

## **Reducing PFAS in Consumer and Other Products**

December 11, 2024 2:50– 4:15PM ET

Moderated by John Schierlmann, IndSpyre Solutions, Inc.

**Speakers:** 

- Kim Hazard, California Department of Toxic Substances Control
- Megan Arnett, University of California at Berkley, Center for Green Chemistry
- Donna Walden, greenUP!
- Katy Wolf, Consultant and Former Director of the Institute for Research and Technical Assistance
- John Schierlmann, IndSpyre Solutions, Inc

epa.gov/p2

# Overview of "PFAS"

>What are "PFAS" Products?

- Perfluorooctoanoic acid PFOA
- Perfluorooctyl sulfonates PFOS
- And others...

>Regulations

- How do we find out these are toxic chemicals?
- How do we regulate and slow down their use?
- How do we find safter alternatives and eliminate toxic ones?
- Chemistry that has been severely regulated
  - Nonylphenolethoxylates (laundry)
  - Alkylphenolethoxylates (detergents, intermediates in chemicals)
  - Bisphenol A (plastics, water bottles)
  - Microplastics in products (cosmetics)



# Overvie w

#### ≻What is being done? –a lot now

- Minnesota
  - Ban of products entering the state Jan 1, 2025, containing "PFAS"
- Maine
  - Required corporations to submit formulations containing "PFAS" Jan 1, 2023
- EPA
- Other States are inquiring with industry and universities
- Product Validators EPA Safer Choice / Green Seal
- NGOs
- Canada



# States with Legislation in Motion







## **PFAS Pollution Prevention in California**

Kimberly Hazard, PhD <u>Kim.Hazard@dtsc.ca.gov</u>

National Pollution Prevention Training and Conference December 11, 2024





Per- and polyfluoroalkyl substances (PFASs) are a class of many thousands of chemicals containing one of the strongest bonds in chemistry.





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#### PFASs are found pretty much everywhere we look...





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## ... and have been associated with multiple hazards

- Carcinogenicity
- Cardiovascular toxicity
- Developmental toxicity
- Endocrine toxicity
- Hepatotoxicity
- Immunotoxicity
- Nephrotoxicity
- Ocular toxicity

- Reproductive toxicity
- Environmental persistence
- Mobility in the environment
- Bioaccumulation
- Lactational and transplacental transfer

- Phytotoxicity and wildlife developmental, reproductive, and survival impairment
- Can cross the blood-brain barrier



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# The root of the problem?

PFASs are **used in products** and released to the environment during manufacturing, use, and at their end of life





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## **PFAS releases from consumer products**

Schellenberger et al. 2019, Environ. Sci. Technol.

• Washing of jackets releases microfibers coated with polymeric PFASs

- Zheng and Salamova 2020, Environ. Sci. Technol.
  - Laundering resulted in 100% removal of measured PFASs



https://pubs.acs.org/doi/abs/10.1021/acs.est.9b04165

https://pubs.acs.org/doi/abs/10.1021/acs.est.0c03035



### **DTSC Safer Consumer Products Program**

- Science-based, 4-part process that identifies specific products containing hazardous chemicals and evaluates safer alternatives
- Reducing hazardous chemicals in products that consumers buy and use
- Prevent regrettable substitutes



<u>Cal. Code Regs., Title 22, Division 4.5, Chapter 55 – Safer Consumer Products</u> SCP website: <u>https://dtsc.ca.gov/scp/</u>



### **Our approach: Regulating PFASs as a Class**

- All PFASs\* are Candidate Chemicals under the SCP regulations
- Class-based regulation prevents regrettable substitution



Published: 17 February 2021 CID: 025001 https://doi.org/10.1289/EHP7431

#### ⊞ Sections 🛃 PDF

🎤 Tools < Share

\*As defined by Buck et al. (2011) Integr. Environ. Assess. Manag. 7(4):513-541



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#### The P-sufficient approach

"if a chemical is **highly persistent**, its continuous release will lead to **continuously increasing contamination** (...) [and] **increasing probabilities of the occurrence of known and unknown effects**." (<u>Cousins et al. 2019</u>)

