toxic substances that are normally liquids at ambient temperature, the owner or operator shall assume that the quantity in the vessel or pipe, as determined under paragraph (b) of this section, is spilled instantaneously to form a liquid pool.

(i) The surface area of the pool shall be determined by assuming that the liquid spreads to 1 centimeter deep unless passive mitigation systems are in place that serve to contain the spill and limit the surface area. Where passive mitigation is in place, the surface area of the contained liquid shall be used to calculate the volatilization rate.

(ii) If the release would occur onto a surface that is not paved or smooth, the owner or operator may take into account administrative controls that limit the maximum quantity.
§ 68.28

account the actual surface characteristics.

(2) The volatilization rate shall account for the highest daily maximum temperature occurring in the past three years, the temperature of the substance in the vessel, and the concentration of the substance if the liquid spilled is a mixture or solution.

(3) The rate of release to air shall be determined from the volatilization rate of the liquid pool. The owner or operator may use the methodology in the RMP Offsite Consequence Analysis Guidance or any other publicly available techniques that account for the modeling conditions and are recognized by industry as applicable as part of current practices. Proprietary models that account for the modeling conditions may be used provided the owner or operator allows the implementing agency access to the model and describes model features and differences from publicly available models to local emergency planners upon request.

(e) Worst-case release scenario—flammables. The owner or operator shall assume that the quantity of the substance, as determined under paragraph (b) of this section, vaporizes resulting in a vapor cloud explosion. A yield factor of 10 percent of the available energy released in the explosion shall be used to determine the distance to the explosion endpoint if the model used is based on TNT-equivalent methods.

(f) Parameters to be applied. The owner or operator shall use the parameters defined in §68.22 to determine distance to the endpoints. The owner or operator may use the methodology provided in the RMP Offsite Consequence Analysis Guidance or any commercially or publicly available air dispersion modeling techniques, provided the techniques account for the modeling conditions and are recognized by industry as applicable as part of current practices. Proprietary models that account for the modeling conditions may be used provided the owner or operator allows the implementing agency access to the model and describes model features and differences from publicly available models to local emergency planners upon request.

(g) Consideration of passive mitigation. Passive mitigation systems may be considered for the analysis of worst case provided that the mitigation system is capable of withstanding the release event triggering the scenario and would still function as intended.

(h) Factors in selecting a worst-case scenario. Notwithstanding the provisions of paragraph (b) of this section, the owner or operator shall select as the worst case for flammable regulated substances or the worst case for regulated toxic substances, a scenario based on the following factors if such a scenario would result in a greater distance to an endpoint defined in §68.22(a) beyond the stationary source boundary than the scenario provided under paragraph (b) of this section:

(1) Smaller quantities handled at higher process temperature or pressure; and

(2) Proximity to the boundary of the stationary source.

§ 68.28 Alternative release scenario analysis.

(a) The number of scenarios. The owner or operator shall identify and analyze at least one alternative release scenario for each regulated toxic substance held in a covered process(es) and at least one alternative release scenario to represent all flammable substances held in covered processes.

(b) Scenarios to consider. (1) For each scenario required under paragraph (a) of this section, the owner or operator shall select a scenario:

(i) That is more likely to occur than the worst-case release scenario under §68.25; and

(ii) That will reach an endpoint offsite, unless no such scenario exists.

(2) Release scenarios considered should include, but are not limited to, the following, where applicable:

(i) Transfer hose releases due to splits or sudden hose uncoupling;

(ii) Process piping releases from failures at flanges, joints, welds, valves and valve seals, and drains or bleeds;

(iii) Process vessel or pump releases due to cracks, seal failure, or drain, bleed, or plug failure;

(iv) Vessel overfilling and spill, or overpressurization and venting through relief valves or rupture disks; and
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(v) Shipping container mishandling and breakage or puncturing leading to a spill.

(c) Parameters to be applied. The owner or operator shall use the appropriate parameters defined in §68.22 to determine distance to the endpoints. The owner or operator may use either the methodology provided in the RMP Offsite Consequence Analysis Guidance or any commercially or publicly available air dispersion modeling techniques, provided the techniques account for the specified modeling conditions and are recognized by industry as applicable as part of current practices. Proprietary models that account for the modeling conditions may be used provided the owner or operator allows the implementing agency access to the model and describes model features and differences from publicly available models to local emergency planners upon request.

(d) Consideration of mitigation. Active and passive mitigation systems may be considered provided they are capable of withstanding the event that triggered the release and would still be functional.

(e) Factors in selecting scenarios. The owner or operator shall consider the following in selecting alternative release scenarios:
(1) The five-year accident history provided in §68.42; and
(2) Failure scenarios identified under §68.50 or §68.67.

§ 68.33 Defining offsite impacts—environment.

(a) The owner or operator shall list in the RMP environmental receptors within a circle with its center at the point of the release and a radius determined by the distance to the endpoint defined in §68.22(a) of this part.

(b) Data sources acceptable. The owner or operator may rely on information provided on local U.S. Geological Survey maps or on any data source containing U.S.G.S. data to identify environmental receptors.

68.36 Review and update.

(a) The owner or operator shall review and update the offsite consequence analyses at least once every five years.

(b) If changes in processes, quantities stored or handled, or any other aspect of the stationary source might reasonably be expected to increase or decrease the distance to the endpoint by a factor of two or more, the owner or operator shall complete a revised analysis within six months of the change and submit a revised risk management plan as provided in §68.190.

§ 68.39 Documentation.

The owner or operator shall maintain the following records on the offsite consequence analyses:

(a) For worst-case scenarios, a description of the vessel or pipeline and substance selected as worst case, assumptions and parameters used, and the rationale for selection; assumptions shall include use of any administrative controls and any passive mitigation that were assumed to limit the quantity that could be released. Documentation shall include the anticipated effect of the controls and mitigation on the release quantity and rate.

(b) For alternative release scenarios, a description of the scenarios identified, assumptions and parameters used, and the rationale for the selection of specific scenarios; assumptions shall include use of any administrative controls and any mitigation that were assumed to limit the quantity that could
be released. Documentation shall include the effect of the controls and mitigation on the release quantity and rate.

(c) Documentation of estimated quantity released, release rate, and duration of release.

(d) Methodology used to determine distance to endpoints.

(e) Data used to estimate population and environmental receptors potentially affected.

§ 68.42 Five-year accident history.

(a) The owner or operator shall include in the five-year accident history all accidental releases from covered processes that resulted in deaths, injuries, or significant property damage on site, or known offsite deaths, injuries, evacuations, sheltering in place, property damage, or environmental damage.

(b) Data required. For each accidental release included, the owner or operator shall report the following information:

(1) Date, time, and approximate duration of the release;

(2) Chemical(s) released;

(3) Estimated quantity released in pounds;

(4) The type of release event and its source;

(5) Weather conditions, if known;

(6) On-site impacts;

(7) Known offsite impacts;

(8) Initiating event and contributing factors if known;

(9) Whether offsite responders were notified if known; and

(10) Operational or process changes that resulted from investigation of the release.

(c) Level of accuracy. Numerical estimates may be provided to two significant digits.

Subpart C—Program 2 Prevention Program

Source: 61 FR 31721, June 20, 1996, unless otherwise noted.

§ 68.48 Safety information.

(a) The owner or operator shall compile and maintain the following up-to-date safety information related to the regulated substances, processes, and equipment:

1. Material Safety Data Sheets that meet the requirements of 29 CFR 1910.1200(g);

2. Maximum intended inventory of equipment in which the regulated substances are stored or processed;

3. Safe upper and lower temperatures, pressures, flows, and compositions;

4. Equipment specifications; and

5. Codes and standards used to design, build, and operate the process.

(b) The owner or operator shall ensure that the process is designed in compliance with recognized and generally accepted good engineering practices. Compliance with Federal or state regulations that address industry-specific safe design or with industry-specific design codes and standards may be used to demonstrate compliance with this paragraph.

(c) The owner or operator shall update the safety information if a major change occurs that makes the information inaccurate.

§ 68.50 Hazard review.

(a) The owner or operator shall conduct a review of the hazards associated with the regulated substances, process, and procedures. The review shall identify the following:

1. The hazards associated with the process and regulated substances;

2. Opportunities for equipment malfunctions or human errors that could cause an accidental release;

3. The safeguards used or needed to control the hazards or prevent equipment malfunction or human error; and

4. Any steps used or needed to detect or monitor releases.

(b) The owner or operator may use checklists developed by persons or organizations knowledgeable about the process and equipment as a guide to conducting the review. For processes designed to meet industry standards or Federal or state design rules, the hazard review shall, by inspecting all equipment, determine whether the process is designed, fabricated, and operated in accordance with the applicable standards or rules.
(c) The owner or operator shall document the results of the review and ensure that problems identified are resolved in a timely manner.
(d) The review shall be updated at least once every five years. The owner or operator shall also conduct reviews whenever a major change in the process occurs; all issues identified in the review shall be resolved before startup of the changed process.

§ 68.52 Operating procedures.
(a) The owner or operator shall prepare written operating procedures that provide clear instructions or steps for safely conducting activities associated with each covered process consistent with the safety information for that process. Operating procedures or instructions provided by equipment manufacturers or developed by persons or organizations knowledgeable about the process and equipment may be used as a basis for a stationary source's operating procedures.
(b) The procedures shall address the following:
(1) Initial startup;
(2) Normal operations;
(3) Temporary operations;
(4) Emergency shutdown and operations;
(5) Normal shutdown;
(6) Startup following a normal or emergency shutdown or a major change that requires a hazard review;
(7) Consequences of deviations and steps required to correct or avoid deviations; and
(8) Equipment inspections.
(c) The owner or operator shall ensure that the operating procedures are updated, if necessary, whenever a major change occurs and prior to startup of the changed process.

§ 68.54 Training.
(a) The owner or operator shall prepare and implement procedures to ensure that each employee presently operating a process, and each employee newly assigned to a covered process have been trained or tested competent in the operating procedures provided in § 68.52 that pertain to their duties. For those employees already operating a process on June 21, 1999, the owner or operator may certify in writing that the employee has the required knowledge, skills, and abilities to safely carry out the duties and responsibilities as provided in the operating procedures.
(b) Refresher training. Refresher training shall be provided at least every three years, and more often if necessary, to each employee operating a process to ensure that the employee understands and adheres to the current operating procedures of the process. The owner or operator, in consultation with the employees operating the process, shall determine the appropriate frequency of refresher training.
(c) The owner or operator may use training conducted under Federal or state regulations or under industry-specific standards or codes or training conducted by covered process equipment vendors to demonstrate compliance with this section to the extent that the training meets the requirements of this section.
(d) The owner or operator shall ensure that operators are trained in any updated or new procedures prior to startup of a process after a major change.

§ 68.56 Maintenance.
(a) The owner or operator shall prepare and implement procedures to maintain the on-going mechanical integrity of the process equipment. The owner or operator may use procedures or instructions provided by covered process equipment vendors or procedures in Federal or state regulations or industry codes as the basis for stationary source maintenance procedures.
(b) The owner or operator shall train or cause to be trained each employee involved in maintaining the on-going mechanical integrity of the process. To ensure that the employee can perform the job tasks in a safe manner, each such employee shall be trained in the hazards of the process, in how to avoid or correct unsafe conditions, and in the procedures applicable to the employee's job tasks.
(c) Any maintenance contractor shall ensure that each contract maintenance employee is trained to perform the maintenance procedures developed under paragraph (a) of this section.
§ 68.58 Compliance audits.
(a) The owner or operator shall certify that they have evaluated compliance with the provisions of this subpart at least every three years to verify that the procedures and practices developed under the rule are adequate and are being followed.
(b) The compliance audit shall be conducted by at least one person knowledgeable in the process.
(c) The owner or operator shall develop a report of the audit findings.
(d) The owner or operator shall promptly determine and document an appropriate response to each of the findings of the compliance audit and document that deficiencies have been corrected.
(e) The owner or operator shall retain the two (2) most recent compliance audit reports. This requirement does not apply to any compliance audit report that is more than five years old.

§ 68.60 Incident investigation.
(a) The owner or operator shall investigate each incident which resulted in, or could reasonably have resulted in, a catastrophic release.
(b) An incident investigation shall be initiated as promptly as possible, but not later than 48 hours following the incident.
(c) A summary shall be prepared at the conclusion of the investigation which includes at a minimum:
(1) Date of incident;
(2) Date investigation began;
(3) A description of the incident;
(4) The factors that contributed to the incident; and,
(5) Any recommendations resulting from the investigation.
(d) The owner or operator shall promptly address and resolve the investigation findings and recommendations. Resolutions and corrective actions shall be documented.
(e) The findings shall be reviewed with all affected personnel whose job tasks are affected by the findings.
(f) Investigation summaries shall be retained for five years.

Subpart D—Program 3 Prevention Program

SOURCE: 61 FR 31722, June 20, 1996, unless otherwise noted.

§ 68.65 Process safety information.
(a) In accordance with the schedule set forth in § 68.67, the owner or operator shall complete a compilation of written process safety information before conducting any process hazard analysis required by the rule. The compilation of written process safety information is to enable the owner or operator and the employees involved in operating the process to identify and understand the hazards posed by those processes involving regulated substances. This process safety information shall include information pertaining to the hazards of the regulated substances used or produced by the process, information pertaining to the technology of the process, and information pertaining to the equipment in the process.
(b) Information pertaining to the hazards of the regulated substances in the process. This information shall consist of at least the following:
(1) Toxicity information;
(2) Permissible exposure limits;
(3) Physical data;
(4) Reactivity data;
(5) Corrosivity data;
(6) Thermal and chemical stability data; and
(7) Hazardous effects of inadvertent mixing of different materials that could foreseeably occur.

NOTE TO PARAGRAPH (b): Material Safety Data Sheets meeting the requirements of 29 CFR 1910.1200(g) may be used to comply with this requirement to the extent they contain the information required by this subparagraph.
(c) Information pertaining to the technology of the process.
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§ 68.67 Process hazard analysis.

(a) The owner or operator shall perform an initial process hazard analysis (hazard evaluation) on processes covered by this part. The process hazard analysis shall be appropriate to the complexity of the process and shall identify, evaluate, and control the hazards involved in the process. The owner or operator shall determine and document the priority order for conducting process hazard analyses based on a rationale which includes such considerations as extent of the process hazards, number of potentially affected employees, age of the process, and operating history of the process. The process hazard analysis shall be conducted as soon as possible, but not later than June 21, 1999. Process hazard analyses completed to comply with 29 CFR 1910.119(e) are acceptable as initial process hazard analyses. These process hazard analyses shall be updated and revalidated, based on their completion date.

(b) The owner or operator shall use one or more of the following methodologies that are appropriate to determine and evaluate the hazards of the process being analyzed.

(i) What-if;
(ii) Checklist;
(iii) What-if/Checklist;
(iv) Hazard and Operability Study (HAZOP);
(v) Failure Mode and Effects Analysis (FMEA);
(vi) Fault Tree Analysis; or
(vii) An appropriate equivalent methodology.

(c) The process hazard analysis shall address:

(i) The hazards of the process;
(ii) The identification of any previous incident which had a likely potential for catastrophic consequences;
(iii) Engineering and administrative controls applicable to the hazards and their interrelationships such as appropriate application of detection methodologies to provide early warning of releases. (Acceptable detection methods might include process monitoring and control instrumentation with alarms, and detection hardware such as hydrocarbon sensors.);
(iv) Consequences of failure of engineering and administrative controls;
(v) Stationary source siting;
(vi) Human factors; and
(vii) A qualitative evaluation of a range of the possible safety and health effects of failure of controls.

(d) The process hazard analysis shall be performed by a team with expertise in engineering and process operations, and the team shall include at least one employee who has experience and knowledge specific to the process being

(1) Information concerning the technology of the process shall include at least the following:

(i) A block flow diagram or simplified process flow diagram;
(ii) Process chemistry;
(iii) Maximum intended inventory;
(iv) Safe upper and lower limits for such items as temperatures, pressures, flows or compositions; and,
(v) An evaluation of the consequences of deviations.

(2) Where the original technical information no longer exists, such information may be developed in conjunction with the process hazard analysis in sufficient detail to support the analysis.

(d) Information pertaining to the equipment in the process.

(1) Information pertaining to the equipment in the process shall include:

(i) Materials of construction;
(ii) Piping and instrument diagrams (P&ID’s);
(iii) Electrical classification;
(iv) Relief system design and design basis;
(v) Ventilation system design;
(vi) Design codes and standards employed;
(vii) Material and energy balances for processes built after June 21, 1999; and
(viii) Safety systems (e.g. interlocks, detection or suppression systems).

(2) The owner or operator shall document that equipment complies with recognized and generally accepted good engineering practices.

(3) For existing equipment designed and constructed in accordance with codes, standards, or practices that are no longer in general use, the owner or operator shall determine and document that the equipment is designed, maintained, inspected, tested, and operating in a safe manner.
§ 68.69 Operating procedures.

(a) The owner or operator shall develop and implement written operating procedures that provide clear instructions for safely conducting activities involved in each covered process consistent with the process safety information and shall address at least the following elements.

(1) Steps for each operating phase:
   (i) Initial startup;
   (ii) Normal operations;
   (iii) Temporary operations;
   (iv) Emergency shutdown including the conditions under which emergency shutdown is required, and the assignment of shutdown responsibility to qualified operators to ensure that emergency shutdown is executed in a safe and timely manner.
   (v) Emergency operations;
   (vi) Normal shutdown; and,
   (vii) Startup following a turnaround, or after an emergency shutdown.

(2) Operating limits:
   (i) Consequences of deviation; and
   (ii) Steps required to correct or avoid deviation.

(3) Safety and health considerations:
   (i) Properties of, and hazards presented by, the chemicals used in the process;
   (ii) Precautions necessary to prevent exposure, including engineering controls, administrative controls, and personal protective equipment;
   (iii) Control measures to be taken if physical contact or airborne exposure occurs;
   (iv) Quality control for raw materials and control of hazardous chemical inventory levels; and,
   (v) Any special or unique hazards.

(4) Safety systems and their functions.

(b) Operating procedures shall be readily accessible to employees who work in or maintain a process.

(c) The operating procedures shall be reviewed as often as necessary to assure that they reflect current operating practice, including changes that result from changes in process chemicals, technology, and equipment, and changes to stationary sources. The owner or operator shall certify annually that these operating procedures are current and accurate.

(d) The owner or operator shall develop and implement safe work practices to provide for the control of hazards during operations such as lockout/tagout; confined space entry; opening process equipment or piping; and control over entrance into a stationary source by maintenance, contractor, laboratory, or other support personnel. These safe work practices shall apply to employees and contractor employees.

§ 68.71 Training.

(a) Initial training. (1) Each employee presently involved in operating a process, and each employee before being involved in operating a newly assigned process, shall be trained in an overview
of the process and in the operating pro­
cedures as specified in §68.69. The
training shall include emphasis on the
specific safety and health hazards,
emergency operations including shut­
down, and safe work practices applica­
table to the employee's job tasks.
(2) In lieu of initial training for those
employees already involved in operat­
ing a process on June 21, 1999 an owner
or operator may certify in writing that
the employee has the required knowl­
dge, skills, and abilities to safely
carry out the duties and responsibil­
ities as specified in the operating pro­
cedures.
(b) Refresher training. Refresher train­
ing shall be provided at least every
three years, and more often if nec­
essary, to each employee involved in
operating a process to assure that the
employee understands and adheres to
the current operating procedures of the
process. The owner or operator, in con­
sultation with the employees involved
in operating the process, shall deter­
mine the appropriate frequency of re­
resher training.
(c) Training documentation. The owner
or operator shall ascertain that each
employee involved in operating a proc­
ess has received and understood the
training required by this paragraph.
The owner or operator shall prepare a
record which contains the identity of
the employee, the date of training, and
the means used to verify that the em­
ployee understood the training.
§ 68.73 Mechanical integrity.
(a) Application. Paragraphs (b) through (f)
of this section apply to the
following process equipment:
(1) Pressure vessels and storage
tanks;
(2) Piping systems (including piping
components such as valves);
(3) Relief and vent systems and de­
tices;
(4) Emergency shutdown systems;
(5) Controls (including monitoring
devices and sensors, alarms, and inter­
locks) and,
(6) Pumps.
(b) Written procedures. The owner or
operator shall establish and implement
written procedures to maintain the on­
going integrity of process equipment.
(c) Training for process maintenance
activities. The owner or operator shall
train each employee involved in main­
taining the on-going integrity of proc­
ess equipment in an overview of that
process and its hazards and in the pro­
cedures applicable to the employee's
job tasks to assure that the employee
can perform the job tasks in a safe
manner.
(d) Inspection and testing. (1) Inspec­
tions and tests shall be performed on
process equipment.
(2) Inspection and testing procedures
shall follow recognized and generally
accepted good engineering practices.
(3) The frequency of inspections and
tests of process equipment shall be con­
sistent with applicable manufacturers' 
recommendations and good engineering
practices, and more frequently if deter­
mined to be necessary by prior operat­
ing experience.
(4) The owner or operator shall docu­
ment each inspection and test that has
been performed on process equipment.
The documentation shall identify the
date of the inspection or test, the name
of the person who performed the in­
spection or test, the serial number or
other identifier of the equipment on
which the inspection or test was per­
formed, a description of the inspection
or test performed, and the results of
the inspection or test.
(e) Equipment deficiencies. The owner
or operator shall correct deficiencies in
equipment that are outside acceptable
limits (defined by the process safety in­
formation in §68.65) before further use
or in a safe and timely manner when
necessary means are taken to assure
safe operation.
(f) Quality assurance. (1) In the con­
struction of new plants and equipment,
the owner or operator shall assure that
equipment as it is fabricated is suit­
able for the process application for
which they will be used.
(2) Appropriate checks and inspec­
tions shall be performed to assure that
equipment is installed properly and
consistent with design specifications
and the manufacturer's instructions.
(3) The owner or operator shall as­
sure that maintenance materials, spare
parts and equipment are suitable for
the process application for which they
will be used.
§ 68.75 Management of change.

(a) The owner or operator shall establish and implement written procedures to manage changes (except for “replacements in kind”) to process chemicals, technology, equipment, and procedures; and, changes to stationary sources that affect a covered process.

(b) The procedures shall assure that the following considerations are addressed prior to any change:

(1) The technical basis for the proposed change;

(2) Impact of change on safety and health;

(3) Modifications to operating procedures;

(4) Necessary time period for the change; and,

(5) Authorization requirements for the proposed change.

(c) Employees involved in operating a process and maintenance and contract employees whose job tasks will be affected by a change in the process shall be informed of, and trained in, the change prior to start-up of the process or affected part of the process.

(d) If a change covered by this paragraph results in a change in the process safety information required by §68.65 of this part, such information shall be updated accordingly.

(e) If a change covered by this paragraph results in a change in the operating procedures or practices required by §68.69, such procedures or practices shall be updated accordingly.

§ 68.77 Pre-startup review.

(a) The owner or operator shall perform a pre-startup safety review for new stationary sources and for modified stationary sources when the modification is significant enough to require a change in the process safety information.

(b) The pre-startup safety review shall confirm that prior to the introduction of regulated substances to a process:

(1) Construction and equipment is in accordance with design specifications;

(2) Safety, operating, maintenance, and emergency procedures are in place and are adequate;

(3) For new stationary sources, a process hazard analysis has been performed and recommendations have been resolved or implemented before startup; and modified stationary sources meet the requirements contained in management of change, §68.75.

(4) Training of each employee involved in operating a process has been completed.

§ 68.79 Compliance audits.

