2.4 MUNICIPAL SOLID WASTE LANDFILLS 2.4 Municipal Solid Waste Landfills

Disclaimer: Emission factors in AP-42 are neither EPA-recommended emission limits (e.g., best available control technology or BACT, or lowest achievable emission rate or LAER) nor standards (e.g., National Emission Standard for Hazardous Air Pollutants or NESHAP, or New Source Performance Standards or NSPS). Use of these factors as source-specific permit limits and/or as emission regulation compliance determinations is NOT recommended by EPA. Becaus emission factors essentially represent an average of a range of emission rates, approximately half of the subject sources are expected to have emission rates greater than the emission factor, and the other half are expected to have emission rates less than the emission factor. As such, EPA does not recommend using emission factors as limits or standards. This could cause, for example, a permit limit using an AP-42 emission factor resulting in approximately half of the sources being in noncompliance. We recommend source testing be done for the best possible emission values. For more information on the use of emission factors, please refer to the AP-42 Introduction.

2.4.1 General1-4

A municipal solid waste (MSW) landfill unit is a discrete area of land or an excavation that receives household waste, and that is not a land application unit, surface impoundment, injection well, or waste pile. An MSW landfill unit may also receive other types of wastes, such as commercial solid waste, nonhazardous sludge, and industrial solid waste. The municipal solid waste types potentially accepted by MSW landfills include (most landfills accept only a few of the following categories):

- MSW.
- · Household hazardous waste,
- Municipal sludge,
- · Municipal waste combustion ash,
- · Infectious waste,
- Waste tires,
- Industrial non-hazardous waste,
- · Conditionally exempt small quantity generator (CESQG) hazardous waste,
- Construction and demolition waste,
- · Agricultural wastes,
- · Oil and gas wastes, and
- Mining wastes.

In the United States in 2018, approximately 57 percent50% of solid waste iswas landfilled, 16 percent is incinerated12% was combusted for energy recovery, and 27 percent is 32% was recycled or composted. There were an estimated 2,5001,274 active MSW landfills in the United States in 19952021. These landfills were estimated to receive 189339 million megagrams (Mg) (208373 million tons) of waste annually, with 6 In 1998, 55 to 60 percent 65% of MSW was reported as household waste, and 35 to 45 percent of MSW was reported as commercial waste.

References for this AP-42 section are available electronically here. The reader is referred to Sections 13.2.2 (Unpaved Roads), and 11.2.4 (Heavy Construction Operations) of the Electronic AP-42: Compilation of Air Emissions Factors from Stationary Sources, and Section II-7 (Construction Equipment) of Volume II, of the AP-42 document for determination of associated fugitive dust and exhaust emissions from these emission sources at MSW landfills. In addition to this, Section 3.1 (Stationary Gas Turbines) and Section 13.5 (Industrial Flares) of the Electronic AP-42: Compilation of Air Emissions Factors from Stationary Sources also contains emission factors for landfill gas (LFG) fired turbines and open flares, respectively.

2.4.2 Process Description^{2,5}8

There are three major designs for municipal landfills. These are the area, trench, and ramp methods. All of these methods utilize a three_step process, which includes spreading the waste, compacting the waste, and covering the waste with soil. The trench and ramp methods are not commonly used; and are not the preferred methods when liners and leachate collection systems are utilized or required by law. The area fill method involves placing waste on the ground surface or landfill liner, spreading it in layers, and compacting with heavy equipment. A daily soil cover is spread over the compacted waste. The trench method entails excavating trenches designed to receive a day's worth of waste. The soil from the excavation is often used for cover material and wind breaks. The ramp method is typically employed on sloping land, where waste is spread and compacted similar to the area method, however, the cover material obtained is generally from the front of the working face of the filling operation.

Modern landfill design often incorporates liners constructed of soil (i.e., recompacted clay), or synthetics (i.e., high density polyethylene), or both to provide an impermeable barrier to leachate (i.e., water that has passed through the landfill) and gas migration from the landfill.

2.4.3 Control Technology 1,2,69

The Resource Conservation and Recovery Act (RCRA) Subtitle D regulations promulgated on October 9, 1991, require that the concentration of methane generated by MSW landfills not exceed 25 percent% of the lower explosive limit (LEL) in on-site structures, such as scale houses, or the LEL at the facility property boundary.

The original New Source Performance Standards (NSPS) and Emission Guidelines (EG) for air emissions from MSW landfills for certain new and existing landfills were published in the Federal Register on March ±12, 1996. Since then, the MSW NSPS/EG were updated on August 29, 2016. Additionally, a National Emission Standard for Hazardous Air Pollutants (NESHAP) was promulgated on January 16, 2003, and the residual risk and technology review (RTR) was promulgated on March 26, 2020, with technical corrections to the RTR promulgated on February 3, 2022. A history of MSW landfills can be found on the EPA's Municipal Solid Waste Landfills: New Source Performance Standards (NSPS), Emission Guidelines (EG) and Compliance Times website. The regulation requires NSPS, EG, and NESHAP for MSW landfills are similar in that Best Demonstrated Technology (BDT) be used to reduce MSW landfill they regulate emissions from affected new and existing MSW landfills emitting greater than or equal to 50 Mg/yr (55 tons/yr) of of landfill gas using non-methane organic compounds (NMOCs). The MSW landfills that are affected by the NSPS/Emission Guidelines are each new MSW landfill, and each existing MSW landfill that has accepted waste since November 8, 1987, or that has capacity available for future use. The NSPS/Emission Guidelines affect landfills with NMOC) as an estimate for VOC emissions. These regulations established a design capacity of 2.5 million Mg (2.75 million tons) or more and 2.5 million cubic meters and NMOC emission rate thresholds that if exceeded require landfills to install a gas collection and control system (GCCS). Control systems require: (1) a well-designed and well-operated gas collection systemGCCS, and (2) a control device capable of reducing NMOCs in the collected gas by 98 weight-percent.

Landfill gas (LFG) collection systems, <u>also referred to as GCCS</u>, are either active or passive systems. Active collection systems provide a pressure gradient in order to extract LFG by use of mechanical blowers or compressors. Passive systems allow the natural pressure gradient created by the increase in pressure created by LFG generation within the landfill to mobilize the gas for collection.

LFG control and treatment options include (1) combustion of the LFG, and (2) purification of the LFG. Combustion techniques include techniques that do not recover energy (i.e., flares and thermal incinerators), and techniques that recover energy (i.e., gas turbines and internal combustion engines) and generate electricity from the combustion of the LFG. Boilers can also be employed to recover energy from LFG in the form of steam. Flares involve an open combustion process that requires oxygen for combustion; and can be open or enclosed. Thermal incinerators heat an organic chemical to a high enough temperature in the presence of sufficient oxygen to oxidize the chemical to carbon dioxide (CO₂) and water. Purification techniques can also be used to process raw landfill gas to pipeline quality natural gas by using adsorption, absorption, and membranes.

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2.4.4 Emissions^{2,7}10

Methane (CH₄) and CO₂ are the primary constituents of landfill gas; and are produced by microorganisms within the landfill under anaerobic conditions. Transformations of CH₄ and CO₂ are mediated by microbial populations that are adapted to the cycling of materials in anaerobic environments. Landfill gas generation, including rate and composition, proceeds through four phases. The first phase is aerobic [i.e., with oxygen (O₂) available] and the primary gas produced is CO₂. The second phase is characterized by O₂ depletion, resulting in an anaerobic environment, where large amounts of CO₂ and some hydrogen (H₂) are produced. In the third phase, CH₄ production begins, with an accompanying reduction in the amount of CO₂ produced. Nitrogen (N₂) content is initially high in landfill gas in the first phase; and declines sharply as the landfill proceeds through the second and third phases. In the fourth phase, gas production of CH₄, CO₂, and N₂ becomes fairly steady. The total time and phase duration of gas generation varies with landfill conditions (i.e., waste composition, design management, and anaerobic state).

Typically, LFG also contains a small amount of non-methane organic compounds (NMOC). NMOC. This NMOC fraction often contains various organic hazardous air pollutants (HAP), greenhouse gases (GHG), and compounds associated with stratospheric ozone depletion. The NMOC fraction also contains volatile organic compounds (VOC). The weight fraction of VOC can be determined by subtracting the weight fractions of individual compounds that are non-photochemically reactive (i.e., negligibly -reactive organic compounds as defined in 40 CFR 51.100).

Other emissions associated with MSW landfills include combustion products from LFG control and utilization equipment (i.e., flares, engines, turbines, and boilers). These include carbon monoxide (CO), oxides of nitrogen (NO_X), sulfur dioxide (SO₂), hydrogen chloride (HCl), particulate matter (PM) and other combustion products (including HAPs). PM emissions can also be generated in the form of fugitive dust created by mobile sources (i.e., garbage trucks) traveling along paved and unpaved surfaces. The reader should consult AP-42 Volume I Sections 13.2.1 and 13.2.2 for information on estimating fugitive dust emissions from paved and unpaved roads.

The rate of emissions from a landfill is governed by gas production and transport mechanisms. Production mechanisms involve the production of the emission constituent in its vapor phase through vaporization, biological decomposition, or chemical reaction. Transport mechanisms involve the transportation of a volatile constituent in its vapor phase to the surface of the landfill, through the air boundary layer above the landfill, and into the atmosphere. The three major transport mechanisms that enable transport of a volatile constituent in its vapor phase are diffusion, convection, and displacement.

2.4.4.1 Uncontrolled Emissions —

EPA notes that fugitive emissions include those emissions not gathered by collection devices (uncollected). Here, fugitive, uncollected, and uncontrolled emissions are synonymous terms. One way to estimate uncontrolled emissions of the various compounds presenta pollutant in landfill gas, total landfill gas emissions must first be estimated. Uncontrolled CH₄ emissions—is to begin by determining the annual volume of landfill methane generation, accounting for air infiltration as necessary, and to use the ideal gas law to provide pollutant mass per year. Methane generation may be estimated for individual landfills by using amultiplying the result of Equation HH-1, found at 40 CFR 98.343(a)(1), by 1474.83 to obtain methane generation for the reporting year for which emissions are calculated in terms of cubic meters per year. The equation is as follows:

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$$Q_{CH4} = G_{CH4} \frac{1000}{(0.0192)(35.3147)} = G_{CH4} \times 1474.83 \tag{1}$$

where: Q_{CH_4}

Methane generation rate for the reporting year, m³/yr;

 G_{CH_4} = Result of Equation HH-1, metric tons CH₄/yr;

1000 = Conversion, kilograms to metric tons;

0.0192 = Density of methane at 60° F and 14.7 psia, kg/ft³; and

35.3147 = conversion, ft³ to m³.