## The SCP regulatory framework allowed us to take a "P-sufficient approach"



California Code of Regulations, Title 22, Division 4.5, Chapters 54 and 55

There are potential
exposures to a
Candidate Chemical in
the product

#### AND

One or more exposures have the potential to contribute to or cause significant or widespread adverse impacts



### **Priority Product: Carpets and rugs containing PFASs**

- Carpets in California contained up to 60 tons of PFASs as of 2017
- Likely source of human exposure and PFASs in landfill leachate
  - PFAS emissions detected in indoor air and dust
  - Children may be exposed at levels 5x greater than adults



https://pubs.acs.org/doi/abs/10.1021/acs.est.9b06956 https://pubs.rsc.org/en/content/articlelanding/2005/em/b507731c https://pubs.acs.org/doi/abs/10.1021/acs.est.5b06237 https://pubs.acs.org/doi/abs/10.1021/acs.est.9b06956



## **Priority Product: Carpets and rugs containing PFASs**

- Regulation effective 7/1/21
- Manufacturers of carpets and rugs containing PFASs were required to notify DTSC by 8/30/21
- No notifications were received
- DTSC conducted compliance testing
  - Testing for total fluorine by combustion-ion chromatography (CIC)

https://dtsc.ca.gov/scp/carpets-and-rugs-withperfluoroalkyl-and-polyfluoroalkyl-substances-pfass/





## Compliance Testing: Total F in Carpet & Rug Fibers





#### **Priority Product: Treatments containing PFASs for Use on Converted Textiles or Leathers**

• Products sold for use on carpets, furnishings, clothing, shoes, etc.

Including: aerosols, wipes, liquids, gels





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#### **Priority Product: Treatments containing PFASs for Use on Converted Textiles or Leathers**

- Regulation effective 4/1/22
- Aftermarket treatments only; not products sold for use during manufacturing
- 30 manufacturers have confirmed replacement of PFASs or removal of products from California market
- DTSC conducting compliance testing





https://dtsc.ca.gov/scp/treatments-with-pfass/

#### SCP's regulations are not the same as a product ban



The focus is on alternatives analysis and having manufacturers answer key questions:

- Is it necessary?
- Is there a safer alternative?
- What are the tradeoffs?

#### Goal:

Encourage innovation and green chemistry

https://dtsc.ca.gov/scp/alternatives-analysis-guide-version-1-0-downloads/



## **California PFAS Legislative Bans**

- Plant fiber-based food packaging AB 1200\* (ban effective 2023)
- Juvenile products AB 652\* (ban effective 2023)
  - Highchairs, playpens, strollers, crib mattresses, etc.
- Cosmetics and personal care products AB 2771 (beginning in 2025)
  - DTSC is not responsible for implementation or enforcement.
- Textiles AB 1817\* (beginning in 2025)
  - Apparel, handbags, backpacks, shower curtains, furnishings, upholstery, towels, etc.
- Take All Menstrual Product-PFAS Out Now (T.A.M.P.O.N.) Act AB 2515\*
- AB 347: Grants DTSC authority for enforcement of juvenile products, textile articles, and food packaging



\*DTSC granted compliance & enforcement authority



## More work to do!





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#### What's next? Priority Product Work Plan

- Identifies product categories DTSC will evaluate during a given three-year period
- For quarterly updates, see the <u>SCP timeline</u>
- Early-stage projects:
  - Candidate Chemicals in Artificial Turf
  - Floor Polishes and Waxes Containing PFASs



Priority Product Work Plan | Department of Toxic Substances Control



#### **Collaborating on the quest for safer consumer products**





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#### Let's stay in touch!

My contact: <u>kim.hazard@dtsc.ca.gov</u>

SCP home page: <u>dtsc.ca.gov/scp</u>

Sign up for the SCP E-List to get the latest program updates!



We are working toward safer California households, workplaces, and products.