(a) The owner or operator shall certify that they have evaluated compliance with the provisions of this section at least every three years to verify that the procedures and practices developed under the standard are adequate and are being followed.

(b) The compliance audit shall be conducted by at least one person knowledgeable in the process.

(c) A report of the findings of the audit shall be developed.

(d) The owner or operator shall promptly determine and document an appropriate response to each of the findings of the compliance audit, and document that deficiencies have been corrected.

(e) The owner or operator shall retain the two (2) most recent compliance audit reports.

§ 68.81 Incident investigation.

(a) The owner or operator shall investigate each incident which resulted in, or could reasonably have resulted in a catastrophic release of a regulated substance.

(b) An incident investigation shall be initiated as promptly as possible, but not later than 48 hours following the incident.

(c) An incident investigation team shall be established and consist of at least one person knowledgeable in the process involved, including a contract employee if the incident involved work of the contractor, and other persons with appropriate knowledge and experience to thoroughly investigate and analyze the incident.

(d) A report shall be prepared at the conclusion of the investigation which includes at a minimum:

(1) Date of incident;

(2) Date investigation began;

(3) A description of the incident;

(4) The factors that contributed to the incident; and,
§ 68.83 Employee participation.

(a) The owner or operator shall develop a written plan of action regarding the implementation of the employee participation required by this section.

(b) The owner or operator shall consult with employees and their representatives on the conduct and development of process hazards analyses and on the development of the other elements of process safety management in this rule.

(c) The owner or operator shall provide to employees and their representatives access to process hazard analyses and to all other information required to be developed under this rule.

§ 68.85 Hot work permit.

(a) The owner or operator shall issue a hot work permit for hot work operations conducted on or near a covered process.

(b) The permit shall document that the fire prevention and protection requirements in 29 CFR 1910.252(a) have been implemented prior to beginning the hot work operations; it shall indicate the date(s) authorized for hot work; and identify the object on which hot work is to be performed. The permit shall be kept on file until completion of the hot work operations.

§ 68.87 Contractors.

(a) Application. This section applies to contractors performing maintenance or repair, turnaround, major renovation, or specialty work on or adjacent to a covered process. It does not apply to contractors providing incidental services which do not influence process safety, such as janitorial work, food and drink services, laundry, delivery or other supply services.

(b) Owner or operator responsibilities.

(1) The owner or operator, when selecting a contractor, shall obtain and evaluate information regarding the contract owner or operator’s safety performance and programs.

(2) The owner or operator shall inform contract owner or operator of the known potential fire, explosion, or toxic release hazards related to the contractor’s work and the process.

(3) The owner or operator shall explain to the contract owner or operator the applicable provisions of subpart E of this part.

(4) The owner or operator shall develop and implement safe work practices consistent with §68.69(d), to control the entrance, presence, and exit of the contract owner or operator and contract employees in covered process areas.

(5) The owner or operator shall periodically evaluate the performance of the contract owner or operator in fulfilling their obligations as specified in paragraph (c) of this section.

(c) Contract owner or operator responsibilities.

(1) The contract owner or operator shall assure that each contract employee is trained in the work practices necessary to safely perform his/her job.

(2) The contract owner or operator shall assure that each contract employee is instructed in the known potential fire, explosion, or toxic release hazards related to his/her job and the process, and the applicable provisions of the emergency action plan.

(3) The contract owner or operator shall document that each contract employee has received and understood the training required by this section. The contract owner or operator shall prepare a record which contains the identity of the contract employee, the date of training, and the means used to verify that the employee understood the training.

(4) The contract owner or operator shall assure that each contract employee follows the safety rules of the stationary source including the safe work practices required by §68.69(d).

(5) The contract owner or operator shall advise the owner or operator of...
§ 68.90  Applicability.

(a) Except as provided in paragraph (b) of this section, the owner or operator of a stationary source with Program 2 and Program 3 processes shall comply with the requirements of § 68.95.

(b) The owner or operator of stationary source whose employees will not respond to accidental releases of regulated substances need not comply with § 68.95 of this part provided that they meet the following:

(1) For stationary sources with any regulated toxic substance held in a process above the threshold quantity, the stationary source is included in the community emergency response plan developed under 42 U.S.C. 11003;

(2) For stationary sources with only regulated flammable substances held in a process above the threshold quantity, the owner or operator has coordinated response actions with the local fire department; and

(3) Appropriate mechanisms are in place to notify emergency responders when there is a need for a response.

§ 68.95  Emergency response program.

(a) The owner or operator shall develop and implement an emergency response program for the purpose of protecting public health and the environment. Such program shall include the following elements:

(1) An emergency response plan, which shall be maintained at the stationary source and contain at least the following elements:

(  i ) Procedures for informing the public and local emergency response agencies about accidental releases;

(  ii ) Documentation of proper first-aid and emergency medical treatment necessary to treat accidental human exposures; and

(  iii ) Procedures and measures for emergency response after an accidental release of a regulated substance;

(2) Procedures for the use of emergency response equipment and for its inspection, testing, and maintenance;

(3) Training for all employees in relevant procedures; and

(4) Procedures to review and update, as appropriate, the emergency response plan to reflect changes at the stationary source and ensure that employees are informed of changes.

(b) A written plan that complies with other Federal contingency plan regulations or is consistent with the approach in the National Response Team's Integrated Contingency Plan Guidance (''One Plan'') and that, among other matters, includes the elements provided in paragraph (a) of this section, shall satisfy the requirements of this section if the owner or operator also complies with paragraph (c) of this section.

(c) The emergency response plan developed under paragraph (a)(1) of this section shall be coordinated with the community emergency response plan developed under 42 U.S.C. 11003. Upon request of the local emergency planning committee or emergency response officials, the owner or operator shall promptly provide to the local emergency response officials information necessary for developing and implementing the community emergency response plan.

Subpart F—Regulated Substances for Accidental Release Prevention

§ 68.100  Purpose.

This subpart designates substances to be listed under section 112(r)(3), (4), and (5) of the Clean Air Act, as amended, identifies their threshold quantities, and establishes the requirements for petitioning to add or delete substances from the list.

§ 68.115  Threshold determination.

(a) A threshold quantity of a regulated substance listed in § 68.130 is
present at a stationary source if the total quantity of the regulated substance contained in a process exceeds the threshold.

(b) For the purposes of determining whether more than a threshold quantity of a regulated substance is present at the stationary source, the following exemptions apply:

(1) Concentrations of a regulated toxic substance in a mixture. If a regulated substance is present in a mixture and the concentration of the substance is below one percent by weight of the mixture, the amount of the substance in the mixture need not be considered when determining whether more than a threshold quantity is present at the stationary source. Except for oleum, toluene 2,4-diisocyanate, toluene 2,6-diisocyanate, and toluene diisocyanate (unspecified isomer), if the concentration of the regulated substance in the mixture is one percent or greater by weight, but the owner or operator can demonstrate that the partial pressure of the regulated substance in the mixture (solution) under handling or storage conditions in any portion of the process is less than 10 millimeters of mercury (mm Hg), the amount of the substance in the mixture in that portion of the process need not be considered when determining whether more than a threshold quantity is present at the stationary source. The owner or operator shall document this partial pressure measurement or estimate.

(2) Concentrations of a regulated flammable substance in a mixture. (i) General provision. If a regulated substance is present in a mixture and the concentration of the substance is below one percent by weight of the mixture, the mixture need not be considered when determining whether more than a threshold quantity of the regulated substance is present at the stationary source. Except as provided in paragraph (b)(2) (ii) and (iii) of this section, if the concentration of the substance is one percent or greater by weight of the mixture, then, for purposes of determining whether a threshold quantity is present at the stationary source, the entire weight of the mixture shall be treated as the regulated substance unless the owner or operator can demonstrate that the mixture itself does not have a National Fire Protection Association flammability hazard rating of 4. The demonstration shall be in accordance with the definition of flammability hazard rating 4 in the NFPA 704, Standard System for the Identification of the Hazards of Materials for Emergency Response, National Fire Protection Association, Quincy, MA, 1996. Available from the National Fire Protection Association, 1 Battery March Park, Quincy, MA 02269-9101. This incorporation by reference was approved by the Director of the Federal Register in accordance with 5 U.S.C. 552(a) and 1 CFR part 51. Copies may be inspected at the Environmental Protection Agency Air Docket (6102), Attn: Docket No. A-96-08, Waterside Mall, 401 M. St. SW., Washington DC; or at the Office of Federal Register at 800 North Capitol St., NW, Suite 700, Washington, DC. Boiling point and flash point shall be defined and determined in accordance with NFPA 30, Flammable and Combustible Liquids Code, National Fire Protection Association, Quincy, MA, 1996. Available from the National Fire Protection Association, 1 Battery March Park, Quincy, MA 02269-9101. This incorporation by reference was approved by the Director of the Federal Register in accordance with 5 U.S.C. 552(a) and 1 CFR part 51. Copies may be inspected at the Environmental Protection Agency Air Docket (6102), Attn: Docket No. A-96-08, Waterside Mall, 401 M. St. SW., Washington DC; or at the Office of Federal Register at 800 North Capitol St., NW, Suite 700, Washington, DC. The owner or operator shall document the National Fire Protection Association flammability hazard rating.

(ii) Gasoline. Regulated substances in gasoline, when in distribution or related storage for use as fuel for internal combustion engines, need not be considered when determining whether more than a threshold quantity is present at a stationary source.

(iii) Naturally occurring hydrocarbon mixtures. Prior to entry into a natural gas processing plant or a petroleum refining process unit, regulated substances in naturally occurring hydrocarbon mixtures need not be considered when determining whether more than a
threshold quantity is present at a stationary source. Naturally occurring hydrocarbon mixtures include any combination of the following: condensate, crude oil, field gas, and produced water, each as defined in § 68.3 of this part.

(3) Articles. Regulated substances contained in articles need not be considered when determining whether more than a threshold quantity is present at the stationary source.

(4) Uses. Regulated substances, when in use for the following purposes, need not be included in determining whether more than a threshold quantity is present at the stationary source:

(i) Use as a structural component of the stationary source;

(ii) Use of products for routine janitorial maintenance;

(iii) Use by employees of foods, drugs, cosmetics, or other personal items containing the regulated substance; and

(iv) Use of regulated substances present in process water or non-contact cooling water as drawn from the environment or municipal sources, or use of regulated substances present in air used either as compressed air or as part of combustion.

(5) Activities in laboratories. If a regulated substance is manufactured, processed, or used in a laboratory at a stationary source under the supervision of a technically qualified individual as defined in § 720.3(ee) of this chapter, the quantity of the substance need not be considered in determining whether a threshold quantity is present. This exemption does not apply to:

(i) Specialty chemical production;

(ii) Manufacture, processing, or use of substances in pilot plant scale operations; and

(iii) Activities conducted outside the laboratory.


§ 68.120 Petition process.

(a) Any person may petition the Administrator to modify, by addition or deletion, the list of regulated substances identified in § 68.130. Based on the information presented by the petitioner, the Administrator may grant or deny a petition.

(b) A substance may be added to the list if, in the case of an accidental release, it is known to cause or may be reasonably anticipated to cause death, injury, or serious adverse effects to human health or the environment.

(c) A substance may be deleted from the list if adequate data on the health and environmental effects of the substance are available to determine that the substance, in the case of an accidental release, is not known to cause and may not be reasonably anticipated to cause death, injury, or serious adverse effects to human health or the environment.

(d) No substance for which a national primary ambient air quality standard has been established shall be added to the list. No substance regulated under title VI of the Clean Air Act, as amended, shall be added to the list.

(e) The burden of proof is on the petitioner to demonstrate that the criteria for addition and deletion are met. A petition will be denied if this demonstration is not made.

(f) The Administrator will not accept additional petitions on the same substance following publication of a final notice of the decision to grant or deny a petition, unless new data becomes available that could significantly affect the basis for the decision.

(g) Petitions to modify the list of regulated substances must contain the following:

(1) Name and address of the petitioner and a brief description of the organization(s) that the petitioner represents, if applicable;

(2) Name, address, and telephone number of a contact person for the petition;

(3) Common chemical name(s), common synonym(s), Chemical Abstracts Service number, and chemical formula and structure;

(4) Action requested (add or delete a substance);

(5) Rationale supporting the petitioner's position; that is, how the substance meets the criteria for addition and deletion. A short summary of the rationale must be submitted along with a more detailed narrative; and
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(6) Supporting data; that is, the petition must include sufficient information to scientifically support the request to modify the list. Such information shall include:

(i) A list of all support documents;

(ii) Documentation of literature searches conducted, including, but not limited to, identification of the database(s) searched, the search strategy, dates covered, and printed results;

(iii) Effects data (animal, human, and environmental test data) indicating the potential for death, injury, or serious adverse human and environmental impacts from acute exposure following an accidental release; printed copies of the data sources, in English, should be provided; and

(iv) Exposure data or previous accident history data, indicating the potential for serious adverse human health or environmental effects from an accidental release. These data may include, but are not limited to, physical and chemical properties of the substance, such as vapor pressure; modeling results, including data and assumptions used and model documentation; and historical accident data, citing data sources.

(h) Within 18 months of receipt of a petition, the Administrator shall publish in the FEDERAL REGISTER a notice either denying the petition or granting the petition and proposing a listing.

§ 68.125 Exemptions.

Agricultural nutrients. Ammonia used as an agricultural nutrient, when held by farmers, is exempt from all provisions of this part.

§ 68.130 List of substances.

(a) Regulated toxic and flammable substances under section 112(r) of the Clean Air Act are the substances listed in Tables 1, 2, 3, and 4. Threshold quantities for listed toxic and flammable substances are specified in the tables.

(b) The basis for placing toxic and flammable substances on the list of regulated substances are explained in the notes to the list.

<table>
<thead>
<tr>
<th>Chemical name</th>
<th>CAS No.</th>
<th>Threshold quantity (lbs)</th>
<th>Basis for listing</th>
</tr>
</thead>
<tbody>
<tr>
<td>Acrolein [2-</td>
<td>107-02-8</td>
<td>5,000 b</td>
<td></td>
</tr>
<tr>
<td>Propenial]</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Acrylonitrile [2-</td>
<td>107-13-1</td>
<td>20,000 b</td>
<td></td>
</tr>
<tr>
<td>Propenitrile]</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Acryloyl chloride [2-</td>
<td>814-68-6</td>
<td>5,000 b</td>
<td></td>
</tr>
<tr>
<td>Propenoyl chlo-</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>ride]</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Allyl alcohol [2-</td>
<td>107-18-61</td>
<td>15,000 b</td>
<td></td>
</tr>
<tr>
<td>Propen-1-ol]</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Allylamine [2-</td>
<td>107-11-9</td>
<td>10,000 b</td>
<td></td>
</tr>
<tr>
<td>Propen-1-amine]</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Ammonia (anhy-</td>
<td>7664-41-7</td>
<td>10,000 a, b</td>
<td></td>
</tr>
<tr>
<td>drous)</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Ammonia (conc</td>
<td>7664-41-7</td>
<td>20,000 a, b</td>
<td></td>
</tr>
<tr>
<td>20% or greater)</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Arsenous tri-</td>
<td>7784-34-1</td>
<td>15,000 b</td>
<td></td>
</tr>
<tr>
<td>chloride]</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Arsine ..........</td>
<td>7784-42-1</td>
<td>1,000 b</td>
<td>b</td>
</tr>
<tr>
<td>Boron trichloride [Boran, trichloro-]</td>
<td>10284-34-5</td>
<td>5,000 b</td>
<td></td>
</tr>
<tr>
<td>Boron trifluoride [Boran, trifluoro-]</td>
<td>7637-07-2</td>
<td>5,000 b</td>
<td></td>
</tr>
<tr>
<td>Boron trifluoride compound with methyl ether (1:1) [Boron, trifluoro[oxybis [methane]], T-4]</td>
<td>353-42-4</td>
<td>15,000 b</td>
<td></td>
</tr>
<tr>
<td>Bromine ..........</td>
<td>7726-95-6</td>
<td>10,000 a, b</td>
<td></td>
</tr>
<tr>
<td>Carbon disulfide ...</td>
<td>75-15-0</td>
<td>20,000 b</td>
<td></td>
</tr>
<tr>
<td>Chlorine ..........</td>
<td>7782-50-5</td>
<td>2,500 a, b</td>
<td></td>
</tr>
<tr>
<td>Chlorine dioxide [Chlorine oxide (ClO2)]</td>
<td>10049-04-4</td>
<td>1,000 c</td>
<td></td>
</tr>
<tr>
<td>Chloroform [Methane, trichloro-]</td>
<td>67-66-3</td>
<td>20,000 b</td>
<td></td>
</tr>
<tr>
<td>Chloromethyl ether [Methane, oxychloro-]</td>
<td>542-88-1</td>
<td>1,000 b</td>
<td></td>
</tr>
<tr>
<td>Chloromethyl methyl ether [Methane, chloromethoxy-]</td>
<td>107-30-2</td>
<td>5,000 b</td>
<td></td>
</tr>
<tr>
<td>Crotonaldehyde [2-Butenial]</td>
<td>4170-30-3</td>
<td>20,000 b</td>
<td></td>
</tr>
<tr>
<td>Crotonaldehyde, (E)- [2-Butenial, (E)-]</td>
<td>123-73-9</td>
<td>20,000 b</td>
<td></td>
</tr>
<tr>
<td>Cyanogen chloride</td>
<td>506-77-4</td>
<td>10,000 c</td>
<td></td>
</tr>
<tr>
<td>Cyclohexylamine [Cyclohexanam-</td>
<td>108-91-8</td>
<td>15,000 b</td>
<td></td>
</tr>
<tr>
<td>ine]</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Diborane ..........</td>
<td>19287-45-7</td>
<td>2,500 b</td>
<td></td>
</tr>
<tr>
<td>Dimethyl dichlorosilane [Silane, dichlorodimeth-</td>
<td>75-78-5</td>
<td>5,000 b</td>
<td></td>
</tr>
<tr>
<td>yl]</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Dimethylhydrazine [Hydrazine, 1,1-dimethyl-]</td>
<td>57-14-7</td>
<td>15,000 b</td>
<td></td>
</tr>
</tbody>
</table>
### Table 1 to §68.130—List of Regulated Toxic Substances and Threshold Quantities for Accidental Release Prevention—Continued

[Alphabetical Order—77 Substances]

<table>
<thead>
<tr>
<th>Chemical name</th>
<th>CAS No.</th>
<th>Threshold quantity (lbs)</th>
<th>Basis for listing</th>
</tr>
</thead>
<tbody>
<tr>
<td>Epichlorohydrin [Oxirane, (chloromethyl)-]</td>
<td>106–89–8</td>
<td>20,000</td>
<td>b</td>
</tr>
<tr>
<td>Ethylenediamine [1,2- Ethanediamine], Ethyleneimine [Aziridine]</td>
<td>107–15–3</td>
<td>20,000</td>
<td>b</td>
</tr>
<tr>
<td>Ethylene oxide [Oxirane]</td>
<td>75–21–8</td>
<td>10,000</td>
<td>a, b</td>
</tr>
<tr>
<td>Fluorine</td>
<td>7782–41–4</td>
<td>1,000</td>
<td>b</td>
</tr>
<tr>
<td>Formaldehyde (solution)</td>
<td>110–00–9</td>
<td>5,000</td>
<td>b</td>
</tr>
<tr>
<td>Hydrazine</td>
<td>302–01–2</td>
<td>15,000</td>
<td>b</td>
</tr>
<tr>
<td>Hydrochloric acid [anhydrous]</td>
<td>7647–01–0</td>
<td>10,000</td>
<td>d</td>
</tr>
<tr>
<td>Hydrogen fluoride/ Hydrofluoric acid (anhydrous) [Hydrofluoric acid]</td>
<td>8014–95–7</td>
<td>10,000</td>
<td>a</td>
</tr>
<tr>
<td>Hydrogen sulfide/ Hydrogen sulfide (anhydrous) [Hydrogen sulfide]</td>
<td>7783–06–4</td>
<td>10,000</td>
<td>a, b</td>
</tr>
<tr>
<td>Isopropyl mercaptan [Methanethiol]</td>
<td>74–93–1</td>
<td>10,000</td>
<td>b</td>
</tr>
</tbody>
</table>
| Methane, chloro-
| Methyl chloride                                    | 74–87–3  | 10,000                   | a                 |
| Methane, chloro-
| Methyl chloroformate [Carbonochloridic acid, 1- methylthyl ester] | 126–98–7 | 10,000                   | b                 |
| Methacrylonitrile [2- Propenitrile, 2-methyl-]     | 74–87–3  | 10,000                   | a                 |
| Methyl thiocyanate [Thiocyanic acid, methyl ester]  | 7783–06–4| 5,000                    | b                 |
| Methyl trichlorosilane [Silane, trichloromethyl]   | 75–79–6  | 5,000                    | b                 |
| Nickel carbonyl                                     | 13463–39–3| 1,000                    | b                 |
| Nitric acid (conc 80% or greater)                   | 7697–37–2| 15,000                   | b                 |
| Nitric oxide [Nitrogen oxide (NO)]                  | 10102–43–9| 10,000                   | b                 |
| Oleum (Fuming Sulfuric acid) [Sulfuric acid, mixture with sulfur trioxide] | 8014–95–7| 10,000                   | a                 |
| Perchloromethylmercaptan [Methanesulfenyl chloride, trichloro-] | 75–44–5  | 500                      | a, b              |
| Phosphine                                           | 7803–51–2| 5,000                    | b                 |
| Phosphorus                                          | 10025–87–3| 5,000                    | b                 |
| Phosphorus trihalide [Phosphorus trihalide]        | 7719–12–2| 15,000                   | b                 |
| Piperidine                                          | 110–89–4 | 15,000                   | b                 |
| Propylene oxide [Oxirane, methylene chloride]       | 74–46–9  | 5,000                    | a, b              |
| Propylene oxide [Oxirane, methylene chloride]       | 7783–06–4| 5,000                    | b                 |
| Propylene oxide [Oxirane, methylene chloride]       | 74–46–9  | 5,000                    | a, b              |
| Sulfur dioxide (anhydrous)                         | 7446–09–5| 5,000                    | a, b              |
| Sulfur dioxide (anhydrous)                         | 7446–11–9| 10,000                   | a, b              |
| Tetramethyllead [Plumbane, tetramethyl-]            | 75–74–1  | 10,000                   | b                 |
| Tetranitromethane [Methane, tetranitro-]           | 509–14–6 | 10,000                   | b                 |
### TABLE 1 TO § 68.130.—LIST OF REGULATED TOXIC SUBSTANCES AND THRESHOLD QUANTITIES FOR ACCIDENTAL RELEASE PREVENTION—Continued  
*Alphabetical Order—77 Substances*

<table>
<thead>
<tr>
<th>Chemical name</th>
<th>CAS No.</th>
<th>Threshold quantity (lbs)</th>
<th>Basis for listing</th>
</tr>
</thead>
<tbody>
<tr>
<td>Titanium tetra-chloride (TiCl4) (T-4)</td>
<td>7550-45-0</td>
<td>2,500</td>
<td>b</td>
</tr>
<tr>
<td>Toluene 2,4-diisocyanate (Benzene, 2,4-diisocyanato-1-methyl)</td>
<td>584-84-9</td>
<td>10,000</td>
<td>a</td>
</tr>
<tr>
<td>Toluene 2,6-diisocyanate (Benzene, 1,3-diisocyanato-2-methyl)</td>
<td>91-08-7</td>
<td>10,000</td>
<td>a</td>
</tr>
<tr>
<td>Toluene diisocyanate (unspecified isomer) (Benzene, 1,3-diisocyanatomethyl)</td>
<td>26471-62-5</td>
<td>10,000</td>
<td>a</td>
</tr>
<tr>
<td>Acrylonitrile</td>
<td>75-05-8</td>
<td>2,500</td>
<td>a</td>
</tr>
<tr>
<td>Butadiene</td>
<td>106-99-0</td>
<td>500</td>
<td></td>
</tr>
<tr>
<td>Benzene</td>
<td>71-43-2</td>
<td>5,000</td>
<td></td>
</tr>
<tr>
<td>Chloroform</td>
<td>67-66-3</td>
<td>10,000</td>
<td></td>
</tr>
<tr>
<td>1-Methyl-2-propanol</td>
<td>106-34-9</td>
<td>15,000</td>
<td></td>
</tr>
<tr>
<td>Methyl ethyl ketone</td>
<td>78-93-3</td>
<td>10,000</td>
<td></td>
</tr>
<tr>
<td>Tetrahydrofuran</td>
<td>109-88-4</td>
<td>2,500</td>
<td>a, b</td>
</tr>
</tbody>
</table>

**NOTE:** Basis for Listing:
- a) Mandated for listing by Congress.
- b) On EHS list, vapor pressure 10 mmHg or greater.
- c) Toxic gas.
- d) Toxicity of hydrogen chloride, potential to release hydrogen chloride, and history of accidents.
- e) Toxicity of sulfur trioxide and sulfuric acid, potential to release sulfur trioxide, and history of accidents.