The Landfill Gas Emissions Model (LandGEM) is an automated estimation tool with a Microsoft Excel interface that can be used to estimate generation or emissions rates for total landfill gas, methane, carbon dioxide, and NMOCs, and individual air pollutants from municipal solid waste landfills. Version 3.1, available from the following EPA website: https://www.epa.gov/system/files/other-files/2023-12/landgem-v3.1beta-dec-2023.xlsm, was updated in December 2023 and includes Equation HH-1 from 40 CFR 98.343(a)(1) and its selectable parameters, as well as the theoretical first-order kinetic model of methane production found in LandGEM Version 3.03. Note that the regulatory defaults for the equations in LandGEM Version 3.03 must be used when required by programs such as 40 CFR Part 60 subpart XXX, 40 CFR Part 60 Subpart Cf (Emission Guideline), or 40 CFR Part 63 AAAA.

developed by the EPA. This model is known as the Landfill Air Emissions Estimation model, and can be accessed from the Office of Air Quality Planning and Standards Technology Transfer Network Website (OAQPS TTN Web) in the Clearinghouse for Inventories and Emission Factors (CHIEF) technical area (URL http://www.epa.gov/ttn/chief). The Landfill Air Emissions Estimation model equation is as follows:

$$Q_{CH_{+}} = L_{e} R - (e^{-8ke} - & e^{-8kt})$$
 (1)

where:

 $\begin{array}{ll} Q_{CH_4} & = & \frac{\text{Methane generation rate at time t, m}^3/\text{yr;}}{\text{E}_{\Theta}} & = & \frac{\text{Methane generation potential, m}^3 \text{CH}_4/\text{Mg refuse;}}{\text{Methane generation potential, m}^3} \end{array}$

R = Average annual refuse acceptance rate during active life, Mg/yr;

e = Base log, unitless;

k = Methane generation rate constant, yr⁻¹;

c = Time since landfill closure, yrs (c = 0 for active landfills); and

Time since the initial refuse placement, yrs.

It should be noted that the model above was designed to estimate LFG generation and not LFG emissions to the atmosphere. Other fates may exist for the gas generated in a landfill, including capture and subsequent microbial degradation within the landfill's surface layer. Currently, there are no data that adequately address this fate. It is generally accepted that the bulk of the gas generated will be emitted through cracks or other openings in the landfill surface.

Site-specific landfill information is generally available for variables R, c, and t. When refuse acceptance rate information is scant or unknown, R can be determined by dividing the refuse in place by the age of the landfill. If a facility has documentation that a certain segment (cell) of a landfill received *only* nondegradable refuse, then the waste from this segment of the landfill can be excluded from the calculation of R. Nondegradable refuse includes concrete, brick, stone, glass, plaster, wallboard, piping, plastics, and metal

objects. The average annual acceptance rate should only be estimated by this method when there is inadequate information available on the actual average acceptance rate. The time variable, t, includes the total number of years that the refuse has been in place (including the number of years that the landfill has accepted waste and, if applicable, has been closed).

Values for variables L_{φ} and k must be estimated. Estimation of the potential CH_4 generation capacity of refuse (L_{φ}) is generally treated as a function of the moisture and organic content of the refuse. Estimation of the CH_4 generation constant (k) is a function of a variety of factors, including moisture, pH, temperature, and other environmental factors, and landfill operating conditions. Specific CH_4 generation constants can be computed by the use of EPA Method 2E (40 CFR Part 60 Appendix A).

The Landfill Air Emission Estimation model includes both regulatory default values and recommended AP-42 default values for L_{Θ} and k. The regulatory defaults were developed for compliance purposes (NSPS/Emission Guideline). As a result, the model contains conservative L_{Θ} and k default values in order to protect human health, to encompass a wide range of landfills, and to encourage the use of site-specific data. Therefore, different L_{Θ} and k values may be appropriate in estimating landfill emissions for particular landfills and for use in an emissions inventory.

Recommended AP-42 defaults include a k-value of 0.04/yr for areas recieving 25 inches or more of rain per year. A default k of 0.02/yr should be used in drier areas (<25 inches/yr). An $\rm L_{\odot}$ -value of 100 m³/Mg (3,530 ft³/ton) refuse is appropriate for most landfills. Although the recommended default k and $\rm L_{\odot}$ are based upon the best fit to 21 different landfills, the predicted methane emissions ranged from 38 to 492% of actual, and had a relative standard deviation of 0.85. It should be emphasized that in order to comply with the NSPS/Emission Guideline, the regulatory defaults for k and $\rm L_{\odot}$ must be applied as specified in the final rule.

When gas generation reaches steady state conditions, LFG consists of approximately 40 percent% by volume CO_2 , 55 percent% CH_4 , 5 percent% N_2 (and other gases), and trace amounts of NMOCs. Therefore, the estimate derived for CH_4 generation using the Landfill Air Emissions Estimation modelLandGEM can also be used to represent CO_2 generation. Addition of the CH_4 and CO_2 emissions will yield an estimate of total landfill gas emissions. If site-specific information is available to suggest that the CH_4 content of landfill gas is not 55 percent,%, then the site-specific information should be used, and the CO_2 emission estimate should be adjusted accordingly.

For landfills, volatile organic compound (VOC) emissions are equivalent to NMOC emissions minus the emissions from compounds with low to no photochemical reactivity. Predominant compounds with low to no photochemical reactivity found in landfills include methyl chloroform, acetone, methylene chloride, tetrachloroethylene, chlorodifluoromethane, dichlorodifluoromethane, and ethane. When the contribution of emissions from these compounds with low to no photochemical reactivity is low, then NMOC emissions are a good surrogate for VOC emissions. Recent data review shows that the contribution of these seven predominant compounds with low to no photochemical reactivity to be less than 6.5% of NMOC.¹¹⁻³⁸

Most of the NMOC emissions result from the volatilization of organic compounds contained in the landfilled waste. Small amounts may be created by biological processes and chemical reactions within the landfill. The current version of the Landfill Air Emissions Estimation model_LandGEM contains a proposed regulatory default value for total NMOC of 4,000 ppmv, expressed as hexane. However, available data show that there is a range of over 4,400 ppmv for total NMOC values from landfills. The proposedThe regulatory default value for NMOC concentration was developed for regulatory compliance purposes (40 CFR Part 60, Subpart XXX) and to provide the most cost-effective default values on a national basis. For emissions inventory purposes, site-specific information should be taken into account when determining the total NMOC concentration. In the absence of site-specific information, a value of 2,420400 ppmv as hexane is suggested for landfills known to have co-disposal of MSW and non-residential waste. If the landfill is known to contain only MSW or have very little organic commercial/industrial wastes, then a total NMOC value of 595for default values before 1992, 600 ppmv as hexane should be used, and for default values on and after 1992, 550 ppmv as hexane should be used. In addition, as with the landfill model defaults, the regulatory default value for NMOC content must be used in order to

comply with the NSPS/Emission Guideline. According to NSPS (40 CFR Part 60, Subpart XXX) and Emission Guideline (40 CFR Part 60, Subpart Cf), the landfills with annual NMOC emissions greater than 34 megagrams must consider further emission measurement efforts or installation of a gas collection system.

If Before a site-specific total default pollutant concentration is available (i.used (e., as measured by EPA Reference Method 25C.g. from Table 2.4-1), it must be corrected reviewed for potential air infiltration which correction. from Table 2.4-1), it must be reviewed for potential air infiltration correction. Air infiltration can occur by two different mechanisms: LFG sample dilution, and air intrusion into the landfill. These corrections require site-specific data for the LFG CH4, CO2, nitrogen (N2), and oxygen (O2) content. If the ratio of N2 to O2 is less than or equal to 4.0 (as found in ambient air), then the total pollutant concentration is adjusted for sample dilution by assuming that CO2 and CH4 are the primary (100-percent)%) constituents of landfill gas, and the following equation is used:

$$C_{\underline{p}}(\underline{ppmv}) \text{ (corrected for air infiltration)} = \frac{C_{\underline{p}}(\underline{ppmv}) \cdot (1 - x - 10^6)}{C_{\underline{C}\Theta_2}(\underline{ppmv}) + C_{\underline{C}H_2}(\underline{ppmv})}$$
(2)

where:

$$\begin{array}{lll} \text{Cp} & = & \text{Concentration of pollutant P in landfill gas (i.e., NMOC as hexane), ppmv;} \\ \text{C}_{\text{CO}_2} & = & \text{CO}_2\text{-concentration in landfill gas, ppmv;} \\ \text{C}_{\text{CH}_4} & = & \text{CH}_4\text{-Concentration in landfill gas, ppmv; and} \\ \text{1} \times 10^6 & = & \text{Constant used to correct concentration of P to units of ppmv.} \end{array}$$

$$C_{P}(ppmv) (corrected for air infiltration) = \frac{c_{P}(ppmv)(1 \times 10^{6})}{c_{CO_{2}}(ppmv) + c_{CH_{4}}(ppmv)}$$
(2)

where:

 C_P = Concentration of pollutant P in landfill gas (e.g., NMOC as hexane), ppmv;

 $C_{CO_2} = CO_2$ concentration in landfill gas, ppmv; $C_{CH_4} = CH_4$ Concentration in landfill gas, ppmv; and

 $1x10^6$ = Constant used to correct concentration of P to units of ppmv.

If the ratio of N_2 to O_2 concentrations (i.e., C_{N_2} , C_{O_2}) C_{N_2} , C_{O_2} is greater than 4.0, then the total pollutant concentration should be adjusted for air intrusion into the landfill by using equation 2 and adding the concentration of N_2 (i.e., C_{N_2}) to the denominator. Values for C_{CO_2} , C_{CH_4} , C_{N_2} , and C_{O_2} , can usually be found in the source test report for the landfill along with the total pollutant concentration data. concentration of N_2 (i.e., C_{N_2}) to the denominator. Values for C_{CO_2} , C_{CH_4} , C_{N_2} , C_{O_2} , can usually be found in the source test report for the particular landfill along with the total pollutant concentration data.

To estimate emissions of NMOC or other landfill gas constituents, the following equation should be used:

$$Q_{p} = 1.82 \cdot Q_{CH_4} * \frac{C_p}{(1 \times 10^6)}$$
(3)

where:

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Multiplication factor (assumes that approximately 55 percent of landfill gas is CH₄ and 45 percent is CO₂, N₂, and other constituents).

$$Q_P = \frac{1}{F} Q_{CH_4} \times \frac{C_P}{(1 \times 10^6)} \tag{3}$$

where:

 Q_P Emission rate of pollutant P (e.g., NMOC), m3/yr;

Fraction by volume of CH4 in landfill gas from measurement data for the current reporting year, if available (fraction, dry basis, corrected to 0% oxygen); otherwise, use the default of 0.5;

 \mbox{CH}_4 generation rate, \mbox{m}^3/\mbox{yr} (from equation 1 or LandGEM); and Q_{CH_4}

Concentration of P in landfill gas, ppmv. C_P

Uncontrolled mass emissions per year of total NMOC (as hexane), CO2, CH4, and speciated organic and inorganic compounds can be estimated by the following equation:

$$UM_{p} = Q_{p} * \left[\frac{MW_{p} * -1 - atm}{(8.205 \times 10^{-5} - m^{3} - atm/gmol - {}^{2}K)(1000g/kg)(273 + T^{2}K)} \right]$$
(4)

where:

Uncontrolled mass emissions of pollutant P (i.e., NMOC), kg/yr; Molecular weight of P, g/gmol (i.e., 86.18 for NMOC as hexane); $UM_{\mathbf{p}}$

 $MW_{\mathbb{P}}$

NMOC emission rate of P, m³/yr; and Qр

T = Temperature of landfill gas, ^oC.

$$UM_{P} = Q_{P} \times \frac{MW_{P} \times 1 atm}{(8.205 \times 10^{-5} \frac{m^{3} atm}{gmol {}^{\circ}K})(1000 \frac{g}{kg})(273 + T {}^{\circ}K)}$$
(4)

where:

 UM_P = Uncontrolled mass emissions of pollutant P (e.g., NMOC), kg/yr; MW_P = Molecular weight of P, g/gmol (e.g., 86.18 for NMOC as hexane); Q_P = NMOC emission rate of P, m³/yr; and

 Q_P = NMOC emission rate of P, m³/yr; and T = Temperature of landfill gas, °C.