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# Safer by Design: An Industry + Academic Collaboration to find PFAS Alternatives

Megan Arnett, PhD

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National Pollution Prevention Training and Conference December 11, 2024



#### BERKELEY CENTER for GREEN CHEMISTRY

#### **Innovation for Safer Materials**

The nation's first center for interdisciplinary studies in green chemistry



Interdisciplinary education in green chemistry



Research to identify safer alternatives to hazardous chemicals and materials



#### Engagement

providing technical support to decision-makers, community groups and advocates to promote the design and use of safer products

#### Greener Solutions Not Just an Academic Exercise PBHLTH 271H

Berkeley Center for Green Chemistry - University of California, Berkeley



## **Greener Solutions Course**



# BCGC has identified safer alternatives to PFASs in a range of products



PFAS in Firefighter Turnouts Partner: International Association of Firefighters



PFASs in compostable food packaging Partner: Method



PFAS removal for carpet recycling Partner: DTSC, GSPI, XT Green



PFASs in home product packaging Partner: Method



PFAS in Floor Polishes Partner: Defend Our Health



PFAS in Aftermarket Carpet Treatments Partner: DTSC

# Safer Consumer Products Program Alternatives Assessment

- Is it necessary?
- Is there a safer alternative?
- What are the tradeoffs?



## BCGC's search for a safer alternative:

- What are the **mechanisms of performance?**
- What are the **mechanisms of toxicity?**
- How does biology (organisms) accomplish similar functions?

## Case Study #1: Alternatives to PFAS in Floor Polish

# Alternatives to PFAS in Floor Polish



Yuning Xu

MPH in Environmental Health Sciences Class of 2022



Jenna Tan

\_\_\_\_\_

PhD in Chemistry Class of 2022



Tessa Wardle

MS in Environmental Health Sciences Class of 2022

(slide credit 10-15: modified from floor polish student team final presentations)

## State of the Floor Polish Industry

- Fluorochemicals commonly used in floor polish: Capstone FS-60 and FS-65
- 2008 survey revealed that nearly every floor polish on the market contained a fluorochemical
  - Existing drive from within industry to remove PFAS
- Primary commercial users: schools, hospitals, retail and grocery stores







# PFAS in floor polish highly effective - posing challenge for comparable replacement



## Strategy 1: Rhamnolipids - bacterially produced surfactants



- Use renewable feedstocks by fermentation
- Stable in wide range of conditions (pH, temp)
- Biodegradable and low toxicity

#### Current commercial applications





Cosmetics

Pharmaceuticals





#### Agriculture

Bioremediation

Liepins, J. *et al.*. Glycolipid Biosurfactant Production from Waste Cooking Oils by Yeast: Review of Substrates, Producers and Products. *Fermentation* **2021**, *7*, 136. Image credits: PxHere, Pixabay, Flickr, The Science Explorer
Strategy 2: Amino acids - versatile building blocks to make a variety of surfactants



- Anionic cleaning
- Cationic anti-microbial agent

Image credits: BioNinja, Pixabay, Flickr



- Produce from renewable and raw feedstock (e.g. vegetable oils)
- High surface activity
- Low toxicity and quick biodegradation

Current commercial applications



Cleaning

# Rhamnolipids and amino acids form environmentally friendly surfactants.



## Safer Consumer Products Program Alternatives Assessment

- Is it necessary? *PFASs offer critical advantages in floor polish formulation, but its worth evaluating the necessity and timing of floor polishing*
- Is there a safer alternative? Based on our research, safer alternatives for PFASs used in floor polishes exist
- What are the tradeoffs?

Cost, feedstocks

## Case Study #2: Alternatives to PFAS in Firefighter Turnout Gear

#### **The Team**











Grace Campbell Environmental Health Sciences MS Candidate

Sophia Glazer Environmental Health Sciences MPH Candidate Brittany Stinger Chemistry PhD Candidate

McKenna Thompson Environmental Health Sciences MPH Candidate Sophie Thompson Environmental Health Sciences MPH Candidate

(slide credit 18-20: firefighter student team final presentation)

#### **Current Turnout Gear**

#### Typically 3 layers in firefighter turnout gear: Thermal Layer, Moisture Barrier, Outer Shell



The Moisture Barrier contains a Polytetrafluoroethylene (PTFE) layer.

**PTFE** is ....

- a synthetic fluoropolymer (a polymer with multiple fluorine carbon bonds)
- one of the most well-known and applied per- and polyfluoroalkyl substances (PFAS) - also used in the coating of nonstick cookware.