### TABLE 2 TO § 68.130.—LIST OF REGULATED TOXIC SUBSTANCES AND THRESHOLD QUANTITIES FOR ACCIDENTAL RELEASE PREVENTION  
*CAS Number Order—77 Substances*

<table>
<thead>
<tr>
<th>CAS No.</th>
<th>Chemical name</th>
<th>Threshold quantity (lbs)</th>
<th>Basis for listing</th>
</tr>
</thead>
<tbody>
<tr>
<td>50-00-0</td>
<td>Formaldehyde (solution)</td>
<td>15,000</td>
<td>b</td>
</tr>
<tr>
<td>57-14-7</td>
<td>1,1-Dimethylydrazine (Hydrazine, 1,1-dimethyl-</td>
<td>15,000</td>
<td>b</td>
</tr>
<tr>
<td>60-34-4</td>
<td>Methyl hydrazine (Hydrazine, methyl-)</td>
<td>15,000</td>
<td>b</td>
</tr>
<tr>
<td>67-66-3</td>
<td>Chloroform (Methane, trichloro-)</td>
<td>20,000</td>
<td></td>
</tr>
<tr>
<td>74-87-3</td>
<td>Methyl chloride (Methane, chloro-)</td>
<td>10,000</td>
<td>a</td>
</tr>
<tr>
<td>74-90-8</td>
<td>Hydrocyanic acid</td>
<td>2,500</td>
<td>a, b</td>
</tr>
<tr>
<td>74-93-1</td>
<td>Methyl mercaptan (Methanethiol)</td>
<td>10,000</td>
<td>b</td>
</tr>
<tr>
<td>75-15-0</td>
<td>Carbon disulfide</td>
<td>20,000</td>
<td></td>
</tr>
<tr>
<td>75-21-8</td>
<td>Ethylene oxide (Oxirane)</td>
<td>10,000</td>
<td>a, b</td>
</tr>
<tr>
<td>74-44-5</td>
<td>Phosgene (Carbonic dichloride)</td>
<td>500</td>
<td>a, b</td>
</tr>
<tr>
<td>75-55-8</td>
<td>Propylene imine (Aziridine, 2-methyl-)</td>
<td>10,000</td>
<td>b</td>
</tr>
<tr>
<td>75-56-9</td>
<td>Propylene oxide (Oxirane, methyl-)</td>
<td>10,000</td>
<td></td>
</tr>
<tr>
<td>75-74-1</td>
<td>Tetramethylead (Plumbane, tetramethyl-)</td>
<td>10,000</td>
<td></td>
</tr>
<tr>
<td>75-77-4</td>
<td>Trimethylchlorosilane (Silane, chlorotrimethyl-)</td>
<td>10,000</td>
<td>b</td>
</tr>
<tr>
<td>75-78-6</td>
<td>Dimethylchlorosilane (Silane, dichloromethyl-)</td>
<td>5,000</td>
<td></td>
</tr>
<tr>
<td>78-62-0</td>
<td>Isobutynitrile (Propenitrile, 2-methyl-)</td>
<td>20,000</td>
<td></td>
</tr>
<tr>
<td>79-21-0</td>
<td>Peracetic acid (Ethaneperoxoic acid)</td>
<td>10,000</td>
<td></td>
</tr>
<tr>
<td>79-22-1</td>
<td>Methyl chlorofromate (Carbonochloridic acid, methyl ester)</td>
<td>5,000</td>
<td></td>
</tr>
<tr>
<td>91-08-7</td>
<td>Toluene 2,6-disocyanate (Benzene, 1,3-diisocyanato-2-methyl-)</td>
<td>10,000</td>
<td></td>
</tr>
<tr>
<td>106-89-8</td>
<td>Epichlorohydrin (Oxirane, chloromethyl-)</td>
<td>20,000</td>
<td></td>
</tr>
<tr>
<td>107-02-8</td>
<td>Acrolein (2-Propanal)</td>
<td>5,000</td>
<td>b</td>
</tr>
<tr>
<td>107-11-9</td>
<td>Allylamine (2-Propen-1-amine)</td>
<td>10,000</td>
<td></td>
</tr>
<tr>
<td>107-12-0</td>
<td>Propionitrile (Propenitrile)</td>
<td>10,000</td>
<td></td>
</tr>
<tr>
<td>107-13-1</td>
<td>Acrylonitrile (2-Prophenitrile)</td>
<td>20,000</td>
<td></td>
</tr>
<tr>
<td>107-15-3</td>
<td>Ethylenediamine (1,2-Ethanenediamine)</td>
<td>20,000</td>
<td></td>
</tr>
<tr>
<td>107-18-6</td>
<td>Allyl alcohol (2-Propen-1-ol)</td>
<td>15,000</td>
<td></td>
</tr>
<tr>
<td>107-30-2</td>
<td>Chloromethyl methyl ether (Methane, chloromethoxy-)</td>
<td>5,000</td>
<td></td>
</tr>
<tr>
<td>108-05-4</td>
<td>Vinyl acetate monomer (Acetic acid ethenyl ester)</td>
<td>15,000</td>
<td></td>
</tr>
<tr>
<td>108-23-6</td>
<td>Isopropyl chlorofromate (Carbonochloridic acid, 1-methyl ethyl ester)</td>
<td>15,000</td>
<td></td>
</tr>
<tr>
<td>108-91-8</td>
<td>Cyclohexylamine (Cyclohexanamine)</td>
<td>15,000</td>
<td></td>
</tr>
<tr>
<td>109-61-5</td>
<td>Propyl chlorofromate (Carbonochloridic acid, propylester)</td>
<td>15,000</td>
<td></td>
</tr>
<tr>
<td>110-00-9</td>
<td>Furan</td>
<td>5,000</td>
<td></td>
</tr>
<tr>
<td>110-89-4</td>
<td>Piperidine</td>
<td>15,000</td>
<td></td>
</tr>
<tr>
<td>123-73-9</td>
<td>Crotonaldehyde (E)- [2-Butenal, (E)]</td>
<td>20,000</td>
<td></td>
</tr>
</tbody>
</table>

---

Toluene 2,4-diisocyanate (Benzene, 2,4-diisocyanato-1-methyl) | 584-84-9 | 10,000 | a |
**Table 2 to §68.130.—List of Regulated Toxic Substances and Threshold Quantities for Accidental Release Prevention—Continued**

<table>
<thead>
<tr>
<th>CAS No.</th>
<th>Chemical name</th>
<th>Threshold quantity (lbs)</th>
<th>Basis for listing</th>
</tr>
</thead>
<tbody>
<tr>
<td>126–98–7</td>
<td>Methacrylonitrile [2-Propenenitrile, 2-methyl-]</td>
<td>10,000</td>
<td>b</td>
</tr>
<tr>
<td>151–56–4</td>
<td>Ethylenemine [Azidine]</td>
<td>10,000</td>
<td>b</td>
</tr>
<tr>
<td>302–01–2</td>
<td>Hydrazine</td>
<td>15,000</td>
<td>b</td>
</tr>
<tr>
<td>353–42–4</td>
<td>Boron trifluoride compound with methyl ether (1:1) [Boron, trifluoroxy oxy(borane)]</td>
<td>15,000</td>
<td>b</td>
</tr>
<tr>
<td>506–77–4</td>
<td>Cyanogen chloride</td>
<td>10,000</td>
<td>c</td>
</tr>
<tr>
<td>509–14–8</td>
<td>Tetratriomethane [Methane, tetrahydro-]</td>
<td>10,000</td>
<td>b</td>
</tr>
<tr>
<td>542–88–1</td>
<td>Chloromethyl ether [Methane, oxy(borane)]</td>
<td>1,000</td>
<td>b</td>
</tr>
<tr>
<td>556–64–9</td>
<td>Methyl thiocyanate [Thiocyanic acid, methyl ester]</td>
<td>20,000</td>
<td>b</td>
</tr>
<tr>
<td>584–84–9</td>
<td>Toluene, 2,4-disocyanate [Benzene, 2,4-disocyanat-1-methyl-]</td>
<td>10,000</td>
<td>a</td>
</tr>
<tr>
<td>594–42–3</td>
<td>Perchloromethylmercaptan [Methanesulfenyl chloride, trichloro-]</td>
<td>10,000</td>
<td>b</td>
</tr>
<tr>
<td>624–83–9</td>
<td>Methyl isocyanate [Methane, isocyanato]</td>
<td>10,000, a, b</td>
<td></td>
</tr>
<tr>
<td>816–68–6</td>
<td>Acryl chloride [2-Propenyl chloride]</td>
<td>5,000</td>
<td>b</td>
</tr>
<tr>
<td>4170–30–3</td>
<td>Crotonaldehyde [2-Butenal]</td>
<td>20,000</td>
<td>b</td>
</tr>
<tr>
<td>7446–09–5</td>
<td>Sulfur dioxide (anhydrous)</td>
<td>5,000, a, b</td>
<td></td>
</tr>
<tr>
<td>7446–11–9</td>
<td>Sulfur trioxide</td>
<td>10,000</td>
<td>a, b</td>
</tr>
<tr>
<td>7550–45–0</td>
<td>Titanium tetrachloride [Titanium chloride (TiCl4) (T-4)]</td>
<td>2,500</td>
<td>b</td>
</tr>
<tr>
<td>7637–07–2</td>
<td>Boron trifluoride [Borane, trifluoro-]</td>
<td>5,000</td>
<td>b</td>
</tr>
<tr>
<td>7647–01–0</td>
<td>Hydrochloric acid (conc 37% or greater)</td>
<td>15,000</td>
<td>d</td>
</tr>
<tr>
<td>7647–01–0</td>
<td>Hydrogen chloride (anhydrous) [Hydrochloric acid]</td>
<td>5,000</td>
<td>a</td>
</tr>
<tr>
<td>7648–39–3</td>
<td>Hydrogen fluoride/Hydrofluoric acid (conc 50% or greater) [Hydrofluoric acid]</td>
<td>15,000</td>
<td>a, b</td>
</tr>
<tr>
<td>7664–41–7</td>
<td>Ammonia (anhydrous)</td>
<td>10,000</td>
<td>a</td>
</tr>
<tr>
<td>7679–37–2</td>
<td>Nitric acid (conc 60% or greater)</td>
<td>15,000</td>
<td>b</td>
</tr>
<tr>
<td>7726–95–6</td>
<td>Bromine</td>
<td>10,000</td>
<td>a, b</td>
</tr>
<tr>
<td>7782–41–4</td>
<td>Fluorine</td>
<td>1,000</td>
<td>b</td>
</tr>
<tr>
<td>7782–50–5</td>
<td>Chlorine</td>
<td>2,500</td>
<td>a, b</td>
</tr>
<tr>
<td>7783–06–4</td>
<td>Hydrogen sulfide</td>
<td>10,000</td>
<td>a, b</td>
</tr>
<tr>
<td>7783–07–5</td>
<td>Hydrogen selenide</td>
<td>500</td>
<td>b</td>
</tr>
<tr>
<td>7783–60–0</td>
<td>Sulfur tetrafluoride [Sulfur fluoride (SF4), (T-4)]</td>
<td>2,500</td>
<td>b</td>
</tr>
<tr>
<td>7784–34–1</td>
<td>Arsenous trichloride</td>
<td>15,000</td>
<td>b</td>
</tr>
<tr>
<td>7784–42–1</td>
<td>Arsenic</td>
<td>1,000</td>
<td>b</td>
</tr>
<tr>
<td>8014–95–7</td>
<td>Oleum (Fuming Sulfuric acid) [Sulfuric acid, mixture with sulfur trioxide]</td>
<td>10,000</td>
<td>e</td>
</tr>
<tr>
<td>10026–87–3</td>
<td>Phosphorus oxychloride [Phosphoryl chloride]</td>
<td>5,000</td>
<td>b</td>
</tr>
<tr>
<td>10049–04–4</td>
<td>Chlorine dioxide [Chlorine oxide (ClO2)]</td>
<td>1,000</td>
<td>c</td>
</tr>
<tr>
<td>10102–43–9</td>
<td>Nitric oxide [Nitrogen oxide (NO)]</td>
<td>10,000</td>
<td>b</td>
</tr>
<tr>
<td>10294–34–6</td>
<td>Boron trifluoride [Borane, trichloro]</td>
<td>5,000</td>
<td>b</td>
</tr>
<tr>
<td>13463–39–3</td>
<td>Nickel carbonyl</td>
<td>1,000</td>
<td>b</td>
</tr>
<tr>
<td>13463–40–6</td>
<td>Iron, pentacarbonyl [Iron carbonyl (Fe(CO)5), (TB-5-11)]</td>
<td>2,500</td>
<td>b</td>
</tr>
<tr>
<td>19827–45–7</td>
<td>Dilorane</td>
<td>2,500</td>
<td>b</td>
</tr>
<tr>
<td>26471–62–5</td>
<td>Toluene disocyanate (unspecified isomer) [Benzene, 1,3-disocyanatohexamethylene-1]:</td>
<td>10,000</td>
<td>a</td>
</tr>
</tbody>
</table>

* The mixture exemption in §68.115(b)(1) does not apply to the substance.

**NOTE:** Basis for Listing:

- a Mandated for listing by Congress.
- b On EHS list, vapor pressure 10 mmHg or greater.
- c Toxic gas.
- d Toxicity of hydrogen chloride, potential to release hydrogen chloride, and history of accidents.
- e Toxicity of sulfur trioxide and sulfuric acid, potential to release sulfur trioxide, and history of accidents.

---

**Table 3 to §68.130.—List of Regulated Flammable Substances and Threshold Quantities for Accidental Release Prevention**

<table>
<thead>
<tr>
<th>Chemical name</th>
<th>CAS No.</th>
<th>Threshold quantity (lbs)</th>
<th>Basis for listing</th>
</tr>
</thead>
<tbody>
<tr>
<td>Acetaldehyde</td>
<td>75–07–0</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>Acetylene [Ethene]</td>
<td>74–96–2</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Bromotrifluoroethylene [Ethene, bromotrifluoro-]</td>
<td>598–73–2</td>
<td>10,000</td>
<td>l</td>
</tr>
<tr>
<td>1,3-Butadiene</td>
<td>106–99–0</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Butane</td>
<td>106–97–6</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>1-Butene</td>
<td>106–98–9</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>2-Butene</td>
<td>107–01–7</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Butane</td>
<td>25167–67–3</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Chemical name</td>
<td>CAS No.</td>
<td>Threshold quantity (lbs)</td>
<td>Basis for listing</td>
</tr>
<tr>
<td>---------------------------------------------------</td>
<td>----------</td>
<td>--------------------------</td>
<td>-------------------</td>
</tr>
<tr>
<td>2-Butene-cis</td>
<td>590-18-1</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>2-Butene-trans [2-Butene, (E)]</td>
<td>624-64-6</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Carbon oxysulfide [Carbon oxide sulfide (COS)]</td>
<td>463-58-1</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Chlorine monoxide [Chlorine oxide]</td>
<td>7791-21-1</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>2-Chloropropylene [1-Propene, 2-chloro-]</td>
<td>557-98-2</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>1-Chloropropylene [1-Propene, 1-chloro-]</td>
<td>590-21-6</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Cyanogen [Ethanedinitrile]</td>
<td>460-19-5</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Cyclopropane</td>
<td>75-19-4</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Dichlorosilane [Silane, dichloro-]</td>
<td>4109-96-0</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Difluoroethane [Ethane, 1,1-difluoro-]</td>
<td>75-37-6</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Dimethylamine [Methanamine, N-methyl-]</td>
<td>124-40-3</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>2,2-Dimethylpropane [Propane, 2,2-dimethyl-]</td>
<td>463-82-1</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Ethane</td>
<td>74-84-0</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Ethyl acetylene [1-Butyne]</td>
<td>107-00-6</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Ethylamine [Ethanamine]</td>
<td>75-04-7</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Ethyl chloride [Ethane, chloro-]</td>
<td>75-00-3</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Ethylene [Ethene]</td>
<td>74-65-1</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Ethyl ether [Ethane, 1,1'-oxybis-]</td>
<td>60-29-7</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>Ethyl mercaptan [Ethanethiol]</td>
<td>75-08-1</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>Ethyl nitrite [Nitric acid, ethyl ester]</td>
<td>109-95-5</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Hydrogen</td>
<td>1333-74-0</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Isobutane [Propane, 2-methyl-]</td>
<td>75-29-6</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Isopentane [Butane, 2-methyl-]</td>
<td>78-78-4</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>Isoprene [1,3-Butadiene, 2-methyl-]</td>
<td>78-79-5</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>Isopropylamine [2-Propanamine]</td>
<td>75-31-0</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>Isopropyl chloride [Propane, 2-chloro-]</td>
<td>75-29-6</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>Methane</td>
<td>74-82-8</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Methylamine [Methanamine]</td>
<td>74-89-5</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>3-Methyl-1-butene</td>
<td>563-45-1</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>2-Methyl-1-butene</td>
<td>563-46-2</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>Methyl ether [Methane, oxybis-]</td>
<td>115-10-6</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Methyl formate [Formic acid, methyl ester]</td>
<td>107-31-3</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>2-Methylpropane [1-Propene, 2-methyl-]</td>
<td>115-11-7</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>1,3-Pentadiene</td>
<td>504-60-9</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Pentane</td>
<td>109-66-0</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>1-Pentene</td>
<td>109-67-1</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>2-Pentene, (E)</td>
<td>646-04-8</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>2-Pentene, (Z)</td>
<td>627-20-3</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>Propadiene [1,2-Propadiene]</td>
<td>463-49-0</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Propane</td>
<td>74-98-6</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Propylene [1-Propene]</td>
<td>115-07-1</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Propyne [1-Propane]</td>
<td>74-99-7</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Silane</td>
<td>7803-62-5</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Tetrafluoroethene [Ethene, tetrafluoro-]</td>
<td>116-14-3</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Tetramethylsilane [Silane, tetramethyl-]</td>
<td>75-76-3</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>Trichlorosilane [Silane, trichloro-]</td>
<td>10025-78-2</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>Trifluorochloroethylene [Ethene, chlorotrifluoro-]</td>
<td>79-38-9</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Trimethylamine [Methanamine, N,N-dimethyl-]</td>
<td>75-50-3</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Vinyl acetylene [1-Buten-3-yne]</td>
<td>689-97-4</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Vinyl chloride [Ethene, chloro-]</td>
<td>75-01-4</td>
<td>10,000</td>
<td>a, f</td>
</tr>
<tr>
<td>Vinyl ethyl ether [Ethene, ethoxy-]</td>
<td>109-92-2</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>Vinyl fluoride [Ethene, fluoro-]</td>
<td>75-02-5</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Vinylidene chloride [Ethene, 1,1-dichloro-]</td>
<td>75-35-4</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>Vinylidene fluoride [Ethene, 1,1-difluoro-]</td>
<td>75-38-7</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>Vinyl methyl ether [Ethene, methoxy-]</td>
<td>107-25-5</td>
<td>10,000</td>
<td>f</td>
</tr>
</tbody>
</table>

**NOTE:** Basis for Listing:

- a Mandated for listing by Congress.
- f Flammable gas.
- g Volatile flammable liquid.
### § 68.130

**40 CFR Ch. I (7-1-98 Edition)**

#### TABLE 4 TO § 68.130—LIST OF REGULATED FLAMMABLE SUBSTANCES AND THRESHOLD QUANTITIES FOR ACCIDENTAL RELEASE PREVENTION