This equation assumes that the operating pressure of the system is approximately 1 atmosphere. If the temperature of the landfill gas is not known, a temperature of 25° C $(77^{\circ}$ F25°C $(77^{\circ}$ F) is recommended.

Uncontrolled default concentrations of speciated organics along with some inorganic compounds are presented in Table 2.4-1. These default concentrations have already been corrected for air infiltration and can be used as input parameters to equation 3 or the Landfill Air Emission Estimation model for estimating speciated emissions from landfills when site-specific data are not available. An analysis of the data, based on the codisposal history (with non-residential wastes) of the individual landfills from which the concentration data were derived, indicates that for benzene, NMOC, and toluene, there is a difference in the uncontrolled concentrations. Table 2.4-2 presents the corrected concentrations for benzene, NMOC, and toluene to use based on the site's codisposal history.

It is important to note that the compounds listed in Tables 2.4-1 and 2.4-2 are not the only compounds likely to be present in LFG. The listed compounds are those that were identified through a review of the available literature. The reader should be aware that additional compounds are likely present, such as those associated with consumer or industrial products. Given this information, extreme caution should be exercised in the use of the default VOC weight fractions and concentrations given at the bottom of Table 2.4-2. These default VOC values are heavily influenced by the ethane content of the LFG. Available data have shown that there is a range of over 1,500 ppmv in LFG ethane content among landfills.

2.4.4.12.4.4.2 Controlled Emissions

—Emissions from landfills are typically controlled by installing a gas collection system; and combusting the collected gas through the use of internal combustion engines, flares, or turbines. Gas collection systems are not 100-percent% efficient in collecting landfill gas, so emissions of CH4 and NMOC at a landfill with a gas recovery system still occur. To estimate controlled emissions of CH4, NMOC, and other constituents in landfill gas, the collection efficiency of the system must first be estimated. Reported collection efficiencies typically range from 60 to 85 percent, with an average of 75 percent most commonly assumed-should first be estimated. Different models exist to provide a better estimate for uncontrolled emissions that are more appropriate for emission inventory development (e.g. LandGEM) that incorporates waste degradation parameters which include the potential to generate methane and the waste degradation decay rate. The potential to generate methane is a function of waste type, and age of waste. The waste degradation decay rate is a function of waste type, age of waste, and waste moisture. Waste moisture might be changed by leachate recirculation and rainfall rates. Higher collection efficiencies may be achieved at some sites (i.e., those engineered to control gas emissions). If site-specific collection efficiencies are available (i.e., through a comprehensive surface sampling program), then they should be used instead of the 75 percent average. If a user lacks site-specific collection efficiencies, use the

appropriate values in Table HH-3 to Subpart HH of Part 98 - Landfill Gas Collection Efficiencies for calculations. Se section III.T.2 of the preamble of the GHGRP final rule (89 FR 31853 April 25, 2024) for more information on the finalized collection efficiencies in Table HH-3 and default collection efficiency.

Controlled emission estimates should also need to take into account consideration the control efficiency of the control device. Control efficiencies based on test data for the combustion of CH₄, NMOC, and some speciated organics with differing control devices are presented in Table 2.4-3. Emissions from the control devices need to be added to the uncollected emissions to estimate total controlled emissions.

Controlled CH₄, NMOC, and speciated emissions can be calculated with equation 5. It is assumed that the landfill gas collection and control systemGCCS operates 100-percent% of the time. Minor durations of system downtime associated with routine maintenance and repair (i.e., 5 to 7-percent) will%) should not appreciably effect affect emission estimates. The first term in equation 5 accounts for emissions from uncollected landfill gas, while the second term accounts for emissions of the pollutant that were collected but not combusted in the control or utilization device:

$$CM_{p} = \left[UM_{p} \div \left(1 - \frac{\eta_{eol}}{100}\right)\right] + \left[UM_{p} \div \frac{\eta_{eol}}{100} \div \left(1 - \frac{\eta_{ent}}{100}\right)\right] \tag{5}$$

where:

CMpControlled mass emissions of pollutant P, kg/yr;

Uncontrolled mass emissions of P, kg/yr (from equation 4 or the Landfill Air

Emissions Estimation Model);

Collection efficiency of the landfill gas collection system, percent; and

Control efficiency of the landfill gas control or utilization device, percent.

$$CM_P = \left[UM_P \times \left(1 - \frac{\eta_{col}}{100} \right) \right] + \left[UM_P \times \frac{\eta_{col}}{100} \times \left(1 - \frac{\eta_{cnt}}{100} \right) \right] \tag{5}$$

 CM_P = Controlled mass emissions of pollutant P, kg/yr;

 $UM_P = Uncontrolled mass emissions of P, kg/yr (from equation 4 or LandGEM); <math>\eta_{col} = Collection efficiency of the landfill gas collection system, percent; and <math>\eta_{cnt} = Control efficiency of the landfill gas control or utilization device, percent.$

Emission factors for the secondary compounds, CO and NO_x, exiting the control device are presented in Tables 2.4-4 and 2.4-5. These For convenience, emission factors shouldare also presented for NMOC, although most of this NMOC is presumed to be from incomplete combustion of NMOC generated from the landfill rather than NMOC generated by combustion. These default values can be used when equipment vendor guarantees are not available.

Consistent with the language in the Introduction to AP-42, using source-specific data is preferred for estimating a source's emissions, while controlled emissions of CO_2 and sulfur dioxide (SO_2) are best estimated using site-specific landfill gas constituent concentrations—and, along with mass balance methods. ⁶⁸³⁹ If site-specific data are not available, the data in Tables 2.4-1 through 2.4-3 can be used with the mass balance methods that follow.

Controlled CO_2 emissions include emissions from the CO_2 component of landfill gas (equivalent to uncontrolled emissions) and additional CO_2 formed during the combustion of landfill gas. The bulk of the CO_2 formed during landfill gas combustion comes from the combustion of the CO_4 formed during the combustion of the NMOC fraction. Small quantities will be formed during the combustion of the NMOC fraction. The typically amounts to less than 1-percent. Of total CO_2 emissions by weight. Also, the formation of CO_4 through incomplete combustion of

landfill gas will result in small quantities of CO_2 not being formed. This contribution to the overall mass balance picture is also very small and does not have a significant impact on overall $\frac{68_{39}}{200}$ emissions.

$$CM_{CO_2} = UM_{CO_2} + \left[UM_{CH_4} \stackrel{\text{def}}{=} \frac{\hat{H}_{col}}{100} \stackrel{\text{def}}{=} 2.75 \right]$$
 (6)

where:

CM_{CO₂} = Controlled mass emissions of CO₂, kg/yr;

UMCO₂ = Uncontrolled mass emissions of $\tilde{\text{CO}}_2$, kg/yr (from equation 4 or the Landfill Air

Emission Estimation Model);

UMCH₄ = Uncontrolled mass emissions of CH₄, kg/yr (from equation 4 on the Landfill Air

Emission Estimation Model);

Figure 1 Efficiency of the landfill gas collection system, percent; and

2.75 = Ratio of the molecular weight of CO_2 to the molecular weight of CH_{d} .

$$CM_{CO_2} = UM_{CO_2} + (UM_{CH_4} \times \frac{\eta_{col}}{100} \times 2.75)$$
 (6)

 $CM_{CO_2} = UM_{CO_2} =$ Controlled mass emissions of CO₂, kg/yr;

Uncontrolled mass emissions of CO_2 , kg/yr (from equation 4 or LandGEM);

 $UM_{CH_4} =$ Uncontrolled mass emissions of CH₄, kg/yr (from equation 4 or LandGEM);

Efficiency of the landfill gas collector on system, percent; and η_{col}

2.75 Ratio of the molecular weight of CO₂ to the molecular weight of CH₄.

To prepare estimates of SO₂ emissions, data on the concentration of reduced sulfur compounds within the landfill gas are needed. The best way to prepare this estimate is with site-specific information on the total reduced sulfur content of the landfill gas. Often these data are expressed in ppmv as sulfur (S). Equations 3 and 4 should be used first to determine the uncontrolled mass emission rate of reduced sulfur compounds as sulfur. Then, the following equation can be used to estimate SO₂ emissions:

$$CM_{SO_2} = UM_S * \frac{r_{eol}}{100} * 2.0$$
 (7)

where:

 $CMSO_2$ Controlled mass emissions of SO2, kg/yr;

Uncontrolled mass emissions of reduced sulfur compounds as sulfur, kg/yr (from **UM**S

equations 3 and 4);

Efficiency of the landfill gas collection system, percent; and

Ratio of the molecular weight of SO2 to the molecular weight of S.

$$CM_{SO_2} = UM_S \times \frac{\eta_{col}}{100} \times 2.0 \tag{7}$$

where:

 CM_{SO_2} Controlled mass emissions of SO2, kg/yr;

 UM_S Uncontrolled mass emissions of reduced sulfur compounds as sulfur, kg/y (from

equations 3 and 4);

Efficiency of the landfill gas collection system, percent; and η_{col} 2.0 Ratio of the molecular weight of SO₂ to the molecular weight of S.

The next best method to estimate SO₂ concentrations, if site-specific data for total reduced sulfur compounds as sulfur are not available, is to use site-specific data for speciated reduced sulfur compound concentrations. These data can be converted to ppmv as S with equation 8. After the total reduced sulfur as S has been obtained from equation 8, then equations 3, 4, and 7 can be used to derive SO₂ emissions.

$$C_{\underline{S}} = \sum_{\underline{P}} {}^{\dagger} \qquad C_{\underline{P}} = \sum_{\underline{P}} {}^{\dagger} \qquad C_{\underline{P}} = \sum_{\underline{P}} {}^{\dagger} \qquad (8)$$

Cs = Concentration of total reduced sulfur compounds, ppmv as S (for use in equation 3);

Cp = Concentration of each reduced sulfur compound, ppmv;

 $Sp \hspace{0.5cm} = \hspace{0.5cm} Number of moles of S \hspace{0.1cm} produced \hspace{0.1cm} from \hspace{0.1cm} the \hspace{0.1cm} combustion \hspace{0.1cm} of \hspace{0.1cm} each \hspace{0.1cm} reduced \hspace{0.1cm} sulfur \hspace{0.1cm} and \hspace{0.$

compound (i.e., 1 for sulfides, 2 for disulfides); and

n = Number of reduced sulfur compounds available for summation.

$$C_S = \sum_{i=1}^n C_P \times S_P \tag{8}$$

where:

 C_S = Concentration of total reduced sulfur compounds, ppmv as S (for use in equation 3);

 C_P = Concentration of each reduced sulfur compound, ppmv;

 S_P = Number of moles of S produced from the combustion of each reduced sulfur

compound (e.g., 1 for sulfides, 2 for disulfides); and

n = Number of reduced sulfur compounds available for summation.