(Peeling Back the Fire Suit "Onion", Layer #2, n.d.)

**PFAS** were detected in all three layers of the firefighter turnout gear. **PFAS chemicals present on firefighter turnout gear pose a risk to human and environment health.** (Graham et al., 2020)

#### **Conclusions and Recommendations**



#### **Policy Recommendations**

- We advise the NFPA to *drop the light degradation and viral penetration tests* and keep those that remain.
  - Light degradation & viral penetration tests result in over-engineered moisture barrier
  - Standards are limiting viable PFAS-free options

#### NFPA Standards Critique Light Degradation Resistance: The UV Test

- **Unnecessary** for moisture barrier: the middle layer of turnout gear does not come into contact with UV light.
- PTFE is the **only material** that meets this requirement → Removing the UV test allows for PFAS-free moisture barriers.
- May 2021, a Tentative Interim Amendment was filed by IAFF to remove UV test from the standards but NFPA denied it and subsequent appeals



(Graham et. al., 2020)

"The UV light degradation test for moisture barriers is illogical, not supported by science, stands as a hurdle to advancing the state of the art in firefighter PPE"

- IAFF TIA Appeal

"The UV light test is stopping fire departments from moving towards procurement of PPE that doesn't contain forever chemicals. **Firefighters should be fighting fires, not NFPA** and the gear companies."

Retired battalion chief of New Haven CT Fire Department

## Safer Consumer Products Program Alternatives Assessment

• Is it necessary?

Based on our research, PFASs in firefighter turnout gear are not necessary for optimal performance and firefighter protection

• Is there a safer alternative?

By removing the NFPA requirements for UV protection in the middle layer of turnout gear, the industry would open the door to safer materials innovation

• What are the tradeoffs?

"[Chemicals] are the connection between cancer and climate change, so if you really want to make an impact, focus on chemistry." (Marty Mulvihill, 2019)