<table>
<thead>
<tr>
<th>CAS No.</th>
<th>Chemical name</th>
<th>CAS No.</th>
<th>Threshold quantity (lbs)</th>
<th>Basis for listing</th>
</tr>
</thead>
<tbody>
<tr>
<td>60–29–7</td>
<td>Ethyl ether [Ethane, 1,1'-oxybis]</td>
<td>60–29–7</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>74–82–8</td>
<td>Methane</td>
<td>74–82–8</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>74–84–0</td>
<td>Ethane</td>
<td>74–84–0</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>74–85–1</td>
<td>Ethylene [Ethene]</td>
<td>74–85–1</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>74–86–2</td>
<td>Acetylene [Ethyne]</td>
<td>74–86–2</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>74–89–5</td>
<td>Methylamine [Methanamine]</td>
<td>74–89–5</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>74–96–6</td>
<td>Propane</td>
<td>74–96–6</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>74–99–7</td>
<td>Propyne [1-Propyne]</td>
<td>74–99–7</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>75–00–3</td>
<td>Ethyl chloride [Ethane, chloro-]</td>
<td>75–00–3</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>75–01–4</td>
<td>Vinyl chloride [Ethene, chloro-]</td>
<td>75–01–4</td>
<td>10,000</td>
<td>a, f</td>
</tr>
<tr>
<td>75–02–5</td>
<td>Vinyl fluoride [Ethene, fluoro-]</td>
<td>75–02–5</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>75–04–7</td>
<td>Ethylamine [Ethanamine]</td>
<td>75–04–7</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>75–07–0</td>
<td>Acetaldehyde</td>
<td>75–07–0</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>75–08–1</td>
<td>Ethyl mercaptan [Ethanethiol]</td>
<td>75–08–1</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>75–19–4</td>
<td>Cyclopropane</td>
<td>75–19–4</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>75–28–6</td>
<td>Isobutane [Propane, 2-methyl-]</td>
<td>75–28–6</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>75–31–0</td>
<td>Isopropyl chloride [Propene, 2-chloro-]</td>
<td>75–31–0</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>75–35–4</td>
<td>Vinylidine chloride [Ethene, 1,1-dichloro-]</td>
<td>75–35–4</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>75–37–6</td>
<td>Difluoromethane [Ethene, 1,1-difluoro-]</td>
<td>75–37–6</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>75–38–7</td>
<td>Vinylfluoride [Ethene, 1,1-difluoro-]</td>
<td>75–38–7</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>75–50–3</td>
<td>Trimethylamine [Methanamine, N,N-dimethyl-]</td>
<td>75–50–3</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>75–75–3</td>
<td>Tetramethyilsilane [Silane, tetramethyl-]</td>
<td>75–75–3</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>78–78–4</td>
<td>Isopentane [Butane, 2-methyl-]</td>
<td>78–78–4</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>78–79–6</td>
<td>Isoprene [1,3-Butadiene, 2-methyl-]</td>
<td>78–79–6</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>79–38–9</td>
<td>Trifluorochloroethylene [Ethene, chlorotrifluoro-]</td>
<td>79–38–9</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>106–97–8</td>
<td>Butane</td>
<td>106–97–8</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>106–98–9</td>
<td>1-Butene</td>
<td>106–98–9</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>106–99–0</td>
<td>1,3-Butadiene</td>
<td>106–99–0</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>107–00–6</td>
<td>Ethyl acetylene [1-Butyne]</td>
<td>107–00–6</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>107–01–7</td>
<td>2-Butene</td>
<td>107–01–7</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>107–25–5</td>
<td>Vinyl methyl ether [Ethene, methoxy-]</td>
<td>107–25–5</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>107–31–3</td>
<td>Methyl formate [Formic acid, methyl ester]</td>
<td>107–31–3</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>109–66–0</td>
<td>Pentane</td>
<td>109–66–0</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>109–67–1</td>
<td>1-Pentene</td>
<td>109–67–1</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>109–92–2</td>
<td>Vinyl ethyl ether [Ethene, ethoxy-]</td>
<td>109–92–2</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>109–95–5</td>
<td>Ethyl nitrite [Nitrous acid, ethyl ester]</td>
<td>109–95–5</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>115–07–1</td>
<td>Propylene [1-Propene]</td>
<td>115–07–1</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>115–10–6</td>
<td>Methyl ether [Methane, oxybis-]</td>
<td>115–10–6</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>115–11–7</td>
<td>2-Methylpropane [1-Propene, 2-methyl-]</td>
<td>115–11–7</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>116–14–3</td>
<td>Tetrafluoroethylene [Ethene, tetrafluoro-]</td>
<td>116–14–3</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>124–40–3</td>
<td>Dimethylamine [Methanamine, N-methyl-]</td>
<td>124–40–3</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>460–19–5</td>
<td>Cyanogen [Ethanedinitrile]</td>
<td>460–19–5</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>463–49–0</td>
<td>Propadiene [1,2-Propadiene]</td>
<td>463–49–0</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>463–58–1</td>
<td>Carbon disulfide [Carbon oxide sulfide (COS)]</td>
<td>463–58–1</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>463–82–1</td>
<td>2,2-Dimethylpropane [Propane, 2,2-dimethyl-]</td>
<td>463–82–1</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>504–60–9</td>
<td>1,3-Pentadiene</td>
<td>504–60–9</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>557–98–2</td>
<td>2-Chloropropylene [1-Propene, 2-chloro-]</td>
<td>557–98–2</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>563–45–1</td>
<td>Methyl-1-butene</td>
<td>563–45–1</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>563–46–2</td>
<td>2-Methyl-1-butene</td>
<td>563–46–2</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>590–18–1</td>
<td>2-Butene</td>
<td>590–18–1</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>590–21–6</td>
<td>1-Chloropropylene [1-Propene, 1-chloro-]</td>
<td>590–21–6</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>598–73–2</td>
<td>Bromotrifluoromethane [Ethene, bromotrifluoro-]</td>
<td>598–73–2</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>624–64–6</td>
<td>2-Butene-trans [2-Butene, (E)]</td>
<td>624–64–6</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>627–20–3</td>
<td>2-Pentene, (Z)</td>
<td>627–20–3</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>646–04–8</td>
<td>2-Pentene, (E)</td>
<td>646–04–8</td>
<td>10,000</td>
<td>g</td>
</tr>
<tr>
<td>689–97–4</td>
<td>Vinyl acetylene [1-Buten-3-yne]</td>
<td>689–97–4</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>698–99–9</td>
<td>Hydrogen</td>
<td>698–99–9</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>730–74–0</td>
<td>Dichlorosilane [Silane, dichloro-]</td>
<td>730–74–0</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>779–21–1</td>
<td>Chlorine monoxide [Chlorine oxide]</td>
<td>779–21–1</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>7803–62–5</td>
<td>Silane</td>
<td>7803–62–5</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>10025–78–2</td>
<td>Dichlorosilane [Silane, dichloro-]</td>
<td>10025–78–2</td>
<td>10,000</td>
<td>f</td>
</tr>
<tr>
<td>25167–67–3</td>
<td>Butene</td>
<td>25167–67–3</td>
<td>10,000</td>
<td>g</td>
</tr>
</tbody>
</table>

Note: Basis for Listing.  
\( a, f \) Mandated for listing by Congress.  
\( f \) Flammable gas.  
\( g \) Volatile flammable liquid.

Subpart G—Risk Management Plan

SOURCE: 61 FR 31726, June 20, 1996, unless otherwise noted.

§ 68.150 Submission.
(a) The owner or operator shall submit a single RMP that includes the information required by §§ 68.155 through 68.185 for all covered processes. The RMP shall be submitted in a method and format to a central point as specified by EPA prior to June 21, 1999.
(b) The owner or operator shall submit the first RMP no later than the latest of the following dates:
(1) June 21, 1999;
(2) Three years after the date on which a regulated substance is first listed under § 68.130; or
(3) The date on which a regulated substance is first present above a threshold quantity in a process.
(c) Subsequent submissions of RMPs shall be in accordance with § 68.190.
(d) Notwithstanding the provisions of §§ 68.155 to 68.190, the RMP shall exclude classified information. Subject to appropriate procedures to protect such information from public disclosure, classified data or information excluded from the RMP may be made available in a classified annex to the RMP for review by Federal and state representatives who have received the appropriate security clearances.

§ 68.155 Executive summary.
The owner or operator shall provide in the RMP an executive summary that includes a brief description of the following elements:
(a) The accidental release prevention and emergency response policies at the stationary source;
(b) The stationary source and regulated substances handled;
(c) The worst-case release scenario(s) and the alternative release scenario(s), including administrative controls and mitigation measures to limit the distances for each reported scenario;
(d) The general accidental release prevention program and chemical-specific prevention steps;
(e) The five-year accident history;
(f) The emergency response program;
(g) Planned changes to improve safety.

§ 68.160 Registration.
(a) The owner or operator shall complete a single registration form and include it in the RMP. The form shall cover all regulated substances handled in covered processes.
(b) The registration shall include the following data:
(1) Stationary source name, street, city, county, state, zip code, latitude, and longitude;
(2) The stationary source Dun and Bradstreet number;
(3) Name and Dun and Bradstreet number of the corporate parent company;
(4) The name, telephone number, and mailing address of the owner or operator;
(5) The name and title of the person or position with overall responsibility for RMP elements and implementation;
(6) The name, title, telephone number, and 24-hour telephone number of the emergency contact;
(7) For each covered process, the name and CAS number of each regulated substance held above the threshold quantity in the process, the maximum quantity of each regulated substance or mixture in the process (in pounds) to two significant digits, the SIC code, and the Program level of the process;
(8) The stationary source EPA identifier;
(9) The number of full-time employees at the stationary source;
(10) Whether the stationary source is subject to 29 CFR 1910.119;
(11) Whether the stationary source is subject to 40 CFR part 355;
(12) Whether the stationary source has a CAA Title V operating permit;
(13) The date of the last safety inspection of the stationary source by a Federal, state, or local government agency and the identity of the inspecting entity.

§ 68.165 Offsite consequence analysis.
(a) The owner or operator shall submit in the RMP information:
§ 68.168 Five-year accident history.

The owner or operator shall submit in the RMP the information provided in §68.42(b) on each accident covered by §68.42(a).

§ 68.170 Prevention program/Program 2.

(a) For each Program 2 process, the owner or operator shall provide in the RMP the information indicated in paragraphs (b) through (k) of this section. If the same information applies to more than one covered process, the owner or operator may provide the information only once, but shall indicate to which processes the information applies.

(b) The SIC code for the process.

(c) The name(s) of the chemical(s) covered.

(d) The date of the most recent review or revision of the safety information and a list of Federal or state regulations or industry-specific design codes and standards used to demonstrate compliance with the safety information requirement.

(e) The date of completion of the most recent hazard review or update.

(f) The expected date of completion of any changes resulting from the hazard review:

(1) Major hazards identified;

(2) Process controls in use;

(3) Mitigation systems in use;

(4) Monitoring and detection systems in use; and

(5) Changes since the last hazard review.

(g) The date of the most recent review or revision of operating procedures.

(h) The date of the most recent review or revision of training programs:

(1) The type of training provided—classroom, classroom plus on the job, on the job; and

(2) The type of competency testing used.

(i) The date of the most recent review or revision of maintenance procedures and the date of the most recent equipment inspection or test and the equipment inspected or tested.

(j) The date of the most recent incident investigation and the expected date of completion of any changes resulting from the investigation.

(k) The date of the most recent change that triggered a review or revision of safety information, the hazard review, operating or maintenance procedures, or training.
§ 68.175 Prevention program/Program 3.

(a) For each Program 3 process, the owner or operator shall provide the information indicated in paragraphs (b) through (p) of this section. If the same information applies to more than one covered process, the owner or operator may provide the information only once, but shall indicate to which processes the information applies.

(b) The SIC code for the process.

(c) The name(s) of the substance(s) covered.

(d) The date on which the safety information was last reviewed or revised.

(e) The date of completion of the most recent PHA or update and the technique used.

(1) The expected date of completion of any changes resulting from the PHA;

(2) Major hazards identified;

(3) Process controls in use;

(4) Mitigation systems in use;

(5) Monitoring and detection systems in use; and

(6) Changes since the last PHA.

(f) The date of the most recent review or revision of operating procedures.

(g) The date of the most recent review or revision of training programs.

(1) The type of training provided—classroom, classroom plus on the job, on the job; and

(2) The type of competency testing used.

(h) The date of the most recent review or revision of maintenance procedures and the date of the most recent equipment inspection or test and the equipment inspected or tested.

(i) The date of the most recent change that triggered management of change procedures and the date of the most recent review or revision of management of change procedures.

(j) The date of the most recent prestartup review.

(k) The date of the most recent compliance audit and the expected date of completion of any changes resulting from the compliance audit;

(l) The date of the most recent incident investigation and the expected date of completion of any changes resulting from the investigation;

(m) The date of the most recent review or revision of employee participation plans;

(n) The date of the most recent review or revision of hot work permit procedures;

(o) The date of the most recent review or revision of contractor safety procedures; and

(p) The date of the most recent evaluation of contractor safety performance.

§ 68.180 Emergency response program.

(a) The owner or operator shall provide in the RMP the following information:

(1) Do you have a written emergency response plan?

(2) Does the plan include specific actions to be taken in response to an accidental releases of a regulated substance?

(3) Does the plan include procedures for informing the public and local agencies responsible for responding to accidental releases?

(4) Does the plan include information on emergency health care?

(5) The date of the most recent review or update of the emergency response plan;

(6) The date of the most recent emergency response training for employees.

(b) The owner or operator shall provide the name and telephone number of the local agency with which the plan is coordinated.

(c) The owner or operator shall list other Federal or state emergency plan requirements to which the stationary source is subject.

§ 68.185 Certification.

(a) For Program 1 processes, the owner or operator shall submit in the RMP the certification statement provided in §68.12(b)(4).

(b) For all other covered processes, the owner or operator shall submit in the RMP a single certification that, to the best of the signer’s knowledge, information, and belief formed after reasonable inquiry, the information submitted is true, accurate, and complete.

§ 68.190 Updates.

(a) The owner or operator shall review and update the RMP as specified in paragraph (b) of this section and submit it in a method and format to a
§ 68.200

central point specified by EPA prior to June 21, 1999.

(b) The owner or operator of a stationary source shall revise and update the RMP submitted under §68.150 as follows:

(1) Within five years of its initial submission or most recent update required by paragraphs (b)(2) through (b)(7) of this section, whichever is later.

(2) No later than three years after a newly regulated substance is first listed by EPA;

(3) No later than the date on which a new regulated substance is first present in an already covered process above a threshold quantity;

(4) No later than the date on which a regulated substance is first present above a threshold quantity in a new process;

(5) Within six months of a change that requires a revised PHA or hazard review;

(6) Within six months of a change that requires a revised offsite consequence analysis as provided in §68.36; and

(7) Within six months of a change that alters the Program level that applied to any covered process.

(c) If a stationary source is no longer subject to this part, the owner or operator shall submit a revised registration to EPA within six months indicating that the stationary source is no longer covered.

Subpart H—Other Requirements

SOURCE: 61 FR 31728, June 20, 1996, unless otherwise noted.

§ 68.200 Recordkeeping.

The owner or operator shall maintain records supporting the implementation of this part for five years unless otherwise provided in subpart D of this part.

§ 68.210 Availability of information to the public.

(a) The RMP required under subpart G of this part shall be available to the public under 42 U.S.C. 7414(c).

(b) The disclosure of classified information by the Department of Defense or other Federal agencies or contractors of such agencies shall be controlled by applicable laws, regulations, or executive orders concerning the release of classified information.

§ 68.215 Permit content and air permitting authority or designated agency requirements.

(a) These requirements apply to any stationary source subject to this part 68 and parts 70 or 71 of this chapter. The 40 CFR part 70 or part 71 permit for the stationary source shall contain:

(1) A statement listing this part as an applicable requirement;

(2) Conditions that require the source owner or operator to submit:

(i) A compliance schedule for meeting the requirements of this part by the date provided in §68.10(a); or

(ii) As part of the compliance certification submitted under 40 CFR 70.6(c)(5), a certification statement that the source is in compliance with all requirements of this part, including the registration and submission of the RMP.

(b) The owner or operator shall submit any additional relevant information requested by the air permitting authority or designated agency.

(c) For 40 CFR part 70 or part 71 permits issued prior to the deadline for registering and submitting the RMP and which do not contain permit conditions described in paragraph (a) of this section, the owner or operator or air permitting authority shall initiate permit revision or reopening according to the procedures of 40 CFR 70.7 or 71.7 to incorporate the terms and conditions consistent with paragraph (a) of this section.

(d) The state may delegate the authority to implement and enforce the requirements of paragraph (e) of this section to a state or local agency or agencies other than the air permitting authority. An up-to-date copy of any delegation instrument shall be maintained by the air permitting authority. The state may enter a written agreement with the Administrator under which EPA will implement and enforce the requirements of paragraph (e) of this section.

(e) The air permitting authority or the agency designated by delegation or agreement under paragraph (d) of this section shall, at a minimum:
(1) Verify that the source owner or operator has registered and submitted an RMP or a revised plan when required by this part;

(2) Verify that the source owner or operator has submitted a source certification or in its absence has submitted a compliance schedule consistent with paragraph (a)(2) of this section;

(3) For some or all of the sources subject to this section, use one or more mechanisms such as, but not limited to, a completeness check, source audits, record reviews, or facility inspections to ensure that permitted sources are in compliance with the requirements of this part; and

(4) Initiate enforcement action based on paragraphs (e)(1) and (e)(2) of this section as appropriate.

§ 68.220 Audits.

(a) In addition to inspections for the purpose of regulatory development and enforcement of the Act, the implementing agency shall periodically audit RMPs submitted under subpart G of this part to review the adequacy of such RMPs and require revisions of RMPs when necessary to ensure compliance with subpart G of this part.

(b) The implementing agency shall select stationary sources for audits based on any of the following criteria:

(1) Accident history of the stationary source;

(2) Accident history of other stationary sources in the same industry;

(3) Quantity of regulated substances present at the stationary source;

(4) Location of the stationary source and its proximity to the public and environmental receptors;

(5) The presence of specific regulated substances;

(6) The hazards identified in the RMP; and

(7) A plan providing for neutral, random oversight.

(c) Exemption from audits. A stationary source with a Star or Merit ranking under OSHA’s voluntary protection program shall be exempt from audits under paragraph (b)(2) and (b)(7) of this section.

(d) The implementing agency shall have access to the stationary source, supporting documentation, and any area where an accidental release could occur.

(e) Based on the audit, the implementing agency may issue the owner or operator of a stationary source a written preliminary determination of necessary revisions to the stationary source’s RMP to ensure that the RMP meets the criteria of subpart G of this part. The preliminary determination shall include an explanation for the basis for the revisions, reflecting industry standards and guidelines (such as AIChE/CCPS guidelines and ASME and API standards) to the extent that such standards and guidelines are applicable, and shall include a timetable for their implementation.

(f) Written response to a preliminary determination. (1) The owner or operator shall respond in writing to a preliminary determination made in accordance with paragraph (e) of this section. The response shall state the owner or operator will implement the revisions contained in the preliminary determination in accordance with the timetable included in the preliminary determination or shall state that the owner or operator rejects the revisions in whole or in part. For each rejected revision, the owner or operator shall explain the basis for rejecting such revision. Such explanation may include substitute revisions.

(2) The written response under paragraph (f)(1) of this section shall be received by the implementing agency within 90 days of the issue of the preliminary determination or a shorter period of time as the implementing agency specifies in the preliminary determination as necessary to protect public health and the environment. Prior to the written response being due and upon written request from the owner or operator, the implementing agency may provide in writing additional time for the response to be received.

(g) After providing the owner or operator an opportunity to respond under paragraph (f) of this section, the implementing agency may issue the owner or operator a written final determination of necessary revisions to the stationary source’s RMP. The final determination may adopt or modify the revisions contained in the preliminary
§ 68.220

(determination under paragraph (e) of this section or may adopt or modify the substitute revisions provided in the response under paragraph (f) of this section. A final determination that adopts a revision rejected by the owner or operator shall include an explanation of the basis for the revision. A final determination that fails to adopt a substitute revision provided under paragraph (f) of this section shall include an explanation of the basis for finding such substitute revision unreasonable.

(h) Thirty days after completion of the actions detailed in the implementation schedule set in the final determination under paragraph (g) of this section, the owner or operator shall be in violation of subpart G of this part and this section unless the owner or operator revises the RMP prepared under subpart G of this part as required by the final determination, and submits the revised RMP as required under §68.150.

(i) The public shall have access to the preliminary determinations, responses, and final determinations under this section in a manner consistent with §68.210.

(j) Nothing in this section shall preclude, limit, or interfere in any way with the authority of EPA or the state to exercise its enforcement, investigatory, and information gathering authorities concerning this part under the Act.
**APPENDIX A TO PART 68—TABLE OF TOXIC ENDPOINTS**

[As defined in § 68.22 of this part]

<table>
<thead>
<tr>
<th>CAS No.</th>
<th>Chemical name</th>
<th>Toxic endpoint (mg/L)</th>
</tr>
</thead>
<tbody>
<tr>
<td>107-02-8</td>
<td>Acrolein [2-Propenal]</td>
<td>0.0011</td>
</tr>
<tr>
<td>107-13-1</td>
<td>Acrylonitrile [2-Propenenitrile]</td>
<td>0.076</td>
</tr>
<tr>
<td>814-68-6</td>
<td>Acryl chloride [2-Propenoic chloride]</td>
<td>0.0099</td>
</tr>
<tr>
<td>107-18-6</td>
<td>Allyl alcohol [2-Propen-1-ol]</td>
<td>0.036</td>
</tr>
<tr>
<td>107-11-9</td>
<td>Allylamine [2-Propen-1-amine]</td>
<td>0.0032</td>
</tr>
<tr>
<td>7664-41-7</td>
<td>Ammonia (anhydrous)</td>
<td>0.14</td>
</tr>
<tr>
<td>7664-41-7</td>
<td>Ammonia (conc 20% or greater)</td>
<td>0.14</td>
</tr>
<tr>
<td>7784-36-1</td>
<td>Anhydrous trichloroethane</td>
<td>0.010</td>
</tr>
<tr>
<td>7784-42-1</td>
<td>Aniline</td>
<td>0.0019</td>
</tr>
<tr>
<td>10394-34-5</td>
<td>Boron trihalide [Borane, trichloro-]</td>
<td>0.010</td>
</tr>
<tr>
<td>7693-07-2</td>
<td>Boron trifluoride [Borane, trifluoro-]</td>
<td>0.028</td>
</tr>
<tr>
<td>335-42-4</td>
<td>Boron trifluoride compound with methyl ether (1:1) [Boron, trifluoro] [oxybis(methane)], T-4</td>
<td>0.023</td>
</tr>
<tr>
<td>7775-95-6</td>
<td>Bromine</td>
<td>0.0005</td>
</tr>
<tr>
<td>75-15-0</td>
<td>Carbon disulfide</td>
<td>0.016</td>
</tr>
<tr>
<td>7782-50-5</td>
<td>Chlorine</td>
<td>0.0087</td>
</tr>
<tr>
<td>10049-04-4</td>
<td>Chlorine dioxide [Chlorine oxide (OCl2)]</td>
<td>0.0028</td>
</tr>
<tr>
<td>67-66-3</td>
<td>Chloroform [Methane, trichloro-]</td>
<td>0.49</td>
</tr>
<tr>
<td>542-88-1</td>
<td>Chloromethyl ether [Methane, oxymethyl-]</td>
<td>0.00025</td>
</tr>
<tr>
<td>107-30-2</td>
<td>Chloromethyl methyl ether [Methane, chloromethoxy-]</td>
<td>0.0018</td>
</tr>
<tr>
<td>4170-30-3</td>
<td>Crotonaldehyde [2-Butenal]</td>
<td>0.029</td>
</tr>
<tr>
<td>123-73-9</td>
<td>Crotonaldehyde [Et-1, 2-Butenal, (Et-)]</td>
<td>0.029</td>
</tr>
<tr>
<td>516-77-4</td>
<td>Cyanogen chloride</td>
<td>0.030</td>
</tr>
<tr>
<td>108-91-8</td>
<td>Cyclohexylamine [Cyclohexanamine]</td>
<td>0.16</td>
</tr>
<tr>
<td>19287-45-7</td>
<td>Diborane</td>
<td>0.0011</td>
</tr>
<tr>
<td>75-78-5</td>
<td>Dimethylchlorosilane [Silane, dichlorodimethyl-]</td>
<td>0.025</td>
</tr>
<tr>
<td>57-14-7</td>
<td>1,1-Dimethylhydrazine [Hydrazine, 1,1-dimethyl-]</td>
<td>0.012</td>
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<tr>
<td>106-86-8</td>
<td>Epichlorohydrin [Oxirane, chloromethyl-]</td>
<td>0.076</td>
</tr>
<tr>
<td>107-15-3</td>
<td>Ethyleneimine [1,2-Ethanediamine]</td>
<td>0.49</td>
</tr>
<tr>
<td>151-56-4</td>
<td>Ethylenamine [Aziridine]</td>
<td>0.018</td>
</tr>
<tr>
<td>75-21-8</td>
<td>Ethylene oxide [Oxirane]</td>
<td>0.090</td>
</tr>
<tr>
<td>7782-41-4</td>
<td>Fluorine</td>
<td>0.0039</td>
</tr>
<tr>
<td>50-00-0</td>
<td>Formaldehyde (solution)</td>
<td>0.012</td>
</tr>
<tr>
<td>110-00-9</td>
<td>Furan</td>
<td>0.0012</td>
</tr>
<tr>
<td>302-01-2</td>
<td>Hydrazine</td>
<td>0.011</td>
</tr>
<tr>
<td>7647-01-0</td>
<td>Hydrochloric acid (conc 37% or greater)</td>
<td>0.030</td>
</tr>
<tr>
<td>74-90-8</td>
<td>Hydrochloric acid</td>
<td>0.011</td>
</tr>
<tr>
<td>7647-01-0</td>
<td>Hydrogen chloride (anhydrous) [Hydrochloric acid]</td>
<td>0.030</td>
</tr>
<tr>
<td>7664-33-3</td>
<td>Hydrogen fluoride [Hydrofluoric acid (conc 50% or greater) [Hydrofluoric acid]</td>
<td>0.016</td>
</tr>
<tr>
<td>7783-07-5</td>
<td>Hydrogen selenide</td>
<td>0.00066</td>
</tr>
<tr>
<td>7783-06-4</td>
<td>Hydrogen sulfide</td>
<td>0.042</td>
</tr>
<tr>
<td>13460-40-0</td>
<td>Iron, pentachloride [Iron bis(5-chloro-1-pentyl)</td>
<td>0.00644</td>
</tr>
<tr>
<td>78-82-0</td>
<td>Isobutyronitrile [Propenamide, 2-methyl-]</td>
<td>0.14</td>
</tr>
<tr>
<td>108-23-6</td>
<td>Isopropyl chloroformate [Carbonochloride acid, 1-methyl ester]</td>
<td>0.10</td>
</tr>
<tr>
<td>126-98-7</td>
<td>Methylacrylonitrile [2-Propenenitrile, 2-methyl-]</td>
<td>0.0007</td>
</tr>
</tbody>
</table>
### APPENDIX A TO PART 68—TABLE OF TOXIC ENDPOINTS—Continued