If no site-specific data are available, a value of 46.9 ppmv can be assumed for C_s (for use in equation 3). This value was obtained by using the default concentrations presented in Table 2.4-1 for reduced sulfur compounds and equation 8.

Hydrochloric acid [Hydrogen Chloride (HCl)] emissions are formed when chlorinated compounds in LFG are combusted in control equipment. The best methods to estimate emissions are mass balance methods that are analogous to those presented above for estimating SO_2 emissions. Hence, the best source of data to estimate HCl emissions is site-specific LFG data on total chloride [expressed in ppmv as the chloride ion (Cl)]. If these data are not available, then total chloride can be estimated from data on individual chlorinated

species using equation 9 below. However, emission estimates may be underestimated, since not every chlorinated compound in the LFG will be represented in the laboratory report (i.e., only those that the analytical method specifies).

$$C_{Cl} = \sum_{n} C_{p} \cdot Cl_{p} \tag{9}$$

where:

CCI = Concentration of total chloride, ppmv as CI (for use in equation 3);

Cp = Concentration of each chlorinated compound, ppmv;

Clp = Number of moles of Cl⁻produced from the combustion of each chlorinated

compound (i.e., 3 for 1,1,1-trichloroethane); and

n = Number of chlorinated compounds available for summation.

$$C_{Cl} = \sum_{i=1}^{n} C_P \times Cl_P \tag{9}$$

 C_{Cl} = Concentration of total chloride, ppmv as Cl⁻ (for use in equation 3);

 C_P = Concentration of each chlorinated compound, ppmv;

 Cl_P = Number of moles of Cl⁻ produced from the combustion of each chlorinated

compound (e.g., 3 for 1,1,1-trichloroethane); and

n = Number of chlorinated compounds available for summation.

After the total chloride concentration ($C_{\rm Cl}$) has been estimated, equations 3 and 4 should be used to determine the total uncontrolled mass emission rate of chlorinated compounds as chloride ion (UM $_{\rm Cl}$). This value is then used in equation 10 below to derive HCl emission estimates:

$$CM_{HCI} = UM_{CI} * \frac{r_{leol}}{100} * 1.03 * (10)$$

where:

CM_{HCl} = Controlled mass emissions of HCl, kg/yr;

UMCI = Uncontrolled mass emissions of chlorinated compounds as chloride, kg/yr (from

equations 3 and 4);
= Efficiency of the landfill gas collection system, percent;

1.03 = Ratio of the molecular weight of HCl to the molecular weight of Cl⁻; and

η_{cnt} = Control efficiency of the landfill gas control or utilization device, percent.

$$CM_{HCl} = UM_{Cl} \times \frac{\eta_{col}}{100} \times 1.03 \times \frac{\eta_{cnt}}{100}$$
 (10)

 CM_{HCl} = Controlled mass emissions of HCl, kg/yr;

 UM_{Cl} = Uncontrolled mass emissions of chlorinated compounds as chloride, kg/yr (from equations 3 and 4):

 η_{col} = Efficiency of the landfill gas collection system, percent;

1.03 = Ratio of the molecular weight of HCl to the molecular weight of Cl⁻; and

 η_{cnt} = Control efficiency of the landfill gas control or utilization device, percent.

In estimating HCl emissions, it is assumed that all of the chloride ion from the combustion of chlorinated LFG constituents is converted to HCl. If an estimate of the control efficiency, $\theta_{\rm em, llen}$, is not available, then the high end of the control efficiency range for the equipment listed in Table 92.4-3 should be used. This assumption is recommended to assume that HCl emissions are not under-estimated.

If site-specific data on total chloride or speciated chlorinated compounds are not available, then a default value of 42.0 ppmv can be used for C_{Cl}. This value was derived from the default LFG constituent concentrations presented in Table 2.4-1. As mentioned above, use of this default may produce underestimates of HCl emissions since it is based only on those compounds for which analyses have been performed. The constituents listed in Table 2.4-lare1 are likely not all of the chlorinated compounds present in LFG.

The reader is referred to Sections 11.2-1 (Unpaved Roads, SCC 50100401), and 11-2.4 (Heavy Construction Operations) of Volume I, and Section II-7 (Construction Equipment) of Volume II, of the AP-42 document for determination of associated fugitive dust and exhaust emissions from these emission sources at MSW landfills.

2.4.5 Source Classification Codes

The Source Classification Codes for Municipal Solid Waste Landfills are:

- 50100401 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Unpaved Road Traffic
- 50100402 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Fugitive Emissions
- 50100403 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Area Method
- 50100404 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Trench Method
- 50100405 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Ramp Method
- 50100406 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Gas Collection System: Other
- 50100407 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Storage Piles
- 50100408 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Conveying of Cover material
- 50100409 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Spreading of Daily Cover
- 50100410 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Dump: Waste Gas Destruction: Waste Gas Flares
- 50100411 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Destruction: Incinerator
- 50100412 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Destruction: Other Not Elsewhere Classified
- 50100420 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Energy Recovery: Turbine

- 50100421 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG) **Energy Recovery: Internal Combustion Engine**
- 50100422 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Other Not Elsewhere Classified
- 50100423 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG Energy Recovery: Boiler
- 50100424 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG **Energy Recovery: Microturbine**
- 50100425 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG **Energy Recovery: Direct Use**
- 50100426 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG Energy Recovery: Industrial Use
- 50100427 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG Energy Recovery: Automobile Fuel
- 50100433 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Landfill Gas (LFG Purification
- 50100440 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Bioreactor
- 50100441 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Final Cover
- 50100442 Waste Disposal; Solid Waste Disposal Government; Municipal Solid Waste Landfill; Hazardous **Fugitive Emissions**
- 50300601 Waste Disposal; Solid Waste Disposal Industrial; Solid Waste Landfill; Waste Gas Destruction
- 50300602 Waste Disposal; Solid Waste Disposal Industrial; Solid Waste Landfill; Other Not Elsewhere Classifie
- 50300603 Waste Disposal; Solid Waste Disposal Industrial; Solid Waste Landfill; Hazardous; Fugitive Emission
- 50300604 Waste Disposal; Solid Waste Disposal Industrial; Solid Waste Landfill; Hazardous Fugitive Emissions
- 50300607 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Storage Piles
- 50300608 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Conveying of Cove
- material 50300609 - Waste Disposal; Solid Waste Disposal - Industrial; Municipal Solid Waste Landfill; Spreading of Daily
- Cover 50301001 – Waste Disposal; Solid Waste Disposal – Industrial; Municipal Solid Waste Landfill; Unpaved Road
- 50301002 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Area method 50301003 - Waste Disposal; Solid Waste Disposal - Industrial; Municipal Solid Waste Landfill; Trench Method
- 50301004 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Ramp Method
- 50301005 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Gas Collection System: Other Not Elsewhere Classified
- 50301010 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) **Destruction: Incinerator**
- 50301011 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) <u>Destruction: Other Not Elsewhere Classified</u>
- 50301020 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Turbine
- 50301021 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Internal Combustion Engine
- 50301022 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) **Energy Recovery: Other Not Elsewhere Classified**
- 50301023 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Boiler
- 50301024 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) **Energy Recovery: Microturbine**
- 50301025 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Direct Use
- 50301026 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Solid Waste Disposal

- Energy Recovery: Industrial Use
- 50301027 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Energy Recovery: Automobile Fuel
- 50301030 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Purification
- 50301040 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Bioreactor
- 50301041 Waste Disposal; Solid Waste Disposal Industrial; Municipal Solid Waste Landfill; Final Cover
- 50600601 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Fugitive Emissions
- 50600602 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Hazardous Fugitive Emissions
- 50600603 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Unpaved Road
 Traffic Area Method
- 50600604 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Area Method
- 50600605 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Trench Method
- 50600606 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Ramp Method
- 50600607 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Gas Collection
 System: Other Not Elsewhere Classified
- 50600610 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Destruction: Incinerator
- 50600611 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Destruction: Other Not Elsewhere Classified
- 50600620 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Energy Recovery: Turbine
- 50600621 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Energy Recovery: Internal Combustion Engine
- 50600622 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Energy Recovery: Other Not Elsewhere Classified
- 50600623 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Energy Recovery: Boiler
- 50600624 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG) Energy Recovery: Microturbine
- 50600625 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Energy Recovery: Direct Use
- 50600626 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Energy Recovery: Industrial Use
- 50600627 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Energy Recovery: Automobile Fuel
- 50600630 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Purification
- 50600640 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Bioreactor
- 50600641 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Final Cover
- 50600642 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Conveying of <u>Cover Material</u>
- 50600643 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Storage Piles
- 50600644 Waste Disposal; Solid Waste Disposal Commercial; Municipal Solid Waste Landfill; Spreading of Daily Cover
- 50700601 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Fugitive Emissions
- 50700602 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Hazardous Fugitive Emissions
- 50700603 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Unpaved Road
 Traffic Area Method

- 50700604 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Area Method
- 50700605 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Trench Method
- 50700606 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Ramp Method
- 50700607 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Gas Collection
 System: Other Not Elsewhere Classified
- 50700610 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG Destruction: Incinerator
- 50700611 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG Destruction: Other Not Elsewhere Classified
- 50700620 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG Energy Recovery: Turbine
- 50700621 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG Energy Recovery: Internal Combustion Engine
- 50700622 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG Energy Recovery: Other Not Elsewhere Classified
- 50700623 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG Energy Recovery: Boiler
- 50700624 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG Energy Recovery: Microturbine
- 50700625 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Energy Recovery: Direct Use
- 50700626 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG Energy Recovery: Industrial Use
- 50700627 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG Energy Recovery: Automobile Fuel
- 50700630 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Landfill Gas (LFG)
 Purification
- 50700640 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Bioreactor
- 50700641 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Final Cover
- 50700642 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Conveying of Cover Material
- 50700643 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Storage Piles
- 50700644 Waste Disposal; Solid Waste Disposal Institutional; Municipal Solid Waste Landfill; Spreading of Daily Cover

2.4.52.4.6 Major Updates Since the Fifth Edition

The Fifth Edition was released in January 1995. Supplemnt D (8/98) is a major revision of the text and recommended emission factors contained in the section. The most significant revisions to this section since publication in the Fifth Edition are summarized below.

August 1998 (Supplement D):

- The equations to calculate the CH₄, CO₂ and other constituents were simplified.
- ullet The default L_0 and k were revised based upon an expanded base of gas generation data.
- The default ratio of CO2 to CH4 was revised based upon averages observed in available source test reports.
- The default concentrations of LFG constituents were revised based upon additional data.
- Additional control efficiencies were included, and existing efficiencies were revised based upon additional
 emission test data.
- Revised and expanded-The recommended emission factors for secondary compounds emitted from typical
 control devices were revised and expanded.

November 1998 (Supplement E (11/98) includes):

- A correction inwas made to equation 10 and a very minor change.
- Minor changes in the molecular weights for 1,1,1-Trichloroethane (methyl chloroform), 1,1-Dichloroethane, 1,2-Dichloropropane, and Trichloroethylene (trichloroethene) presented in Table 2.4-1 were made to agree with values presented in Perry's Handbook.

2.4-18 EMISSION FACTORS 11/98

Table 2.4-1.

August 2024:

A disclaimer was added on the use of AP-42 emission factors.