## Extra Slides

## **Conclusions and Recommendations Technical Assessment**



#### **Cellulose Fibers**

Meets Tears Resistance Low Water Contact angle

#### **Short Chain PHAs**

High Water Contact Angle Data Gaps

#### Graphene

Low Water Contact Angle Low Thermal Resistance Coatings

Silicon Based Flame Retardants

> Increase Water Contact Angle Data Gaps

#### Phosphorus Based Flame Retardants

Increase Thermal Decomposition Temperature Data Gaps



#### Polyethylene

High Water Contact Angle Data Gaps on Thermal Resistance

#### Polyurethane

Low Water Contact Angle Meets other standards

## **Conclusions and Recommendations** Hazard Assessment



Manufacturing Risk Data Gaps

**Fibers** 

#### **Short Chain PHAs**

Biodegradable Data Gaps

#### Graphene

Concerns of Persistence Data Gaps Coatings

Silicon Based Flame Retardants

> Few Group 1 Endpoints Potential Manufacturer Risk

#### Phosphorus Based Flame Retardants

Many Group 1 Endpoints Data Gaps for Novel PFRs



#### Polyethylene

Monomer Potential Carcinogen Polymer Persistence

#### Polyurethane

Monomer Potential Carcinogen Potential Manufacturer Risk

## Technical performance is highly dependent on concentration and surface choice

|  | Baseline Surfactants             |                        |                                       | Alternative Surfactants              |  |                                   |
|--|----------------------------------|------------------------|---------------------------------------|--------------------------------------|--|-----------------------------------|
| Technical properties                         | FS-60                            | 0 FS-65                |                                       | Sodium dodecyl<br>sulfate (SDS)      | Rhamnolipid                                  | Sodium lauroyl<br>glutamate (SLG) |
| Water contact angle (°)                      | <b>M</b><br>~51-56°*             |                        | <b>H</b><br>~20° at 3.48 mM on<br>PVC | <b>M</b><br>~20° at 1.5 mM on<br>PVC | <b>L</b><br>138.69° on sericite<br>(mineral) |                                   |
| Surface tension<br>(mN/m)                    | <b>H</b><br>19 at 0.05%          | <b>H</b><br>18 at 0.05 | %                                     | Н<br>23.8 - 34.6                     | <b>Н</b><br>26-29                            | Н<br><30                          |
| Critical micelle concentration (mM)          | <b>H</b><br>0.23*                |                        | <b>M</b><br>8-8.5                     | <b>Н</b><br>0.41                     | <b>Н</b><br>0.48                             |                                   |
| LogKow (octanol water partition coefficient) | <b>M</b><br>2.51 (0.276 - 5.99)* |                        | <b>Н</b><br>1.69                      | <b>L</b><br>5.77 (4.22 - 7.38)       | <b>H</b><br>0.597*                           |                                   |
| High efficacy (H)                            | loderate efficacy (M)            |                        | ow efficacy <b>(L)</b>                | *=data for simi                      | ar compound                                  |                                   |

#### The surfactant has assistance in floor polish formulas!

### Hazard Table

|                   |                      | N-EtFOSAA  | PFBS     | PFOS      | Rhamnolipid | Sodium<br>lauroyl<br>glutamate |  |
|-------------------|----------------------|------------|----------|-----------|-------------|--------------------------------|--|
|                   |                      | 67584-51-4 | 375-73-5 | 1763-23-1 | 4348-76-9   | 29923-31-7                     | Key                                    |
|                   | Carcinogen           | 3          | 3        | 2         | 5           | DG                             | LC = Low concern                       |
|                   | Mutagen              | LC         | LC       | 3         | 4           | LC                             | PC= Potential concern                  |
| Group I endpoints | Repro/Dev            | DG         | 2        | 1         | DG          | РС                             | *prediction based on similar compounds |
|                   | Endocrine disruptor  | DG         | 1        | 1         | DG          | DG                             | Color scale:                           |
|                   | Mammalian Toxicity   | 4*         | 2        | 4         | 4           | 5                              | 1= Very High Hazard                    |
|                   | Systemic Toxicity    | DG         | 2        | 1         | LC          | LC                             | 2= High Hazard                         |
| Group II          | Neurotoxicity        | DG         | 3        | 1         | DG          | DG                             | 3 = Moderate Hazard                    |
| endpoints         | Respiratory Toxicant | DG         | PC       | PC        | 4           | LC                             | 4 = Low Hazard                         |
|                   | Skin irritation      | PC         | 1        | LC        | 4           | 2                              | 5 = Very Low Hazard                    |
|                   | Eye irritation       | DG         | 2        | 2         | 1           | 2                              | LC = Low Concern                       |
| Eco toxicity      | Aquatic Toxicity     | 2*         | 3        | 2         | 2           | LC                             | PC = Potential Concern                 |
| Eato              | Persistence          | 1*         | 1        | 1         | 5           | 5                              | <b>Bold</b> = confidence in score      |
| Fate              | Bioaccumulation      | PC*        | PC       | 1         | PC          | PC                             | <i>Italicized</i> = potential score    |

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|                   | Endocrine disruptor  | DG         | 1        | 1         | DG          | DG                             | Color scale:                              |
|                   | Mammalian Toxicity   | 4*         | 2        | 4         | 4           | 5                              | 1= Very High Hazard                       |
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| endpoints         | Respiratory Toxicant | DG         | PC       | PC        | 4           | LC                             | 4 = Low Hazard                            |
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|                   | Eye irritation       | DG         | 2        | 2         | 1           | 2                              | LC = Low Concern                          |
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| rate              | Bioaccumulation      | PC*        | PC       | 1         | PC          | PC                             | <i>Italicized</i> = potential score       |

# Case Study #3: Removing PFAS from recycled carpets

## Recycled carpets are contaminated with PFAS

- Nearly 4 billions pounds of post consumer carpets are disposed of annually
- CA recycles >20% of carpets but these are contaminated with PFAS
- U.S states without carpet stewardship programs send 96% of carpets to landfills/burning
  - PFAS released through air emissions and landfill leachate

#### Products from PCC Recycling



![](_page_54_Picture_7.jpeg)

**Recovered Nylon** 

Recovered Polypropylene

![](_page_54_Picture_10.jpeg)

Image credit ACS GC consortium 2021

## Additional alternatives considered

![](_page_55_Figure_1.jpeg)

Sułek, M.W., et al. "Alkyl polyglucosides as components of water based lubricants." Journal of surfactants and detergents (2013): 369-375.