[As defined in §68.22 of this part]

<table>
<thead>
<tr>
<th>CAS No.</th>
<th>Chemical name</th>
<th>Toxic endpoint (mg/L)</th>
</tr>
</thead>
<tbody>
<tr>
<td>74–87–3</td>
<td>Methyl chloride [Methane, chloro-]</td>
<td>0.82</td>
</tr>
<tr>
<td>79–22–1</td>
<td>Methyl chlorofomate [Carbonochloridic acid, methyl ester]</td>
<td>0.0019</td>
</tr>
<tr>
<td>60–34–4</td>
<td>Methyl hydrazine [Hydrazine, methyl-]</td>
<td>0.0094</td>
</tr>
<tr>
<td>624–83–9</td>
<td>Methyl isocyanate [Methane, isocyanato-]</td>
<td>0.0012</td>
</tr>
<tr>
<td>74–90–1</td>
<td>Methyl mercaptan [Methanethiol]</td>
<td>0.049</td>
</tr>
<tr>
<td>556–64–9</td>
<td>Methyl thioformate [Thioformic acid, methyl ester]</td>
<td>0.085</td>
</tr>
<tr>
<td>75–79–6</td>
<td>Methyltrichlorosilane [Silane, trichloromethyl-]</td>
<td>0.018</td>
</tr>
<tr>
<td>13463–39–3</td>
<td>Nickel carbonyl</td>
<td>0.00967</td>
</tr>
<tr>
<td>7697–37–2</td>
<td>Nitric acid (conc 80% or greater)</td>
<td>0.028</td>
</tr>
<tr>
<td>10102–43–9</td>
<td>Nitric oxide (Nitrogen oxide (NO))</td>
<td>0.031</td>
</tr>
<tr>
<td>8014–95–7</td>
<td>Oleum (Fuming Sulfuric acid) [Sulfuric acid, mixture with sulfur trioxide]</td>
<td>0.010</td>
</tr>
<tr>
<td>79–21–0</td>
<td>Peracetic acid [Ethanesperoic acid]</td>
<td>0.010</td>
</tr>
<tr>
<td>594–42–3</td>
<td>Perchloromethyl mercaptan [Methanesulfenyl chloride, trichloro-]</td>
<td>0.0036</td>
</tr>
<tr>
<td>75–44–5</td>
<td>Phosgene (Carbonic dichloride)</td>
<td>0.00011</td>
</tr>
<tr>
<td>7803–51–2</td>
<td>Phosphine</td>
<td>0.0035</td>
</tr>
<tr>
<td>10025–87–3</td>
<td>Phosphorus oxychloride (Phosphoryl chloride)</td>
<td>0.0030</td>
</tr>
<tr>
<td>7719–12–2</td>
<td>Phosphorus trichloride (Phosphorous trichloride)</td>
<td>0.028</td>
</tr>
<tr>
<td>110–89–4</td>
<td>Piperidine</td>
<td>0.022</td>
</tr>
<tr>
<td>107–12–0</td>
<td>Propionitrile [Propanenitrile]</td>
<td>0.00037</td>
</tr>
<tr>
<td>109–61–5</td>
<td>Propyl chloroformate [Carbonochloridic acid, propylester]</td>
<td>0.010</td>
</tr>
<tr>
<td>75–55–8</td>
<td>Propyleneimine [Aziridine, 2-methyl-]</td>
<td>0.12</td>
</tr>
<tr>
<td>75–56–9</td>
<td>Propylene oxide [Oxirane, methyl-]</td>
<td>0.59</td>
</tr>
<tr>
<td>7446–09–5</td>
<td>Sulfur dioxide (anhydrous)</td>
<td>0.0078</td>
</tr>
<tr>
<td>7783–60–0</td>
<td>Sulfur tetrafluoride (Sulfur fluoride (SF4), (T-4))</td>
<td>0.0092</td>
</tr>
<tr>
<td>7446–11–9</td>
<td>Sulfur trioxide</td>
<td>0.010</td>
</tr>
<tr>
<td>75–74–1</td>
<td>Tetramethyllead [Plumbane, tetramethyl-]</td>
<td>0.0040</td>
</tr>
<tr>
<td>509–14–8</td>
<td>Tetranitromethane [Methane, tetranitro-]</td>
<td>0.0040</td>
</tr>
<tr>
<td>7790–45–0</td>
<td>Titanium tetrafluoride [Titanium chloride (TiCl4) (T-4)]</td>
<td>0.023</td>
</tr>
<tr>
<td>584–84–9</td>
<td>Toluene 2,4-disocyanate [Benzene, 2,4-disocyanato-1-methyl-]</td>
<td>0.0070</td>
</tr>
<tr>
<td>91–08–7</td>
<td>Toluene 2,6-disocyanate [Benzene, 1,3-disocyanato-2-methyl-]</td>
<td>0.0070</td>
</tr>
<tr>
<td>26471–62–5</td>
<td>Toluene disocyanate (unspecified isomer) [Benzene, 1,3-disocyanomethyl-]</td>
<td>0.0070</td>
</tr>
<tr>
<td>75–77–4</td>
<td>Trimethylchlorosilane [Silane, chlorotrimethyl-]</td>
<td>0.050</td>
</tr>
<tr>
<td>108–05–4</td>
<td>Vinyl acetate monomer [Acetic acid ethenyl ester]</td>
<td>0.26</td>
</tr>
</tbody>
</table>

Environmental Protection Agency

40 CFR Part 68
Accidental Release Prevention Requirements; Risk Management Programs Under Clean Air Act Section 112(r)(7), Amendments; Final Rule
ENVIORMENrAL PROTECTION AGENCY

40 CFR Part 68
[FRL—6214–9]
RIN 2050–AE46

Accidental Release Prevention Requirements; Risk Management Programs Under Clean Air Act Section 112(r)(7); Amendments

AGENCY: Environmental Protection Agency (EPA).

ACTION: Final rule.

SUMMARY: This action modifies the chemical accident prevention rule codified in 40 CFR Part 68. The chemical accident prevention rule requires owners and operators of stationary sources subject to the rule to submit a risk management plan (RMP) by June 21, 1999, to a central location specified by EPA. In this action, EPA is amending the rule to: add four mandatory and five optional RMP data elements, establish specific procedures for protecting confidential business information when submitting RMPs, adopt the government’s use of a new industry classification system, and make technical corrections and clarifications to Part 68. However, as stated in the proposed rule for these amendments, this action does not address issues concerning public access to offsite consequence analysis data in the RMP. DATES: The rule is effective February 5, 1999.

ADDRESSES: The docket is available for public inspection and copying between 8:00 a.m. and 5:30 p.m., Monday through Friday (except government holidays) at Room 1500, 401 M Street SW, Washington, DC 20460. A reasonable fee may be charged for copying.

FOR FURTHER INFORMATION CONTACT: Sicy Jacob or John Ferris, Chemical Emergency Preparedness and Prevention Office, Environmental Protection Agency (5104), 401 M Street SW, Washington, DC 20460, (202) 260–7249 or (202) 260–4043, respectively; or the Emergency Planning and Community Right-to-Know Hotline at 800–424–9346 (in the Washington, DC metropolitan area, (703) 412–9810). You may wish to visit the Chemical Emergency Preparedness and Prevention Office (CEPPO) Internet site, at www.epa.gov/ceppo.

SUPPLEMENTARY INFORMATION:

Regulated Entities

Entities potentially regulated by this action are those stationary sources that have more than a threshold quantity of a regulated substance in a process. Regulated categories and entities include:

<table>
<thead>
<tr>
<th>Category</th>
<th>Examples of regulated entities</th>
</tr>
</thead>
<tbody>
<tr>
<td>Chemical Manufacturers</td>
<td>Basic chemical manufacturing, petrochemicals, resins, agricultural chemicals, pharmaceuticals, paints, cleaning compounds.</td>
</tr>
<tr>
<td>Petroleum</td>
<td>Refineries.</td>
</tr>
<tr>
<td>Other Manufacturing</td>
<td>Paper, electronics, semiconductors, fabricated metals, industrial machinery, food processors.</td>
</tr>
<tr>
<td>Agriculture</td>
<td>Agricultural retailers.</td>
</tr>
<tr>
<td>Public Sources</td>
<td>Drinking water and waste water treatment systems.</td>
</tr>
<tr>
<td>Utilities</td>
<td>Electric utilities.</td>
</tr>
<tr>
<td>Other</td>
<td>Military and energy installations.</td>
</tr>
</tbody>
</table>

This table is not meant to be exhaustive, but rather provides a guide for readers to indicate those entities likely to be regulated by this action. The table lists entities EPA is aware of that could potentially be regulated by this action. Other entities not listed in the table could also be regulated. To determine whether a stationary source is regulated by this action, carefully examine the provisions associated with the list of substances and thresholds under § 68.130 and the applicability criteria under § 68.10. If you have questions regarding applicability of this action to a particular entity, consult the hotline or persons listed in the preceding FOR FURTHER INFORMATION CONTACT section.

I. Introduction and Background

A. Statutory Authority

These amendments are being promulgated under sections 112(r) and 301(a)(1) of the Clean Air Act (CAA) as amended (42 U.S.C. 7412(r), 7601(a)(1)).

B. Background

The 1990 CAA Amendments added section 112(r) to provide for the prevention and mitigation of accidental chemical releases. Section 112(r) mandates that EPA promulgate a list of "regulated substances," with threshold quantities. Processes at stationary sources that contain a threshold quantity of a regulated substance are subject to accidental release prevention regulations promulgated under CAA section 112(r)(7). EPA promulgated the list of regulated substances on January 31, 1994 (59 FR 4478) (the "List Rule") and the accidental release prevention regulations creating the risk management program requirements on June 20, 1996 (61 FR 31668) (the "RMP Rule"). Together, these two rules are codified as 40 CFR Part 68. EPA amended the List Rule on August 25, 1997 (62 FR 45132), to change the listed concentration of hydrochloric acid. On January 6, 1998 (63 FR 640), EPA amended the List Rule to delist Division 1.1 explosives (classified by DOT), to clarify certain provisions related to regulated flammable substances and to clarify the transportation exemption.

Part 68 requires that sources with more than a threshold quantity of a regulated substance in a process develop and implement a risk management program that includes a five-year accident history, offsite consequence analyses, a prevention
program, and an emergency response program. In Part 68, processes are divided into three categories (Programs 1 through 3). Processes that have no potential impact on the public in the case of accidental releases have minimal requirements (Program 1). Processes in Programs 2 and 3 have additional requirements based on the potential for offsite consequences associated with the worst-case accidental release and their accident history. Program 3 is also triggered if the processes are subject to OSHA’s Process Safety Management (PSM) Standard. By June 21, 1999, sources must submit to a location designated by EPA, a risk management plan (RMP) that summarizes their implementation of the risk management program.

When EPA promulgated the risk management program regulations, it stated that it intended to work toward electronic submission of RMPs. The Accident Prevention Subcommittee of the CAA Advisory Committee convened an Electronic Submission Workgroup to examine the technical and practical issues associated with creating a national electronic repository for RMPs. Based on workgroup recommendations, EPA is in the process of developing two systems, a user-friendly PC-based submission system (RMP*Submit) and a database of RMPs (RMP*Info).

The Electronic Submission Workgroup also recommended that EPA add some mandatory and optional data elements to the RMP and asked EPA to clarify how confidential business information (CBI) submitted in the RMP would be handled. Based on these recommendations and requests for clarifications, EPA proposed amendments to Part 68 on April 17, 1998 (63 FR 19216). These amendments proposed to replace the use of Standard Industrial Classification (SIC) codes with the North American Industry Classification System (NAICS) codes, add four mandatory data elements to the RMP, add five optional data elements to the RMP, establish specific requirements for an announcement of information claimed CBI, and make technical corrections and clarifications to the rule. EPA received 47 written comments on the proposed rule. Today’s rule reflects EPA’s consideration of all comments; major issues raised by commenters and EPA’s responses are discussed in Section III of this preamble. A summary of all comments submitted and EPA’s responses can be found in a document entitled, Accidental Release Prevention Requirements for Risk Management Programs Under Clean Air Act Section 112(r)(7); Amendments: Summary and Response to Comments, in the Docket (see ADDRESSES).

II. Summary of the Final Rule

NAICS Codes

On January 1, 1997, the U.S. Government, in cooperation with the governments of Canada and Mexico, adopted a new industry classification system, the North American Industry Classification System (NAICS), to replace the Standard Industrial Classification (SIC) codes (April 9, 1997, 62 FR 17288). The applicability of some Part 68 requirements (i.e., Program 3 prevention requirements) is determined, in part, by SIC codes, and Part 68 also requires the reporting of SIC codes in the RMP. Therefore, EPA is revising Part 68 to replace all references to “SIC code” with “NAICS code.” In addition, EPA is replacing, as proposed, the nine SIC codes subject to Program 3 prevention program requirements with ten NAICS codes, as follows:

- **NAICS Sector**
  - 3221 Pulp mills
  - 3241 Petroleum refineries
  - 3251 Petrochemical manufacturing
  - 325181 Alkalies and chlorine
  - 325188 All other inorganic chemical manufacturing
  - 325192 Other cyclic crude and intermediate manufacturing
  - 325199 All other basic organic chemical manufacturing
  - 325211 Plastics and resins
  - 325311 Nitrogen fertilizer
  - 32532 Pesticide and other agricultural chemicals

NAICS codes are either five or six digits, depending on the degree to which the sector is subdivided.

RMP Data Elements

As proposed, EPA is adding four new data elements to the RMP: latitude/longitude method and description, CAA Title V permit number, percentage weight of a toxic substance in a liquid mixture, and NAICS code for each process that had an accidental release reported in the five-year accident history. EPA is also adding five optional data elements: local emergency planning committee (LEPC) name, source or parent company e-mail address, source homepage address, phone number at the source for public inquiries, and status under OSHA’s Voluntary Protection Program (VPP).

Prevention Program Reporting

EPA is not revising Sections 68.170 and 68.175 as proposed. Prevention program reporting, therefore, will not be changed. EPA requires a prevention program for each portion of a process for which a Process Hazard Analysis (PHA) or hazard review was conducted. Instead, EPA plans to create functions within RMP*Submit to provide stationary sources with a flexible way of explaining the scope and content of each prevention program they implement at their facility.

Confidential Business Information

EPA is clarifying how confidential business information (CBI) submitted in the RMP will be handled. EPA has determined that the information required by certain RMP data elements does not meet the criteria for CBI and therefore may not be claimed as such. The Agency is also requiring submission of substantiation at the time a CBI claim is filed.

Finally, EPA is promulgating several of the technical corrections and clarifications, as proposed in the Federal Register, April 17, 1998 (63 FR 19216).

III. Discussion of Issues

EPA received 47 comments on the proposed rule. The commenters included chemical manufacturers, petroleum refineries, environmental groups, trade associations, a state agency, and members of the public. The major issues raised by commenters are addressed briefly below. The Agency’s complete response to comments received on this rulemaking is available in the docket (see ADDRESSES). The document is titled Accidental Release Prevention Requirements: Risk Management Programs Under Clean Air Act Section 112(r)(7); Amendments: Summary and Response to Comments.

A. NAICS Codes

Two commenters asked that sources be given the option to use either SIC codes or NAICS codes, or both, in their initial RMP because the NAICS system is new and may not be familiar to sources. EPA disagrees with this suggestion. EPA intends to provide several outreach mechanisms to assist sources in identifying their new NAICS code. RMP*Submit will provide a “pick list” that will make it easier for sources to find the appropriate code. Also, selected NAICS codes are included in the General Guidance for Risk Management Programs (July 1998) and in the industry-specific guidance documents that EPA is developing. EPA will also utilize the Emergency Planning and Community Right-to-Know Hotline at 800–424–9346 (or 703–412–9810) and its web site at www.epa.gov/ceppo/, to assist sources in determining the correct NAICS code. EPA also notes that the Internal Revenue Service is planning to require businesses to...
provide NAICS-based activity codes on their 1998 tax returns, so many sources will have become familiar with their NAICS codes by the June 1999 RMP deadline. EPA believes it is necessary and appropriate to change from SIC codes to NAICS codes at this time. EPA recognizes that NAICS codes were developed for statistical purposes by the Office of Management and Budget (OMB). In the notice of April 9, 1997 (62 FR 17288) OMB stated that the "[u]se of NAICS for nonstatistical purposes (e.g., administrative, regulatory, or taxation) will be determined by the agency or agencies that have chosen to use the SIC for nonstatistical purposes." EPA has determined that NAICS is appropriate in this rule for several reasons. First, the reason the SIC codes were replaced by NAICS codes is because the SIC codes no longer accurately represent today's industries. The SIC codes will become more obsolete over time because OMB will no longer be supporting the SIC codes; therefore, no new or modified SIC codes will be developed to reflect future changes in industries. Second, as the SIC codes become obsolete, most users of SIC codes will likely change to NAICS codes over time, so future data sharing and consistency will be enhanced by use of NAICS codes in the RMP program. Third, through this rulemaking process, EPA has analyzed specific conversions of SIC codes to NAICS codes for the RMP program and was able to identify NAICS codes that were applicable to fulfilling the RMP program purposes. Finally, because the RMP reporting requirement is new, it is reasonable to begin the program with NAICS codes now rather than converting to them later.

Three commenters expressed support for the ten NAICS codes that EPA proposed to use in place of the nine SIC codes referenced in section 68.10(d)(1) of Part 68 and one commenter partially objected. Section 68.10(d)(1) provides that processes in the referenced codes are subject to Program 3 requirements (if not excluded for reasons). One commenter objected to EPA's proposal to replace the SIC code for pulp and paper mills with only the NAICS code for pulp mills that do not also produce paper or board. The commenter asked EPA to reexamine the accident history of paper and paperboard mills. As discussed in the preamble of the proposed rule, EPA reviewed the accident history data prior to proposing the new NAICS codes. Neither facilities that classify themselves as paper mills (NAICS Code 322121) nor paperboard mills (NAICS code 32213) met the accident history criteria that EPA used to select industrial sectors for Program 3.

EPA notes that a pulp process at a paper or a paperboard mill may still be subject to Program 3 as long as the process contains more than a threshold quantity of a regulated substance and is not eligible for Program 1. Section 68.10(d)(1) uses industrial codes to classify processes, not facilities as a whole. Since section 68.10(d)(1) will continue to list the code for pulp mills, pulpmaking processes will continue to be subject to Program 3. In addition, under section 68.10(d)(2), paper processes will be in Program 3 (unless eligible for Program 1) if they are subject to OSHA's Process Safety Management (PSM) standard. Most pulp and paper processes are, in fact, subject to this standard.

One commenter objected to assigning NAICS codes to a process rather than the source as a whole. EPA first notes that the requirement to assign a SIC code to a process was adopted in the original RMP rule two years ago. Today's rule does not change that requirement except to substitute NAICS for SIC codes. In any event, EPA is today modifying Part 68 to clarify that sources provide the NAICS code that "most closely corresponds to the process." EPA believes that assigning an industry code to a process will help implementing agencies and the public understand what the covered process does; using the code makes it possible to provide this information without requiring a detailed explanation from the source. In addition, the primary NAICS code for a source as a whole may not reflect the activity of the covered process.

B. RMP Data Elements

EPA proposed to add, as optional RMP data elements: local emergency planning committee (LEPC), source (or parent company) E-mail address, source homepage address, phone number at the source for public inquiries, and OSHA Voluntary Protection Program (VPP) status. EPA also proposed to add, as mandatory data elements: method and description of latitude/longitude, Title V permit number, percent weight of a toxic substance in a liquid mixture, and NAICS code (only in the five-year accident history section).

Commenters generally supported the new optional data elements. One commenter requested that the optional elements be made mandatory. EPA disagrees with this comment. While the elements are useful, many sources coverages will not have e-mail addresses or home pages. The RMP will provide both addresses and phone numbers so that the public will have methods to reach the source. EPA has learned that in some areas there are no functioning LEPCs, therefore, at this time, EPA will not add this as a mandatory data element. However, in most cases, the LEPC for an area can be determined by contacting the local government or the State Emergency Response Commission (SERC) for which the area is located. Therefore, reporting these data elements will remain optional at this time.

One commenter supported adding the listing of local emergency planning committee in the RMP data elements as an optional data element. The commenter stated that, although it is an optional data element, this listing will enhance the ability of local responders and emergency planners to adequately prepare and train for emergency events. Of the data elements that were proposed to be mandatory, one commenter objected to the addition of latitude/longitude method and description. The commenter stated that it was not clear in the proposal why the method and description information is needed. EPA is seeking latitude/longitude method and description in accordance with its Locational Data Policy. Several EPA regulations require sources to provide their latitude and longitude, so that EPA can more readily locate facilities and communicate data between Agency offices. Sharing of data between EPA offices reduces duplication of information. Latitude/longitude method and description information provided by EPA, and other users of the data, to rectify discrepancies that may appear in the latitude and longitude information provided by the source under various EPA requirements. Documentation of the method by which the latitude and longitude are determined and a description of the location point referenced by the latitude and longitude (e.g., administration building) will permit data users to evaluate the accuracy of those coordinates, thus addressing EPA data sharing and integration objectives.

EPA believes this information will also facilitate EPA-State coordination of environmental programs, including the chemical accident prevention rule. The State/EPA Data Management Program is a successful multi-year initiative linking State environmental regulatory agencies and EPA in cooperative action. The Program's goals include improvements in data quality and data integration based on location identification. The latitude/longitude method and description will be added to the existing RMP data...
elements. RMP*Submit will provide a list of methods and descriptions from which sources may choose.

EPA also proposed to require that sources report the percentage weight (weight percent) of a toxic substance in a mixture in the offsite consequence analysis (OCA) and the accident history sections of the RMP. This information is necessary for users of RMP data to understand worst case and alternative release scenarios that have been modeled. EPA has decided to require reporting of the weight percent of toxic substance in a liquid mixture because this information is necessary to understand the volatilization rate, which determines the downwind dispersion distance of the substance. The volatilization rate is affected by the vapor pressure of the substance in the mixture. For example, a spill of 70 percent hydrofluoric acid (HF) will volatilize more quickly than a spill of the same quantity of HF in a 50 percent solution; consequently, over a 10-minute period, the 70 percent solution will lose more hydrofluoric acid than the 50 percent solution. Reviewers of the RMP data, including local emergency planning committees, need to know the weight percent to be able to evaluate the results reported in the offsite consequence analysis and the impacts reported in the accident history. Without knowing the weight percent of the substance in the mixture, users of the data may compare scenarios or incidents that appear to involve the same chemical in the same physical state, but in fact involve the same chemical held in a different physical state.