 Equation 1 was replaced by Equation HH 1 fro 			ctor so that
consistent values for methane generation are	provided across pro	grams. Default	
Equation 3 was updated to account for a site-s	pecific fraction of vo	olume of the the in la	ntimission Factor
The default average conficiency was confidence with the confidence was confidence with the confidence was confidence was confidence with the confidence was confidence was confidence with the conf			
1,1,1 The complete list of Source Classification Code	s (SCCs) for MSW la	ndfills were specified a	B and undated in
The complete list of Source Classification Code 1,1,2,2 - tetraction code the chapter.	167.85	1.11	ina apaaaca iii
1 Dichlargeth was rewsidens dichlaride ation HH-	1 and its parameter	choices. 2.35	₽
1, d - D is the resolution of the second contract d	96.94	0.20	B
1. Denfault-noncentrations were rounded to two s		Tables 2.4-1 and 2.4-2	to be consistent
1,2 Diction of the chapter and with LandGEN	- <u>117 qq</u>	0.18	Đ
The default concentration for CO was 2 Propanol (isopropyl alcohol)	updated (DEFAULT	CONCENTRATIONS FO	IR LFG E
Acetone	NTS**(\$ <mark>58.08</mark> 01004	2 5030 060 3)	₿
Acrylonitrile ^a	53.06	6.33	Ð
Bromodichloromethane	163.83	3.13	€
Butane	58.12	5.03	E
Carbon disulfide ^a	76.13	0.58	C
Carbon monoxide ^b	28.01	141	E
Carbon tetrachloride ^a	153.84	0.004	B
Carbonyl sulfide ^a	60.07	0.49	Ð
Chlorobenzene ^a	112.56	0.25	E
Chlorodifluoromethane	86.47	1.30	E
Chloroethane (ethyl chloride) ^a	64.52	1.25	₽
Chloroform ^a	119.39	0.03	B
Chloromethane	50.49	1.21	₽
Dichlorobenzene ^e	147	0.21	E
Dichlorodifluoromethane	120.91	15.7	A
Dichlorofluoromethane	102.92	2.62	Ð
Dichloromethane (methylene chloride) ^{tt}	84.94	14.3	A
Dimethyl sulfide (methyl sulfide)	62.13	7.82	€
Ethane	30.07	889	E
Ethanol	46.08	27.2	E
Ethyl mercaptan (ethanethiol)	62.13	2.28	Ð
Ethylbenzene ^a	106.16	4.61	B
Ethylene dibromide	187.88	0.001	E
Fluorotrichloromethane	137.38	0.76	B
Hexane ^a	86.18	6.57	B
Hydrogen sulfide	34.08	35.5	₽
Mercury (total) ^{a,d}	200.61	2.92×10^{-4}	E

- Table 2.4-1. (Concluded)).
- NMOC default concentrations were split into pre 1992 and 1992 or later (1992+) to account for codisposal in landfills in Table 2.4-2 and added an additional default concentration was added for NMOC 1992+.
- Affirmed that NMOC is a good surrogate for VOC emissions when NMOC compounds with negligible
 photochemical reactivity are low, removed default reference to 39% ratio of VOC to NMOC compounds
 for non-regulatory programs, and reminded users to develop their own ratios or to use extreme caution if
 default values were selected (footnote c in Table 2.4-2).
- Tables 2.4-4 and 2.4-5 were rearranged and factors for CO from flares and internal combustion engines have been updated, the NO₂ factors from flares and internal combustion engines have been replaced with factors for NO_X from flares and internal combustion engines, a factor for NMOC, as hexane, for flares has been added and 4 factors for NMOC, as hexane, varying by percent load have been added for internal combustion engines.
- Footnote e in Table 2.4-4 was corrected.
- New quality ratings have been given to new/revised factors based on approaches contained in the revised Emissions Factors Procedures Document (January 2023). Factors are given quality ratings based on representativeness of factor (e.g., Highly, Moderately, or Minimally Representative).

2.4-20 EMISSION FACTORS 8/24

<u>Table 2.4-1.</u> <u>DEFAULT CONCENTRATIONS FOR LFG CONSTITUENTS</u>^a (SCC 50100402, 50300603, 50600601, 50700601)