#### Siloxanes/silicones

- Higher surface activity than hydrocarbons (lowers surface tension to values similar to fluorosurfactants)
- But persistent, • bioaccumulative, and toxic

#### Alkyl Polyglucosides

- Can be low-cost and lowecological impact
- Current commercial • applications
- Has been tested by floor • polish industry and is not successful when mixed into formulation

![](_page_55_Picture_10.jpeg)

### Pine oil

Turpentine oil

Active component: turpentine oil Low mammalian toxicity

- Current application in detergents/cleaning products
- Concern with skin and eye irritation/corrosivity
- Mixture of chemical constituents

#### XT Green's Search for PFAS Solutions

#### How much PFAS did we remove from the carpet fiber?

- Round #4 of testing: Determine the residual PFAS in fiber & % removal
  - Strong base in aqueous solution plus organic solvent #4 removes 99 to > 99% of PFAS from carpet fiber

| PFAS # of I<br>Carbons |   | PFAS in fiber<br>(ppt) | 2 M NaOH +<br>50% Solvent #4<br>(ppt) | Residual PFAS in<br>Treated Fiber<br>(ppt) | % Removal |
|------------------------|---|------------------------|---------------------------------------|--|-----------|
| PFBA                   | 4 | ND                     | 591,524                               | 1020                                       | >99       |
| PFBS                   | 4 | ND                     | ND                                    | ND   | -         |
| PFPeA                  | 5 | 143000                 | ND                                    | ND   | -         |
| PFHxA                  | 6 | 257,100                | 160,088                               | 1698                                       | 99        |
| PFHxSaAm               | 6 | Surprise!              | 3,498,975                             | ND   | >99       |
| PFHpA                  | 7 | 49,000                 | 131,438                               | 2  | >99       |
| 6:2 FtS                | 8 | 8,804,400              | ND                                    | ND   | -         |
| PFOA                   | 8 | 125,000                | 91,050                                | 1035                                       | 99        |
| PFOS                   | 8 | 83,800                 | 209,738                               | 630  | >99       |

Determine the concentration of PFAS in the PC4

Concentration of PFAS in PC4 similar to fiber, short-chain PFAS in PC4

*Slide credit Ned Antell, ACS GC consortium 2021* 

| AS # of PFAS in PC4 2 M NaOH +<br>Carbons (ppt) 50% Solvent #4<br>(ppt) |   | Residual PFAS in<br>Treated PC4<br>(ppt)  | % Removal  |  |
|---|---|---|--|--|
| 4   | 265,920   | TBD   | TBD  | TBD  |
| 4   | 40,588  | TBD   | TBD  | TBD  |
| 5   | 158,330   | TBD   | TBD  | TBD  |
| 6   | 47,833  | TBD   | TBD  | TBD  |
| 6   | 2,695,133   | TBD   | TBD  | TBD  |
| 7   | 38,830  | TBD   | TBD  | TBD  |
| 8   | ND  | TBD   | TBD  | TBD  |
| 8   | 46,535  | TBD   | TBD  | TBD  |
| 8   | 19,243  | TBD   | TBD  | TBD  |
| 9   | 46,368  | TBD   | TBD  | TBD  |
|   | # of<br>Carbons<br>4<br>4<br>5<br>6<br>6<br>6<br>7<br>8<br>8<br>8<br>8<br>8<br>8<br>9 | H of<br>Carbons PFAS in PC4<br>(ppt)   4 265,920   4 40,588   5 158,330   6 47,833   6 2,695,133   7 38,830   8 ND   8 46,535   8 19,243   9 46,368 | # of<br>Carbons PFAS in PC4<br>(ppt) 2 M NaOH +<br>50% Solvent #4<br>(ppt)   4 265,920 TBD   4 40,588 TBD   5 158,330 TBD   6 47,833 TBD   6 2,695,133 TBD   7 38,830 TBD   8 ND TBD   8 19,243 TBD   9 46,368 TBD | # of<br>Carbons PFAS in PC4<br>(ppt) 2 M NaOH +<br>50% Solvent #4 Residual PFAS in<br>Treated PC4<br>(ppt)   4 265,920 TBD TBD   4 40,588 TBD TBD   5 158,330 TBD TBD   6 47,833 TBD TBD   7 38,830 TBD TBD   8 ND TBD TBD   8 19,243 TBD TBD   9 46,368 TBD TBD |