One commenter stated that for gas mixtures, percentage by volume (or volume percent) should be required to be reported rather than weight percent. In this final rule, EPA does not require reporting of the weight percent (or volume percent) of a regulated substance in a gas mixture. If a source handles regulated substances in a gaseous mixture (e.g., chlorine with hydrogen chloride), the quantity of a particular regulated substance in the mixture is what is reported in the RMP, since that is what would be released into the air. Its percentage weight in the mixture is irrelevant.

Another commenter objected to this data element, claiming that it could result in reverse engineering and create a competitive disadvantage. EPA does not believe that this requirement would create a competitive disadvantage, since similar information is available to the public under Emergency Planning and Community Right-to-Know Act (EPCRA) of 1986. Even so, if it were to have such an effect, sources can claim this element as CBI if it can meet the criteria for CBI claims in 40 CFR Part 2. Another commenter stated that the public would be concerned if the percentages did not add to 100, in the event that the source handles both regulated and non-regulated substances. EPA believes that because a source must model only one substance in a release scenario, the source need not report the percentages of the other substances in the mixture. Therefore, it is expected that the weight percent for mixtures would not always add up to 100, because the mixture could contain non-regulated substances.

A third commenter suggested that requiring sources to report percentage weight of a toxic substance in a liquid mixture would create confusion with the reporting of mixtures containing flammable regulated substances. In the January 6, 1998 rule (63 FR 640), EPA clarified that flammable regulated substances in mixtures are only covered by the RMP rule if the entire mixture meets the National Fire Protection Association (NFPA) criteria of 4, thus the entire mixture becomes the regulated substance. As a result, the percentage of flammables in a mixture is not relevant under the rule and the requirement to report the percentage weight will only apply to toxic substances in a liquid mixture.

Finally, in the Federal Register notice of June 20, 1996 (61 FR 31688), EPA clarified the relationship between the risk management program and the air permit program under Title V of the CAA for sources subject to both requirements. Under section 502(b)(5)(A), permitting authorities must have the authority to assure compliance by all covered sources with each applicable CAA standard, regulation or requirement, including the regulations implementing section 112(r)(7). Requiring sources covered by Title V and section 112(r) to provide their Title V permit number will help Title V permitting authorities assure that each source is complying with the RMP rule.

In summary, with the exception of adding the phrase “that most closely corresponds to the process” in sections 68.42(b)(4), 68.160(b)(7), 68.170(b), and 68.175(b), EPA has decided to finalize the optional and mandatory data elements as they were proposed.

C. Prevention Program Reporting

The final RMP rule, issued June 20, 1996 (61 FR 31668), requires sources to report their prevention program for each “process.” Because the applicable definition of “process” is broad, multiple production and storage units might be a single, complex “process.” However, the Agency realizes that some elements of a source’s prevention program for a process may not be applicable to every portion of the process. In such a situation, reporting prevention program information for the process as a whole could be misleading without an explanation of which prevention program element applies to which part of the process. In order to get more specific information on which prevention program practices apply to different production and storage units within a process, EPA proposed to revise the rule to require prevention program reporting for each part of the process for which a separate process hazard analysis (PHA) or hazard review was conducted. EPA further proposed deleting the second sentence from both sections 68.170(a) and 68.175(a), which presently states that, “[i]f the same information applies to more than one covered process, the owner or operator may provide the information only once, but shall indicate to which process the information applies.”

A number of industry commenters objected to the proposed revisions as wrongly assuming that a one-to-one relationship exists between a prevention program and a PHA. The commenters asserted that EPA’s proposed revision did not reflect how facilities conduct PHAs or implement prevention measures and would cause significant duplicate reporting, creating unnecessary extra work for facility personnel. One commenter explained that depending on a source’s circumstances, it might conduct a PHA for each production line, including all of its different units, or it might conduct a PHA for each common element of its different production lines. Accordingly, the commenters claimed that EPA’s proposal to require the owner/operator to submit separate prevention program information for every portion of a process covered by a PHA would result in multiple submissions of much of the same material, and would add no value to process safety or accidental release prevention. Commenters also opposed the deletion of the second sentence in sections 68.170(a) and 68.175(a). One commenter noted that many of the elements of the prevention program will not only be common to a process, but will be common to an entire stationary source. Thus commenters argued that EPA’s proposals would result in redundant submittals and place an unjustified burden on the regulated community.

EPA acknowledges that PHAs do not necessarily determine the scope of prevention program measures. Moreover, EPA agrees that duplicative
reporting should be reduced as much as possible. At the same time, EPA, implementing agencies, and other users of RMP data need to have information that is detailed enough to understand the hazards posed by, and the safety practices used for, particular parts of processes and equipment. EPA recognizes that some aspects of prevention programs are likely to be implemented facility-wide, rather than on a process or unit basis, whereas other aspects may apply to a particular process or only to particular units within a process. For example, most sources are likely to develop an employee participation plan and a system for hot work permits facility-wide, rather than on a process or unit basis. For sources having processes that include several units (e.g., multiple reactors or purification systems), the hazards, process controls, and mitigation systems may vary among the individual units. For example, one may have a deluge fire control system while another may have a runaway reaction quench system.

EPA has concluded that its proposed changes to prevention program reporting would not lead sources to prepare RMPs that accurately and efficiently communicate the hazards posed by different aspects of covered processes and the safety practices used to address those hazards. The Agency now believes that no rule changes are necessary to ensure that RMPs convey that information. The current rule already requires prevention program reporting to accurately reflect the changes that have been how to efficiently convey that information in sufficient detail. EPA believes that its electronic program for submitting RMPs can be designed to provide for sufficient specificity in prevention program reporting without requiring duplicative reporting. In particular, the Agency plans to create a comment/text field in RMP*Submit for specifying which parts of a prevention program apply to which portions of a particular process. For example, if a deluge system only applies to a certain part of the overall process, the source would indicate in the comment/text screen the portions of the process to which the deluge system applies.

To reduce the burden of reporting, EPA also plans to create a function in RMP*Submit which will allow a source to automatically copy prevention program data previously entered for one process to fill blank fields in another process's prevention program. The source could then edit any of the data elements that are different. For example, where the prevention programs for two processes are identical (e.g., two identical storage tanks that are considered separate processes), the source could copy the data entered for one to fill in the blank field for the other. If some of the data elements vary between the prevention programs, the source will be able to autofill and change only those items that vary among processes or units.

Although the autofill option will minimize the burden of reporting common data elements for those sources filing electronically, EPA has decided not to delete the sentence, in both sections 68.170(a) and 68.175(a), which states, "[i]f the same information applies to more than one covered process, the owner or operator may provide the information only once, but shall indicate to which processes the information applies ," as proposed.

D. Confidential Business Information (CBI)

1. Background

A central element of the chemical accident prevention program as established by the Clean Air Act and implemented by Part 68 is providing state and local governments and the public with information about the risk of chemical accidents in their communities and what stationary sources are doing to prevent such accidents. As explained in the preamble to the final RMP rule (61 FR 31318, June 20, 1996), every covered stationary source is required to develop and implement a risk management program and provide information about that program in its RMP. Under CAA section 112(r)(7)(B)(iii), a source’s RMP must be registered with EPA and also submitted to the Federal Chemical Safety and Hazard Investigation Board ("the Board"), the state in which the source is located, and any local entity responsible for emergency response or planning. That section also provides that RMPs “shall be available to the public under section 114(a)” of the CAA. Section 114(c) gives the public access to information obtained under the Clean Air Act except for information (other than emission data) that would divulge trade secrets.

As noted previously, in the final RMP rule EPA announced its plan to develop a centralized system for submitting electronic versions of RMPs that would reduce the paperwork burden on both industry and receiving agencies and provide ready public access to RMP data. Under the system, a covered source would submit its RMP on a computer diskette, which would be entered into a central database that all interested parties could access electronically. The system would thus make it possible for a single RMP submission to reach all interested parties, including those identified in section 112(r)(7)(B)(iii).

An important assumption underlying the Agency’s central submission plan was that RMPs would rarely, if ever, contain confidential business information (CBI). Following publication of the final rule, concerns were raised that at least some of the information required to be reported in RMPs could be CBI in the case of particular sources. While the June 20, 1996 rule provided for protection of CBI under section 114(c) (see section 68.210(a)), EPA was asked to address how CBI would be protected in the context of the electronic programs being developed for RMP submission and public access.

In the April 17, 1998 proposal to revise the RMP rule, EPA made several proposals concerning protection of CBI. It first reviewed the information requirements for RMPs (sections 68.155–185) and proposed to find that certain required data elements would not entail divulging information that could meet the test for CBI set forth in the Agency’s comprehensive CBI regulations at 40 CFR Part 2.

Information provided in response to those requirements could not be claimed CBI. EPA also requested comment on whether some information that might be claimed as CBI (e.g., worst-case release rate or duration) would be “emission data” and thus publicly available under section 114(c) even if CBI.

EPA administers a variety of statutes pertaining to the protection of the environment, each with its own data collection requirements and requirements for disclosure of information to the public. In the implementation of these statutes, the Agency collects emission, chemical, process, waste stream, financial, and other data from facilities in many, if not most, sectors of American business. Companies may consider some of this information vital to their competitive position. 40 CFR section 2.208.
In the course of implementing statutes, the Agency may have a need to communicate some or all of the information it collects to the public as the basis for a rulemaking, to its contractors, or in response to requests pursuant to the Freedom of Information Act (FOIA). Information found to be CBI is exempt from disclosure under FOIA. To manage both CBI claims and FOIA requests, EPA has promulgated in 40 CFR Part 2, Subpart B a set of procedures for reviewing CBI claims, releasing information found not to be CBI, and where authorized, disclosing CBI. Subpart B lists the criteria that information must meet in order to be considered CBI, as well as the special handling requirements the Agency must follow when disclosing CBI to authorized representatives.

For RMP requirements that might entail divulging CBI, EPA proposed that a source be required to substantiate a CBI claim to EPA at the time that it makes its annual RMP under EPA’s Part 2 regulations, a source claiming CBI generally is required to substantiate the claim only when EPA needs to make the information part of some proceeding (e.g., a rulemaking) or EPA receives a request from the public (e.g., under the Freedom of Information Act (FOIA)) for the information. In view of the public information function of RMPs and the interest already expressed by members of the public in them, EPA proposed “up-front substantiation” of CBI claims to ensure that information not meeting CBI criteria would be made available to the public as soon as possible. This approach of requiring up-front substantiation is the same as that used for trade secret claims filed under the Emergency Planning and Community Right-to-Know Act (EPCRA) of 1986.  

In addition, EPA proposed that any source claiming CBI submit two versions of its RMP: (1) a redacted (“sanitized”), electronic version, which would become part of RMP+Info, and (2) an unsanitized (unredacted) paper copy of the RMP (see proposed section 68.151(c)). The electronic database of RMPs would contain only the redacted version unless and until EPA ruled against all or part of the source’s CBI claim, in keeping with the Part 2 procedures. In this way, the public would have access only to the non-CBI elements of sources’ RMPs. EPA further stated that state and local agencies could receive the unredacted RMPs by requesting them from EPA under the Part 2 regulations. Those regulations authorize EPA to provide CBI to an agency having implementation responsibilities under the CAA if the agency either demonstrates that it has the authority under state or local law to compel such information directly from the source or that it will “provide adequate protection to the interests of affected businesses” (40 CFR 2.301(h)(3)).

The following sections of this preamble summarize and respond to the comments EPA received on the CBI-related aspects of its proposal. At the outset, however, EPA wants to emphasize that it does not anticipate many CBI claims being made in connection with RMPs. The Agency developed the RMP data elements with the issue of CBI in mind. It sought to define data elements that would provide basic information about a source’s risk management program without requiring it to reveal CBI. To have done otherwise would have risked creating RMPs that were largely unavailable to the public. EPA continues to believe that the required RMP data elements will rarely require that a business divulge CBI. The Agency will carefully monitor the CBI claims made. If it appears that the number of claims being made is jeopardizing the public information objective of the chemical accident prevention program, EPA will consider ways of revising RMPs, including rulemakings or revising the underlying program, to ensure that important health and safety information is available to the public.

2. RMP Data Elements Found Not CBI

Fifteen commenters representing environmental groups and members of the public opposed allowing some or all RMP data to be claimed as CBI in light of the public’s interest in the information RMPs will provide. A number of commenters urged EPA not to allow the following RMP data elements (and supporting documents) to be claimed as CBI:

- Mitigation measures considered by the firm in its offsite consequence analysis
- Major process hazards identified by the firm
- Process controls in use
- Mitigation systems in use
- Monitoring and detection systems in use, and
- Changes since the last hazard review.

In addition, one commenter contended that even chemical identity and quantity should be ineligible for CBI protection, since the requirement to submit an RMP only applies to facilities using a few well-known, extremely hazardous chemicals, and the public’s right to know should always outweigh a company’s claim to CBI.

Along the same lines, a number of commenters urged EPA to develop a “corporate sunshine rule” that would allow confidentiality concerns to be overridden if the protected information is needed by the public and experts to understand and assess safety issues. Another commenter recommended that a business claiming a chemical’s identity as CBI should be required to provide the generic name of the chemical and information about its adverse health effects so the public can determine the potential risks.

One commenter argued that some of the RMP data that EPA suggested could reveal CBI, (e.g., release rate), were not “emission data,” because the worst case scenario data are theoretical estimates, and do not represent any real emissions, past or present.

Representatives of the chemical and petroleum industries disagreed with EPA’s proposal to list the data elements that EPA believed could not reveal CBI in any case. These commenters asserted that EPA could not anticipate all the ways in which information required by a data element might reveal CBI, and accordingly urged the Agency to make...
case-by-case determinations on CBI claims. They also contended that “emission data” under section 114(c) does not extend to data on possible, as opposed to actual, emissions, and thus that RMP information concerning potential accidental releases would not qualify as “emission data,” which must be made available to the public.

As pointed out above, an important purpose of the chemical accident prevention program required by section 112(r) is to inform the public of the risk of accidents in their communities and the methods sources are employing to reduce such risks. EPA therefore believes that as much RMP data as possible should be available to the public as soon as possible. However, section 112(r)(7)(B)(iii) requires that RMPs be made “available to the public under section 114(c),” which provides for protection of trade secret information (other than emission data). Given the statute’s direction to protect whatever trade secret information is contained in an RMP, EPA is not authorized to release such information even when the public’s need for such information arguably outweighs a business’ interest in its confidentiality. The Agency also cannot issue a “corporate sunshine rule” that conflicts with existing law requiring EPA (and other agencies) to protect trade secret information.

As explained above (and in more detail in the proposed rule), EPA examined each RMP data element to determine which would require information that might, depending on a business’ circumstances, meet the CBI criteria set forth in EPA’s regulations implementing section 114(c) and other information-related legal requirements. The purpose of this exercise was to both protect potential trade secret information and promote the public information purpose of RMPs by identifying which RMP information might reveal CBI in a particular case and by precluding CBI claims for information that could not reveal CBI in any case. EPA presented the results of its analysis and an explanation of why certain data elements could entail the reporting of CBI depending on a business’ circumstances and why others could not. No commenter provided any specific examples or explanations that contradicted the Agency’s rationale for its determinations of which data elements could or could not result in reporting of CBI.

However, EPA is deleting from the list of 40 CFR Part 68.151(b)(1) the reference to 40 CFR Part 68.160(b)(9), to allow for the possibility of the number of full-time employees at the stationary source to be claimed as CBI. Upon further review, EPA was unable to determine that providing the number of employees at the stationary source could never entail divulging information that could meet the test for CBI set forth in the Agency’s comprehensive CBI regulations at 40 CFR Part 2. Therefore, EPA has removed this element from the list of data elements that can not be claimed CBI in Part 68. With this exception, EPA is promulgating the list of RMP data elements for which CBI claims are precluded, as proposed (Section 68.151(b)).

EPA’s justifications for its specific CBI findings appear in an appendix to this preamble. A more detailed analysis of all RMP data elements and CBI determinations is available in the docket (see ADDRESSES). The Agency continues to find no reasonable basis for anticipating that the listed elements will in any case require a business to reveal CBI that is not “emission data.” The information required by each of the listed data elements either fails to meet the criteria for CBI set forth in EPA’s regulations at Part 2 or meets the Part 2 definition of “emission data.” In many cases, the information is available to the public through other reports filed with EPA, states, or local agencies (e.g., reports required by Emergency Planning and Community Right-to-Know Act (EPCRA) sections 312 and 313 provide general facility identification information and reports of most accidental releases are available through several Federal databases including EPA’s Emergency Release Notification System and Accidental Release Information Program databases). In order to preclude CBI claims for other data elements, the Agency would have to show that the information required by a data element either was “emission data” under section 114(c) or could not, under any circumstances, reveal CBI. As explained below, EPA does not believe such a showing can be made for any of the data elements not on the list. Therefore, CBI claims made for information required by data elements not on the list will be evaluated on a case-by-case basis according to the procedures contained in 40 CFR Part 2 (except that substantiation will have to accompany the claims, as discussed below).

The Agency agrees with the commenters who argued that information about potential accidental releases is not “emission data” under section 114(c). EPA’s existing policy statement (see 56 FR 7042, Feb. 21, 1991) on what information may be considered “emission data” was developed to implement sections 110 and 114(a) of the CAA, which the Agency generally invokes when it seeks to gather technical data from a source about its actual emissions to the air. While the policy is not explicitly limited in its scope, EPA believes it would be inappropriate to apply it to RMP data elements concerning hypothetical, as opposed to actual, releases to the air. Under the definition of “emission data” contained in Part 2, information is “emission data” if it is (1) “necessary to determine the identity, amount, frequency, concentration, or other characteristics * * * of any emission which has been emitted by the source,” (2) “necessary to determine the identity, amount, frequency, concentration, or other characteristics * * * of the emissions which, under an applicable standard or limitation, the source was authorized to emit;” or (3) general facility identification information regarding the source which distinguishes it from other sources (40 CFR section 2.301(a)(2)(i) (emphasis added)). Under these criteria, EPA has concluded that only the RMP data elements relating to source-level registration information (sections 68.160(b)(1)–(6), (8)–(13)) and the five-year accident history (section 68.168) are “emission data.” Of the RMP data elements, only the five-year accident history involves actual, past emissions to the environment; the other data elements would not, therefore, qualify as “emission data” under the first prong of the Part 2 definition. Moreover, the data elements relating to a source’s offsite consequence analysis, prevention program and emergency response program do not attempt to identify or otherwise reflect “authorized” emissions; the data elements instead reflect the source’s potential for accidental releases. Accordingly, these data elements would not be “emission data” under the second prong of the definition. As for the third prong, some of the source-level data are “emission data” because they help identify a source. Most other RMP data elements are reported on a process level and are not generally used to distinguish one source from another.

The Agency believes it is unable to show that the remaining data elements could not, under any circumstances, reveal CBI. EPA continues to believe that it is theoretically possible for the remaining data elements (the elements not listed in section 68.151(b)) to reveal CBI either directly or through reverse engineering, depending on the circumstances of a particular case. At the same time, EPA believes that, in practice, the remaining data elements will rarely reveal CBI. The purpose of
the data in the RMP is for a source to articulate its hazards, and the steps it takes to prevent accidental releases. In general, the kinds of information specifying the source’s hazards and risk management program are not likely to be competitively sensitive.

In particular, covered processes at the vast majority of stationary sources subject to the RMP rule are too common and well-known to support a CBI claim for information related to such processes. For example, covered public drinking water and wastewater treatment plants generally use common regulated substances in standard processes (i.e., chlorine used for disinfection). Also, covered processes at many sources involve the storage of regulated substances that the sources sell (e.g., propane, ammonia), so the processes are already public knowledge. Other covered processes involve the use of well-known combinations of regulated substances such as refrigerants. RMP information regarding these types of processes should not include CBI.

Even in the case of unusual or unique processes, it is generally unlikely that RMP information could be used to reveal CBI through reverse engineering. To begin with, required RMP information is general enough that it is unlikely to provide a basis for reverse engineering a process. For example, a source must report in its RMP whether overpressurization is a hazard and whether relief valves are used to control pressure, but it is not required to report information on actual pressures used, flow rates, chemical composition, or the configuration of equipment. Moreover, while RMP information may provide some data that could be used in an attempt to discover CBI information through reverse engineering, it typically will not provide enough data for such an attempt to succeed, because the source is not required to provide a detailed description of the chemistry or production volume of the process. Businesses claiming CBI based on the threat of reverse engineering will be required to show how reverse engineering could in fact succeed with the information that the RMP would otherwise make public, together with other publicly available information. A business unable to do so will have its claim denied.

While EPA is requiring that a source claiming a chemical’s identity as CBI provide the generic category or class name of the chemical, the RMP does not require sources to provide information about how to provide the effects of the chemical. Chemicals were included in the section 112(r) program because they are acutely toxic or flammable; health effects related to chronic exposure were not considered because they are addressed by other rules (see List Rule at 59 FR 4481). EPA believes that generic names are sufficient to indicate the general health concerns from short-term exposures. Should a member of the public desire more information, EPA encourages the use of EPCRA section 322(h), which provides a means for the public to obtain information about the adverse health effects of a chemical covered by that statute, where the chemical’s identity has been claimed a trade secret. The public will find this provision of EPCRA useful because most sources subject to the RMP rule are also subject to EPCRA.

3. Up-front Substantiation of CBI Claims

One commenter supported the proposal to require CBI claims to be substantiated at the time they are made. Another commenter stated that there is no compelling need to require up-front substantiation of CBI claims when up-front substantiation would place a sizable burden on both industry and EPA and would be in direct conflict with the Paperwork Reduction Act. The commenter claimed that, with the exception of EPCRA, where a submitter is allowed to claim only one data element—chemical identity—as CBI, it is EPA’s standard procedure not to require submitters to provide written substantiation unless a record has been requested. Further, the commenter stated that the Agency has not shown any reason for departing from that procedure in this rule. EPA believes that requiring up-front substantiation of CBI claims made for RMP data has ample precedent, is fully consistent with the Agency’s CBI regulations and the Paperwork Reduction Act, and is critical to achieving the public information purposes of the accident prevention program. EPCRA is not the only example of up-front substantiation requirement. The Agency has also required up-front substantiation in several other regulatory contexts, including those where, like here, providing the public with health and safety information is an important objective (see e.g., 40 CFR section 725.94, 40 CFR section 710.38, and 40 CFR section 720.85 (regulations promulgated under Toxic Substances Control Act)).