Methyl-ethyl-ketone*1,1,1-Trichloroethane (methyl chloroform)*	Compound	Molecular Weight	Default Concentration (ppmv)	Emission Factor Rating
Tetrachloroethane		72.11 133.41	7.090.48	<u>AB</u>
1.1-Dichloroethene (vinylidene chloride)* 96.94 0.20 8 1.2-Dichloroethane (ethylene dichloride)* 98.96 0.41 8 1.2-Dichloropropane (propylene dichloride)* 112.99 0.18 D Methyl-mercaptan2-Propanol (isopropyl alcohol) 4860.11 2.4950 GE Acetone 58.08 7.0 8 PentaneAcrylonitrile* 72.1553.06 63.329 CD Perchloroethylene 165163.83 3.731 BC Retrachloroethylene 165163.83 3.731 BC Perchloroethylene 165163.83 3.731 BC Retrachloroethylene 165163.83 3.731 BC Carbon disulfide* 58.12 5.0 C Carbon disulfide* 76.13 0.58 C Carbon tetrachloride* 153.84 4.0x10³ 8 Carbon tetrachloride* 153.84 4.0x10³ 8 Carbon tetrachloride* 112.56 0.25 C Chlorodifluoromethane 86.47 1.3 C PropaneChloroethane (ethyl chloride)* 44.0964.52 44.13 B Chloroform* 119.39 3.0x10² B L-1,2-dichloroetheneChloromethane 50.496.94 1.284 B Dichlorobenzene* 147 0.21 E Dichloroethylene 131.40102.92 2.826 BD Crischloroethane (methylene (hloride)* 84.94 14 A A Vinyl-chloride*Dimorthyla sulfide (methyl sulfide) 62.5913 7.348 BC Ethylene dibromide 187.88 1.0x10³ E Ethylene dibromide 187.88 1.0x10³ E		100.16 <u>167.85</u>	1. 87 1	<u>BC</u>
1,2-Dichloroethane (ethylene dichloride)* 98.96 0.41 B 1,2-Dichloropropane (propylene dichloride)* 112.99 0.18 D Methyl-mercaptane_Propanol (isopropyl alcohol) 4860.11 2.4950 CE Acetone 58.08 7.0 B PentaneAcrylonitrile* 72.1553.06 63.329 CD Perchloroethylene (tetrachloroethylene)*Bromodichloromethane 165163.83 3.731 BC Carbon disulfide* 76.13 0.58 C Carbon disulfide* 76.13 0.58 C Carbon monoxide* 110° Minimally Representative* Minimally Representative* 60.07 0.49 D D Chlorodenzene* 112.56 0.25 C C C C C C C C C	1,1-Dichloroethane (ethylidene dichloride) ^a	<u>98.97</u>	2.4	<u>B</u>
1,2-Dichloropropane (propylene dichloride)a 112.99 0.18 D	1,1-Dichloroethene (vinylidene chloride) ^a	96.94	0.20	<u>B</u>
Methyl-mercaptan2-Propanol (isopropyl alcohol) 4860.11 2.4950 CE Acetone 58.08 7.0 B PentaneAcrylonitrile³ 72.4553.06 6.3.29 CD Perchloroethylene 465163.83 3.731 BC (tetrachloroethylene)*Bromodichloromethane Butane C Butane 58.12 5.0 C Carbon disulfide³ 76.13 0.58 C Carbon tetrachloride³ 28.01 110° Minimally Representative¹ Carbon tetrachloride³ 153.84 4.0x10³ B Carbonyl sulfide³ 60.07 0.49 D Chlorobenzene³ 112.56 0.25 C Chlorodenzene³ 112.56 0.25 C Chloroffluoromethane 86.47 1.3 B PropaneChloroethane (ethyl chloride)³ 44.0964.52 1+1.3 B Chloroform³ 119.39 3.0x10² B L-1,2-dichloroethene(chloromethane 50.4996.94 1,2.84 B Dichlorodifluorometh	1,2-Dichloroethane (ethylene dichloride) ^a	98.96	0.41	<u>B</u>
Acetone	1,2-Dichloropropane (propylene dichloride) ^a	112.99	0.18	<u>D</u>
PertaneAcrylonitrile® 72.1553.06 63.29 GD	Methyl mercaptan2-Propanol (isopropyl alcohol)	48 <u>60</u> .11	2.49 50	C <u>E</u>
Perchloroethylene	Acetone	<u>58.08</u>	7.0	<u>B</u>
(tetrachloroethylene) ^a Bromodichloromethane S8.12 5.0 C Carbon disulfide ^a 76.13 0.58 C Carbon monoxide ^b 28.01 110° Minimally Representative ^f Carbon tetrachloride ^a 153.84 4.0x10³ B Carbonyl sulfide ^a 60.07 0.49 D Chlorodifluoromethane 86.47 1.3 C PropaneChloroethane (ethyl chloride) ^a 44.0964.52 11.1,3 B Chloroform ^a 119.39 3.0x10° ² B t-1,2-dichloroetheneChloromethane 50.4996.94 1,2.84 B Dichlorobenzene ^c 147 0.21 E Dichlorodifluoromethane 120.91 16 A Trichloroethylene 131.40102.92 2.826 BD (trichloroethene) ^a Dichlorofluoromethane 10 14 A Dichloromethane (methylene chloride) ^a 84.94 14 A Vinyl-chloride ^a Dimethyl sulfide (methyl sulfide) 62.5013 7.348 BC Ethane 30.07	Pentane Acrylonitrile ^a	72.15 <u>53.06</u>	<u>6.</u> 3 .29	<u>€</u> <u>D</u>
Butane 58.12 5.0 C Carbon disulfide³ 76.13 0.58 C Carbon monoxide⁵ 28.01 110° Minimally Representative¹ Carbon tetrachloride³ 153.84 4.0x10³ B Carbonyl sulfide³ 60.07 0.49 D Chlorobenzene³ 112.56 0.25 C Chlorodifluoromethane 86.47 1.3 C PropaneChloroethane (ethyl chloride)³ 44.0964.52 11.1,3 B Chloroform³ 119.39 3.0x10² B t-1,2-diehloroetheneChloromethane 50.4996.94 1,2.84 B Dichlorodifluoromethane 120.91 16 A Triehloroethylene 131.40102.92 2.826 BD trichloroethene)³Dichlorofluoromethane 10.14 A Dichloromethane (methylene chloride)³ 84.94 14 A Vinyl-ehloride³Dimethyl sulfide (methyl sulfide) 62.5913 7.348 BC Ethanol 46.08 27 E Ethyl mercap	Perchloroethylene	165 163.83	3. 73 1	<u>BC</u>
Carbon disulfide³ 76.13 0.58 C Carbon monoxide⁰ 28.01 110⁰ Minimally Representative¹ Carbon tetrachloride³ 153.84 4.0x10³ B Carbonyl sulfide³ 60.07 0.49 D Chlorobenzene³ 112.56 0.25 C Chlorodifluoromethane 86.47 1.3 C PropaneChloroethane (ethyl chloride)³ 44.0964.52 1+1.3 B Chloroform³ 119.39 3.0x10² B t-1,2-dichloroetheneChloromethane 50.4996.94 1,2.84 B Dichlorobenzeneҫ 147 0.21 E Dichlorodifluoromethane 120.91 16 A Trichloroethene)³Dichlorofluoromethane 131.40102.92 2.826 BD (trichloroethene)³Dichlorofluoromethane 34.94 14 A Dichlorodifleo* 84.94 14 A Vinyl-chloride³Dichlorofluoromethane 30.07 890 C Ethanol 46.08 27 E Et	$\frac{(tetrachloroethylene)^{a}}{Bromodichloromethane}$			
Carbon monoxide ^b 28.01 110 ^e Minimally Representative ^f Carbon tetrachloride ^a 153.84 4.0x10 ^a B Carbonyl sulfide ^a 60.07 0.49 D Chlorobenzene ^a 112.56 0.25 C Chlorodifluoromethane 86.47 1.3 C PropaneChloroethane (ethyl chloride) ^a 44.0964.52 H+.1.3 B Chloroform ^a 119.39 3.0x10 ^a B t-1,2-dichloroethaneChloromethane 50.4996.94 1.2.84 B Dichlorobenzene ^c 147 0.21 E Dichlorodifluoromethane 120.91 16 A Trichloroethylene 131.40102.92 2.826 BD (trichloroethene) ^a Dichlorofluoromethane 120.91 16 A Vinyl chloride ^a Dimethyl sulfide (methyl sulfide) 84.94 14 A Vinyl chloride ^a Dimethyl sulfide (methyl sulfide) 62.5013 7.348 BC Ethane 30.07 890 C Ethanol 46.08 27	<u>Butane</u>	<u>58.12</u>	<u>5.0</u>	<u>C</u>
Carbon tetrachloride³ 153.84 4.0x10³ B Carbonyl sulfide³ 60.07 0.49 D Chlorobenzene³ 112.56 0.25 C Chlorodifluoromethane 86.47 1.3 C PropaneChloroethane (ethyl chloride)³ 44.0964.52 11-1.3 B Chloroform³ 119.39 3.0x10² B t-1,2-dichloroetheneChloromethane 50.4996.94 1.2.84 B Dichlorobenzene⁴ 147 0.21 E Dichlorodifluoromethane 120.91 16 A Trichloroethylene 131.40102.92 2.826 BD (trichloroethene)³Dichlorofluoromethane 10.000 14 A Vinyl-chloride³Dimethyl sulfide (methyl sulfide) 84.94 14 A Vinyl-chloride³Dimethyl sulfide (methyl sulfide) 62.5013 7.348 BC Ethane 30.07 890 C Ethyl mercaptan (ethanethiol) 62.13 2.3 D Xylenes³Ethylbenzene³ 106.16 12.14.6 B </td <td>Carbon disulfide^a</td> <td><u>76.13</u></td> <td>0.58</td> <td><u>C</u></td>	Carbon disulfide ^a	<u>76.13</u>	0.58	<u>C</u>
Carbonyl sulfide³ 60.07 0.49 D Chlorobenzene³ 112.56 0.25 C Chlorodifluoromethane 86.47 1.3 C PropaneChloroethane (ethyl chloride)³ 44.0964.52 11-1.3 B Chloroform³ 119.39 3.0x10² B t-1,2 dichloroetheneChloromethane 50.4986.94 1,2-84 B Dichlorobenzene² 147 0.21 E Dichlorodifluoromethane 120.91 16 A Trichloroethylene 131.40102.92 2.826 BD (trichloromethane)abichlorofluoromethane 141 A Dichloromethane (methylene chloride)³ 84.94 14 A Vinyl-chloride³Dimethyl sulfide (methyl sulfide) 62.5913 7.348 BC Ethane 30.07 890 C Ethanol 46.08 27 E Ethyl mercaptan (ethanethiol) 62.13 2.3 D Xylenes³Ethylbenzene³ 106.16 12.14.6 B Eluorotrichloromethane	<u>Carbon monoxide</u> ^b	<u>28.01</u>	110 ^e	
Chlorobenzene³ 112.56 0.25 C Chlorodifluoromethane 86.47 1.3 C Propane Chloroethane (ethyl chloride)³ 44.0964.52 11.1.3 B Chloroform³ 119.39 3.0x10² B t-1,2 dichloroethene Chloromethane 50.4926.94 1.2.84 B Dichlorobenzene² 147 0.21 E Dichlorodifluoromethane 120.91 16 A Trichloroethylene 131.40102.92 2.826 BD (trichloroethene)³Dichlorofluoromethane 84.94 14 A Vinyl chloride³Dimethyl sulfide (methyl sulfide) 62.5913 7.348 BC Ethane 30.07 890 C Ethyl mercaptan (ethanethiol) 62.13 2.3 D Xylenes³Ethylbenzene³ 106.16 12.14.6 B Ethylene dibromide 187.88 1.0x10³ E Eluorotrichloromethane 137.38 0.76 B Hexane³ 86.18 6.6 B Hydrogen su	Carbon tetrachloride ^a	<u>153.84</u>	4.0x10 ⁻³	<u>B</u>
Chlorodifluoromethane 86.47 1.3 C Propane Chloroethane (ethyl chloride)³ 44.0964.52 11.1.3 B Chloroform³ 119.39 3.0x10² B t-1,2-dichloroethene Chloromethane 50.4996.94 1,2.84 B Dichlorobenzene² 147 0.21 E Dichlorodifluoromethane 120.91 16 A Trichloroethylene 131.40102.92 2.826 BD (trichloroethene)³Dichlorofluoromethane 62.5913 7.348 BC Ethane 30.07 890 C Ethane 30.07 890 C Ethyl mercaptan (ethanethiol) 62.13 2.3 D Xylenes³Ethylbenzene³ 106.16 12.14.6 B Ethylene dibromide 187.88 1.0x10³ E Fluorotrichloromethane 137.38 0.76 B Hexane³ 86.18 6.6 B Hydrogen sulfide 34.08 36 B	Carbonyl sulfide ^a	60.07	0.49	<u>D</u>
Propane Chloroethane (ethyl chloride)³ 44.0964.52 11.3 B Chloroform³ 119.39 3.0x10² B t 1,2 dichloroetheneChloromethane 50.4996.94 1,2.84 B Dichlorobenzenec 147 0.21 E Dichlorodifluoromethane 120.91 16 A Trichloroethylene 131.40102.92 2.826 BD (trichloromethane (methylene chloride)³ 84.94 14 A Vinyl-chloride³Dimethyl sulfide (methyl sulfide) 62.5013 7.348 BC Ethane 30.07 890 C Ethanol 46.08 27 E Ethyl mercaptan (ethanethiol) 62.13 2.3 D Xylenes³Ethylbenzene³ 106.16 12.14.6 B Ethylene dibromide 187.88 1.0x10³³ E Fluorotrichloromethane 137.38 0.76 B Hexane³ 86.18 6.6 B Hydrogen sulfide 34.08 36 B	<u>Chlorobenzene</u> ^a	112.56	0.25	<u>C</u>
Chloroform³ 119.39 3.0x10³² B t-1,2-dichloroetheneChloromethane 50.49€6.94 1,2.84 B Dichlorobenzene⁻ 147 0.21 E Dichlorodifluoromethane 120.91 16 A Trichloroethylene 131.40102.92 2.826 BD (trichloroethene)³Dichlorofluoromethane BE BD Dichloromethane (methylene chloride)³ 84.94 14 A Vinyl-chloride³Dimethyl sulfide (methyl sulfide) 62.5013 7.348 BC Ethane 30.07 890 C Ethanol 46.08 27 E Ethyl mercaptan (ethanethiol) 62.13 2.3 D Xylenes³Ethylbenzene³ 106.16 12.14.6 B Ethylene dibromide 187.88 1.0x10³ E Fluorotrichloromethane 137.38 0.76 B Hexane³ 86.18 6.6 B Hydrogen sulfide 34.08 36 B	Chlorodifluoromethane	86.47	1.3	<u>C</u>
t-1,2-dichloroetheneChloromethane 50.4996.94 1_2.84 B Dichlorobenzene ^c 147 0.21 E Dichlorodifluoromethane 120.91 16 A Trichloroethylene 131.40102.92 2.826 BD (trichloroethene) ^a Dichlorofluoromethane 84.94 14 A Vinyl-chloride ^a Dimethyl sulfide (methyl sulfide) 62.5013 7.348 BC Ethane 30.07 890 C Ethanol 46.08 27 E Ethyl mercaptan (ethanethiol) 62.13 2.3 D Xylenes ^a Ethylbenzene ^a 106.16 12.14.6 B Ethylene dibromide 187.88 1.0x10 ⁻³ E Fluorotrichloromethane 137.38 0.76 B Hexane ^a 86.18 6.6 B Hydrogen sulfide 34.08 36 B	Propane Chloroethane (ethyl chloride) ^a	44.09 <u>64.52</u>	11. 1 <u>.3</u>	В
Dichlorobenzene ^c 147 0.21 E Dichlorodifluoromethane 120.91 16 A Trichloroethylene 131.40102.92 2.826 BD (trichloroethene) ^a Dichlorofluoromethane 84.94 14 A Vinyl-chloride ^a Dimethyl sulfide (methyl sulfide) 62.5013 7.348 BC Ethane 30.07 890 C Ethanol 46.08 27 E Ethyl mercaptan (ethanethiol) 62.13 2.3 D Xylenes ^a Ethylbenzene ^a 106.16 12.14.6 B Ethylene dibromide 187.88 1.0x10 ⁻³ E Fluorotrichloromethane 137.38 0.76 B Hexane ^a 86.18 6.6 B Hydrogen sulfide 34.08 36 B	<u>Chloroform</u> ^a	119.39	3.0x10 ⁻²	<u>B</u>
Dichlorodifluoromethane 120.91 16 A Trichloroethylene 131.40102.92 2.826 BD (trichloroethene) ^a Dichlorofluoromethane 0 431.40102.92 2.826 BD (trichloroethene) ^a Dichlorofluoromethane 0 44.00 14 A Vinyl chloride ^a Dimethyl sulfide (methyl sulfide) 62.5913 7.348 BC Ethane 30.07 890 C Ethanol 46.08 27 E Ethyl mercaptan (ethanethiol) 62.13 2.3 D Xylenes ^a Ethylbenzene ^a 106.16 12.14.6 B Ethylene dibromide 187.88 1.0x10 ³ E Fluorotrichloromethane 137.38 0.76 B Hexane ^a 86.18 6.6 B Hydrogen sulfide 34.08 36 B	t 1,2 dichloroethene Chloromethane	50.49 96.94	<u>1.</u> 2.84	В
Trichloroethylene 131.40102.92 2.826 BD (trichloroethene) ^a Dichlorofluoromethane 141.40102.92 2.826 BD Dichloromethane (methylene chloride) ^a 84.94 14 A Vinyl-chloride ^a Dimethyl sulfide (methyl sulfide) 62.5013 7.348 BC Ethane 30.07 890 C Ethylane 46.08 27 E Ethyl mercaptan (ethanethiol) 62.13 2.3 D Xylenes ^a Ethylbenzene ^a 106.16 12.14.6 B Ethylene dibromide 187.88 1.0x10 ⁻³ E Fluorotrichloromethane 137.38 0.76 B Hexane ^a 86.18 6.6 B Hydrogen sulfide 34.08 36 B	<u>Dichlorobenzene</u> ^c	<u>147</u>	0.21	<u>E</u>
(trichloroethene) ^a Dichlorofluoromethane Best Name Dichloromethane (methylene chloride) ^a 84.94 14 A Vinyl-chloride ^a Dimethyl sulfide (methyl sulfide) 62.5013 7.348 BC Ethane 30.07 890 C Ethanol 46.08 27 E Ethyl mercaptan (ethanethiol) 62.13 2.3 D Xylenes ^a Ethylbenzene ^a 106.16 12.14.6 B Ethylene dibromide 187.88 1.0x10 ⁻³ E Fluorotrichloromethane 137.38 0.76 B Hexane ^a 86.18 6.6 B Hydrogen sulfide 34.08 36 B	<u>Dichlorodifluoromethane</u>	120.91	<u>16</u>	<u>A</u>
Vinyl chloride ^a Dimethyl sulfide (methyl sulfide) 62.5913 7.348 BC Ethane 30.07 890 C Ethanol 46.08 27 E Ethyl mercaptan (ethanethiol) 62.13 2.3 D Xylenes ^a Ethylbenzene ^a 106.16 12.14.6 B Ethylene dibromide 187.88 1.0x10 ⁻³ E Fluorotrichloromethane 137.38 0.76 B Hexane ^a 86.18 6.6 B Hydrogen sulfide 34.08 36 B		131.40 102.92	2. 82 6	<u>₿</u> <u>D</u>
Ethane 30.07 890 C Ethanol 46.08 27 E Ethyl mercaptan (ethanethiol) 62.13 2.3 D Xylenes [®] Ethylbenzene ^a 106.16 12.14.6 B Ethylene dibromide 187.88 1.0x10 ⁻³ E Fluorotrichloromethane 137.38 0.76 B Hexane ^a 86.18 6.6 B Hydrogen sulfide 34.08 36 B	Dichloromethane (methylene chloride) ^a	84.94	14	<u>A</u>
Ethane 30.07 890 C Ethanol 46.08 27 E Ethyl mercaptan (ethanethiol) 62.13 2.3 D Xylenes ^a Ethylbenzene ^a 106.16 12.14.6 B Ethylene dibromide 187.88 1.0x10 ⁻³ E Fluorotrichloromethane 137.38 0.76 B Hexane ^a 86.18 6.6 B Hydrogen sulfide 34.08 36 B	Vinyl chloride ^a Dimethyl sulfide (methyl sulfide)	62. 50 <u>13</u>	7. 34 <u>8</u>	<u>BC</u>
Ethyl mercaptan (ethanethiol) 62.13 2.3 D Xylenes ^a Ethylbenzene ^a 106.16 12.14.6 B Ethylene dibromide 187.88 1.0x10 ⁻³ E Fluorotrichloromethane 137.38 0.76 B Hexane ^a 86.18 6.6 B Hydrogen sulfide 34.08 36 B		30.07	890	<u>C</u>
Xylenes ^a Ethylbenzene ^a 106.16 12.14.6 B Ethylene dibromide 187.88 1.0x10 ⁻³ E Fluorotrichloromethane 137.38 0.76 B Hexane ^a 86.18 6.6 B Hydrogen sulfide 34.08 36 B	<u>Ethanol</u>	46.08	<u>27</u>	<u>E</u>
Ethylene dibromide 187.88 1.0x10³ E Fluorotrichloromethane 137.38 0.76 B Hexane³ 86.18 6.6 B Hydrogen sulfide 34.08 36 B	Ethyl mercaptan (ethanethiol)	<u>62.13</u>	2.3	<u>D</u>
Ethylene dibromide 187.88 1.0x10 ⁻³ E Fluorotrichloromethane 137.38 0.76 B Hexane ^a 86.18 6.6 B Hydrogen sulfide 34.08 36 B	Xylenes ^a Ethylbenzene ^a	106.16	<u>12.14.6</u>	В
Hexane ^a 86.18 6.6 B Hydrogen sulfide 34.08 36 B		187.88	1.0x10 ⁻³	<u>E</u>
<u>Hydrogen sulfide</u> <u>34.08</u> <u>36</u> <u>B</u>	Fluorotrichloromethane	137.38	0.76	<u>B</u>
<u>Hydrogen sulfide</u> <u>34.08</u> <u>36</u> <u>B</u>	Hexane ^a	86.18	6.6	<u>B</u>
	Hydrogen sulfide	34.08	<u>36</u>	_
	Mercury (total) ^{a,d}	200.61	2.9x10 ⁻⁴	<u>E</u>