Safer Alternatives to Industrial PFAS Use National Pollution Prevention Training and Conference December 11, 2024

![](_page_57_Picture_1.jpeg)

![](_page_57_Picture_2.jpeg)

Katy Wolf, Ph.D Donna Walden, President, greenUP!

### Background

- Worked on EPA project on ozone depleting substance alternatives for U.S. ban at the Rand Corporation as part of the Montreal Protocol
  - Focused on air conditioning and refrigeration, foam blowing, solvent and fire extinguishant applications
- Established nonprofit organization in Los Angeles with heavy focus on solvents and solvent alternatives
- Worked with flexible foam manufacturers on alternatives in California
- Continued following developments in other applications

![](_page_58_Picture_6.jpeg)

![](_page_58_Picture_7.jpeg)

## Historical Perspective

- CFCs,1,1,1-trichloroethane (TCA) banned worldwide under Montreal Protocol in 1996 because they contributed to ozone depletion
  - Used in refrigeration and air conditioning, foam blowing, solvent, aerosol and fire extinguishant applications
- HCFCs marketed as replacements
  - Phased out later since they also contributed to ozone depletion
  - Note that nPB came on the market at the same time and chlorinated solvents took an increasing market share

## Historical Perspective (Continued)

- HFCs and HFEs marketed as replacements
  - Don't cause ozone depletion
  - Do cause global warming
- HFCs are being phased out but will take many years
- 3M is voluntarily phasing out HFEs
- HFOs are now being marketed as replacements
  - Don't cause ozone depletion or global warming
  - Replacement is well underway

### Chain of Substitution

![](_page_61_Figure_1.jpeg)

## What is the Problem?

- European countries define nearly all HFCs, HFEs, HFOs as PFAS chemicals
  - Definition covers chemicals containing certain structural components
- U.S. has a less comprehensive definition that includes some HFCs and some HFOs based on structural components
- Current U.S. position is that it designates PFAS chemicals on a case-by-case basis which has not included HFCs, HFEs or HFOs
- Issue has become controversial

Hydrofluorocarbons – H-F-C Hydrofluoroethers – H-F-E Hydrofluoroolefins – H-F-O

"F" - Fluorinated gases (F-gases) contain fluorine.

## Emerging PFAS Issue

- Replacement of HFCs and HFEs with HFOs is underway and could end up being regrettable substitution
- Have focused attention on PFAS water problem and detection methods for air are still being worked out
- Regulating PFAS are used in consumer and institutional products but generally not in industrial products
- To avoid a regrettable substitution, need to find not in kind (NIK) alternatives as replacements in industrial products

## New PFAS Project

- Focus is primarily on industrial product PFAS
- Evaluate safer NIK alternatives to HFCs, HFEs and HFOs in several applications
  - Refrigerants
  - Foam blowing agents
  - Solvents
  - Propellants

![](_page_64_Picture_7.jpeg)

## Initial List of Specific Uses

- Refrigerants used in chillers for large commercial buildings
- Refrigerants used in motor vehicle air conditioners
- Foam blowing agents used in spray polyurethane foam for walls and roofs in homes and buildings
- Foam blowing agents for flexible foam used in furniture
- Solvents used in precision, electronics, metal cleaning
- Propellants and solvents used in aerosol products
- May add a few fluoropolymer applications

![](_page_65_Picture_8.jpeg)

## Project Tasks

- Work with manufacturers/users of NIK alternatives
  - Evaluate costs
  - Evaluate feasibility
  - Evaluate health & environmental effects
- Develop at least four case studies
- Arrange and hold webinars
- Prepare final project report

![](_page_66_Picture_8.jpeg)