Even under its general CBI regulations, the Agency need not wait for a request to release data to require business to substantiate their CBI claims. When EPA expects to get a request to release data claimed confidential, the Agency is to initiate “at the earliest practicable time” the regulations’ procedures for making CBI determinations (40 CFR section 2.204(a)(3)). Those procedures include calling on affected businesses to substantiate their claims (see 40 CFR section 2.204(e)). Since state and local agencies, environmental groups, academics and others have already indicated their interest in obtaining complete RMP data (see comments received on this rulemaking, available in the DOCKET), EPA fully expects to get requests for RMP data claimed CBI. Consequently, even if EPA did not establish an up-front substantiation requirement in this rule, under the Agency’s general CBI regulations it could require businesses claiming CBI for RMP data to substantiate their claims without first receiving a request to release the data. Establishing an up-front requirement in this rule will simply allow EPA to obtain substantiation of CBI claims without having to request it in every instance. Requiring up-front substantiation for RMP CBI claims is consistent with the Paperwork Reduction Act. Any burden posed by this requirement has already been evaluated as part of the Information Collection Request (ICR) associated with this rulemaking. EPA disagrees that up-front substantiation will impose a substantial or undue burden. As noted above, under EPA’s current CBI regulations, a source claiming CBI could and probably would be required to provide substantiation for its claim, in view of the pressure to do so in RMP information. A requirement to submit substantiation with the claim should thus make little difference to the source. Moreover, a source presumably does not make any claim of CBI lightly. Before filing a CBI claim, the source must first determine whether the claim meets the criteria specified in 40 CFR section 2.208. Up-front substantiation only requires that the source document that determination at the time it files its claim. Since it would be sensible for a source to document the basis of its CBI claim for its own purposes (e.g., in the case of a request for substantiation), EPA expects that many sources already prepare documentation for their CBI claims by the time they file them. Also, submitting substantiation at the time of claim reduces any additional burden later, such as reviewing the Agency’s request, retrieving the relevant information, etc. Therefore, providing documentation at the time of filing should impose no additional burden. In view of the public information function of RMPs, EPA believes that up-front substantiation is clearly warranted.
for CBI claims made for RMP data. Up-front substantiation will ensure that sources filing claims have carefully considered whether the data they seek to protect in fact meets the criteria for protection. Given the public interest already expressed in RMP data, EPA expects that CBI claims for RMP data will have to be substantiated at some point. Up-front substantiation will save EPA and the public time and resources that would otherwise be required to respond to each CBI claim with a request for substantiation. EPA is therefore promulgating the up-front substantiation requirement as proposed.  

4 State and Local Agency Access to Unredacted RMPs

One commenter objected to EPA's statement in the proposal that it would provide unredacted (unsanitized) versions of the RMPs to a state and local agency only upon meeting the criteria required by the EPA's CBI rules at 40 CFR Part 2. The commenter, an association of fire fighters, argued that the Agency's position was inconsistent with CAA section 112(r)(7)(B)(iii), which provides that RMPs "shall . . . be submitted to the Chemical Safety and Hazard Investigation Board (a federal agency), to the State in which the stationary source is located, and to any local agency or entity having responsibility for planning for or responding to accidental releases which may occur at such source...." The commenter claimed that this provision entitles the specified entities, including local fire departments, to receive unredacted RMPs without having to make the showings required by EPA's CBI regulations.

EPA is not resolving this issue today. The Agency has reviewed the relevant statutory text and legislative history, as well as analogous provisions of EPCRA, and believes that arguments can be made on both sides of this issue. While section 112(r)(7)(B)(iii) calls for RMPs to be submitted to states, local entities and the Board, it is not clear that Congress intended CBI contained in RMPs to be provided to those entities without ensuring appropriate protection of CBI.

At stake in resolving this issue are two important interests—local responders' interest in unrestricted access to information that may be critical to their safety and effectiveness in responding to emergencies and businesses' interest in protecting sensitive information from their competitors. Before making a final decision on this issue, EPA believes it would benefit from further public input. Because EPA stated that it would not provide unredacted RMPs to states and local agencies, those interested in protecting CBI may not have considered it necessary to lay out the legal and policy arguments supporting their views. State and local agencies, many of which in the past have expressed concern about the potential administrative burden of receiving RMPs directly from sources, also did not comment on the issue. EPA has therefore decided to accept additional comments on this issue alone. (Additional comments on any other issues addressed in this rulemaking will not be considered or addressed, since the Agency is taking final action on them here.) Comments should be mailed to the persons listed in the preceding FOR FURTHER INFORMATION CONTACT section. In the meantime, unredacted RMPs will be available to states, local agencies and the Board under the terms of the Agency's existing CBI regulations at 40 CFR section 2.301(h)(3) (for state and local agencies) and 40 CFR section 2.209(c) (for the Board).

Section 112(r)(7)(B)(iii) states in relevant part:

[RMPs] shall also be submitted to the Chemical Safety and Hazard Investigation Board, to the Committee on the Judiciary, to the State in which the stationary source is located, and to any local agency or entity having responsibility for planning for or responding to accidental releases which may occur at such source, and shall be available to the public under section 114(c) of the Act.

Section 114(c) provides for the public availability of any information obtained by EPA under the Clean Air Act, except for information (other than emissions data) that would divulge trade secrets.

From a public policy perspective, there are some obvious advantages to reading section 112(r)(7)(B)(iii) in the way the commenter suggests. Local fire departments and other local responders are typically the first to arrive at the scene of chemical accidents in their jurisdictions. RMP information that first responders could find helpful include chemical identity, chemical quantity, and potential source of an accident. Under EPA's regulations, however, any or all of this information could be claimed CBI. In addition, state and local authorities are often in the best position to assess the adequacy of a source's risk management program and to initiate a dialogue with the facility should its RMP indicate a need for improvement. However, state and local authorities' ability to provide this contribution to community safety would be impeded to the extent a source claimed key information as CBI. While states and local agencies may obtain information claimed CBI under EPA's CBI regulations (assuming they can make the requisite showing), the time required to obtain the necessary authority or findings from state or local and EPA officials could be substantial.

At the same time, there are also public policy reasons for ensuring protection of CBI contained in RMPs. Congress has in many statutes, including the CAA and EPCRA, provided for the protection of trade secrets to safeguard the competitive position of private businesses. Businesses' ability to maintain the confidentiality of trade secrets helps ensure competition in the U.S. economy and U.S. businesses' competitive position in the world economy. Protection of trade secrets also encourages innovation, which is an important contributor to economic growth.

A reading of section 112(r)(7)(B)(iii) that demands submission of unredacted RMPs to states, local entities, and the Board may lead to widespread public access to information claimed CBI. For purposes of section 112(r)(7)(B)(iii), "any local agency or entity having responsibility for planning for or responding to accidental releases" includes local emergency planning committees (LEPCs) established under EPCRA. Section 301(c) of EPCRA provides that LEPCs must include representatives from both the public and private sectors, including the media and facilities subject to EPCRA requirements. Submission of an unredacted RMP to an LEPC would thus entail release of CBI to some members of the public and potentially even competitors. More generally, local agencies may not be subject to any legal requirement to protect CBI and may lack the knowledge and resources to address CBI claims. Arguably, it would be
anomalous for Congress to require EPA to protect trade secrets contained in RMPs against release to the public only to risk divulging the same information by requiring submission of unredacted RMPs to a broad range of entities that may not have the need or capacity to protect CBI themselves. It would also appear inconsistent with the approach Congress took to protecting trade secrets in EPCRA, where Congress did not provide for release of trade secret chemical identity information to local agencies.

Relatedly, many state and local agencies objected to EPA’s original proposal in the RMP proposed rulemaking (58 FR 54190, October 20, 1993) that sources submit RMPs directly to States, local agencies, and the Board, as well as EPA. They noted that managing the information contained in RMPs would be difficult without a significant expenditure of typically scarce resources. Many states and local agencies thus supported EPA’s final decision to develop an electronic submission and distribution system that would allow covered sources to submit their RMPs to EPA, which would make them available to states, local agencies, and the Board, as well as the general public. If the statute is read to require submission of RMP information to state and local agencies, and the Board, to the extent it is claimed as CBI, the resource concerns raised by State and local agencies commenters likely would be raised to that extent again.

EPA also questions the extent to which states, local entities and the Board would be disadvantaged if they did not receive unredacted RMPs without making the showings required by EPA’s CBI regulations. As noted earlier, EPA expects that relatively little RMP information will be CBI. RMP data will only rarely contain CBI, and the up-front substantiation will minimize the number of CBI claims it receives by ensuring that sources carefully examine the basis for any claims before submitting them. Consequently, the Agency believes that a state or local agency will rarely confront a redacted RMP.

Moreover, EPCRA provides state and local entities, including fire departments, with access to much of the pertinent data already. EPA’s regulations under EPCRA cover a universe of sources and chemicals that includes most, if not all, the sources and substances covered by the RMP rule. The EPCRA regulations require reporting of some of the same information by the same RMP rule, including chemical identity. EPCRA withholds from public release only chemical identities that are trade secrets and the location of specific chemicals where a facility so requests. In practice, relatively few facilities have requested trade secret protection for a chemical’s identity.

Additionally, EPCRA section 312(f) empowers local fire departments to conduct on-site inspections at facilities subject to EPCRA section 312(a) and obtain information on chemical location. Most facilities subject to EPCRA section 312(a) are also subject to the RMP rule. On-site inspections could also provide information on hazards and mitigation measures. In addition, EPCRA section 303(d)(3) authorizes LEPCs, which include representatives of fire departments, to request from facilities covered by EPCRA section 302(b) such information as may be necessary to prepare an emergency response plan and to include such information in the plan as appropriate. Some sources subject to the RMP rule are also covered by EPCRA section 312(a).

In light of the points made above, EPA questions whether section 112(r)(7)(B)(iii) should be interpreted to require submission of unredacted RMPs containing CBI to the statutorily specified entities without provision being made for protecting CBI. EPA invites the public to provide any additional comment or information relevant to interpreting the submission requirement of section 112(r)(7)(B)(iii).

5. Other CBI Issues

Two commenters disagreed with EPA’s statement that a source cannot make a CBI claim for information available to the public under EPCRA or another statute. They claimed that a request for information under EPCRA cannot supersed the CBI provisions applicable to data collected under the authorities of the CAA or Toxic Substances Control Act or any other regulatory program.

EPA does not agree with this comment. Claims of CBI may not be upheld if the information is properly obtainable or made public under other statutes or authorities. For example, chemical quantity on site is available to the public under EPCRA Tier II reporting. In addition, under EPCRA section 303(d)(3), LEPCs have the authority to request any information they need to develop and implement community emergency response plans. If information obtained through such a request is included in the community plan, it will become available to the public under EPCRA section 324. Information obtainable or made public under EPCRA would not be eligible for CBI protection under 40 CFR section 2.208, which specifically excludes from CBI protection information already available to the public. Filing a CBI claim under the CAA or another statute does not protect information if it is legitimately requested and made public under other federal, state, or local law. Information obtainable or made public (through proper means) under existing statutes cannot be CBI under EPA’s CBI regulations.

6. Actions Taken

In summary, the Agency is adding two sections (68.151 and 68.152) to Part 68. Section 68.151 sets forth the procedures for a source to follow when asserting a CBI claim and lists data elements that can not be claimed as CBI. This section also requires sources filing CBI claims to provide the information claimed confidential, in a format to be specified by EPA, instead of the unsanitized paper copy of the RMP as discussed in the proposal. Section 68.152 sets forth the procedures for substantiating CBI claims. Sources claiming CBI are required to submit their substantiation of their claims at the same time they submit their RMPs.

E. Other Issues

Two commenters asked why EPA had proposed to drop the phrase “if used” in section 68.165(b)(3) where the rule asks for the basis of the offsite consequence analysis results. EPA has decided to retain the language, since sources will have a choice of using either EPA’s RMP guidance documents or a model. Where a model is used, the source will have to provide the name of the model. These commenters also asked why EPA proposed to drop (alternative releases only) from section 68.165(b)(13). EPA has also decided to retain the parenthetical language.

One commenter stated that EPA should allow sources to submit RMPs either electronically or in hard copy. The commenter stated that not allowing hard copy submissions will be burdensome on many sources who have never filed an electronic report to the government before. As stated in the April proposal, EPA is allowing sources to submit RMPs on paper. Paper submitters are asked to fill out a simple paper form to tell EPA why they are unable to file electronically.

Two commenters objected to placing offsite consequence analysis (OCA) data, particularly worst-case release scenarios, on the Internet, for security reasons. Issues related to public access to OCA data are beyond the scope of this rulemaking, as this action is limited to the issues discussed above. It does
not include decisions regarding how the public will access the OCA data elements of the RMPs. Statements in the preamble about EPA providing public access to RMP data are not intended to address which portions of the RMP data will be electronically available.

A number of commenters were concerned about a statement EPA made in the preamble to the proposed rule regarding the definition of “process”, and stated that EPA’s interpretation of “process” is not consistent with the interpretation in the Occupational Safety and Health Administration (OSHA) uses in its process safety management (PSM) standard (29 CFR 1910.119). In this rulemaking, EPA did not propose any changes to the definition of process nor is it adopting any changes to the definition. As EPA stated in the preamble to the final RMP rule, it will interpret “process” consistently with OSHA’s interpretation of that term (29 CFR 1910.119). Therefore, if a source is subject to the PSM rule, the limits of its process(es) for purposes of OSHA PSM will be the limits of its process(es) for purposes of RMP (except in cases involving atmospheric storage tanks containing flammable regulated substances, which are exempt from PSM but not RMP). If a source is not covered by OSHA PSM and is complicated from an engineering perspective, it should consider contacting its implementing agency for advice on determining process boundaries. EPA and OSHA are coordinating the agencies’ approach to common issues, such as the interpretation of “process”.

F. Technical Corrections

When Part 68 was promulgated, the text of section 68.79(a), was drawn from the OSHA PSM standard, but it was not revised to reflect the different structure of EPA’s rule. The OSHA PSM standard is contained in a single section; EPA’s Program 3 prevention program is contained in a subpart. Rather than referencing “this section,” the paragraph should have referenced the “subpart.” Therefore, as proposed, EPA is changing “section” to “subpart” in section 68.79(a).

Under section 68.180(b), EPA intended that all covered sources report the name and telephone number of the agency with which they coordinate emergency response activities, even if the source is not required to have an emergency response plan. However, the rule refers only to coordinating the emergency plan. In this action, EPA is revising this section to refer to the local agency with which emergency response activities and the emergency response plan is coordinated.

IV. Section-by-Section Discussion of the Final Rule

In Section 68.3, Definitions, the definition of SIC is removed and replaced by the definition of NAICS.

Section 68.10, Applicability, is revised to replace the SIC codes with NAICS codes, as discussed above.

Section 68.42, Five-Year Accident History, is revised to require the percentage concentration by weight of regulated toxic substances released in a liquid mixture and the five- or six-digit NAICS code that most closely corresponds to the process that had the release. The phrase “five- or six-digit” has been added before the NAICS code to clarify the level of detail required for NAICS code reporting.

Section 68.79, Compliance Audits, the word “section” in paragraph (a) is replaced by “subpart.”

Section 68.150, Submission, is revised by adding a paragraph to state that procedures for asserting CBI claims and determining the sufficiency of such claims are provided in new Sections 68.151 and 68.152.

Section 68.151 is added to set forth the procedures to assert a CBI claim and list data elements that may not be claimed as CBI, as discussed above.

Section 68.152 is added to set forth procedures for substantiating CBI claims, as proposed.

Section 68.160, Registration, is revised by adding the requirements to report the method and description of latitude and longitude, replacing SIC codes with five- or six-digit NAICS codes, and adding the requirement to report Title V permit number, when applicable. This section is also revised to include optional data elements. The phrase “five- or six-digit” has been added before NAICS code to clarify the level of detail required for NAICS code reporting.

Section 68.165, Offsite Consequence Analysis, is revised by adding the requirement that the percentage weight of a regulated toxic substance in a liquid mixture be reported.

Section 68.170, Prevention Program/ Program 2, is revised to replace SIC codes with five- or six-digit NAICS codes, as is Section 68.175.

Section 68.180, Emergency Response Program, is revised to clarify that paragraph (b) covers both the coordination of response activities and plans, as proposed.

V. Judicial Review

The proposed rule amending the accidental release prevention requirements; under section 112(r)(7) was proposed in the Federal Register on April 17, 1998. This Federal Register action announces EPA’s final decision on the amendments. Under section 307(b)(1) of the CAA, judicial review of this action is available only by filing a petition for review in the U.S. Court of Appeals for the District of Columbia Circuit on or before March 8, 1999. Under section 307(b)(2) of the CAA, the requirements that are the subject of today’s action may not be challenged later in civil or criminal proceedings brought by EPA to enforce these requirements.

VI. Administrative Requirements

A. Docket

The docket is an organized and complete file of all the information considered by the EPA in the development of this rulemaking. The docket is a dynamic file, because it allows members of the public and industries involved to readily identify and locate documents so that they can effectively participate in the rulemaking process. Along with the proposed and promulgated rules and their preambles, the contents of the docket serve as the record in the case of judicial review. (See section 307(d)(7)(A) of the CAA.)

The official record for this rulemaking, as well as the public version, has been established for this rulemaking under Docket No. A–98–08 (including comments and data submitted electronically). A public version of this record, including printed, paper versions of electronic comments, which does not include any information claimed as CBI, is available for inspection from 8:00 a.m. to 5:30 p.m., Monday through Friday, excluding legal holidays. The official rulemaking record is located at the address in ADDRESSES at the beginning of this document.

B. Executive Order 12866

Under Executive Order (E.O.) 12866, [58 FR 51,735 (October 4, 1993)], the Agency must determine whether the regulatory action is “significant”, and therefore subject to OMB review and the requirements of the E.O. The Order defines “significant regulatory action” as one that is likely to result in a rule that may:

(1) Have an annual effect on the economy of $100 million or more or adversely affect in a material way the economy, a sector of the economy, productivity, competition, jobs, the environment, public health or safety, or state, local or tribal government or communities;
(2) Create a serious inconsistency or otherwise interfere with an action taken or planned by another agency;
(3) Materially alter the budgetary impact of entitlements, grants, user fees, or loan programs or the rights and obligations of recipients thereof; or
(4) Raise novel legal or policy issues arising out of legal mandates, the President's priorities, or the principles set forth in the E.O.

Pursuant to the terms of Executive Order 12866, OMB has notified EPA that it considers this a “significant regulatory action” within the meaning of the Executive Order. EPA has submitted this action to OMB for review. Changes made in response to OMB suggestions or recommendations will be documented in the public record.

C. Executive Order 12875

Under Executive Order 12875, EPA may not issue a regulation that is not required by statute and that creates a mandate upon a State, local or tribal government, unless the Federal government provides the funds necessary to pay the direct compliance costs incurred by those governments, or EPA consults with those governments. If EPA complies by consulting, Executive Order 12875 requires EPA to provide to the Office of Management and Budget a description of the extent of EPA's prior consultation with representatives of affected State, local and tribal governments, the nature of their concerns, copies of any written communications from the governments, and a statement supporting the need to issue the regulation. In addition, Executive Order 12875 requires EPA to develop an effective process permitting elected officials and other representatives of State, local and tribal governments “to provide meaningful and timely input to the development of regulatory proposals containing significant unfunded mandates.”

EPA has concluded that this rule may create a nominal mandate on State, local or tribal governments and that the Federal government will not provide the funds necessary to pay the direct costs incurred by these governments in complying with the mandate. Specifically, some public entities may be covered sources and will have to add the new data elements to their RMP. In developing this rule, EPA consulted with state, local and tribal governments to enable them to provide meaningful and timely input in the development of this rule. Even though this rule revises Part 68 in a way that does not significantly change the burden imposed by the underlying rule, EPA has taken efforts to involve state and local entities in this regulatory effort.

Specifically, much of the rule responds to issues raised by the Electronic Submission Workgroup discussed above, which includes State and local government stakeholders. In addition, EPA has recently conducted seminars with tribal governments; however, there were no concerns raised on any issues that are covered in this rule. EPA discussed the need for issuing this regulation in sections II and III in this preamble. Also, EPA provided OMB with copies of the comments to the proposed rule.

D. Executive Order 13045

Executive Order 13045: “Protection of Children from Environmental Health Risks and Safety Risks” (62 FR 19885, April 23, 1997) applies to any rule that: (1) is determined to be “economically significant” as defined under E.O. 12866, and (2) concerns an environmental health or safety risk that EPA has reason to believe may have a disproportionate effect on children. If the regulatory action meets both criteria, the Agency must evaluate the environmental health or safety effects of the planned rule on children, and explain why the planned regulation is preferable to other potentially effective and reasonably feasible alternatives considered by the Agency.

This final rule is not subject to the E.O. 13045 because it is not “economically significant” as defined in E.O. 12866, and because it does not involve decisions based on environmental health or safety risks.

E. Executive Order 13084

Under Executive Order 13084, EPA may not issue a regulation that is not required by statute, that significantly or uniquely affects the communities of Indian tribal governments, and that imposes substantial direct compliance costs on those communities, unless the Federal government provides the funds necessary to pay the direct compliance costs incurred by the tribal governments, or EPA consults with those governments. If EPA complies by consulting, Executive Order 13084 requires EPA to provide to the Office of Management and Budget, in a separately identified section of the preamble to the rule, a description of the extent of EPA’s prior consultation with representatives of affected tribal governments, a summary of the nature of their concerns, and a statement supporting the need to issue the regulation. In addition, Executive Order 13084 requires EPA to develop an effective process permitting elected and other representatives of Indian tribal governments “to provide meaningful and timely input in the development of regulatory policies on matters that significantly or uniquely affect their communities.”

Today’s rule does not significantly or uniquely affect the communities of Indian tribal governments. Two of the amendments made by this rule, the addition of RMP data elements and the conversion of SIC codes to NAICS codes, impose only minimal burden on any sources that may be owned or operated by tribal governments, such as drinking water and waste water treatment systems. The third amendment made by this rule addresses the procedures for submission of confidential business information in the RMP. The sources that are mentioned above handle chemicals that are known to public (e.g., chlorine for use of disinfection, propane used for fuel, etc.). EPA does not, therefore, expect RMP information on these types of processes to include CBI, so any costs related to CBI will not fall on Indian tribal governments. Accordingly, the requirements of section 3(b) of Executive Order 13084 do not apply to this rule.

Notwithstanding the non-applicability of E. O. 13084, EPA has recently conducted seminars with the tribal governments. However, there were no concerns raised on any issues that are covered in this rule.

F. Regulatory Flexibility

EPA has determined that it is not necessary to prepare a regulatory flexibility analysis in connection with this final rule. EPA has also determined that this action will not have a significant economic impact on a substantial number of small entities. Two of the amendments made by this rule, the addition of RMP data elements and the conversion of SIC codes to NAICS codes, impose only minimal burden on small entities. Moreover, those small businesses that claim CBI will not fall on Indian tribal governments. Two of the amendments made by this rule, the addition of RMP data elements and the conversion of SIC codes to NAICS codes, impose only minimal burden on any sources that may be owned or operated by tribal governments, such as drinking water and waste water treatment systems. The third amendment made by this rule addresses the procedures for submission of confidential business information in the RMP. The sources that are mentioned above handle chemicals that are known to public (e.g., chlorine for use of disinfection, propane used for fuel, etc.). EPA does not, therefore, expect RMP information on these types of processes to include CBI, so any costs related to CBI will not fall on Indian tribal governments. Accordingly, the requirements of section 3(b) of Executive Order 13084 do not apply to this rule.

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small entity impacts of CBI submission, see Document Number, IV-B–02, available in the docket for this rulemaking (see ADDRESSES section).

G. Paperwork Reduction

1. General

The information collection requirements in this rule have been submitted for approval to the Office of Management and Budget (OMB) under the Paperwork Reduction Act, 44 U.S.C. 3501 et seq. An Information Collection Request (ICR) document has been prepared by EPA (ICR No. 1656.05) and a copy may be obtained from Sandy Farmer, by mail at Office of Policy, Regulatory Information Division, U.S. Environmental Protection Agency (2137), 401 M St, SW, Washington, DC 20460, by e-mail at farmer.sandy@epamail.epa.gov or by calling (202) 260–2740. A copy may also be downloaded off the Internet at http://www.epa.gov/icr. The information requirements are not effective until OMB approves them.