Table 2.4-1. (Continued)

Compound	Molecular	Default Concentration	Emission Factor
Compound	Weight	(ppmv)	Rating
Methyl ethyl ketone ^a	<u>72.11</u>	<u>7.1</u>	<u>A</u>
Methyl isobutyl ketone ^a	<u>100.16</u>	<u>1.9</u>	<u>B</u>
Methyl mercaptan	48.11	<u>2.5</u>	<u>C</u>
<u>Pentane</u>	<u>72.15</u>	<u>3.3</u>	<u>C</u>
Perchloroethylene (tetrachloroethylene) ^a	<u>165.83</u>	<u>3.7</u>	<u>B</u>
<u>Propane</u>	44.09	<u>11</u>	<u>B</u>
<u>t-1,2-dichloroethene</u>	<u>96.94</u>	<u>2.8</u>	<u>B</u>
Trichloroethylene (trichloroethene) ^a	<u>131.4</u>	<u>2.8</u>	<u>B</u>
Vinyl chloride ^a	<u>62.5</u>	<u>7.3</u>	<u>B</u>
<u>Xylenes^a</u>	<u>106.16</u>	<u>12</u>	<u>B</u>

NOTE: This is not an all-inclusive list of potential LFG constituents, only those for which test data were available at multiple sites. References $\frac{10-6741-97}{2}$. Source Classification Codes in parentheses.

sites where CO was measured, only 2-showed detectable levels of CO. Note that large values – on the order of 1,000 ppm and greater – can

- indicate underground combustion or other atypical conditions.
- ____Source tests did not indicate whether this compound was the para- or ortho- isomer. The para isomer is a Title III-listed HAP.
 - $\underline{\mbox{\tt d}}$ No data were available to speciate total Hg into the elemental and organic forms.
 - e Reference 98.
 - ^f Emission factor is minimally representative of the population. Emission factor quality ratings based on the Emissions Factors Procedures Document (January 2023).

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<u>a</u> Hazardous Air Pollutants listed in Title III of the 1990 Clean Air Act Amendments.

 $^{^{\}theta}$ — $_{-}$ b Carbon monoxide is not a typical constituent of LFG, but does<u>can</u> exist in instances involving landfill (underground) combustion. Therefore, LFG, typically in small quantities. This default value should be used with caution. <u>Just 2</u> of 18

Table 2.4-2. DEFAULT CONCENTRATIONS OF BENZENE, NMOC, AND TOLUENE BASED ON WASTE DISPOSAL HISTORY^a

(SCC 50100402, 50300603, 50600601, 50700601)

Pollutant	nt Molecular Weight		Default Concentration (ppmv)	Emission Factor Rating	g	
Benzene ^b	ā	78.11				1
Benzene ^b - Co-disposal	•	<u>78.1</u>	1	11.1	D	
Benzene ^b - No or Unknown co-disposal		<u>78.11</u>		1. 91 9	В	
NMOC (as hexane) ^c - Co-disposal (pre 1992)	NMOC (as hexane) ^c - Co-disposal (pre 1992) ^d		.8	<u>2400</u>	<u>D</u>	
NMOC (as hexane) ^c - No or Unknown co-disp (pre 1992) ^d	oosal	<u>86.1</u>	.8	2420 600	₽ <u>B</u>	
NMOC (as hexane) c - No or Unknown co-dis			.8	595 <u>550°</u>	BModerately Representativ	
Toluene ^{b-} Co-disposal		92.1	.3	<u>170</u>	<u>D</u>	
Co-disposal				165	Đ	
Toluene ^b - No or Unknown co-disposal		92.1	.3	39 .3	А	

^aReferences 10-5441-85. Source Classification Codes in parentheses.

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b Hazardous Air Pollutants listed in Title III of the 1990 Clean Air Act Amendments.

EFor NSPS/Emission Guideline compliance purposes, the default concentration for NMOC as specified in the final rule must be used. For other purposes not associated with NSPS/Emission Guideline compliance, the default VOC content at co-disposal sites = 85 percent by weight (2,060 ppmv as hexane); at No or Unknown sites = 39 percent by weight 235 ppmv as hexane), users should develop and use their own-site specific data.

d Emission factors are split into pre 1992 and 1992 or later (1992+) to account for co-disposal in landfills. The 1992+ factor is based on data from landfills that did not accept co-disposed waste and from landfills that opened in 1992 or later because current landfills do not allow co-disposal. The pre 1992 factors are based on data from landfills that opene before and during 1992, when some landfills allowed co-disposal.

e Reference 113.

Emission factor is moderately representative of the population. Emission factor quality ratings based on the Emissions Factor Procedures Document (January 2023).

Table 2.4-3. CONTROL EFFICIENCIES FOR LFG CONSTITUENTS^a (50100420)

Table 2.4-3. CONTROL EFFICIENCIES FOR LFG CONSTITUENTS				
(50100420)				
(50100421)				
Control Device	Constituent ^b	Typical	Range of	Emission
Gas Turbine (50100420,	Halogenated Species	99.7	<u>Q897r99</u> +	<u>Eactor</u>
5080g0n20(5500600045210),5500800005210)	Halogenated Species	93.0 Efficiency (%)	9099thcy (%)	<u>Rating</u>
<u>5060062125070062420,</u>	Non-Halogenated	98.2 1 Visical 86.1 Page	97-99+	E
BOBO & 01240 (550016000452210 , 55008X0010062210)	Skpærcikislogenated	86.1 Rang	25-99+	E
EOEngine1(500700045211)50301021,	SPACIE S	97.2	94-99+	E
<u>50600621, 50700621)</u>		Ratin		
		g		
Boiler/Steam Turbine (50100423,	NMOC	98 .0	96-99+	D
50301023, 50600623, 50700623)				
Boiler/Steam Turbine (50100423,	Halogenated Species	99.6	87-99+	D
50301023, 50600623, 50700623				
Boiler/Steam Turbine (50100423,	Non-Halogenated	99.8	67-99+	D
<u>50301023, 50600623, 50700623)</u>	Species			
Flare ^c (50100410 <u>, 50300601</u>)	NMOC	99.2	90-99+	В
(Flare ^c (50100410, 50300601)	Halogenated Species	98 .0	91-99+	С
Flare ^c (50100410, 50300601)	Non-Halogenated	99.7	38-99+	С
	Species			
Gas Turbine (50100420,	NMOC	94.4	90-99+	E
<u>50301020, 50600620, 50700620)</u>				

 $^{\rm a}$ References $\underline{10\text{-}67}\underline{41\text{-}97}.$ Source Classification Codes in parentheses.

Inserted Cells **Inserted Cells**

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b Halogenated species are those containing atoms of chlorine, bromine, fluorine, or iodine. For any equipment, the control efficiency for mercury should be assumed to be 0. See section 2.4.4.2 for methods to estimate emissions of SO₂, CO₂, and HCl.