## Summary and Conclusions

- HFCs, HFEs and HFOs may eventually be designated as PFAS chemicals in the U.S.
- Under an abundance of caution, we need to find safe, viable and cost-effective NIK alternatives for companies to adopt in industrial applications
- The project described here is designed to do this

![](_page_67_Figure_4.jpeg)

## THANK YOU!

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#### WWW.NVGREENBUSINESS.ORG

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![](_page_69_Picture_0.jpeg)

## Ind Spyre Solutions, Inc. John Schierlmann Reducing PFAS in Consumer and Other Products

December 2024

## Overvie w

- □ Where are "PFAS"?
- □ Health Effects
- □ Challenges of Removal
- □ Products Seeing Successful Removal
- □ How does industry know what to do?
- □ Next Steps

![](_page_70_Picture_7.jpeg)

## Products Containing Fluorinated Materials

≻Shoes (water repellency)

≻Clothing (oil resistance)

➢Food wrappers (oil resistance)

>Fire Fighting foam (fire suppression)

Carpets (non-staining)

>Paints and coatings (flow/leveling/UV resistance/chemical resistance)

>Industrial intermediates (emulsifying, others)

![](_page_71_Picture_8.jpeg)
### Human Health Effects

≻High cholesterol

- ≻Ulcerative colitis
- ➢Pregnancy induced hypertension
- ≻Thyroid disease
- ≻Testicular, rectal and kidney cancer
- >Decreased response to vaccines



# Benefits of Fluorine Chemistry in Paints, Coatings and Cleaners

Extreme chemical resistance in coatings
Extreme UV resistance in buildings and exterior surfaces
Outstanding surface wetting and emulsification



Fluorinated acrylic sealers



Fluorinated vinyl coatings for buildings



Coating with poor surface wetting



# Why PFAS are Hard to Replace

- Surface tension is affected by many factors
- Unbalanced surface tension causes wetting problems and adhesion failure
- Typically, fluorinated surfactants must be replaced by two or three different surfactants





#### Testing Gloss and Haze Polishes Silicone/alcoholethoxylate hybrid



Product without PFAS



### Product with silicone replacement



Product with silicone chemistry



### Contact Angle measured by Goniometer: Acrylic Polymer Floor Finish (PFAS)



| Formula B: Acrylic Polymer Floor Finish (PFAS) |    |
|--|----|
| Starting Contact Angle (mN/m) [°]              | 56 |
| Ending Contact Angle (mN/m) [°]                | 35 |



### Contact Angle measured by Goniometer: Acrylic Polish with Silicone/Alcoholethoxylate hybrid



| Formula A: Acyrlic polish with hybrid surfactants | [°] |
|---|-----|
| Starting Contact Angle (mN/m) [°]                 | 40  |
| Ending Contact Angle (mN/m) [°]                   | 21  |



How Does Industry Keep Up?

- > EPA and State regulations
- > Trade Organizations
  - HCPA Household Commercial and Products Association
  - ACI-American Cleaning Institute
  - ISSA Industrial Sanitation
  - ACA American Coatings Association
- Product Validation Organizations
  - EPA Safer Choice
  - Green Seal





Advancing Clean. Driving Innovation.









# Changes to Products

- ≻Paints and Coatings
  - Silicone replacements
- ≻Cleaners
  - Silicone or biobased surfactant replacements
- ➢Carpets
- ≻Fire Fighting Foam
  - Bio alternatives



### Conclusions

PFAS" chemistry has been around for over 60 years
Very hard to displace

States are moving to ban intentionally added PFAS from products

>Industry is acting, replacements are difficult





### **Reducing PFAS in Consumer and Other Products**

December 11, 2024 2:50– 4:15PM ET

Moderated by John Schierlmann, IndSpyre Solutions, Inc.

**Speakers:** 

- Kim Hazard, California Department of Toxic Substances Control
- Megan Arnett, University of California at Berkley, Center for Green Chemistry
- Donna Walden, greenUP!
- Katy Wolf, Consultant and Former Director of the Institute for Research and Technical Assistance
- John Schierlmann, IndSpyre Solutions, Inc

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