The submission of the RMP is mandated by section 112(r)(7) of the CAA and demonstrates compliance with Part 68 consistent with section 114(c) of the CAA. The information collected also will be made available to state and local governments and the public to enhance their preparedness, response, and prevention activities. Certain information in the RMP may be claimed as confidential business information under 40 CFR Part 2 and Part 68. This rule will impose very little burden on affected sources. First, EPA estimates that the new data elements will require only a nominal burden, 25 hours for a typical source, because latitude and longitude method and description will be selected from a list of options, the Title V permit number is available to any source to which Title V applies, and the percentage weight of a toxic substance in a liquid mixture is usually provided by the supplier of the mixture. Second, the NAICS code provision is simply a change from one code to another. Third, as discussed above in the preamble, EPA believes that the CBI provisions of this rule will add no additional burden beyond what sources otherwise would face in complying with the CBI rules in 40 CFR Part 2. The Agency has calculated the burden of substantiations made for purposes of this rule below.

Burden means the total time, effort, or financial resources expended by persons to generate, maintain, retain, or disclose or provide information to or for a Federal agency. This includes the time needed to review instructions; develop, acquire, install, and utilize technology and system for the purposes of collecting, validating, and verifying information, processing and maintaining information, and disclosing and providing information; adjust the existing ways to comply with any previously applicable instructions and requirements; train personnel to be able to respond to a collection of information; search data sources; complete and review the collection of information; and transmit or otherwise disclose the information.

An agency may not conduct or sponsor, and a person is not required to respond to a collection of information unless it displays a currently valid OMB control number. The OMB control numbers for EPA’s regulations are listed in 40 CFR Part 9 and 48 CFR Chapter 12.

2. CBI Burden

In the Notice of Proposed Rulemaking for these amendments, EPA proposed to amend existing 40 CFR Part 68 to add two sections which would clarify the procedures for submitting RMPs that contain confidential business information (CBI). As proposed, CBI would be handled in much the same way as it presently is under other EPA programs, except that EPA would require sources claiming CBI to submit documentation substantiating their CBI claims at the time such claims were made and EPA also would not permit CBI claims for certain data elements which clearly are not CBI. Aside from these procedural changes, however, the proposed rule was substantively identical to the existing rules governing the substantiation of CBI claims, presently codified in 40 CFR Part 2.

At the time it proposed these amendments, EPA estimated the public reporting burden for CBI claims to be 15 hours for chemical manufacturers with Program 3 processes, the only kinds of facilities that EPA expects to be able to claim CBI for any RMP data elements. This estimate was premised upon EPA’s assessment that it would require 8.5 hours per claim to develop and submit the CBI substantiation and 6.5 hours to complete an unsolicited version of the RMP, for a total of 15 hours. EPA also estimated that approximately 20 percent of the 4000 chemical manufacturers (out of 64,200 stationary sources estimated to be covered by the RMP rule) may file CBI claims (800 sources). The 800 sources represent a conservative projection based on the Agency’s experience under EPCRA program. Consequently, the total annual public reporting burden for filing CBI claims was estimated to be approximately 12,000 hours over three years (800 facilities multiplied by an average burden of 15 hours), or an annual burden of 4,000 hours (Information Collection Request No. 1656.04).

a. Comment received. EPA received one comment on the ICR developed for the proposed rule, opposing up-front substantiation of any CBI claims. The commenter stated that “[t]his is a major departure from standard EPA procedure, and would impose a substantial and unjustified burden for several years.” The commenter further added that up-front substantiation would significantly increase the burden of this rule, and that up-front substantiation unnecessarily increases the volume of CBI claims and would impose a substantial and unjustified burden for several years.”

b. EPA response. Burden Estimates: EPA disagrees with these comments. As pointed out above, the requirement to submit up-front substantiation of CBI claims imposes no additional burden. In addition, the total burden of the CBI provisions of this rule are not understated. EPA has re-examined its analysis in light of the commenter’s concerns and has determined—contrary to the commenter’s claim—that its initial estimate of the total burden associated with preparing and claiming CBI was likely too conservative. As explained below, the Agency’s best available information indicates that the process of documenting and submitting a claim of CBI should impose a burden of approximately 9.5 hours per CBI claim.

First, EPA believes that the requirement to submit, at the time a source claims information as CBI,
substantiation demonstrating that the material truly is CBI imposes no burden on sources beyond that which presently exists under EPA’s CBI regulations in Part 2. In order to decide whether they might properly claim CBI for a given piece of information, a source must determine if the criteria stated in section 2.208 of 40 CFR Part 2 are satisfied. Naturally, a source goes through this process before a CBI claim is made. EPA agrees that most programs do not require the information that forms the basis for the substantiation to be submitted at the time of the claim; however, a facility must still determine whether or not a claim can be substantiated. Because existing rules require sources to formulate a legitimate basis for claiming CBI, even if those rules do not require immediate documentation, and because the Agency fully expects requests for RMP information which will necessitate sources’ submitting such documentation, EPA believes that upfront submission will not increase the burden of the regulation.

Second, in response to the commenter’s claim that the Agency had underestimated the total burden associated with CBI claims, EPA undertook a review of recent information collection requests (ICRs) covering data similar to that required to be submitted in an RMP. Initially, EPA examined the ICR prepared for Part 2 itself (ICR No. 1665.02,OMB Control No. 2020-0003). Under an analysis contained in the Statement of Support for the ICR, the Agency estimated that it takes approximately 9.4 hours to substantiate claims of CBI, prepare documentation, and submit such documentation to EPA. Next, the Agency reviewed a survey conducted by the Agency (under Office of Management and Budget clearance #2070-0034), to present the average burden associated with indicating confidential business information claims for certain data elements under the proposed inventory update rule (IUR) amendment under TSCA section 8. This survey specifically asked affected industry how long it would take to prepare CBI claims for two data elements—chemical identity and production volume range information. Part 68 also requires similar information (e.g., chemical identity and maximum quantity in a process) to be included in a source’s RMP and, indeed, EPA anticipates that they will be the data elements most likely to be claimed CBI. The average burden estimates for chemical identity were between 1.82 and 3.13 hours, and the average burden estimates for production volume in ranges were between 0.87 and 2.08 hours. Thus, assuming that the average source claims both chemical identity and the maximum quantity in a process as CBI, a conservative estimate for the reporting burden would be 5.21 hours. Finally, EPA examined the burden estimate upon which it relied at proposal. That estimate predicted that the average CBI claim would take 15 hours, of which 8.5 would be developing and submitting the CBI claim, and 6.5 would be completing an unsanitized version of the RMP. In view of EPA’s current plan not to require a source claiming CBI to submit a full, unsanitized RMP, but instead to submit only the particular elements claimed as CBI, the Agency expects the latter burden to decrease to 1 hour, for a total burden of 9.5 hours.

In light of its extensive research of the burden hours involved in preparing and submitting CBI claims, EPA believes that the total burden estimate was not understated in the April proposal. Rather, other ICRs and the ICR proposal, combined with the changes to the method of documenting CBI claims, indicate that a burden estimate between 5.21 and 9.5 hours is appropriate for this final rule. EPA has selected the most conservative of these, 9.5 hours, in its ICR for this final rule.

EPA rejected one ICR’s burden estimate as being inapplicable to the present rulemaking. Although the commenter urged the Agency to adopt the estimate associated with trade secret elements under EPCRA (28 hours), EPA believes that the estimates discussed above are more accurate for several reasons. First, the EPCRA figures are based upon a survey with a very small sample size, as compared to the TSCA survey cited previously. Second, most (if not all) of the facilities submitting RMPs are likely to already be reporting under sections 311 and 312 or section 313 of EPCRA, and many of the industries submitting an RMP are subject to TSCA reporting requirements; thus, most sources may already claim CBI for an RMP data element will have already done some analysis of whether or not such information would reveal legitimately confidential material.

Other Facilities Can Claim CBI: The Agency does not agree with the commenter’s claim that facilities other than chemical manufacturers might be expected to claim CBI for information contained in their RMPs. The other industries affected by the RMP rule (e.g., propane retailers, publicly owned wastewater treatment plants) generally do not have the competitive position. For example, covered public drinking water and wastewater treatment plants generally use common regulated substances in standard processes (i.e., chlorine used for disinfection). Also, covered processes at many sources involve the storage of regulated substances that are sold (e.g., propane, ammonia), so the processes are already public knowledge. Other covered processes involve the use of well-known combinations of regulated substances such as refrigerants. Therefore, it is not likely that these businesses would claim information as CBI.

As a point of comparison, EPA notes that of the 869,000 facilities that are estimated to be required to report under sections 311 and 312 of EPCRA, approximately 58 facilities have submitted trade secret claims for under those sections. For this reason, EPA believes the estimate of 800 sources may, in fact, be an overestimate of the number of sources claiming CBI.

Reporting Multiple Data Elements: The Agency disagrees with the commenter’s assertion that it has underestimated the reporting burden on sources’ claiming multiple data elements as CBI. The burden figures stated above are based on the Agency’s estimates of the average number of data elements that a typical source will likely claim CBI.

Public reporting of the new RMP data elements is estimated to require an average of .25 hours for all sources (64,200 sources) and substantiating CBI claims is estimated to take approximately 9.5 hours for certain chemical manufacturing sources (800 sources). The aggregate increase in burden over that estimated in the previous Information Collection Request (ICR) for part 68 is estimated to be about 23,650 hours over three years, or an annual burden of 7,883 hours for the three years covered by the ICR.

H. Unfunded Mandates Reform Act

Title II of the Unfunded Mandates Reform Act of 1995 (UMRA), P.L. 104-4, establishes requirements for Federal agencies to assess the effects of their regulatory actions on State, local, and tribal governments and the private sector. Under section 202 of the UMRA, EPA generally must prepare a written statement, including a cost-benefit analysis, for proposed and final rules with “Federal mandates” that may result in expenditures to State, local, and tribal governments, in the aggregate, or to the private sector, of $100 million or more in any given year. Before promulgating an EPA rule for which a written statement is needed, section 205...
of the UMRA generally requires EPA to identify and consider a reasonable number of regulatory alternatives and adopt the least costly, most cost-effective or least burdensome alternative that achieves the objectives of the rule. The provisions of section 205 do not apply when they are inconsistent with applicable law. Moreover, section 205 allows EPA to adopt an alternative other than the least costly, most cost-effective or least burdensome alternative if the Administrator publishes with the final rule an explanation why that alternative was not adopted. Before EPA establishes any regulatory requirements that may significantly or uniquely affect small governments, including tribal governments, it must have developed under section 203 of the UMRA a small government agency plan. The plan must provide for notifying potentially affected small governments, enabling officials of affected small governments to have meaningful and timely input in the development of EPA regulatory proposals with significant Federal intergovernmental mandates, and informing, educating, and advising small governments on compliance with the regulatory requirements.

EPA has determined that this rule does not contain a Federal mandate that may result in expenditures of $100 million or more for state, local, and tribal governments, in the aggregate, or the private sector in any one year. The EPA has determined that the total nationwide capital cost for these rule amendments is zero and the annual nationwide cost for these amendments is less than $1 million. Thus, today’s rule is not subject to the requirements of sections 202 and 205 of the Unfunded Mandates Act.

EPA has determined that this rule contains no regulatory requirements that might significantly or uniquely affect small governments. Small governments are unlikely to claim information confidential, because sources owned or operated by these entities (e.g., drinking water and waste water treatment systems), handle chemicals that are known to public. The new data elements and the conversion of SIC codes to NAICS codes impose only minimal burden on these entities.

### I. National Technology Transfer and Advancement Act

Section 12(d) of the National Technology Transfer and Advancement Act of 1995 ("NTTAA"), Pub L. 104-113, section 12(d)(15 U.S.C. 272 note), directs EPA to use voluntary consensus standards in its regulatory activities unless to do so would be inconsistent with applicable law or otherwise impractical. Voluntary consensus standards are technical standards (e.g., materials specifications, test methods, sampling procedures, business practices) that are developed or adopted by voluntary consensus standards bodies. The NTTAA requires EPA to provide Congress, through OMB, explanations when the Agency decides not to use available and applicable voluntary consensus standards.

This action does not involve technical standards. Therefore, EPA did not consider the use of any voluntary consensus standards.

J. Congressional Review Act

The Congressional Review Act, 5 U.S.C. section 801 et seq., as added by the Small Business Regulatory Enforcement Fairness Act of 1996, generally provides that before a rule may take effect, the agency promulgating the rule must submit a rule report, which includes a copy of the rule, to each House of the Congress and to the Comptroller General of the United States. EPA will submit a report containing this rule and other required information to the U.S. Senate, the U.S. House of Representatives, and the Comptroller General of the United States prior to publication of the rule in the Federal Register. This action is not a “major rule” as defined by 5 U.S.C. section 804(2). This rule will be effective February 5, 1999.

### APPENDIX TO PREAMBLE—DATA ELEMENTS THAT MAY NOT BE CLAIMED AS CBI

<table>
<thead>
<tr>
<th>Rule element</th>
<th>Comment</th>
</tr>
</thead>
<tbody>
<tr>
<td>68.160(b)(1) Stationary source name, street, city, county, state, zip code, latitude, and longitude, method for obtaining latitude and longitude, and description of location that latitude and longitude represent.</td>
<td>This information is filed with EPA and other agencies under other regulations and is made available to the public and, therefore, does not meet the criteria for CBI claims. It is also available in business and other directories.</td>
</tr>
<tr>
<td>68.160(b)(2) Stationary source Dun and Bradstreet number.</td>
<td>This information provides no information that would affect a source’s competitive position.</td>
</tr>
<tr>
<td>68.160(b)(3) Name and Dun and Bradstreet number of the corporate parent company.</td>
<td>This information is filed with state and local agencies under EPCRA and is made available to the public and, therefore, does not meet the criteria for CBI claims.</td>
</tr>
<tr>
<td>68.160(b)(4) The name, telephone number, and mailing address of the owner/operator.</td>
<td>This information provides no information that would affect a source’s competitive position.</td>
</tr>
<tr>
<td>68.160(b)(5) The name and title of the person or position with overall responsibility for RMP elements and implementation.</td>
<td>This information is filed with state and local agencies under EPCRA and is made available to the public and, therefore, does not meet the criteria for CBI claims.</td>
</tr>
<tr>
<td>68.160(b)(6) The name, title, telephone number, and 24-hour telephone number of the emergency contact.</td>
<td>This information provides no information that would affect a source’s competitive position.</td>
</tr>
<tr>
<td>68.160(b)(7) Program level and NAICS code of the process.</td>
<td>This information provides no information that would affect a source’s competitive position.</td>
</tr>
<tr>
<td>68.160(b)(8) The stationary source EPA identifier.</td>
<td>This information provides no information that would affect a source’s competitive position.</td>
</tr>
<tr>
<td>68.160(b)(10) Whether the stationary source is subject to 29 CFR 1910.119.</td>
<td>Sources are required to notify the state and local agencies if they are subject to this rule; this information is available to the public and, therefore, does not meet the criteria for CBI claims.</td>
</tr>
<tr>
<td>68.160(b)(11) Whether the stationary source is subject to 40 CFR Part 355.</td>
<td>This information will be known to state and federal air agencies and is available to the public and, therefore, does not meet the criteria for CBI claims.</td>
</tr>
</tbody>
</table>
### APPENDIX TO PREAMBLE—DATA ELEMENTS THAT MAY NOT BE CLAIMED AS CBI—Continued

<table>
<thead>
<tr>
<th>Rule element</th>
<th>Comment</th>
</tr>
</thead>
<tbody>
<tr>
<td>68.160(b)(13)</td>
<td>The date of the last safety inspection and the identity of the inspecting entity.</td>
</tr>
<tr>
<td>68.165(b)(4)</td>
<td>Basis of the results (give model name if used).</td>
</tr>
<tr>
<td>68.165(b)(9)</td>
<td>Wind speed and atmospheric stability class (toxics only).</td>
</tr>
<tr>
<td>68.165(b)(10)</td>
<td>Topography (toxics only)</td>
</tr>
<tr>
<td>68.165(b)(11)</td>
<td>Distance to an endpoint</td>
</tr>
<tr>
<td>68.165(b)(12)</td>
<td>Public and environmental receptors within the distance.</td>
</tr>
<tr>
<td>68.168</td>
<td>Five-year accident history</td>
</tr>
<tr>
<td>68.170(b), (d), (e)(1), and (l)–(k)</td>
<td>NAICS code, prevention program compliance dates and information.</td>
</tr>
<tr>
<td>68.175(b), (d), (e)(1), and (l)–(p)</td>
<td>NAICS codes and the prevention program compliance dates and information provide no information that would affect a source’s competitive position.</td>
</tr>
<tr>
<td>68.180 Emergency response program</td>
<td>This information provides no information that would affect a source’s competitive position.</td>
</tr>
</tbody>
</table>

#### List of Subjects in 40 CFR Part 68

Environmental protection, Administrative practice and procedure, Air pollution control, Chemicals, Hazardous substances, Intergovernmental relations, Reporting and recordkeeping requirements.

Carol M. Browner, Administrator.

For the reasons set out in the preamble to the Code of Federal Regulations, part 68 of the Code of Federal Regulations is amended as follows:

PART 68—CHEMICAL ACCIDENT PREVENTION PROVISIONS

1. The authority citation for Part 68 continues to read as follows:
   Authority: 42 U.S.C. 7412(r), 7601(a)(1), 7661–7661f.

2. Section 68.3 is amended by removing the definition of SIC and by adding in alphabetical order the definition for NAICS to read as follows:

§ 68.3 Definitions.

* * *

NAICS means North American Industry Classification System.

* * *

3. Section 68.10 is amended by revising paragraph (d)(1) to read as follows:

§ 68.10 Applicability.

* * *

(d) * * *

(1) The process is in NAICS code 32211, 32411, 32511, 325181, 325188, 325192, 325199, 325211, 325311, or 32532; or

* * * * *

4. Section 68.42 is amended by revising paragraph (b)(3), redesignating paragraphs (b)(4) through (b)(10) as paragraphs (b)(5) through (b)(11) and by adding a new paragraph (b)(4) to read as follows:

§ 68.42 Five-year accident history.

* * * * *

(b) * * *

(3) Estimated quantity released in pounds and, for mixtures containing regulated toxic substances, percentage concentration by weight of the released regulated toxic substance in the liquid mixture;

(4) Five- or six-digit NAICS code that most closely corresponds to the process;

* * * * *

5. Section 68.79 is amended by revising paragraph (a) to read as follows:

§ 68.79 Compliance audits.

(a) The owner or operator shall certify that they have evaluated compliance with the provisions of this subpart at least every three years to verify that procedures and practices developed under this subpart are adequate and are being followed.

* * * * *

6. Section 68.150 is amended by adding paragraph (e) to read as follows:

§ 68.150 Submission.

* * * * *

(e) Procedures for asserting that information submitted in the RMP is entitled to protection as confidential business information are set forth in §§ 68.151 and 68.152.

7. Section 68.151 is added to read as follows:

§ 68.151 Assertion of claims of confidential business information.

(a) Except as provided in paragraph (b) of this section, an owner or operator of a stationary source subject to this part may not claim as confidential business information any such information that meets the criteria set forth in 40 CFR 2.301.

(b) Notwithstanding the provisions of 40 CFR part 2, an owner or operator of a stationary source subject to this part may not claim confidential business information the following information:

(1) Registration data required by § 68.160(b)(1) through (b)(6) and (b)(8), (b)(10) through (b)(13) and NAICS code and Program level of the process set forth in § 68.160(b)(7);

(2) Offsite consequence analysis data required by § 68.165(b)(4), (b)(9), (b)(10), (b)(11), and (b)(12).

(3) Accident history data required by § 68.168;

(4) Prevention program data required by § 68.175(b), (d), (e)(1), (f) through (k);

(5) Prevention program data required by § 68.175(b), (d), (e)(1), (f) through (p); and

(6) Emergency response program data required by § 68.180.

(c) Notwithstanding the procedures specified in 40 CFR part 2, an owner or operator asserting a claim of CBI with respect to information contained in its RMP shall submit to EPA the time it submits the RMP the following:

(1) The information claimed confidential, provided in a format to be specified by EPA;
A sanitized (redacted) copy of the RMP, with the notation “CBI” substituted for the information claimed confidential, except that a generic category or class name shall be substituted for any chemical name or identity claimed confidential; and

(3) The document or documents substantiating each claim of confidential business information, as described in § 68.152.

8. Section 68.152 is added to read as follows:

§ 68.152 Substantiating claims of confidential business information.
(a) An owner or operator claiming that information is confidential business information must substantiate that claim by providing documentation that demonstrates that the claim meets the substantive criteria set forth in 40 CFR 2.301.

(b) Information that is submitted as part of the substantiation may be claimed confidential by marking it as confidential business information. Information not so marked will be treated as public and may be disclosed without notice to the submitter. If information that is submitted as part of the substantiation is claimed confidential, the owner or operator must provide a sanitized and unsanitized version of the substantiation.

c) The owner, operator, or senior official with management responsibility of the stationary source shall sign a certification that the signer has personally examined the information submitted and that based on inquiry of the persons who compiled the information, the information is true, accurate, and complete, and that those portions of the substantiation claimed as confidential business information would, if disclosed, reveal trade secrets or other confidential business information.

9. Section 68.160 is amended by revising paragraphs (b)(1), (b)(7), and (b)(12) and adding paragraphs (b)(14) through (b)(18) to read as follows:

§ 68.160 Registration.

* * * * *

(b) * * *

(1) Stationary source name, street, city, county, state, zip code, latitude and longitude, method for obtaining latitude and longitude, and description of location that latitude and longitude represent;

* * * * *

(7) For each covered process, the name and CAS number of each regulated substance held above the threshold quantity in the process, the maximum quantity of each regulated substance or mixture in the process (in pounds) to two significant digits, the five- or six-digit NAICS code that most closely corresponds to the process, and the Program level of the process;

* * * * *

(12) If the stationary source has a CAA Title V operating permit, the permit number; and

* * * * *

(14) Source or Parent Company E-Mail Address (Optional);

(15) Source Homepage address (Optional);

(16) Phone number at the source for public inquiries (Optional);

(17) Local Emergency Planning Committee (Optional);

(18) OSHA Voluntary Protection Program status (Optional);

10. Section 68.165 is amended by revising paragraph (b) to read as follows:

§ 68.165 Offsite consequence analysis.

* * * * *

(b) The owner or operator shall submit the following data:

(1) Chemical name;

(2) Percentage weight of the chemical in a liquid mixture (toxics only);

(3) Physical state (toxics only);

(4) Basis of results (give model name if used);

(5) Scenario (explosion, fire, toxic gas release, or liquid spill and evaporation);

(6) Quantity released in pounds;

(7) Release rate;

(8) Release duration;

(9) Wind speed and atmospheric stability class (toxics only);

(10) Topography (toxics only);

(11) Distance to endpoint;

(12) Public and environmental receptors within the distance;

(13) Passive mitigation considered; and

(14) Active mitigation considered (alternative releases only);

11. Section 68.170 is amended by revising paragraph (b) to read as follows:

§ 68.170 Prevention program/Program 2.

* * * * *

(b) The five- or six-digit NAICS code that most closely corresponds to the process.

* * * * *

12. Section 68.175 is amended by revising paragraph (b) to read as follows:

§ 68.175 Prevention program/Program 3.

* * * * *

(b) The five- or six-digit NAICS code that most closely corresponds to the process.

* * * * *

13. Section 68.180 is amended by revising paragraph (b) to read as follows:

§ 68.180 Emergency response program.

* * * * *

(b) The owner or operator shall provide the name and telephone number of the local agency with which emergency response activities and the emergency response plan is coordinated.

* * * * *

[FR Doc. 99–231 Filed 1–5–99; 8:45 am]