CWhere information on equipment was given in the reference, test data were taken from enclosed flares. Control

 $efficiencies\ are\ assumed\ to\ be\ equally\ representative\ of\ open\ flares.$

Table 2.4-4. (Metric Units) EMISSION FACTORS FOR SECONDARY COMPOUNDS EXITING CONTROL DEVICES^a

Control Device	Pollutant ^b	kg/10 ⁶ dscm Methane	Emission Factor Rating
Flare Esoiler/Steam Turbine (50100423, 50301023, 50600623, 50700623)	Nitrogen dioxide	Nitrogen dioxide 650530	
(50100410)Boiler/Steam Turbine ^c (50100423, 50301023, 50600623, 50700623)	Carbon monoxide	12,000 90	<u>CE</u>
(50300601)Boiler/Steam Turbine ^c (50100423, 50301023, 50600623, 50700623)	Particulate matter	270 130	D
IC Engine Gas Turbine (50100420, 50301020, 50600620, 50700620)	Nitrogen dioxide	4 <u>,000</u> 1 <u>,400</u>	D
(50100421)Gas Turbine (50100420, 50301020, 50600620, 50700620)	Carbon monoxide	7,500 3,600	C <u>E</u>
Gas Turbine (50100420, 50301020, 50600620, 50700620)	Particulate matter	770 350	E
Enclosed Combustor/Flared (50100410, 50300601)	<u>Particulate matter</u>	<u>270</u>	<u>D</u>
Boiler/Steam Turbine ^d Enclosed Combustor/Flare ^d (50100410, 50300601)	Nitrogen dioxideoxides	530 <u>610^{e,f}</u>	DHighly Representative
Enclosed Combustor/Flared (50100410, 50300601)	NMOC, as hexane (VOC) j	<u>66^{e,h}</u>	Highly Representative ^g
(50100423)Enclosed Combustor/Flared (50100410, 50300601)	Carbon monoxide	90 920 ^{e,i}	E <u>Highly</u> Representative
Parti	iculate matter	130	Đ
Gas TurbinelC Engine (50100421, 50301021, 50600621, 50700621)	Nitrogen dioxideoxides	1,400 <u>500e,k</u>	DHighly Representatives
(50100420) IC Engine (50100421, 50301021, 50600621, 50700621)	Carbon monoxide	3,600 <u>4,600^{e,l}</u>	E <u>Highly</u> Representative
IC Engine (50100421, 50301021, 50600621, 50700621)	Particulate matter	350 770	E
IC Engine (50100421, 50301021, 50600621, 50700621)	NMOC, as hexane (VOC) j.o 100% Load	250 ^{e,m}	Moderately Representative
IC Engine (50100421, 50301021, 50600621, 50700621)	NMOC, as hexane (VOC) ^{j,o} 80% Load	250 ^{e,m}	Moderately Representative
IC Engine (50100421, 50301021, 50600621, 50700621)	NMOC, as hexane (VOC) i.º 60% Load	270e,m	Moderately Representative
IC Engine (50100421, 50301021,	NMOC, as hexane (VOC) i.o	140 ^{e,m}	Moderately

 $^{^{}a} Source\ Classification\ Codes\ in\ parentheses.\ \ Divide\ kg/10^{6} dscm\ \underline{methane}\ by\ 16,\overline{700}\underline{666.7}\ to\ obtain\ kg/hr/dscmm\underline{methane}\ by\ 16,\overline{700}\underline{666.7}\ to\ obtain\ by\ 16,\overline{700}\underline{666.7}\$ methane.

30% Load

Representative

^b No data on PM size distributions were available, however for other gas-fired combustion sources, most of the particulate matter is less than 2.5 microns in diameter. Hence, this emission factor can be used to provide estimates of PM-10 or PM-2.5 emissions. See section 2.4.4.2 for methods to estimate CO₂, SO₂, and HCl. As mentioned in Basic Information about NO₂, available at https://www.epa.gov/no2-pollution/basic-information-about-no2, nitrogen dioxide (NO2) is one of a group of highly reactive gases known as oxides of nitrogen or nitrogen oxides (NO_x), and it is used as the indicator for the larger group of nitrogen oxides.

^eWhere information on equipment was given in the reference, test data were taken from enclosed flares.

Control efficiencies are assumed to be equally representative of open flares.

^d All source tests were conducted on boilers, however emission factors should also be representative of steam turbines. Emission $factors \ are \ representative \ of \ boilers \ equipped \ with \ low-NO_X \ burners \ and \ flue \ gas \ recirculation. \ No \ data \ were \ available \ for \ burners \ and \ flue \ gas \ recirculation.$ uncontrolled NO_X emissions.

dest data were taken from enclosed flares. Control efficiencies are assumed to be equally representative of open flares. *Factors were converted from lb/mmbtu. To convert back to lb/mmbtu, divide by 16.02 and then divide by 1020 (the heat 8/24

content of methane in ft³/Btu). Note that these factors will have units of lb/mmbtu in WebFIRE.

- f Reference 99.
- Emission factor is highly representative of the population. Emission factor quality ratings based on the Emissions Factors Procedures Document (January 2023).
- h Reference 100.
- Reference 101.
- NMOC = VOC because review of data from references 11-21, 23-24, 26,27, 29,30, 32,36,38,102-109 affirm the effect of compounds with low or no photochemical reactivity is less than 50 ppm LFG.
- k Reference 110.
- Reference 111.
- ^m Reference 112.
- "Emission factor is moderately representative of the population. Emission factor quality ratings based on the Emissions Factors Procedures Document (January 2023).
- °During its review, the EPA investigated whether the NMOC emission factors at varying loads could be combined but found that they should remain separate. All per-load heat input values were found to be from differing data sets via a two sample t-test at a 95% confidence coefficient.

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Table 2.4-5. (English Units) EMISSION RATES FOR SECONDARY COMPOUNDS EXITING CONTROL DEVICES^a

Control Device	Pollutantb	lb/10 ⁶ dscf Methane	Emission Factor Rating
Flare ⁶ Boiler/Steam Turbine ^c (50100423, 50301023, 50600623, 50700623)	Nitrogen dioxide	4033	<u>C</u> <u>E</u>
(50100410)Boiler/Steam Turbine ^c (50100423, 50301023, 50600623, 50700623)	Carbon monoxide	750 <u>5.7</u>	C <u>E</u>
(50300601)Boiler/Steam Turbine ^c (50100423, 50301023, 50600623, 50700623)	Particulate matter	17 8.2	Đ <u>E</u>
IC EngineGas Turbine (50100420, 50301020, 50600620, 50700620)	Nitrogen dioxide	<u>250</u> <u>87</u>	D
(50100421)Gas Turbine (50100420, 50301020, 50600620, 50700620)	Carbon monoxide	4 70 230	<u>C</u> <u>D</u>
Gas Turbine (50100420, 50301020, 50600620, 50700620)	Particulate matter	48 <u>22</u>	Е
Enclosed Combustor/Flare ^d (50100410, 50300601)	Particulate matter	<u>17</u>	<u>D</u>
Boiler/Steam Turbine ^d Enclosed Combustor/Flare ^d (50100410, 50300601)	Nitrogen dioxideoxides	33 38 ^{e,f}	E <u>Highly</u> Representative ^g
Enclosed Combustor/Flared (50100410, 50300601)	NMOC, as hexane (VOC) ^j	4.1 ^{e,f}	Highly Representative ^g
(50100423)Enclosed Combustor/Flared (50100410, 50300601)	Carbon monoxide	5.7 <u>58^{e,f}</u>	E <u>Highly</u> Representative ^g
Pa	rticulate matter	8.2	E
Gas TurbinelC Engine (50100421, 50301021, 50600621, 50700621)	Nitrogen dioxideoxides	87 96 ^{e,k}	DHighly Representative ^g
(50100420)IC Engine (50100421, 50301021, 50600621, 50700621)	Carbon monoxide	230 290 ^{e,l}	D Highly Representative ^g
IC Engine (50100421, 50301021, 50600621, 50700621)	Particulate matter	22 48	E
IC Engine (50100421, 50301021, 50600621, 50700621)	NMOC, as hexane (VOC) ^{j,o} 100% Load	15 ^{e,m}	Moderately Representative ⁿ
IC Engine (50100421, 50301021, 50600621, 50700621)	NMOC, as hexane (VOC) ^{j,o} 80% Load	15 ^{e,m}	Moderately Representative ⁿ
IC Engine (50100421, 50301021, 50600621, 50700621)	NMOC, as hexane (VOC) ^{j,o} 60% Load	17 ^{e,m}	Moderately Representative ⁿ
IC Engine (50100421, 50301021, 50600621, 50700621)	NMOC, as hexane (VOC)i.o 30% Load	<u>ge,m</u>	Moderately Representative

 $^{^{\}rm a}$ Source Classification Codes in parentheses. Divide lb/10 $^{\rm 6}$ dscf by 16,700 to obtain lb/hr/dscfm.

PM-2.5 emissions. See section 2.4.4.2 for methods to estimate CO₂, SO₂, and HCl.

^b Based on data for other combustion sources, most of the particulate matter will be less than 2.5 microns in diameter. Hence, this emission rate can be used to provide estimates of PM-10 or PM-2.5 emissions. See section 2.4.4.2 for methods to estimate CO₂, SO₂, and HCl. As mentioned in Basic Information about NO₂, available at https://www.epa.gov/no2-pollution/basic-information-about-no2, nitrogen dioxide (NO₂) is one of a group of highly reactive gases known as oxides of nitrogen or nitrogen oxides (NO₃), and it is used as the indicator for the larger group of nitrogen oxides.

- ^{e-}Where information on equipment was given in the reference, test data were taken from enclosed flares. Control efficiencies are assumed to be equally representative of open flares.
- d_{Σ} All source tests were conducted on boilers, however emission factors should also be representative of steam turbines. Emission factors are representative of boilers equipped with low-NO_X burners and flue gas recirculation. No data were available for uncontrolled NO_X emissions.
- d Test data were taken from enclosed flares. Control efficiencies are assumed to be equally representative of open flares.

 Emission Factors were converted from lb/mmbtu. To convert back to lb/mmbtu, divide by 1020. Note that these factors will have units of lb/mmbtu in WebFIRE.
- f Reference 99.
- ⁸ Emission factor is highly representative of the population. Emission factor quality ratings based on the Emissions Factors Procedures Document (January 2023).
- h Reference 100.
- Reference 101.
- $\frac{1}{2}$ NMOC = VOC because review of data from references 11-21, 23-24, 26,27, 29,30, 32,36,38,102-109 affirm the effect of compounds with low or no photochemical reactivity is less than 50 ppm LFG.
- k Reference 110.
- Reference 111.
- m Reference 112.
- "Emission factor is moderately representative of the population. Emission factor quality ratings based on the Emissions Factors Procedures Document (January 2023).
- ^o During its review, the EPA investigated whether the NMOC emission factors at varying loads could be combined but found that they should remain separate. All per-load heat input values were found to be from differing data sets via a two sample t-test at a 95% confidence coefficient.

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* Test data from these reports are in the Emission Factor Documentation for AP-42 Section 2.4 Municipal Solid Waste Landfills Revised, August 1997 (reference 115). The documentation reference number and page citations in the document are listed. The appendices have data from the reports.

 $\underline{References \ are \ available \ electronically \ at: \ https://gaftp.epa.gov/ap42/ch02/s04/reference/2024\%20Section/.}$

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